

Note: This is a reference cited in AP 42, *Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at www.epa.gov/ttn/chief/ap42/

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Background Report Reference

AP-42 Section Number: 11.23

Background Chapter: 4

Reference Number: 39

Title: Results of the August 20, 1985
Particulate and SO₂ Emission Test on
the No. 1 Machine Wind Box and
Hood Exhaust No. 3 Scrubber Stack at
the Hibbing Taconite Company plant

Interpoll, Inc.

Interpoll, Inc.

August 1985

RSF 73 4-39

AP-42 Section	11.23
Reference	—
Report Sect.	4
Reference	39

Interpoll Inc.
4500 Ball Road N.E.
Circle Pines, Minnesota 55014
Telephone (612) 786-6020

**RESULTS OF THE AUGUST 20, 1985,
PARTICULATE AND SO₂ EMISSION TEST
ON THE NO. 1 MACHINE WIND BOX AND HOOD
EXHAUST NO. 3 SCRUBBER STACK AT
THE HIBBING TACONITE COMPANY PLANT**

Submitted to:

HIBBING TACONITE COMPANY
P. O. Box 589
Hibbing, Minnesota 55754

Attention: Bob Ives

Approved by:



Perry Lonnes, Ph.D.
President

Report Number 5-2066
August 30, 1985

TACON 3

Source category: Taconite ore processing Date:
 Plant name : Hibbing Taconite Co. Location:
 Test date : August 20, 1985 Ref. No.:
 Process : Traveling grate Basis for process rate :

Source	Type of control	Pollutant	Run No.	Emission rate, lb/hr	Process rate, ton/hr	Emission factor			
						kg/Mg	lb/ton		
Traveling grate (petroleum coke and natural gas fired) Line 1 Stack 3	Multi-cyclone and wet venturi scrubber	filt. PM	1	24.97		ERR	ERR		
		AVERAGE							
		SO2	1	119.80		ERR	ERR		
		AVERAGE						ERR	ERR
		CO2	1	7227.0		ERR	ERR		
		AVERAGE						ERR	ERR
		AVERAGE						ERR	ERR
		AVERAGE						ERR	ERR

No process data provided.

Traveling grate assumed from TACON 1

- Delete TACON 3

- NO PROCESS DATA

- ONLY 1 RUN

(del) 4/28

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4500 Ball Road N.E.
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RESULTS OF THE AUGUST 20, 1985,
PARTICULATE AND SO₂ EMISSION TEST
ON THE NO. 1 MACHINE WIND BOX AND HOOD
EXHAUST NO. 3 SCRUBBER STACK AT
THE HIBBING TACONITE COMPANY PLANT

TACON 3
4/23

DELETE

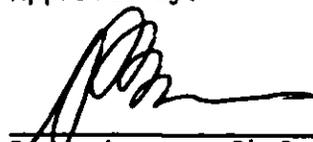
- NO PROCESS DATA
- ONLY 1 RUN

Submitted to:

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P. O. Box 589
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Report Number 5-2066
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ABBREVIATIONS

ACFM	actual cubic feet per minute
cc (ml)	cubic centimeter (milliliter)
DSCFM	standard cubic foot of dry gas per minute
DSML	dry standard milliliter
DEG-F (°F)	degrees Fahrenheit
DIA.	diameter
FT/SEC	feet per second
GPM	gallons per minute
GR/ACF	grains per actual cubic foot
GR/DSCF	grains per dry standard cubic foot
g	gram
HP	horsepower
HRS	hours
IN.	inches
IN. HG.	inches of mercury
IN. WC.	inches of water
LB	pound
LB/DSCF	pounds per dry standard cubic foot
LB/HR	pounds per hour
LB/10 ⁶ BTU	pounds per million British Thermal Units heat input
LB/MMBTU	pounds per million British Thermal Units heat input
MW	megawatt
mg/DSCM	milligrams per dry standard cubic meter
microns (μm)	micrometer
MIN.	minutes
ohm-cm	ohm-centimeter
PPH	pounds per hour
PPM	parts per million
PSI	pounds per square inch
SQ. FT.	square feet
v/v	percent by volume
w/w	percent by weight

Standard conditions are defined as 68 °F (20 °C) and 29.92 IN. of mercury pressure.

1 INTRODUCTION

On August 20, 1985, Interpoll Inc. personnel conducted a particulate and SO₂ emission test on the No. 1 Machine Wind Box and Hood Exhaust No. 3 Scrubber at the Hibbing Taconite Company Plant located in Hibbing, Minnesota. On-site testing was performed by a two-man team under the direction of Dr. P. Lonnes. Coordination between plant operation and testing activities was provided by Steve Rogers and other plant personnel of the Hibbing Taconite Company.

Evaluations were performed in accordance with EPA Methods 1-6, CFR Title 40, Part 60, Appendix A (revised July 1, 1984). Previously collected data was used to select the appropriate nozzle diameter required for isokinetic sample withdrawal. An Interpoll sampling train which meets or exceeds specifications in the above-cited reference was used to collect the samples by means of a heated stainless steel-lined probe. Sulfur dioxide determinations were performed in accordance with the large impinger version of EPA Method 6 using the back half of the Method 5 sampling train.

An integrated flue gas sample was extracted simultaneously with each particulate sample using a specially designed gas sampling system. Integrated flue gas samples were collected in 44-liter Tedlar bags. After sampling was complete, the bags were sealed and returned to the laboratory for Orsat analysis.

Testing on the No. 3 Machine Wind Box and Hood Exhaust No. 4 Scrubber was conducted from two test ports oriented at 90 degrees approximately 76 feet downstream of a transition breeching and 14 feet upstream of the stack outlet. A 12-point traverse was used to collect representative flyash samples from the scrubber stack. Each traverse point was sampled five minutes to give a total sampling time of 60 minutes per run.

