

***EPA Update on the
Development of Alternative Reference Methods
for Mercury
and
Testing Equipment***

**Jeff Ryan
US EPA ORD
ryan.jeff@epa.gov
(919) 933-2998**

**2007 Measurement Technology Workshop
September 11, 2007**

Topics

- Method 30A (IRM) approach
- IRM field testing
- Development of IRM equipment
- Method 30B (sorbent trap) approach
- Sorbent Trap RM laboratory and field testing



Overall Approach to New RMs ...

- Performance-based
- Test program-specific (site) and matrix-specific performance verification
- Use of “Method of Standard Addition” for matrix-specific performance verification
 - Gaseous dynamic spiking for IRM
 - Gaseous static spiking for STRM



Method 30A - Mercury Instrumental Reference Method (IRM)

- Timely (real-time)
- Performance-based
 - Amenable to multiple and new technologies
 - Site/Test program-specific verification of data quality
 - FR Notice of Intent (62 FR 52098, 10/6/97)
- Consistent w/ SO_x & NO_x instrumental methods
- Key elements
 - Interference test (optional)
 - Calibration error/linearity (Hg⁰)
 - System integrity/drift (HgCl₂)
 - System response time
 - Dynamic spiking (HgCl₂)



Dynamic Spiking

- Nothing new
- Included in revisions to Methods 6C and 7E (optional)
- Gaseous method of “standard addition”
- Introduce known quantities of HgCl_2 into probe sampling stack gas
- Spike flow minor (<20%) relative to sample flow
- Requires knowing sample flow rate or dilution ratio



Lehigh Univ. Field Test

- Implement Conceptual (2/28/06) IRM on 4 available systems
- Use one common set of Hg gas standards for our calibrations
 - Vendors still calibrated/configured their systems the way they wanted
- Determine whether performance criteria pass/fail, but without corrective action – get what we get
- Did not perform Part 75 pre-certification tests, but IRM tests very similar



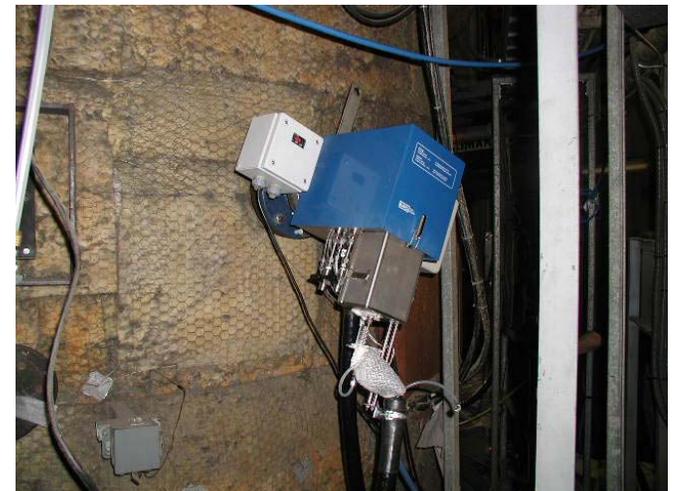
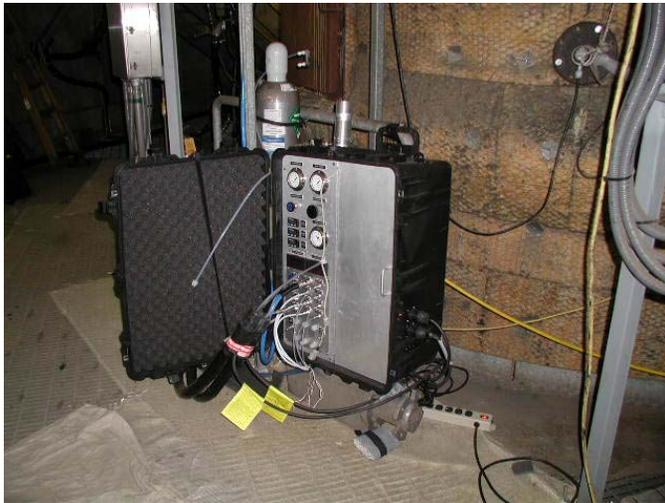
Available Systems

