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Reference	<u>11</u>

United States  
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Agency

Office of Air Quality  
Planning and Standards  
Research Triangle Park, NC 27711

EMB Report 92-NDR-01  
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Air

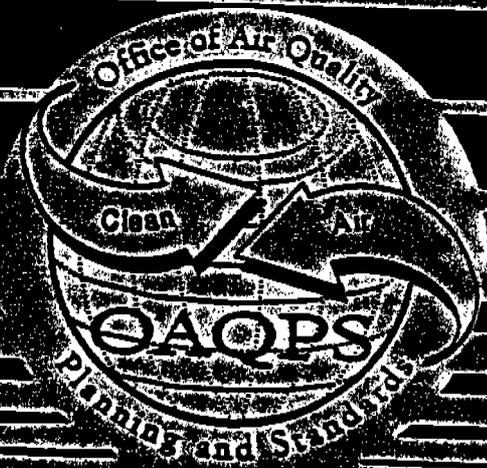
# Wood Treatment Plant Emission Test Report

## Kerr-McGee Chemical Corporation Avoca, Pennsylvania



Note: This is a reference cited in AP 42, *Compilation of Air Pollutant Emission Factors, Volume I Stationary Point and Area Sources*. AP42 is located on the EPA web site at [www.epa.gov/ttn/chief/ap42/](http://www.epa.gov/ttn/chief/ap42/)

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**Final Emissions Test Report**

**Kerr-McGee Chemical Corporation  
Forest Products Division  
Wood Treatment Plant  
Avoca, Pennsylvania**

**U.S. EPA Contract No. 68-D2-0161  
EMB Work Assignment No. 1-2**

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**April 1996**

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# **1. INTRODUCTION**

## **1.1 Summary**

The U.S. Environmental Protection Agency (EPA), Office of Air Quality Planning and Standards (OAQPS), Industrial Studies Branch (ISB), is studying baseline hazardous air pollutant (HAP) emissions and maximum achievable control technology (MACT) for the creosote wood-treating industry.

The Kerr-McGee Chemical Corporation (KMCC) facility located in Avoca, Pennsylvania was selected for study for this project. The KMCC facility has an incinerator that controls emissions from the vacuum system exhaust and the creosote work and storage tanks. The reported efficiency for the incinerator is 99 percent. No other facility uses an incinerator or reports a control efficiency close to the 99 percent reported at KMCC.

The testing program consisted of: (1) monitoring total organic hydrocarbons by CFR 40 Part 60 Appendix A, Method 25A; (2) measuring velocity, moisture, and gas composition measurements with Methods 1 through 4; and, (3) collecting creosote samples for analysis. Analysis for HAPs in the creosote sample was accomplished with TO-13 analysis. Testing with Methods 1 through 4 and Method 25A was conducted on the inlet and outlet of the incinerator. Due to the small dimensions of the inlet port and pipe, and the high temperature of the incinerator outlet, some modifications in the standard Methods 1 through 4 protocols were required. Samples from two incomplete tests were compiled to reflect the emission patterns from the process units and emission control unit (incinerator) over the course of one entire process cycle.

## 1.2. Key Personnel Test Program Organization

The key personnel who coordinated the test program and their phone numbers are:

- Eugene Crumpler, EPA Technical Coordinator (919) 541-0881
- Betty Gatano, RT1, Field Test Coordinator (919) 940-8603
- Jeff Snyder, Kerr-McGee, Plant Contact (717) 457-7446
- James Houck, SAIC, Work Assignment Manager (503) 643-3755
- Roger Bighouse, SAIC, Test Director (503) 643-3755

## **2. Plant and Sampling Location Descriptions**

### **2.1. Process Description and Operation**

The Kerr-McGee Chemical Corporation (KMCC) facility, located in Avoca, Pennsylvania, uses creosote to treat railroad ties, and switch ties made from other hardwoods. In 1991, the facility treated roughly 1.3 million cubic feet of wood products and consumed 900,000 gallons of creosote. The type of creosote used in the facility is P2 creosote, a 60/40 creosote coal tar mixture. The facility operates two treatment cylinders for conditioning and treating wood. Each cylinder has an associated condenser, work tank, storage tank, and vacuum pump. Both Cylinders measure 7 feet in diameter by 150 feet in length. The creosote facility operates 24 hours per day, 5 days per week. Figure 2-1 shows the layout of the creosote treatment process and the air pollution control device.

The facility either uses air seasoned wood or it conditions wood by the Boulton method prior to treatment. Air-seasoning takes months before the wood can be treated, while Boultonizing the wood takes from 12 to 17 hours. Because of the current wood shortage facing the wood treatment industry, the facility cannot afford to have a large amount of inventory being air-seasoned. As a result, the facility is now boultonizing a larger percentage of wood. For example, in the past, the facility treated about 50 percent air-seasoned wood, but because of of the wood shortage, only about 20 percent of the wood treated is now air-seasoned.

The Boulton process is used to condition green wood prior to treatment at the KMCC facility. Initially, the creosote treating cylinder is full. Approximately 3,000 to 6,000 gallons of creosote are then returned to the work tank to ensure that the void space in the cylinder is maintained. The total filling time lasts thirty minutes. Steam coils in the bottom of the cylinder are used for heating, and the cylinder is heated to approximately 180 °F. A 20"-24" Hg vacuum is then pulled on the cylinder to lower the boiling point of water in the wood, causing part of the water to evaporate. The temperature in the cylinder drops approximately 25°F during the initial hours