The important results of the tests are summarized in Section 2. Detailed results are presented in Section 3. Results of preliminary measurements, field data and all other supporting information are presented in the appendices.

SUMMARY AND DISCUSSION

The important results of the particulate emission test is summarized in Table 1. As will be noted, only one particulate concentration determination was performed due to plant difficulties. The results of the sulfur dioxide test is summarized in Table 2.

No difficulties were encountered in the field or in the laboratory evaluation of the flue gas, particulate and sulfur dioxide samples. On the basis of this fact and a complete review of the entire data and results, it is our opinion that the emission rates and emission factors reported herein are accurate and closely reflect the actual values which existed at the time the tests were performed.

TABLE 1. SUMMARY OF THE RESULTS OF THE 8-20-85 PARTICULATE EMISSION TEST ON THE LINE 1 NO. 3 SCRUBBER STACK AT THE HIBBING TACONITE COMPANY LOCATED IN HIBBING, MINNESOTA

ITEM	
DATE OF TEST	8-20-85
TIME OF TEST (HRS)	1535/1640
PROCESS RATE *	
VOLUMETRIC FLOW	
ACTUAL (ACFM)	160000
STANDARD (DSCFM)	132000
GAS TEMPERATURE (DEG-F)	109
GAS MOISTURE CONTENT (% V/V)	7.28
GAS COMPOSITION (% V/V, DRY)	
CARBON DIOXIDE	.80
OXYGEN	19.60
NITROGEN	79.60
ISOKINETIC VARIATION (%)	100.5
PARTICULATE CONCENTRATION	
ACTUAL (GR/ACF)	.018
STANDARD (GR/DSCF)	.022
PARTICULATE EMISSION RATE (LB/HR)	24.97

*Not provided by time of report preparation.

TABLE 2. SUMMARY OF THE RESULTS OF THE AUGUST 20, 1985 SULFUR DIOXIDE EMISSION TEST ON THE LINE 1 NO. 3 SCRUBBER STACK AT THE HIBBING TACONITE COMPANY LOCATED IN HIBBING, MINNESOTA

ITEM	RUN 1
TIME OF TEST (HRS)	1535-1640
S02 CONCENTRATION (PPM/DRY)	91
S02 CONCENTRATION (PPM/WET)	84
S02 MASS RATE (LB/HR)	120
S02 EMISSION FACTOR (LB/MMBTU) * F-FACTOR METHOD	2.31

*Emission factor calculated based on an estimated F-Factor = 9499 DSCF/10⁶ BTU

3 RESULTS

The results of all field and laboratory evaluations are presented in this section. Gas composition results (Orsat and moisture) are presented first followed by the computer printout of the particulate and sulfur dioxide emission determinations. Preliminary measurements including test port locations are given in the appendices.

The results have been calculated on an IBM PC Computer using programs written in Extended BASIC specifically for source testing calculations. EPA-published equations have been used as the basis of the calculation techniques in these programs.

The particulate emission rate has been calculated using the product of the concentration times flow method (as recommended by the EPA) rather than the ratio of areas method. The dry particulate emission factor has been calculated by the Oxygen F-Factor Method using the latest EPA-published dry F-Factor for the stated fuel.

TEST NO. 1
LINE 1 NO. 3 SCRUBBER STACK

3.1 RESULTS OF ORSAT & MOISTURE ANALYSES -- METHOD 3 & 4 (% V/V)

	RUN 1
DATE OF RUN	8-20-85
DRY BASIS (ORSAT)	
CARBON DIOXIDE	.80
OXYGEN	19.60
CARBON MONOXIDE	0
NITROGEN	79.60
WET BASIS (ORSAT)	
CARBON DIOXIDE	.74
OXYGEN	18.17
CARBON MONOXIDE	0
NITROGEN	73.81
MOISTURE CONTENT	7.28
DRY MOLECULAR WEIGHT	28.91
WET MOLECULAR WEIGHT	28.12
SPECIFIC GRAVITY (RELATIVE TO AIR)	.9713
VELOCITY WEIGHTED AND TIME-AVERAGED OXYGEN DURING METHOD 5 RUNS	19.27

TEST NO. 1
LINE 1 NO. 3 SCRUBBER STACK

3.2 RESULTS OF PARTICULATE LOADING DETERMINATIONS -- METHOD 5

RUN 1

DATE OF RUN		8-20-85
TIME RUN START/END(HRS)		1535/1640
PITOT TUBE COEFFICIENT		.840
WATER IN SAMPLE		
CONDENSATE	(ML)	50.6
SILICA GEL	(GRAMS)	25.0
TOTAL PARTICULATE MATER-		
IAL COLLECTED	(GRAMS)	.0651
VM	(MCF)	47.16
Y		1.0094
PBAR	(IN.HG)	28.52
DELTA-H	(IN.WC)	2.11
TM	(DEG-F)	70.70
VMSTD	(DSCF)	45.37
TOTAL SAMPLING TIME	(MIN)	60.0
NOZZLE DIAMETER	(IN)	.251
AVG STACK GAS TEMP	(DEG=F)	109
VOLUMETRIC FLOW		
ACTUAL	(ACFM)	160494
DRY STANDARD ...	(DSCFM)	131610
ISOKINETIC VARIATION	(%)	100.5
PARTICLE CONCENTRATION*		
ACTUAL	(GR/ACF)	.0182
DRY STANDARD .	(GR/DSCF)	.0221
PARTICLE MASS FLOW	(LB/HR)	24.97