- 3 “stationary” Hg CEMS
 - GE/PSA
 - Tekran
 - Thermo

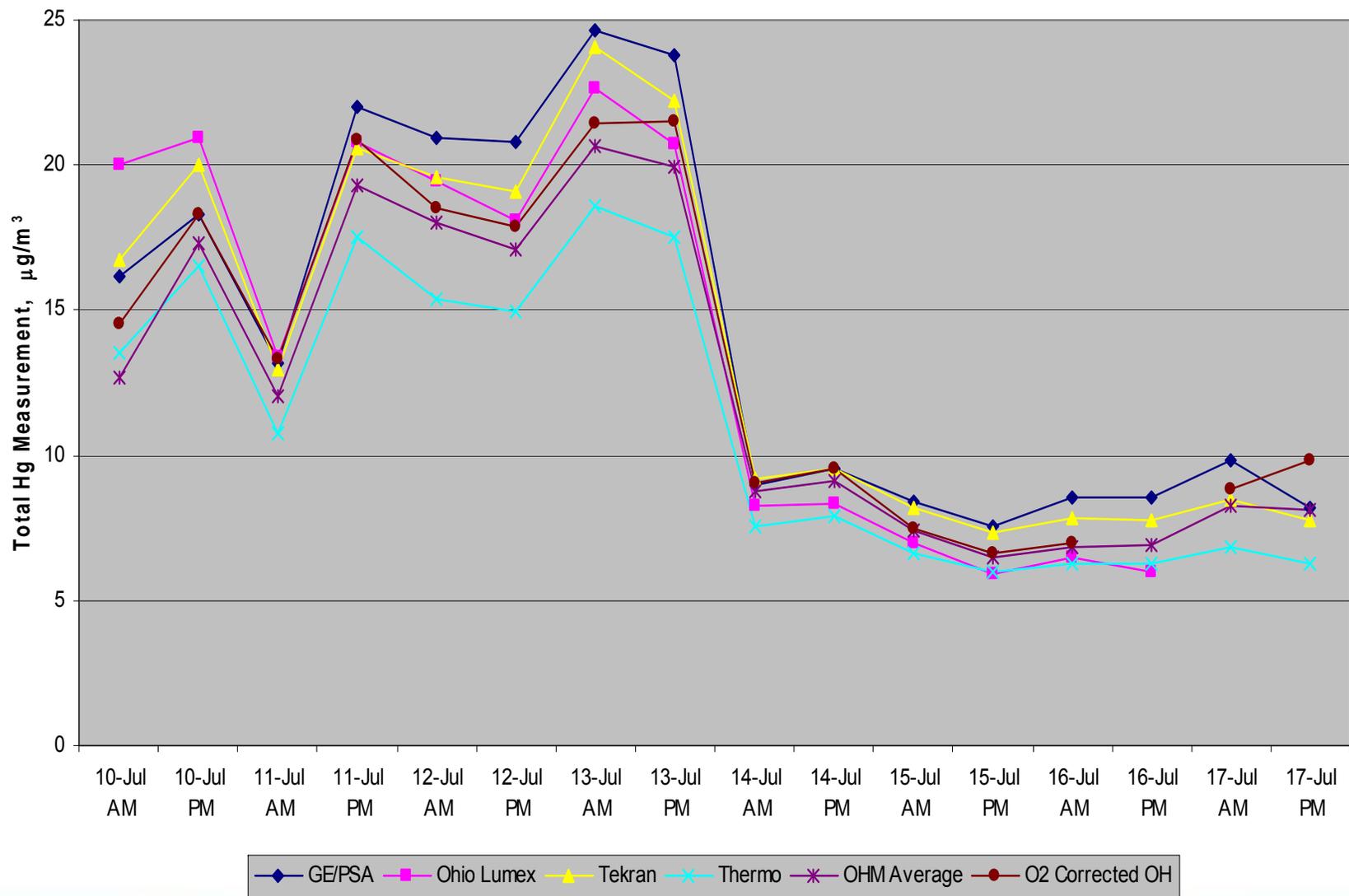


Available Systems

- 1 “portable” Hg CEMS
 - Ohio Lumex (fixed mount)