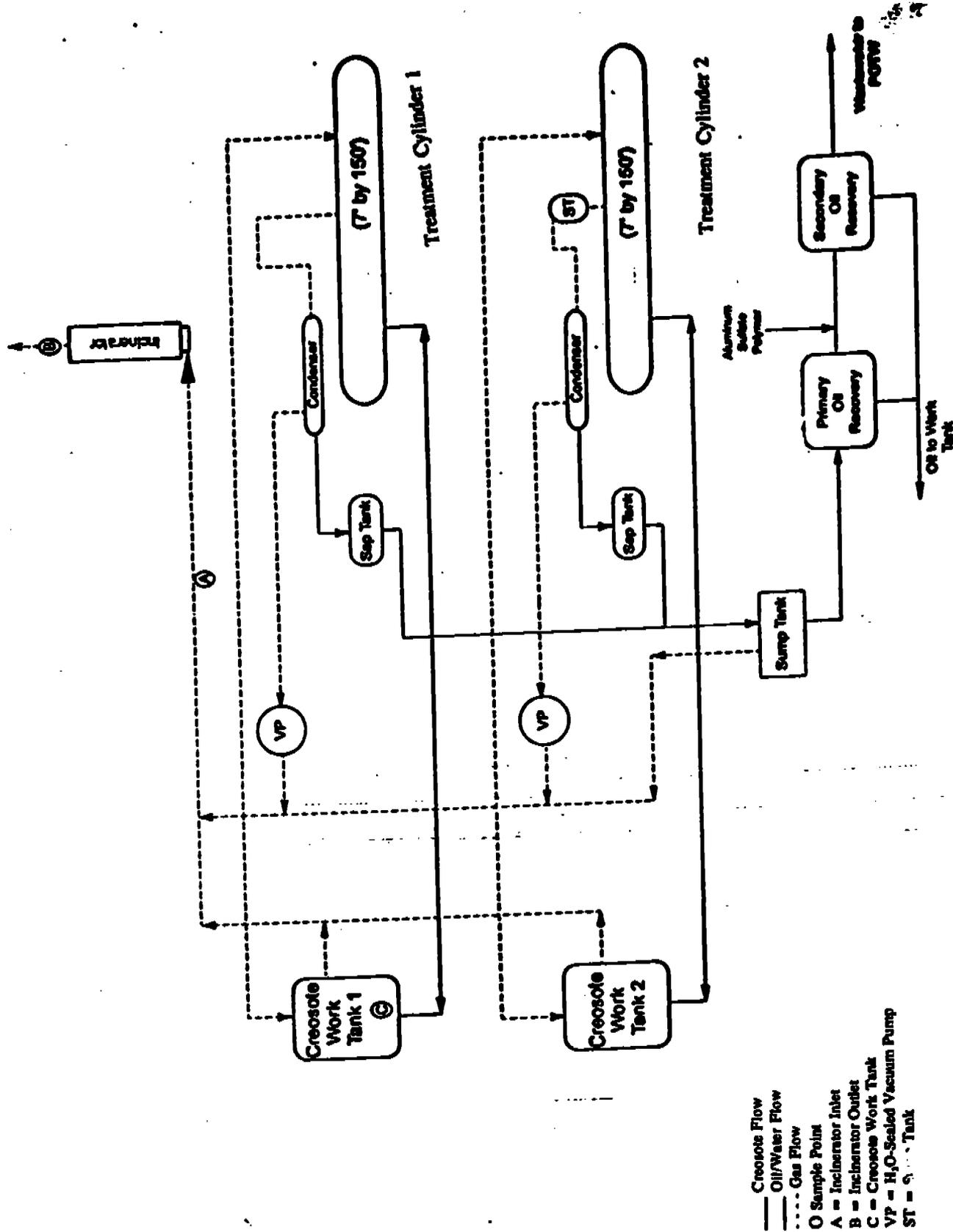


Figure 2-1. Layout and sampling locations for the KMCC facility in Avoca, Pennsylvania

of the vacuum and increased to 180°F in about 3 hours. The operator maintains the cylinder temperature at 180°F for the remainder of Boultonization. Vapors from the cylinder are condensed in a shell-and-tube condenser, and the condensate collects in one of two, 85-gallon sap tanks. During the Boultonization, the operator empties the sap tanks into a sump tank at thirty-minute intervals. The Boulton cycle lasts from 12 to 17 hours, depending on the type of wood being conditioned and the customer requirements.

After the wood has been conditioned, the treatment process begins. The KMCC facility uses the Rueping method to treat the wood. In this process, the vacuum is broken and the creosote in the cylinder is returned to the work tank. A brief period of initial air pressure of 30 psig is applied to the cylinder. The cylinder is then filled with creosote while the pressure on the cylinder is maintained. Once filling is complete, pressurization continues until the cylinder pressure reaches 200 psig. Pressure is maintained for approximately 1 to 4 hours, depending on the type of wood being treated and the specification of the final product. After pressurization, the preservative is drained from the cylinder and returned to the creosote work tank. The total draining time lasts approximately thirty minutes. A final 20" to 24" Hg vacuum is then pulled on the cylinder for two hours

At this point the treating process is completed, and the charge can be removed from the cylinder without affecting the quality of the product. However, on some charges, the facility quenches the ties before the cylinder door is opened, in order to reduce the emissions from the treated wood. After the final 2-hour vacuum, the cylinder is filled with wastewater. Total filling time lasts approximately 30 minutes. The water remains in the cylinder for one hour, and the water is then drained and returned to the wastewater treatment system. A post-quenching vacuum of 22" Hg is pulled on the cylinder for another hour. After vacuuming, the cylinder is opened and the charge removed. The charge remains on the drip pad until all dripping has stopped, and it is then either stored onsite or loaded in rail cars and shipped to the customer.

## **2.2 Control Equipment Description**

The KMCC facility uses a thermal oxidizer (incinerator) to control emissions from the process work tanks, the creosote storage tanks, vacuum system exhaust, and the sump tank. The lines from all of these sources combined to one inlet that enters the bottom of the incinerator. The incinerator was designed and manufactured by the John Zink Company, and the manufacturer's reported efficiency for the incinerator is 99 percent. The thermal incinerator is neither recuperative nor regenerative. Figure 2-2 is a schematic of the incinerator. The following are the design specifications for the incinerator:

### **Waste Stream**

Flow rate: 4000 cfm

Temperature: 100°F

Pressure: Atmospheric

### **Incinerator Operating Conditions**

Excess Air: 25%

Chamber Temperature: 1800°F

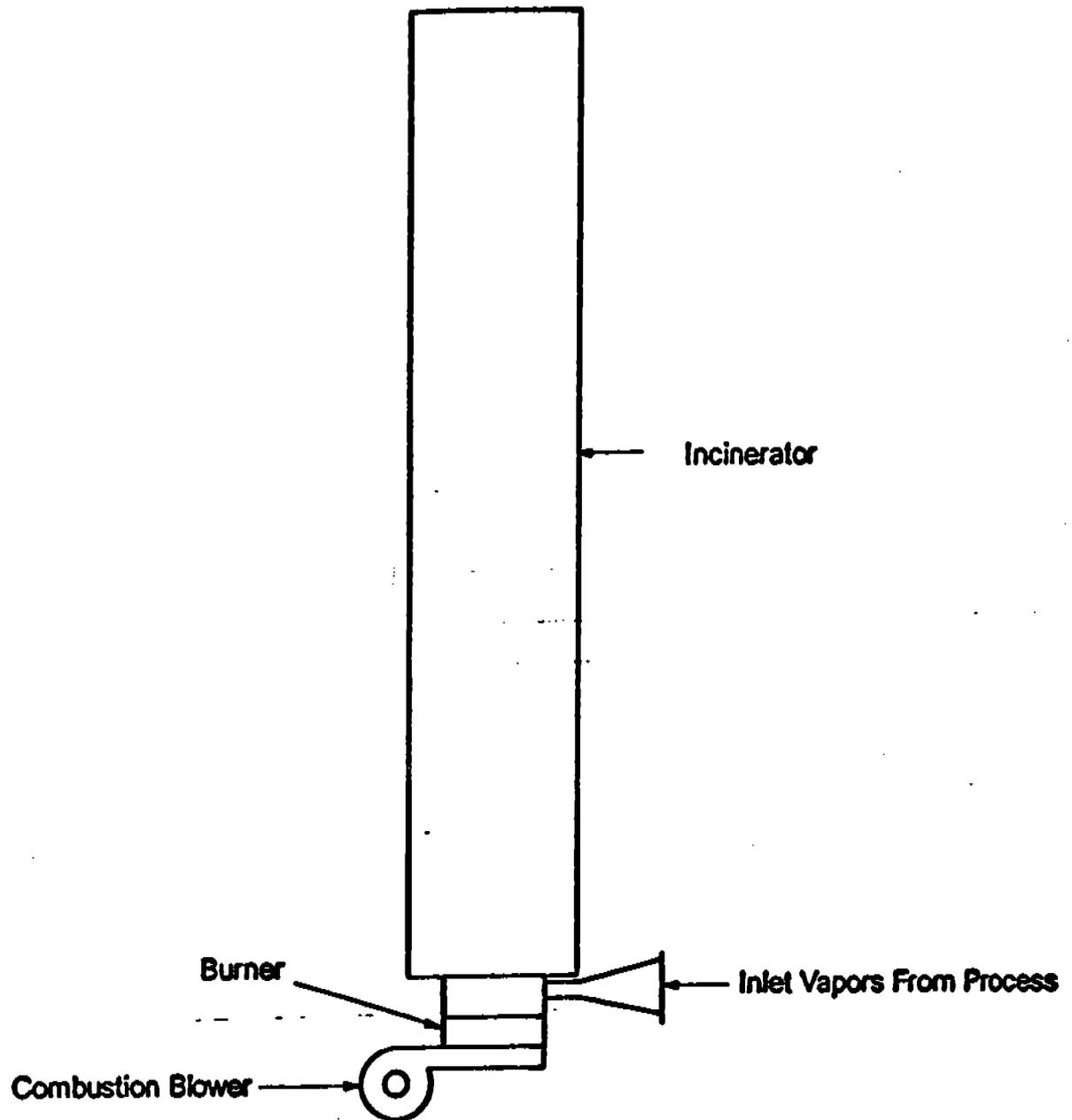
Residence Time: 1.0 sec

The incinerator is fueled by natural gas. However, the fuel usage for the incinerator cannot be measured directly because the natural gas meter services both the incinerator and the facility's boiler. During weekends when the treatment process is not operating, the incinerator operates at reduced temperatures in order to control emissions from the creosote work tanks and the sump tank.

## **2.3. Flue Gas Sampling Locations**

Total hydrocarbon compound sampling (Method 25A) was conducted and associated flow, gas composition, and moisture measurements were taken at the incinerator inlet and the incinerator outlet.

**Figure 2-2. Incinerator at Kerr-McGee Chemical Corporation  
in Avoca, Pennsylvania (not drawn to scale).**



### **2.3.1. Incinerator Inlet**

The existing 1/2" port located in the 3" incinerator inlet was used intermittently for Method 25A and Methods 1 through 4. (Due to the small dimensions, a thermal anemometer probe was used in place of a pitot tube for velocity measurements.) The port is approximately 9" (three diameters) downstream and approximately 5" (1 2/3 diameters) upstream of disturbances.

### **2.3.2. Incinerator Outlet**

A single 2" port was installed in the 30"-diameter stack at a location 72" upstream of where the gas discharges to the atmosphere. This port was used simultaneously for Method 25A and Methods 1 through 4.

## **2.4. Creosote Sampling Locations**

The sample was taken from a tap connected to creosote work tank #1. A single sample was taken during test run #2 with a volume of 273 ml.

### **3. Summary and Discussion Test Results**

#### **3.1. Objectives**

The purpose of this test program is to provide a characterization of overall emissions from creosote wood treatment plants and information on the performance of the control unit. Specific objectives in order of priority are:

- Measure simultaneously total hydrocarbon (THC) emissions (Method 25A) at the incinerator inlet and outlet.
- Determine the efficiency of the incinerator by comparing the THC emissions at the scrubber's inlet and outlet.
- Sample the creosote solution during the total hydrocarbon test run and analyze it for concentrations of HAPs.

Two test runs were conducted. Test #1 was conducted on cylinder #1. The Boulton cycle lasted 14 hours; the final vacuum lasted 4 hours. Test #2 was conducted on cylinder #2. The Boulton cycle lasted 12 hours. Test #2 was terminated before the pressurization phase was completed, due to a gasket failure on the treatment cylinder.

Table 3-1 presents a sampling and analysis matrix.

Table 3-1. Sampling Matrix.