* DRY CATCH ONLY

TEST NO. 1
LINE 1 NO. 3 SCRUBBER STACK

3.3 RESULTS OF SULFUR DIOXIDE DETERMINATIONS -- METHOD 6

	RUN 1
DATE OF RUN	8/20/85
TIME RUN START/END (HRS)	1535/1640
BAROMETRIC PRESSURE (IN.HG.)	28.52
METER TEMPERATURE (DEG-F)	70.70
METER CORRECTION COEFFICIENT	1.0094
VOLUME THROUGH GAS METER	
AT METER CONDITIONS...(CF)	47.160
STANDARD CONDITIONS...(SCF)	45.372
TOTAL SAMPLING TIME (MIN)	60.0
MOISTURE CONTENT (%V/V)	7.28
OXYGEN CONTENT (%V/V DRY)	19.60
MILLIEQUIVALENTS OF SO4 IN GAS SAMPLE	9.7215
SULFUR DIOXIDE CONCENTRATION	
(GR/DSCF)	.1059
(MG/DSCM)	242
(PPM-DRY)	91
(PPM-WET)	84
DRY STANDARD VOLUMETRIC FLOW (DSCFM)	132000
SULFUR DIOXIDE MASS RATE (LB/HR)	119.80

APPENDIX A

SAMPLING TRAIN CALIBRATION DATA

Interpoll Inc.
(612)786-6020

Meter Box Calibration and Usage Status

Date of Report: August 20, 1985

Meter Box No.: 3

Date of Last Calibration: May 22, 1985

Calibration Technician: E. Trowbridge

Wet Test Meter No.: American Meter AI-20

Date of Use	Report No.	Initial Meter Reading	Final Meter Reading	Volume/Job (cu. ft.)	Total Volume* (cu. ft.)
July 16, 1985	5-2040	659.17	824.46	165.29	165.29
July 30, 1985	5-2046	824.80	981.17	156.37	321.66
August 20, 1985	5-2066	981.64	1028.80	47.16	368.82

*Total volume through meter since last calibration

Interpoll Inc.
(612)786-6020
Nozzle Calibration
Data Sheet

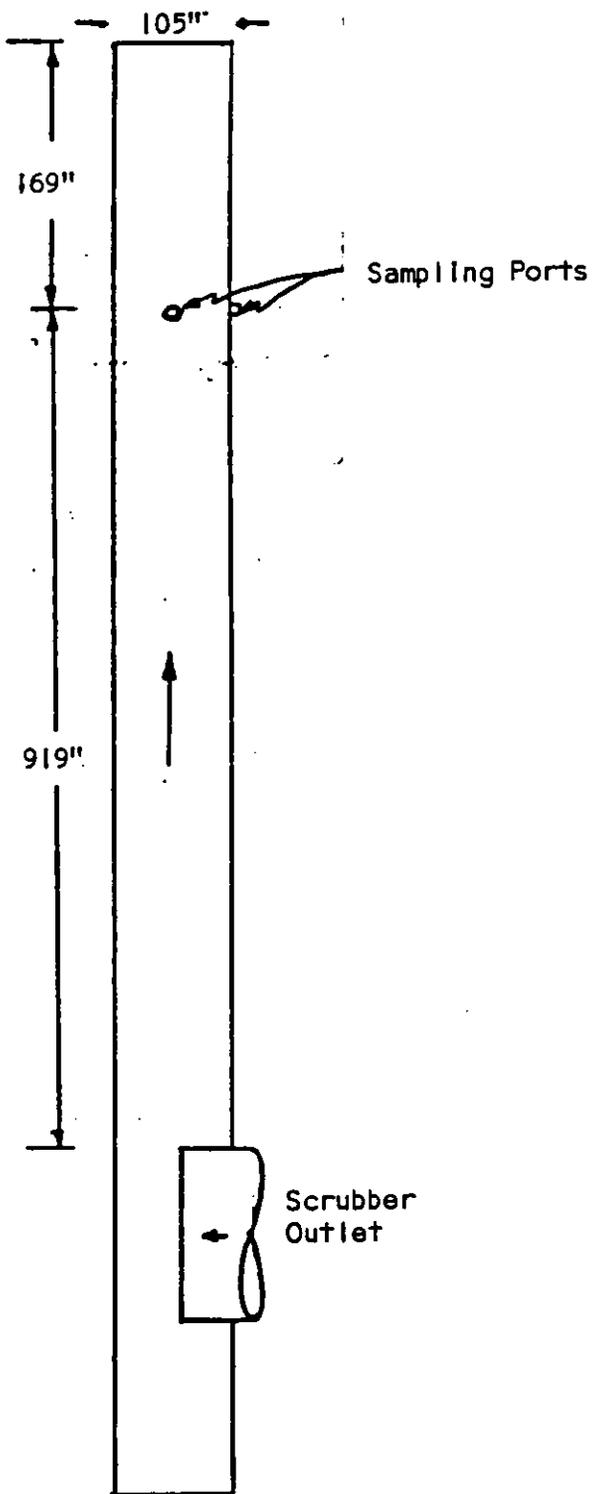
Date of Calibration: August 20, 1985
Technician: E. Trowbridge

Nozzle Number 7-2

Nozzle rotated by 36 degree increments and diameter measured to nearest 0.001 inch. Observed readings and average:

Position	Diameter (inches)
1	.250
2	.251
3	.251
Average:	.251

HOOD AND WINDBOX EXHAUST STACK



NOT TO SCALE

Sample Log Sheet

Job HIBBING TACONITE
 Unit and Sampling Location NO. 1 MACHINE (INDURATION FURNACE) W8 H EXHAUST NO. 3 SCABBER STALK
 Date 8-20-85 Test 1 Run 1

Sample Train: Leak check pretest : < 0.02 cfm at 15 in. Hg. (vac.)
 posttest: 0 cfm at 9 in. Hg. (vac.)