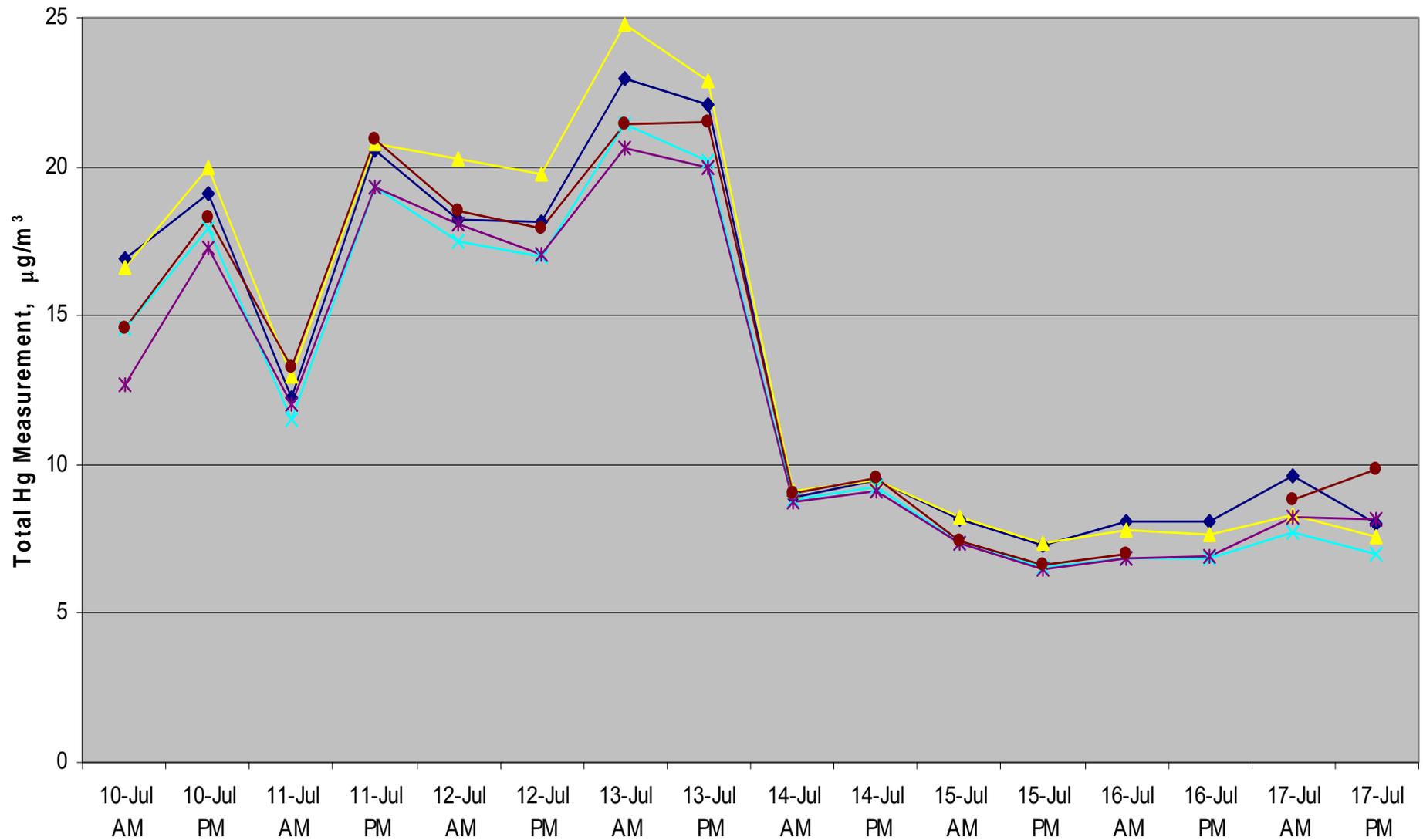
Total Hg Measurements, Normalized for Oxidized Cal Error



RESEARCH & DEVELOPMENT

Building a scientific foundation for sound environmental decisions

Total Hg Measurements, Normalized for Drift



Build

RESEARCH & DEVELOPMENT

◆ GE/PSA
 ▲ Tekran
 ✱ Thermo
 ✱ OHM Average
 ● O2 Corrected OH

Dynamic Spiking Data Sheet (Drift Corrected)