Run	Date	Sample Type	Test Method	Location/Clock Time (hours)		
				Inlet - 4	Outlet - 3	Work Tank -
1	9/8/93	THC	M25A	0800-1545	1545-0620	
1-1	9/8/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1132-1202	
1-2	9/8/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1255-1325	
1-3	9/8/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1413-1443	
1-4	9/8/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1738-1808	
1-5	9/8/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1836-1906	
1-6	9/8/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1947-2018	
1-7	9/8/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		2053-2123	
1-8	9/8/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		2329-2359	
1-9	9/9/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		0042-0112	
1-10	9/9/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		0144-0214	
1-11	9/9/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		0241-0311	
1-12	9/9/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		0340-0413	
1-13	9/9/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		0451-0521	
2	9/10/93	THC Preservative	M25A Tap Sample	1215-2230	0800-2230	1738
2-1	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		0836-0908	
2-2	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		0937-1010	
2-3	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1048-1127	
2-4	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1210-1244	
2-5	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1311-1341	
2-6	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1539-1609	
2-7	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1651-1721	
2-8	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1748-1818	
2-9	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		1850-1929	
2-10	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		2007-2037	
2-11	9/10/93	O <sub>2</sub> /CO <sub>2</sub> /Moist.	M3/M4		2118-2148	

## **3.2. Field Test Changes and Problems**

### **3.2.1. Damaged Analyzer**

One of two THC analyzers was damaged during shipping. Therefore, only the inlet or the outlet could be monitored until a replacement analyzer was acquired. During run 1, the inlet THC concentrations were monitored for the first 7 3/4 hours. At the direction of the Technical Coordinator, the outlet THC concentrations were monitored for the remaining 15 hours of run 1.

A replacement analyzer was acquired and was on-line (at the inlet sampling location) approximately 4 hours into run 2. At that time, both the inlet and outlet THC concentrations were monitored simultaneously until the termination of run 2.

### **3.2.2. Premature Termination of Run 2**

Run 2 was terminated on 9/10/93 at 2230 hours due to plant shutdown. The cylinder gasket failed during the pressurization cycle and could not be repaired until maintenance crews arrived on 9/13/93.

### 3.3. Presentation of Results

#### 3.3.1. THC Emissions

##### 3.3.1.1. *Inlet THC Emissions*

Inlet THC Concentrations The inlet THC concentrations varied cyclically. The inlet concentration dropped when the sump was activated to pump the water from the sump tank into the waste water storage tank (Figure 3-1). This variation is seen only during the vacuum cycles (Boulton cycle and final vacuum). Hourly averages of the inlet THC concentrations during Boultonization were made and are presented in Table 3-2.

Inlet THC concentrations were monitored during the first creosote blowback (after Boulton cycle) of run 2. The entire blowback lasted 30 minutes, with the THC concentration peaking at 4950 ppm THC as propane 4 minutes into the blowback. The THC concentration slowly decreased during the blowback and was at 2910 ppm THC as propane at the end of the blowback. The average concentration during the blowback was 3690 ppm THC as propane. Inlet THC concentrations were also monitored during the pressurization cycle for run 2. In the 30 minutes prior to cylinder gasket failure, the inlet averaged 2620 ppm THC as propane.

Flow Rate Numerous intermittent measurements of flue velocity, temperature, moisture, CO<sub>2</sub> %, and O<sub>2</sub> % were made at the inlet. These results as well as the flow rate are presented in Tables 3-3 and 3-4.

Inlet duct flow rates were intensely monitored on 9/10/93 (Run 2) from 0830 to 1023 (Figure 3-2). The increases and decreases in flow rate correspond to the sump tank pump episodes. Since these episodes of varying flow rate are brief, they will be ignored and the general decreasing trend in flow rate will be used to calculate emission rates. The use of this

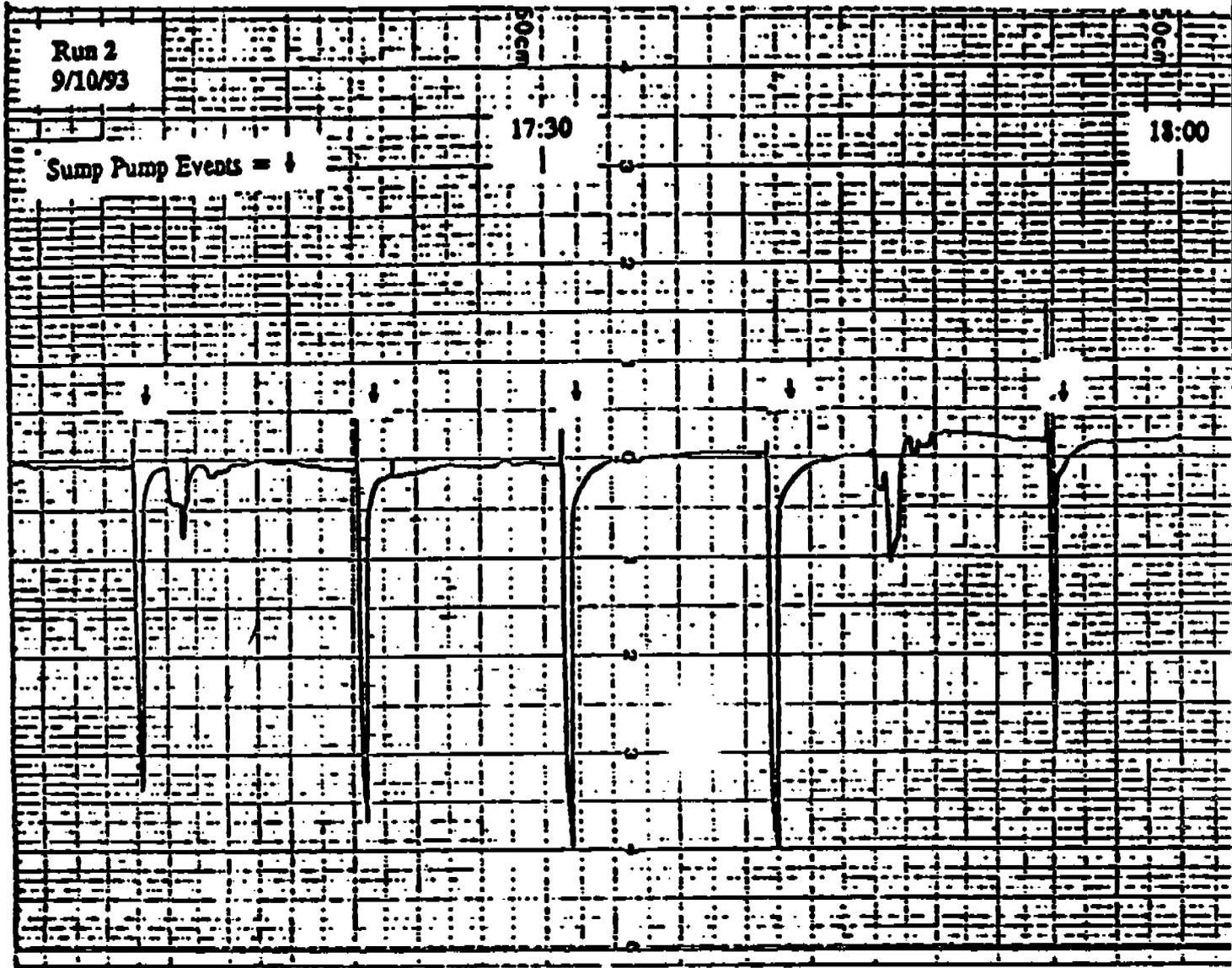


Figure 3-1. Variation of the inlet THC concentration corresponding to sump pump events.