Particulate Sample Type and Number	Weights		
	Final	Tare	Difference
FILTER No. 5572			
PW			
Total (M _n)			0

Impinger Condenser

Impinger Number	Weights		
	Final	Tare	Difference
IMPINGERS	315.0	264.4	50.6
DESICCANT	1327	1302	25
Total (M _{w0})			75.6

Integrated Flue Gas Sampling Data

FG Train No. B-1 Box No. 2 Bag No. 1
 Bag Material TEDLAR Bag Volume 44 L
 Pretest leak check 0 cc/min at 15 in. Hg. 320-P No. 7

Time	cc/min.	Time	cc/min.
(1535)			
	400		
	↓		
(1640)			

Operator R HOLTZ 320-P Reading _____ % v/v

APPENDIX D

LABORATORY DATA SHEETS

7502

Interpoll Inc.
(612)786-6020

Pg ___ of ___

Chain of Custody
Sample Deposition Sheet

Job HIBBING TACONITE

Site LINE 1 No.3 SCAUBBER STACK

Team Leader R HOLTZ

Priority NORMAL

Date Submitted 8-22-85

Received by _____

Test No(s). 1

Number of Runs Completed 1

No. of Samples	Type of Sample	Analysis Required	Date Analyzed	Other Information or Special Instructions
1	Probe Wash Solvent(s):	<input checked="" type="checkbox"/> As per EPA 5 <input type="checkbox"/> _____		
1	Filter: <input checked="" type="checkbox"/> 4" glass fiber <input type="checkbox"/> Stainless steel <input type="checkbox"/> ERC	<input checked="" type="checkbox"/> As per EPA 5 <input type="checkbox"/> _____		
3	Wet Catch	<input type="checkbox"/> Organic <input type="checkbox"/> Inorganic As per EPA 5 Other: _____		
1	Orsat	CO ₂ /O ₂		
1	SO ₂			
	NO _x			
	Fuel Sample			

Source Information

- Type of source: Boiler Asphalt Plant Incinerator Other FURNACE
- Fuel: Coal Wood Gas Oil Other COKE
- Is sample combustible? No Yes
- Does sample need special desiccating? No Yes If yes, explain _____
- Is particle size analysis required No Yes If yes, how many run(s)? _____
run preference _____

INTERPOLL INC.
 EPA Method 5 Probe (Cyclone) Wash
 Gravimetric Analysis Laboratory Data Sheet
 (CFR Title 40 Part 60 Appendix A)

Date of Analysis 8-26-85
 Technician ReV

EPA-M5 Acetone R.B SPEC $\leq 7.8 \mu\text{g/cc}$
 Actual acetone residue blank 6.3 $\mu\text{g/cc}$

1

interpoll

Job HIS TAL Date 8-26-85
 City/State _____ Log # 2502-01
 Source LINEN No 3 SCRUBBER
 Test Site STACK
 Sample type PW Tech RH
 Remarks: _____ Test/Run 1/1
 _____ of _____

Special Handling Required _____

Evaporating Dish No. 66
 Volume of acetone 85 cc
 E. Dish Tare Wt. 84.2729 g
 E. Dish + Sample Wt. 84.3250 g
 Comments _____

2

interpoll

Job _____ Date _____
 City/State _____ Log# _____
 Source _____
 Test Site _____
 Sample type _____ Tech _____
 Remarks: _____ Test/Run _____
 _____ of _____

Special Handling Required _____

Evaporating Dish No. _____
 Volume of acetone _____ cc
 E. Dish Tare Wt. _____ g
 E. Dish + Sample Wt. _____ g
 Comments _____

3

interpoll

Job _____ Date _____
 City/State _____ Log# _____
 Source _____
 Test Site _____
 Sample type _____ Tech _____
 Remarks: _____ Test/Run _____
 _____ of _____

Special Handling Required _____

Evaporating Dish No. _____
 Volume of acetone _____ cc
 E. Dish Tare Wt. _____ g
 E. Dish + Sample Wt. _____ g
 Comments _____

RESULTS:

1/0.0516 D-2 2/

3/

(EPA Method 5) Filter
Gravimetric Analysis Lab
Data Sheet
FR 42(160)

Date of Analysis 8-26-85
Technician REV

1

interpoll

Job Hibbing Tacuote Date 8-20-85
City/State _____ Log# 2502-02
Source Line 1 No 3 Scrubber
Test Site Stac-1C
Sample type _____ Tech RA
Remarks: _____ Test/Run 1/1
_____ of _____

Special Handling Required _____
Filter No. 5572
Filter Type 4" G
Filter Tare Wt. 0.64924 g
Filter + Sample Wt. 0.66278 g
Comments _____

2

interpoll

Job _____ Date _____
City/State _____ Log# _____
Source _____
Test Site _____
Sample type _____ Tech _____
Remarks: _____ Test/Run _____
_____ of _____

Special Handling Required _____
Filter No. _____
Filter Type _____
Filter Tare Wt. _____ g
Filter + Sample Wt. _____ g
Comments _____

3

interpoll

Job _____ Date _____
City/State _____ Log# _____
Source _____
Test Site _____
Sample type _____ Tech _____
Remarks: _____ Test/Run _____
_____ of _____

Special Handling Required _____
Filter No. _____
Filter Type _____
Filter Tare Wt. _____ g
Filter + Sample Wt. _____ g
Comments _____

RESULTS:

1/0.01354

2/

3/

.0651

D-3

INTERPOLL INC.