Facility name:	Armstrong	Date:	18-Jul-06	Time:	9:12
Unit(s) tested:	Unit 2	Test personnel:	JEB, NFR		
Analyzer make & model:	████████	Estimated native Hg concentration:	6.6 $\mu\text{g}/\text{m}^3$		
Serial number:		Estimated unspiked sample flow rate:	48.25 slpm		
Calibration span	20 $\mu\text{g}/\text{m}^3$	Estimated spike gas flow rate:	4.7 slpm		

Preliminary Data

Target Level	C_{target} ($\mu\text{g}/\text{m}^3$)		C_{spike} ($\mu\text{g}/\text{m}^3$)		Selected C_{spike} Value ($\mu\text{g}/\text{m}^3$)	Expected ¹ C_{ss} ($\mu\text{g}/\text{m}^3$)
	Upper	Lower	Upper	Lower		
High	13.2	11.9	75	61	69	12.7
Low	10.6	9.3	47	34	46	10.5

¹Calculated from the selected spike gas concentration, using Equation 6

Spiking Data

Target Level	Q_{probe} (lpm)	Q_{spike} (lpm)	C_{ss} ($\mu\text{g}/\text{m}^3$)	C_{native} ($\mu\text{g}/\text{m}^3$)			% Spike Recovery	
				Pre	Post	Avg.		
High	48.22	4.66	11.55	5.98	6.54	6.26	X 104.2% X	
	48.08	4.66	11.45	6.54	6.76	6.65	X 94.3% X	
	48.29	4.66	12.97	6.76	7.01	6.89	103.10%	
	48.31	4.67	14.09	7.01	7.49	7.25	102.95%	
	48.42	4.67	14.48	7.49	7.67	7.58	103.86%	
	48.26	4.67	13.80	7.67	7.51	7.59	X 93.2% X	
						Avg.	103.3%	
						RSD	0.5%	
Low	48.33	4.66	12.93	7.51	8.42	7.97	X 111.7% X	
	48.34	4.65	12.59	8.42	8.20	8.31	X 96.6% X	
	48.35	4.66	13.05	8.20	8.37	8.29	107.17%	
	48.23	4.66	13.24	8.37	8.85	8.61	103.80%	
	48.37	4.66	14.04	8.85	9.46	9.16	109.41%	
						Avg.	106.8%	
						RSD	2.6%	



RESEARCH & DEVELOPMENT

Building a scientific foundation for sound environmental decisions

RATA of ██████ Normalized to Integrity/Drift Checks

Concentrations corrected to 20°C, wet basis

Test #	Date	Times	██████ μg/wsm ³	OH μg/wsm ³	Difference μg/wsm ³	Status
3	10-Jul	09:33-11:15	16.62	12.66	3.96	Discarded
4	10-Jul	14:00-15:38	19.96	17.29	2.67	Excluded
5	11-Jul	09:00-10:35	12.99	12.02	0.97	Included
6	11-Jul	14:00-15:30	20.78	19.28	1.50	Included
7	12-Jul	09:05-10:45	20.24	18.04	2.20	Included
8	12-Jul	14:00-15:37	19.72	17.08	2.64	Included
9	13-Jul	09:13-10:47	24.77	20.65	4.12	Excluded
10	13-Jul	13:55-15:30	22.85	19.98	2.88	Excluded
11	14-Jul	09:17-11:02	9.12	8.73	0.39	Included
12	14-Jul	14:08-15:45	9.49	9.11	0.38	Included
13	15-Jul	09:30-11:05	8.22	7.38	0.84	Included
14	15-Jul	14:00-15:35	7.35	6.49	0.87	Included
15	16-Jul	09:40-11:20	7.79	6.83	0.96	Included
16	16-Jul	13:10-14:48	7.67	6.90	0.77	Included
17	17-Jul	09:15-10:52	8.30	8.25	0.05	Included
18	17-Jul	14:08-15:46	7.57	8.14	-0.57	Included
Arithmetic Mean of all runs			13.97	12.43	1.54	μg/wsm³