**Table 3-2. Hourly Averages of Inlet THC Concentrations During Boulton Cycle.**

Hour of Boulton Cycle	Average THC Concentration (ppm as propane)		
	Run 1	Run 2	Composite <sup>a</sup>
1	18,260	b	18,260
2	18,260	b	18,260
3	11,490	b	11,490
4	7,910	b	7,910
5	11,490	12,100	11,800
6	19,320	11,760	15,540
7	c	11,100	11,100
8		10,310	10,310
9		9,480	9,480
10		9,700	9,700
11		9,440	9,440
12		9,440	9,440
Average	-	-	11,890

- a. Average of runs 1 and 2.
- b. Inlet analyzer not operational.
- c. Measurement of inlet concentrations terminated.

**Table 3-3. Run 1 Inlet Measurements of Temperature, Oxygen Content, Carbon Dioxide Content, Moisture Content, and Flow Rate During Boulton Cycle.**

<b>Time (Hours)</b>	<b>Temperature (°F)</b>	<b>O<sub>2</sub> (%)</b>	<b>CO<sub>2</sub> (%)</b>	<b>Moisture (%)</b>	<b>Flow Rate (DSCFM)</b>
0848	123	21	0	9.6	29.0
0910	107	18	2	7.8	116.0
0936	107	20	2	7.8	9.7
1007	119	20	2	11.5	19.0
1037	106	20	2	7.1	14.5
1105	106	20	2	6.1	14.5
1134	113	20	1	9.3	9.7
1209	130	20	1	15.1	9.7
1310	118	20	2	10.8	12.1
1545 <sup>a</sup>	—	—	—	—	—

a. Inlet measurements terminated.

**Table 3-4. Run 2 Inlet Measurements of Temperature, Oxygen Content, Carbon Dioxide Content, Moisture Content, and Flow Rate.**

<b>Cycle of Treatment Process</b>	<b>Time (Hours)</b>	<b>Temperature (°F)</b>	<b>O<sub>2</sub> (%)</b>	<b>CO<sub>2</sub> (%)</b>	<b>Moisture (%)</b>	<b>Flow Rate (DSCFM)</b>
<b>Boulton</b>	1025	140	-	-	19.7	-
	1238	137	-	-	19.8	14.5
	1410	132	20	4	15.9	14.5
	1538	116	20	3	10.2	6.8
	1650	120	21	2	11.4	7.3
	1815	114	21	2	8.7	3.9
	1940	111	21	1	8.8	8.7
	2022	109	21	2	7.5	6.8
<b>First Blowback</b>	2046	-	-	-	-	23.2
	2107	-	-	-	-	33.8
<b>Pressurization</b>	2140	-	-	-	-	6.3
	2145 <sup>a</sup>	-	-	-	-	-

a. Gasket failure, run 2 terminated.