EPA Methods 6 & 8 Barium Perchlorate Titrant Standardization Sheet

Date of Standardization 8-26-85
 Technician RCV
 Buret No. 502

The normality of the nominal .01 N barium titrant solution is verified by comparing it against a primary standard of 0.0100 N H₂SO₄ solution using Thorin I as the endpoint indicator.

1. Preparation of Primary Standard (.0100 N H₂SO₄)

- a. Quantitatively transfer the contents of one "Dilut-It" ampule (available from J. T. Baker Coom. No. 4699) to a clean 1 L volumetric flask and bring to volume with distilled water, and mix well by shaking and inverting. Solution is 0.100 N H₂SO₄ and is stable for one year or more if protected from evaporation.
- b. Using a 100 ml volumetric pipet, add 100.00 cc of 0.100 N H₂SO₄ solution to clean 1 L volumetric flask and bring to volume with distilled water and mix well by shaking and inverting. Transfer the contents to a standard reagent bottle and label as follows:

.0100 N H₂SO₄
 Primary Standard

Prepared on 7/03/85 by RCV
 Prepared by dilution from "Dilut-It" ampule

2. Standardization of the Barium Perchlorate Titrant

- a. Pipet 25.00 cc of the 0.0100 N H₂SO₄ primary standard into each of the three clean 250 cc Erlenmeyer flasks. Add 100 cc of checked absolute isopropanol to each flask 4, drops of Thorin I indicating solution and mix by swirling. Titrate these standards with the barium perchlorate titrant recording the starting and ending buret readings below.
- b. Prepare and titrate at least one blank in an identical manner (Pipet 25 ml of distilled water in place of the 0.0100 N H₂SO₄ standard). Record the results of this titration below also.

3. Standardization Data

All buret readings should be read to the nearest 1/100 and an adequately long time period waited before reading to allow the liquid on the walls to flow down.

	Blank	1st Titr.	2nd Titr.	3rd Titr.
Initial reading (ml)	<u>0.00</u>	<u>0.00</u>	<u>0.00</u>	<u>0.00</u>
Final reading (ml)	<u>0.02</u>	<u>25.20</u>	<u>25.20</u>	<u>25.20</u>
Volume of titrant (ml)	V_b <u>0.02</u>	V_1 <u>25.20</u>	V_2 <u>25.20</u>	V_3 <u>25.20</u>

4. Calculations

$$V_t = \frac{V_1 + V_2 + V_3}{3} = \frac{(25.20) + (25.20) + (25.20)}{3} = 25.20 \text{ ml}$$

$$N = \frac{0.25}{V_t - V_b} = \frac{0.25}{(25.20) - (.02)} = 0.00998 \text{ Normality}$$

L-214RRR

Interpoll Inc.
(612)786-6020

EPA Method 6
Barium/Thorin I Sulfate
Titration Data Sheet

Job Name: Hibbing Taconite
Test/Run: 111

Date of Analysis 8.26.85
Technician RAV

Step 1 SAMPLE RECOVERY & PREPARATION

Original sample volume (Samp.+ Bottle)
- (65.5 Bottle) = cc

pH of original sample 2.16 Ion exchanged: Yes No

Note 1: If pH > 3.5 adjust with conc. perchloric acid to pH ~ 2.0

Note 2: If ion exchange is required, must use a 10 cc aliquot and wash this through with a 10 cc aliquot of distilled water. Add 80 cc of isopropyl alcohol and titrate.

Step 2 Orig. sample brought to volume (V_{soln}): 100 cc
 250 cc
 500 cc
 cc

Step 3 Dilution procedure: No dilution (DF = 1)
 10 + 100 (DF = 10)
 10 + 250 (DF = 25)
 5 + 250 (DF = 50)
 1 + 250 (DF = 250)
 + (DF =)

SAMPLE IDENTIFICATION

interpoll

Job Hibbing Taconite Date 8-20-85
City/State Log # 2502-04
Source Line 1 4.3 Scribbles
Test Site Stack Tech RA
Sample type Test/Run 111
Remarks: bt

Step 4 Volume taken for titration (V_a):
 1 cc
 5 cc
 10 cc
 25 cc
 cc

Step 5 TITRATION: Normality of titrant (N) 0.00993 Volume of titrant for blank (V_b) 0.02

	Trial 1	Trial 2	Trial 3	Average
Starting volume of buret	<u>0.00</u> cc	<u>19.60</u> cc	<u>0.00</u> cc	
Final volume of buret	<u>19.60</u> cc	<u>39.20</u> cc	<u>19.60</u> cc	
Volume of titrant for aliquot (V_s)	<u>19.60</u> cc	<u>19.60</u> cc	<u>19.60</u> cc	<u>19.60</u> cc

Interpoll Inc.
(612)786-6020

EPA Method 3 Data Sheet (Orsat Analysis)

Technician RC Date of Analysis 8-20-85
 Orsat Analyzer No. 3 Leak Check Performed YES NO
 Job Hibbing Taconite Source Line 1 No 3 Scrubber Stack
 City/State _____ Fuel Type _____ Test Site _____

Sample Identification	Test/Run	Analysis No.	Buret Readings (cc)		Concentration (% v/v, dry)		F _o
			Zero Point	After CO ₂	After O ₂	CO ₂	
Ambient Air			0.00	0.00	20.90	0.00	20.80
F	1/1	1	0.00	0.80	20.40	0.80	19.60
B		2	0.00	0.80	20.40	0.80	19.60
AVG							0.80
F		1					
B		2					
AVG							
F		1					
B		2					
AVG							

F = Flask (250 cc all-glass); B = Tedlar bag (5-layer)

Note 1 Analyses performed as per CFR title 40, Part 60, Appendix A, Method 3; KOH for CO₂ and reduced methylene blue for O₂ absorption using a Burrell Orsat Analyzer with a 100 cc buret.