Total valid runs	15	----- Arithmetic Means -----		
		██████	OH	Difference
		13.79	12.41	1.38 μg/wsm ³
		Bias Test: Passed		
Factor: 0.90		Confidence Coeff.	0.70 μg/wsm³	
		Relative Accuracy	16.76%	

Runs included in RA calculation	12	----- Arithmetic Means -----		
		██████	OH	Difference
		11.61	10.69	0.92 μg/wsm ³
		Bias Test: Passed		
Factor: 0.92		Confidence Coeff.	0.56 μg/wsm³	
		Relative Accuracy	13.82%	

Runs with Hg below 10 μg/wsm ³	8	----- Arithmetic Means -----		
		██████	OH	Difference
		8.19	7.73	0.46 μg/wsm ³
		Bias Test: Passed		
Factor: 0.94		Confidence Coeff.	0.44 μg/wsm³	
		Relative Accuracy	11.63%	

Runs with Hg above 10 μg/wsm ³	7	----- Arithmetic Means -----		
		██████	OH	Difference
		20.19	17.76	2.43 μg/wsm ³
		Bias Test: Passed		
Factor: 0.88		Confidence Coeff.	0.94 μg/wsm³	
		Relative Accuracy	18.95%	



Study Conclusions ...

- Not all systems passed all IRM performance criteria, but all criteria capable of being achieved
- For the most part, minimal Hg CEM measurement biases relative to OHs ($<\pm 10\%$)
 - Consistent with dynamic spiking recoveries
 - Lack of bias is intended outcome of dynamic spiking procedure – **dynamic spiking served its purpose!**
- Drift correction of data improved IRM data quality
 - Consistent with existing methods (7E, 6C)
- Test results supportive of proposed Method 30A performance criteria



Lehigh Tests - Practical Issues ...

- Spent significant amount of time trying to perform procedures on analyzers that aren't optimized for doing IRM
 - Not conducive to dynamic spiking (probe, flow, spiking, etc)
- Test Equipment:
 - Not capable of traversing
 - Required considerable set-up time
 - Certainly not portable!
- IRM-specific test equipment is needed to make Method 30A viable