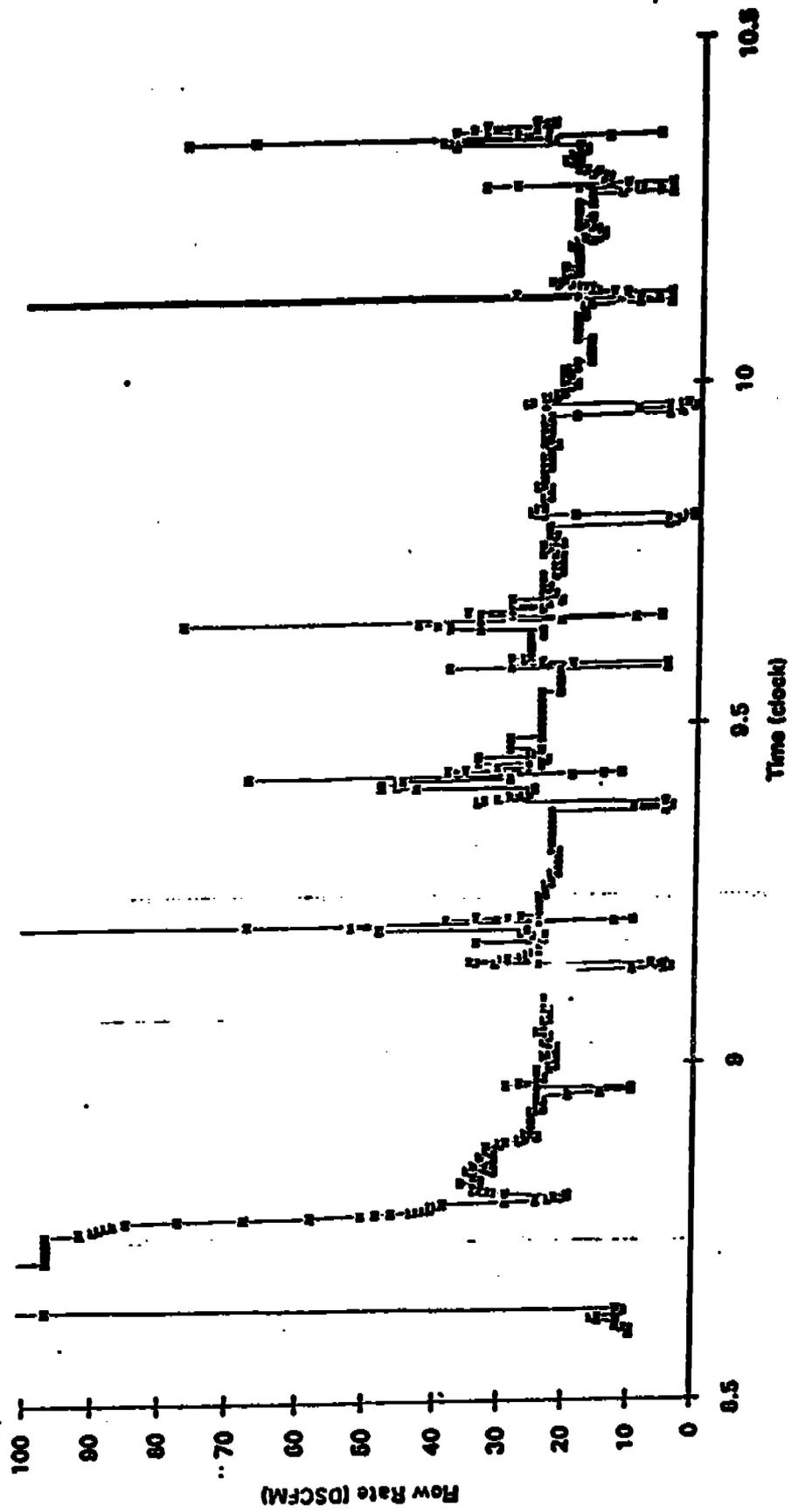


Figure 3-2. Inlet duct flow rates during a portion of the Boulton cycle for run 2.

decreasing trend in flow rates during the Boulton process is supported by the intermittent measurements taken during the Boulton cycle after the intensive flow measurements (Figure 3-3). Flow rates were measured twice during the 30-minute blowback cycle of run 2 and averaged 28.5 DSCFM.

Emission Rate By combining the average flow rate and average THC concentrations measured at the inlet of the incinerator, we can estimate the average emission rates for each cycle of the process for which we have data. Table 3-5 presents the average flow rates, THC concentrations, and emission rates for various cycles of the wood treatment process.

#### *3.3.1.2. Outlet THC Emissions*

Outlet THC Emissions THC concentrations of the outlet were monitored during run 1 from midway through the Boulton cycle to completion of final vacuum. THC concentrations of the outlet were monitored during run 2 from the beginning of the Boulton cycle until gasket failure occurred at the beginning of the pressurization cycle. In all cases, the outlet THC concentration was measured to be <2 ppm as propane except under one circumstance.

Compressed air is used to blow the water from the SAP tank into the sump tank. If the air is left on after all water has been forced out of the SAP tank, THC outlet concentrations go up (Figure 3-4). The duration and magnitude of this "spike" are dependent on the duration and pressure of the "excess" air that is forced through the system. Such spikes ranged from approximately 20 ppm to over 1000 ppm. Each episode lasted less than 30 seconds. Due to the extremely short nature of these spikes, they can be neglected. The THC concentration at the outlet of the incinerator is measured to be <2 ppm as propane.

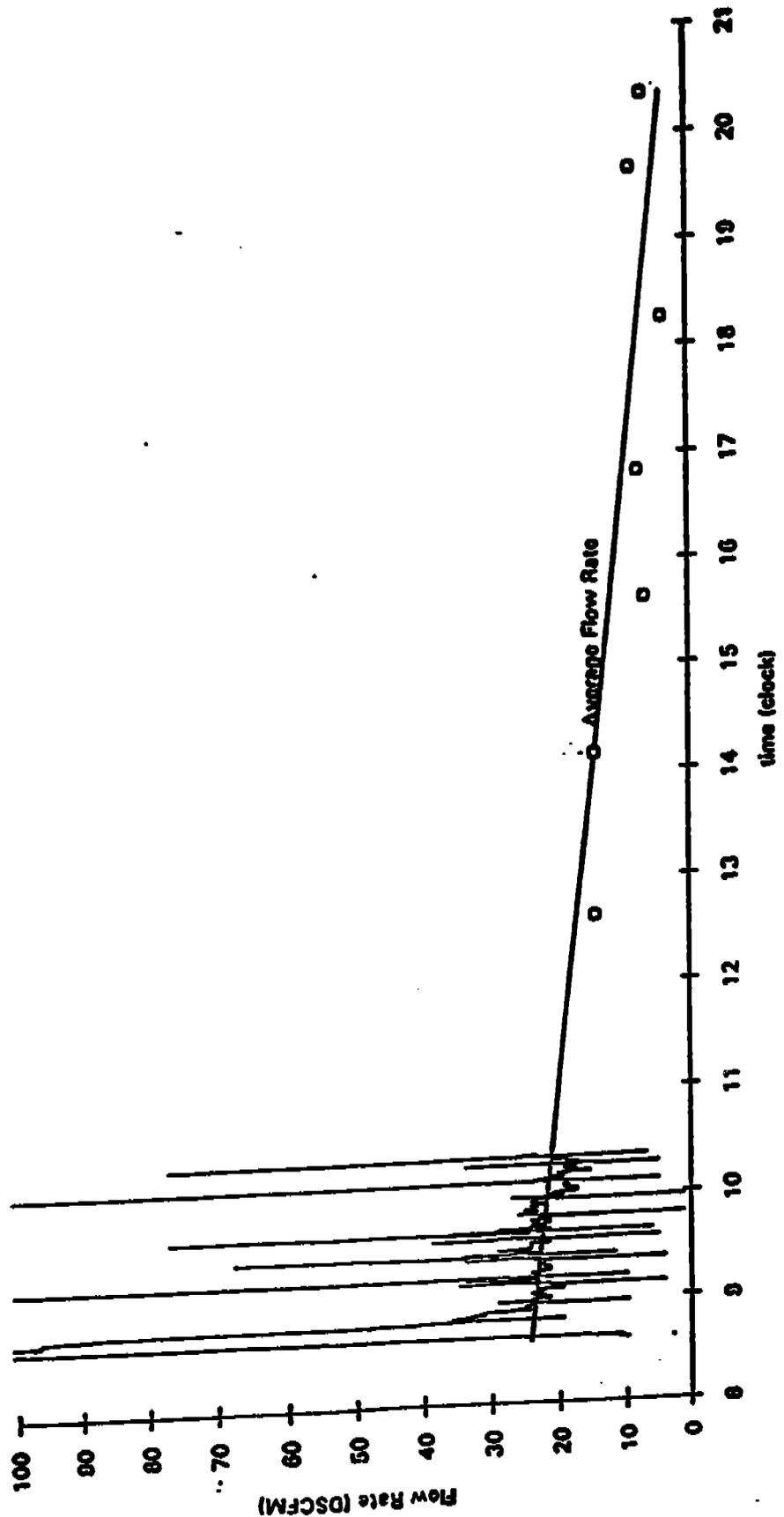


Figure 3-3. Flow rate measurements of inlet duct during entire Boulton cycle for run 2.