Note 2 Use PC-3 Program "ORSAT" Version 6-85.

L-0192R

Particulate Loadings and Emission Rates

The particulate emission rates were determined per EPA Methods 1-5, CFR title 40, Part 60, Appendix A (revised July 1, 1984). In this procedure, a preliminary velocity profile of the gases in the flue is obtained by means of a temperature and velocity traverse. On the basis of these values, sampling nozzles of appropriate diameter are selected to allow isokinetic sampling, a necessary prerequisite for obtaining a representative sample.

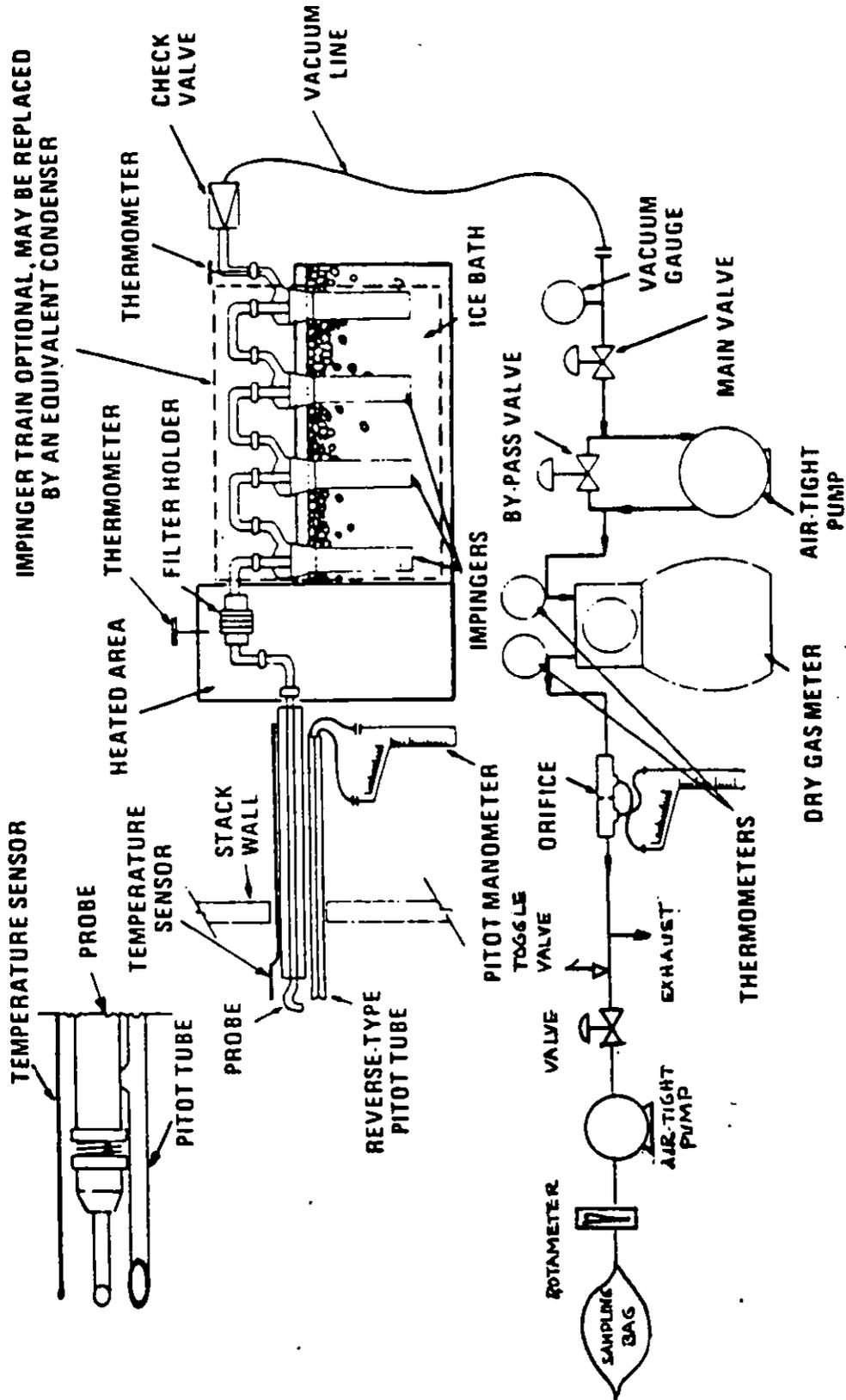
The sampling train consists of a heated stainless steel-lined sampling probe equipped with a Type S pitot and a thermocouple. The probe is attached to a sampling module which houses the all-glass in line filter holder in a temperature controlled oven. In addition, the sampling module also houses the impinger case and a Drierite drying column. The sampling module is connected by means of an umbilical cord to the control module which houses the dry test gasmeter, the calibrated orifice, a leakless pump, two inclined manometers, and all controls required for operating the sampling train.

Particulate samples were collected as follows: The sample gas was drawn in through the sampling probe isokinetically and passed through a 4-inch diameter Gelman Type A/E glass fiber filter. The particulates were removed at this point and collected on the filter. The gases then passed through an ice-cooled impinger train and a desiccant-packed drying column which quantitatively absorb all moisture from the sample gas stream after which the sample gas passes through the pump and the dry test gasmeter which integrates the sample gas flow throughout the course of the test. A calibrated orifice attached to the outlet of the gasmeter provides instantaneous flow rate data.