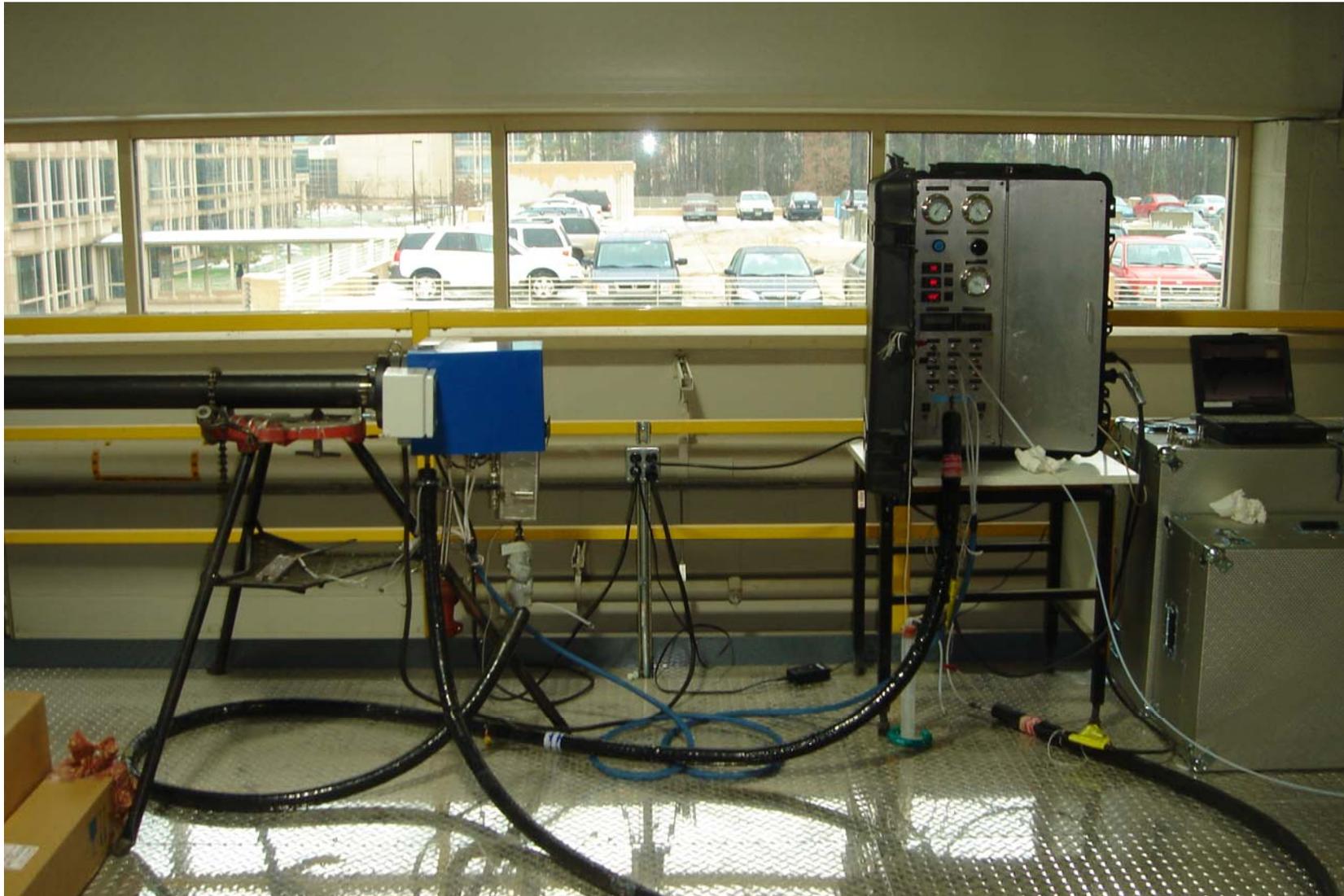


Current Focus

- Develop IRM-Specific Equipment
 - Simple
 - Portable
 - Practical
 - Traversable Probes
 - Less expensive
 - Working with Multiple Vendors
- Perform Stratification Testing
- Perform IRM testing at multiple test sites
- **Objective: Fully perform (successfully) IRM in <2 days**



Lumex/M&C IRM System



RESEARCH & DEVELOPMENT

Building a scientific foundation for sound environmental decisions

EPA/APEX/PSA/THERMO IRM System



RESEARCH & DEVELOPMENT

Building a scientific foundation for sound environmental decisions

Field Testing Status ...



- Well ... We're getting there ...
- Tracer gas approach works well
- Successful dynamic spiking with new probes
- Stratification testing still a challenge due to equipment limitations
- Still struggling with equipment issues – All minor yet annoying
 - Drift
 - Filter
 - Orifice dilution ratio
- Still early in the field testing stage
- We're relying on vendors to develop necessary IRM equipment for us to test!



Development of Hg Method 30B Sorbent Trap Reference Method

- Field data with Appendix K RATA tests very supportive
- Rapid advances of “Thermal” analysis techniques
- Increased understanding of “Wet” analysis techniques
- Reliable Hg⁰ spiking techniques