Table 3-5. Average Inlet THC Concentrations, Flow Rates, and Emission Rates by Process Cycle.

Cycle is Treatment Process	Length of Cycle	Average THC Concentration (ppm as propane)	Average Flow Rate (DSCFM)	Average Emission Rate <sup>a</sup> (lb/hr propane)
Boulton	12 hours	11,890 <sup>b</sup>	14 <sup>c</sup>	1.22
First Blowback	15 minutes	3,690 <sup>c</sup>	28.5 <sup>b</sup>	0.77
Pressurization	30 minutes <sup>c,d</sup>	2,620 <sup>c</sup>	6.3 <sup>c</sup>	0.12
Second Blowback	--	--	--	--
Final Vacuum	--	--	--	--
Removal of Charge	--	--	--	--

- a. Multiply lb/hr propane by 3 to get lb/hr carbon emissions.
- b. From composite of runs 1 and 2.
- c. From run 2 data.
- d. Prior to gasket failure.

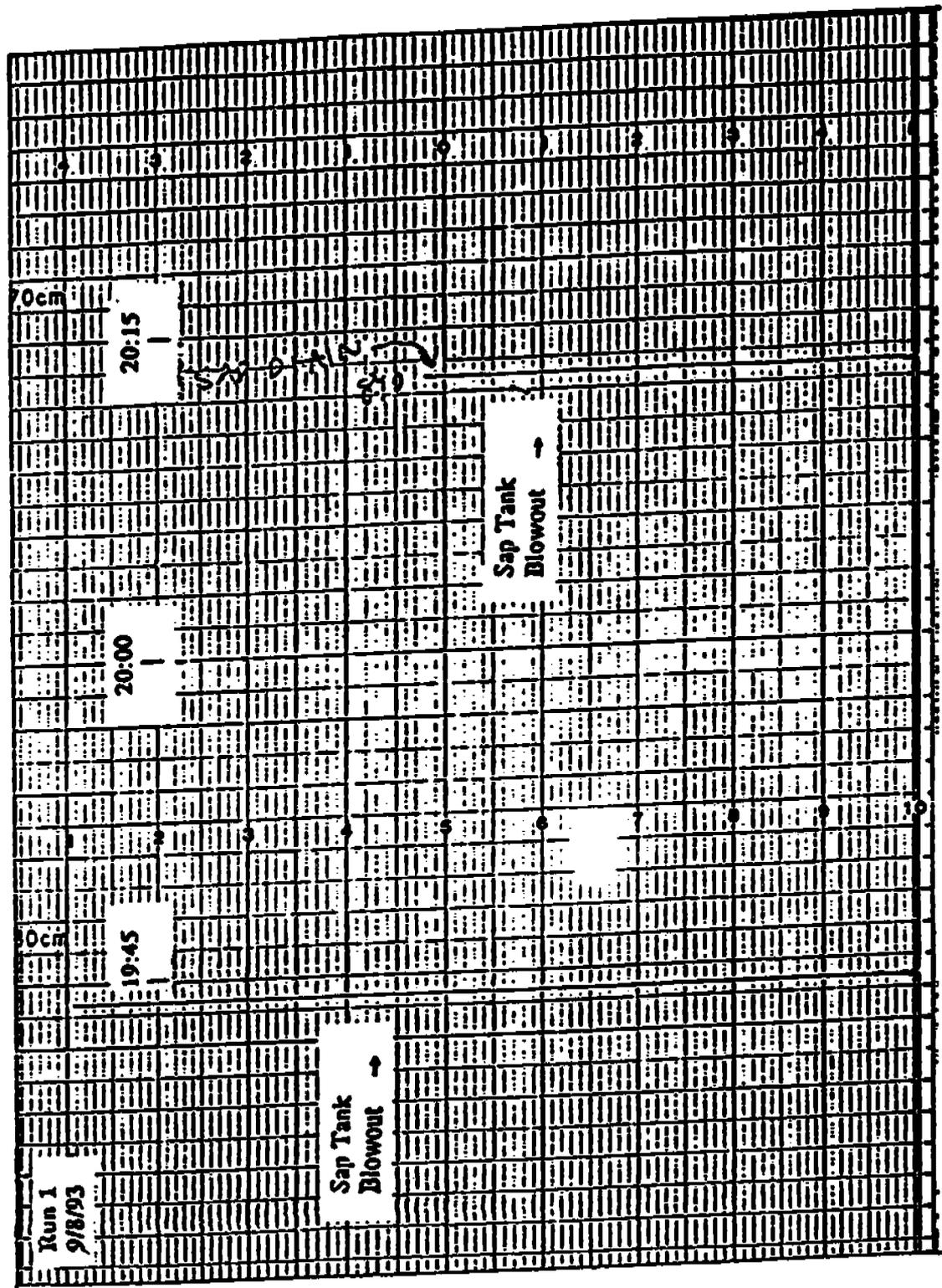


Figure 3-4. Outlet THC concentrations during SAP tank blowouts.

Flow Rate Flow rate, oxygen and carbon dioxide contents, moisture, and temperature measured at the outlet were all very uniform throughout all cycles of the test. Table 3-6 presents the results of each EPA Method 4 test done at the outlet for both runs 1 and 2.

Emission Rates The average emission rates THC (as propane) are presented in Table 3-7.