A representative particulate sample was acquired by sampling for equal periods of time at the centroid of a number of equal area regions in the duct. The sampling rate is adjusted at each site such that an isokinetic sampling condition prevails. Nomographs are used to aid in the rapid determination of the sampling rate.

After sampling is complete, the filter is removed and placed in a clean container. The nozzle and inlet side of the filter holder are quantitatively washed with acetone and the washings are stored in a second container. A brush is often used in the cleaning step to help dislodge deposits. The samples are returned to the laboratory where they are logged in and analyzed. The volume of the acetone rinse ("probe wash") is noted and then the rinse is quantitatively transferred to a tared 120 cc porcelain evaporating dish and the acetone evaporated off at 97-105 °F. This temperature is used to prevent condensation of atmospheric moisture due to the cooling effect induced by the evaporation of acetone. The acetone-free sample is then transferred to an oven and dried at 105 °C for 30 minutes, cooled in a desiccator over Drierite, and then weighed to the nearest .01 mg. The filter sample is quantitatively transferred to a 6-inch watch glass and dried in an oven at 105 °C for two hours. The filter and watch glass are then cooled in a desiccator and the filter weighed to the nearest .01 mg. All weighings are performed in a balance room where the relative humidity is hydrostatted to less than 50% relative humidity. Microscopic examination of the samples is performed if any unusual characteristics are observed. The weight of the acetone rinse is corrected for the acetone blank. The Drierite column is weighed on-site and the water collected by Drierite is added to the condensate so that the total amount of absorbed water may be ascertained.

Integrated gas samples for Orsat analysis were collected at a constant flow rate throughout each particulate run. The gas samples were analyzed using an all-glass Orsat analyzer. Standard commercially prepared solutions were used in the Orsat analyzer (sat. KOH for carbon dioxide and reduced methylene blue for oxygen). In addition to the above, the oxygen content of the flue gas was measured at each traverse during the particulate determinations using a Teledyne Model 320P-4 Portable Oxygen Analyzer to sample the effluent from the Method 5 train.



Particulate sampling train.

Sulfur Dioxide Emissions

The sulfur dioxide concentrations were determined per Method 6, CFR Title 40, Part 60, Appendix A (Revised July 1, 1984). In the procedure employed, the back half of the Method 5 sampling train was used to collect SO₂ samples during the particulate determinations. Each of the first two impingers was filled with 100 cc of freshly prepared 3% hydrogen peroxide. The samples were quantitatively recovered in the field and returned to the laboratory for sulfate analysis by a barium-thorin titration method. Samples were ion-exchanged using a cation exchanger in the hydrogen ion form to remove any interfering cations present.

CALCULATION EQUATIONS

METHOD 2

$$V_s = 85.48 C_p (\sqrt{\Delta p})_{avg} \sqrt{\frac{T_{s(avg)}}{P_s M_s}}$$

$$Q_{s,d} = 60(1 - B_{ws}) V_s A \left(\frac{528}{T_{s(avg)}}\right) \left(\frac{P_s}{29.92}\right)$$

$$Q_a = 60 V_s A$$

$$\dot{m}_g = \frac{4.995 Q_{s,d} G_d}{1 - B_{ws}}$$

$$RH^* = 100 (vp_{twb} - 0.0003641 P_s (T_{db} - T_{wb}))/vp_{tdb}$$

$$B_{ws}^* = RH(vp_{tdb})/P_s$$

$$\rho = \frac{4.585 \times 10^{-2} P_s M_s}{T_s (avg)}$$

*Alternate equations for calculating moisture content from wet bulb and dry bulb data.

CALCULATION EQUATIONS

METHOD 3

$$\%EA = \frac{100(\%O_2 - .5\% CO)}{0.264\% N_2 - \%O_2 + 0.5\% CO}$$

$$M_d = 0.44(\%CO_2) + 0.32 (\%O_2) + 0.28 (\%N_2 + \%CO)$$

$$M_s = M_d (1 - B_{ws}) + 0.18 B_{ws}$$

$$B_{ws} = \frac{V_w(std)}{V_s(std) + V_m(std)}$$

CALCULATION EQUATIONS

METHOD 5

$$V_{m(\text{std})} = 17.65 V_m \gamma \left(\frac{P_{\text{bar}} + \overline{\Delta H}/13.6}{T_{m(\text{avg})}} \right)$$

$$V_{w(\text{std})} = 0.0472 V_{Is}$$

$$B_{ws} = \frac{V_{w(\text{std})}}{V_{w(\text{std})} + V_{m(\text{std})}}$$

$$I = 0.0944 \left(\frac{T_{s(\text{avg})} V_{m(\text{std})}}{P_s V_s A_n \theta (1 - B_{ws})} \right)$$

$$C_s = \frac{15.43 M_p}{V_{m(\text{std})}}$$

$$C_a = \frac{272.3 M_p P_s}{T_{s(\text{avg})} (V_{w(\text{std})} + V_{m(\text{std})})}$$

$$(\dot{m}_p)_1 = 8.5714 \times 10^{-3} C_s Q_{s,d}$$

$$(\dot{m}_p)_2 = \frac{1.3228 \times 10^{-1} M_p A}{O A_n}$$

$$\dot{m}_p = \frac{(\dot{m}_p)_1 + (\dot{m}_p)_2}{2}$$

SYMBOLS

- A = Cross sectional area of stack, SQ. FT.
- A_n = Cross sectional area of nozzle, SQ. FT.
- B_{ws} = Water vapor in gas stream, proportion by volume
- C_p = Pitot tube coefficient, dimensionless
- C_a = Concentration of particulate matter in stack gas, wet basis, GR/ACF
- C_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, GR/DSCF
- EA = Excess air, percent by volume
- γ = Dry test meter correction factor, dimensionless
- G_d = Specific gravity (relative to air), dimensionless
- I = Isokinetic variation, percent by volume
- M_d = Molecular weight of stack gas, dry basis, g/g - mole.
- \dot{m}_g = Mass flow of wet flue gas, LB/HR
- \dot{m}_p = Particulate mass flow, LB/HR
- M_s = Molecular weight of stack gas, wet basis, g/g, mole.
- M_p = Total amount of particulate matter collected, g
- P_{bar} = Atmospheric pressure, IN. HG. (uncompensated)
- P_g = Stack static gas pressure, IN. WC.