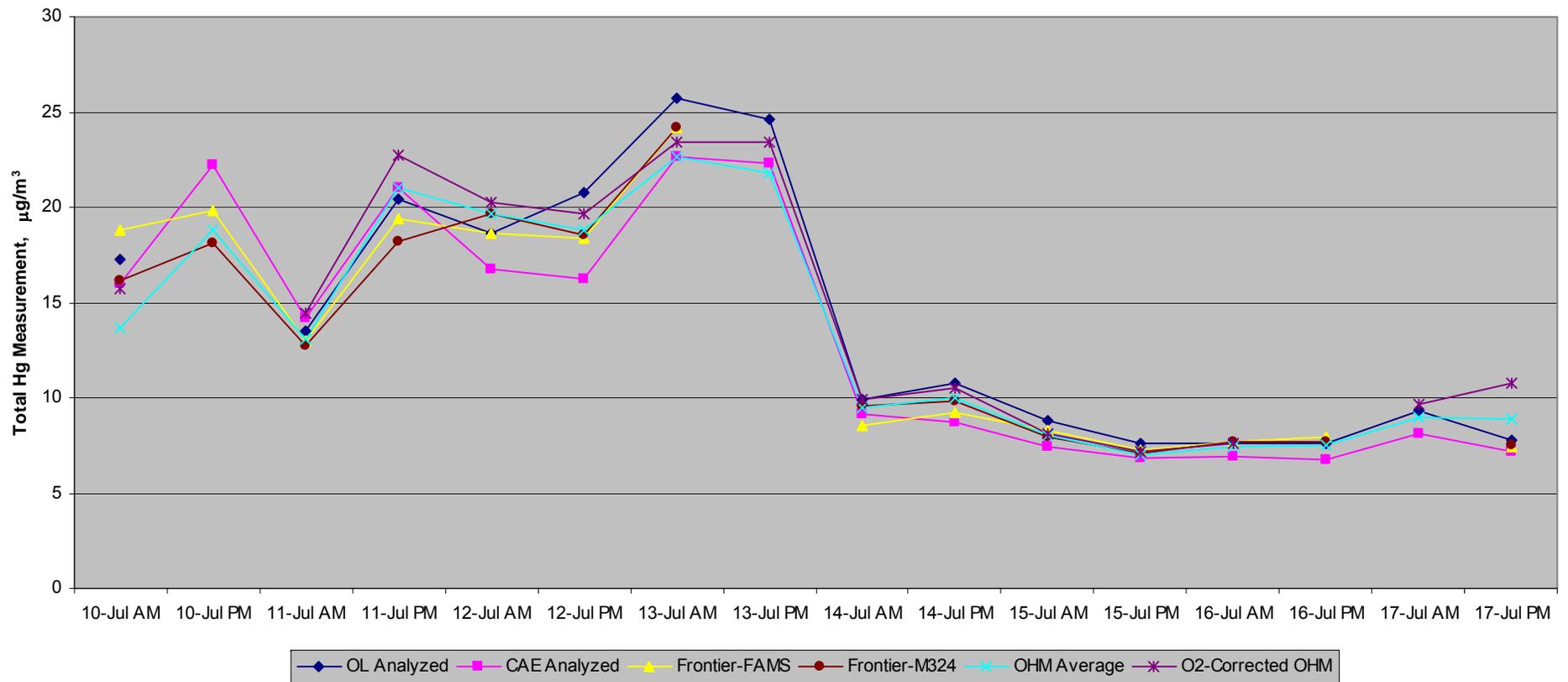


RESEARCH & DEVELOPMENT

Building a scientific foundation for sound environmental decisions

Sorbent Traps vs. OHM

Total Hg Measurements, Sorbent Trap and OHM



RESEARCH & DEVELOPMENT

Building a scientific foundation for sound environmental decisions

Method 30B - Sorbent Trap Reference Method for Mercury

- Performance-based
 - Amenable to new sorbents, equipment, and analytical technologies
 - Lab verification of sorbent performance and analysis
 - Site/Test program-specific verification of data quality
- Capability for timely results
- Minimization of trap spiking, especially in the field
- Description
 - Known volume of stack gas is sampled through paired, in-stack 2-section sorbent traps (e.g., iodated carbon)
 - Analysis by any suitable system that can meet performance criteria (e.g., leaching, digestion, thermal desorption/direct combustion coupled with UV AF, UV AA, XRF)



Method 30B - Sorbent Trap Reference Method for Mercury, cont

- Key QA Elements
 - Laboratory
 - Matrix interference test (for leaching analysis)
 - Minimum sample mass determination
 - Analytical bias test
 - Field (for each test)
 - Field recovery test (assess bias by static spiking)
 - Paired train agreement (assess precision)
 - Trap second section breakthrough



Approach

- Overall focus is how to determine just how much sample must be collected that can be measured reliably
- Largely about characterizing your analytical capabilities
 - How low can you go?
 - Can you measure accurately at that level?
 - Are there any measurement interferences?
 - How does that compare to your background levels?
- Tests and criteria for characterizing measurement performance



Analytical Tests

- Analytical Matrix Interference Test
 - For “wet” analysis technique only
 - Serial dilution test to determine minimum dilution ratio (if any)
 - One time



Analytical Tests

- Minimum Sample Mass Determination
 - Purpose is to identify the lowest Hg mass (or concentration) that you can measure reliably that can be used to target the mass of Hg you need for sampling
 - Falls within your calibration curve ($\geq 2x$ lowest point)
 - Factors in liquid sample volumes, sample dilutions, etc.
 - Considers your detection limits, trap background levels
 - Serves as lower bound for Analytical Bias Test
 - Used to estimate target sample collection volumes and run times



Analytical Tests

- Analytical Bias Test
 - Purpose is to confirm that the minimum sample mass you've identified can be measured accurately by your intended analytical technique
 - Confirm acceptable recovery of Hg^0 and HgCl_2
 - Done at two levels that define the lower and upper bound that actual samples must fall within
 - One time
 - Can expand when needed



Field Testing

- Field Recovery Test
 - Purpose is a test-specific assessment of overall measurement performance (bias)
 - Based on Field Recovery Test in Method 18
 - Confirm acceptable recovery of Hg^0 from 3 pairs of spiked/unspiked trains
 - Analogous to dynamic spiking-the static method of standard addition
 - Potential for Field Recovery Test runs to qualify as RATA runs too
 - Does not have to be performed before collecting field samples, but must be successfully performed for each RATA or field test



Field Testing

- Test Runs
 - Use target sample mass, estimated stack concentration and nominal sample rate to estimate target sample volume and run time
 - Minimum run time of 30 min
 - For RATA testing, run time relief for stack concentrations $<0.5 \mu\text{g}/\text{m}^3$
 - Paired train agreement:
 - $\leq 10\%$ RD for Hg concentrations $>1 \mu\text{g}/\text{m}^3$
 - $\leq 20\%$ RD for Hg concentrations $<1 \mu\text{g}/\text{m}^3$ or $\leq 0.2 \mu\text{g}/\text{m}^3$ absolute difference



Sample Analysis

- Section 1 analyses must be within valid calibrated range
- Sample analyses must be bracketed by valid Continuing Calibration Verification checks
- Section 1 analyses must also be within bounds of Analytical Bias Test
- Sorbent trap Section 2 breakthrough/background
 - $\leq 10\%$ of Section 1 Hg mass for Hg concentrations $> 1 \mu\text{g}/\text{m}^3$
 - $\leq 20\%$ of Section 1 Hg mass for Hg concentrations $> 1 \mu\text{g}/\text{m}^3$
- Additional guidance for estimating Hg levels below lowest point in calibration
 - Section 2
 - Section 1 where stack Hg $< 0.5 \mu\text{g}/\text{m}^3$



Examples and Field Data

- EPA/ORD Thermal Analysis
 - MDL ~1 ng
 - Lowest point in cal curve = 10 ng
 - Minimum sample mass = 20 ng
 - Recoveries within 90-110% for Hg⁰ and HgCl₂
 - Nominal sample flow rates: 400, 800 cc/min
 - Estimated run time for 0.5 µg/m³ stack Hg @ 400 cc/min sample rate = 100 min
 - Estimated run time for 0.5 µg/m³ stack Hg @ 800 cc/min sample rate = 50 min



Field Recovery Tests

- Quad probes, paired spiked/unspiked trains at 400, 800cc/min, 1 hour runs
- Recoveries: avg = 104%, range 97-114%, n = 8
- Average stack Hg concentration = $0.79 \mu\text{g}/\text{m}^3$
 - Average @ 400 cc/min = $0.78 \mu\text{g}/\text{m}^3$
 - Average @ 800 cc/min = $0.75 \mu\text{g}/\text{m}^3$
 - Average unspiked = $0.76 \mu\text{g}/\text{m}^3$
 - Average spike subtracted = $0.81 \mu\text{g}/\text{m}^3$
- RDs: All RDs <10% and < $0.2 \mu\text{g}/\text{m}^3$ absolute dif.
- Breakthroughs <2%
- Analyses were **not** performed in the field!



What Should You Know?

- Focus on RM performance criteria
 - Most is consistent with existing RMs
 - Method 30B is largely analytically-oriented
- Guidance and training will come!
 - CAMD training
 - Web sites for guidance documents and lessons learned
 - Data reports



Questions ...

- IRM?
- IRM Equipment?
- Sorbent Trap RM?
- Other?