- P_{std} = Standard absolute pressure, 29.92 IN. HG.
- A_a = Actual volumetric stack gas flow rate, ACFM
- $Q_{s,d}$ = Dry volumetric stack gas flow rate corrected to standard conditions, DSCFM
- RH = Relative humidity, %
- T_{db} = Dry bulb temperature of stack gas, °F
- T_{wb} = Wet bulb temperature of stack gas, °F
- $T_m(avg)$ = Absolute average dry gas meter temperature, °R
- $T_s(avg)$ = Absolute average stack temperature, °F
- T_{std} = Standard absolute temperature, 528 °F (68 °F)
- θ = Total sampling time, min.
- V_{lc} = Total volume of liquid collected in impingers and silica gel, ml
- V_m = Volume of gas sample as measured by dry gas meter, CF
- $V_m(std)$ = Volume of gas sample measured by the dry gas meter corrected to standard conditions, DSCF
- $V_w(std)$ = Volume of water vapor in the gas sample corrected to standard conditions, SCF
- \bar{V}_s = Average actual stack gas velocity, FT/SEC
- v_{Ptdb} = Vapor pressure at T_{db} , IN. HG.
- v_{Ptwb} = Vapor pressure at T_{wb} , IN. HG

$\overline{\Delta H}$ = Average pressure differential across the orifice meter, IN. WC.

ΔP = Velocity pressure of stack gas, IN. WC.

γ = Dry test meter correction coefficient, dimensionless

ρ = Actual gas density, LB/ACF

CALCULATION EQUATIONS

METHOD 6

$$V_{std} = 17.64 \frac{V_m P_b \gamma}{T_m} \quad (\text{MIDGET IMPINGER VERSION})$$

$$V_{std} = \frac{17.64 V_m (P_b + \frac{\Delta H}{13.6}) \gamma}{T_m} \quad (\text{LARGE IMPINGER VERSION})$$

$$MEQ = (V_t - V_{tb}) N \left(\frac{V_{soln}}{V_a} \right)$$

$$C_s = \frac{7.06 \times 10^{-5} (V_t - V_{tb}) N \left(\frac{V_{soln}}{V_a} \right)}{V_{std}}$$

$$E = \frac{20.90 C_s F_d}{20.90 - \bar{B}'_{O_2}} = \frac{F_c C_s}{\bar{B}'_{CO_2}}$$

$$C_s \text{ (MG/DSCM)} = 1.60186 \times 10^7 C_s$$

$$C_s \text{ (GR/DSCF)} = 7000 C_s$$

$$C_s \text{ (ppm, dry)} = 6.02119 \times 10^6 C_s$$

$$C_s \text{ (ppm, wet)} = 6.02119 \times 10^6 C_s \left(1 - \frac{MC}{100} \right)$$

SYMBOLS

\bar{B}'_{O_2}	=	Average oxygen content in flue gas, % v/v, dry
\bar{B}'_{CO_2}	=	Average carbon dioxide content in flue gas, % v/v, dry
C_s	=	Concentration of sulfur dioxide in flue gas, dry basis, corrected to standard conditions, LB/DSCF
C_s (GR/DSCF)	=	Concentration of sulfur dioxide in flue gas, dry basis, corrected to standard conditions, GR/DSCF
C_s (MG/DSCM)	=	Concentration of sulfur dioxide in flue gas, dry basis, corrected to standard conditions, MG/DSCM
E	=	Emission factor, LB of $SO_2/10^6$ BTU
F_d	=	Dry oxygen F-Factor for given fuel type, DSCF/ 10^6 BTU
F_c	=	Carbon dioxide F-Factor for given fuel type, DSCF/ 10^6 BTU
ΔH	=	Average pressure drop across calibrated orifice, IN. W.C.
γ	=	Dry test meter correction factor, dimensionless
MC	=	Moisture content of flue gas, % v/v
MEQ	=	Total milliequivalents of SO_2 in gas sample
N	=	Normality of barium perchlorate titrant
P_b	=	Barometric pressure at the dry gas meter, IN. HG.

- C_s (ppm-dry) = Concentration of sulfur dioxide in flue gas, dry basis, (v/v), ppm
- C_s (ppm-wet) = Concentration of sulfur dioxide in flue gas, wet basis, (v/v), ppm
- T_m = Absolute average dry gas meter temperature, OR
- V_a = Volume of sample aliquot titrated, cc
- V_m = Dry gas volume as measured by the dry gas meter, DCF
- V_{std} = Dry gas volume as measured by the dry gas meter, corrected to standard conditions (at 68 °F and 1 atmosphere), DSCF
- V_{soln} = Total volume of the solution in which the sulfur dioxide sample is contained, cc
- V_t = Volume of barium perchlorate titrant used for the sample, cc (average of replicate titrations)
- V_{tb} = Volume of barium perchlorate titrant used for the blank, cc