**Table 3-6. Runs 1 and 2 Outlet Measurements of Temperature, Oxygen Content, Carbon Dioxide Content, Moisture Content, and Flow Rate.**

Run	Average Temp. (°F)	Moisture (%)	O <sub>2</sub> (%)	CO <sub>2</sub> (%)	Flow Rate (DSCFM)
1-1	1598	12.9	11.5	4.5	454
1-2	1605	13.0	10.5	4.0	453
1-3	1583	13.4	11.0	5.0	454
1-4	1580	13.8	11	5.0	452
1-5	1580	13.8	12	5.0	451
1-6	1580	13.6	11.0	4.0	453
1-7	1580	13.2	14.0	4.0	454
1-8	1580	12.3	11.0	4.0	459
1-9	1580	13.7	10.5	4.5	453
1-10	1580	15.2	11.5	4.0	446
1-11	1580	13.6	12.5	6.0	451
1-12	1580	13.4	10.0	5.0	507
1-13	1580	13.4	11.0	4.0	454
Average Run 1	1584	13.5	11.3	4.5	464
2-1	1580	14.3	10.5	6.0	448
2-2	1580	13.6	10.0	5.5	452
2-3	1580	13.5	11.0	5.0	453
2-4	1580	12.9	5	5.5	458
2-5	1550	14.3	11.0	5.5	452
2-6	1550	13.1	11.0	5.5	457
2-7	1550	12.9	11.0	5.0	459
2-8	1550	12.7	11.0	6.0	458
2-9	1550	12.8	10.5	6.0	458
2-10	1550	12.3	11.0	6.0	460
2-11	1550	12.8	10.5	5.5	459
Average Run 2	1561	13.2	10.7	5.6	456
Average, Runs 1 & 2	1573	13.4	11.0	5.0	460

**Table 3-7. Average Outlet Emission Rates**

Run	Average THC (ppm as propane)	Average Flow Rate (DSCFM)	THC Emission Rate <sup>a</sup> (lb/hr propane)
1	< 2 ppm	464	<0.0068
2 <sup>b</sup>	< 2 ppm	460	<0.0068
Average	—	462	<0.0068

- a. Multiply THC ppm as propane by 3 to get ppm of carbon.
- b. Run 2 terminated at the beginning of pressurization due to gasket failure.

**3.3.2. Incinerator Efficiency**

Because inlet emission data is limited to the Boulton cycle, first blowback, and beginning of the pressurization cycle, incinerator efficiency can be calculated only for these cycles. Table 3-8 presents the incinerator efficiency results.

**Table 3-8. Incinerator Efficiency Results**

Cycle	Average Inlet Emissions (lb/hr propane)	Average Outlet Emissions (lb/hr propane)	THC Destruction Efficiency (%)
Boulton Cycle	1.22	<0.0068	>99.4
First Blowback	0.77	<0.0068	>99.1
Pressurization <sup>a</sup>	0.12	<0.0068	>94.3

- a. Data limited to first 30 minutes of pressurization due to gasket failure.

The lower detection limit of the THC analyzer measuring the outlet concentration was the limiting factor in measuring the incinerator's THC destruction efficiency.

### **3.3.3. HAPs Concentration in Preservation**

Table 3-9 summarizes the analytical data for HAPs in the creosote preservative and a sample blank. The laboratory report for the sample blank (H<sub>2</sub>O blank) are located in Appendix A.

**Table 3-9. Creosote Analytical Results (µg/ml)**  
**Samples Collected 9-16-93**

Compound	Sample ID	
	AV-C-1	H <sub>2</sub> O Blank
<b>Lower Detection Limit</b>	<b>2.0</b>	<b>0.0015</b>
Phenol	ND	ND
bis(2-Chloroethyl) Ether	ND	ND
2-Chlorophenol	ND	ND
1,3-Dichlorobenzene	ND	ND
1,4-Dichlorobenzene	ND	ND
1,2-Dichlorobenzene	ND	ND
bis(2-Chloroisopropyl) Ether	ND	ND
N-Nitroso-di-n-Propylamine	ND	ND
Hexachloroethane	ND	ND
Nitrobenzene	NT	ND
Isophorone	N	ND
2-Nitrophenol	ND	ND
2,4-Dimethylphenol	ND	ND
Benzoic Acid	ND	ND
bis(2-Chloroethoxy) Methane	ND	ND
2,4-Dichlorophenol	ND	ND
1,2,4-Trichlorobenzene	ND	ND
Naphthalene	223.44	ND

**Table 3-9. Creosote Analytical Results ( $\mu\text{g}/\text{ml}$ ) (continued)**

Compound	Sample ID	
	AV-C-1	H <sub>2</sub> O Blank
Lower Detection Limit	2.0	0.0015
4-Chloroaniline	ND	ND
Hexachlorobutadiene	ND	ND
4-Chloro-3-Methylphenol	ND	ND
2-Methylnaphthalene	150.18	ND
Hexachlorocyclopentadiene	ND	ND
2,4,6-Trichlorophenol	ND	ND
2,4,5-Trichlorophenol	ND	ND
2-Chloronaphthalene	ND	ND
2-Nitroaniline	ND	ND
Dimethylphthalate	ND	ND
Acenaphthylene	216.12	ND
2,6-Dinitrotoluene	ND	ND
3-Nitroaniline	ND	ND
Acenaphthene	ND	ND
2,4-Dinitrophenol	ND	ND
4-Nitrophenol	ND	ND
2,4,2,4-Dinitrotoluene	ND	ND
Dibenzofuran	131.87	ND
Diethylphthalate	ND	ND
Fluorene	128.21	ND

**Table 3-9. Creosote Analytical Results ( $\mu\text{g}/\text{ml}$ ) (continued)**

Compound	Sample ID	
	AV-C-1	H <sub>2</sub> O Blank
Lower Detection Limit	2.0	0.0015
4-Chlorophenyl-Phenyl Ether	ND	ND
4-Nitroaniline	ND	ND
4,6-Dinitro-2-Methylphenol	ND	ND
N-Nitrosodiphenylamine	ND	ND
4-Bromophenyl-Phenyl Ether	ND	ND
Hexachlorobenzene	ND	ND
Pentachlorophenol	ND	ND
Phenanthrene	300.37	ND
Anthracene	40.29	ND
di-n-Butylphthalate	ND	ND
Fluoranthene	227.11	ND
Pyrene	16.50	ND
Butylbenzylphthalate	N	ND
3,3'-Dichlorobenzidine	ND	ND
Chrysene	51.28	ND
Benzo(a)anthracene	40.29	ND
bis(2-Ethylhexyl)phthalate	ND	ND
Di-n-Octylphthalate	ND	ND
Benzo(b)fluoranthene	21.98	ND
Benzo(k)fluoranthene	13.55	ND

**Table 3-9. Cresote Analytical Results ( $\mu\text{g}/\text{ml}$ ) (continued)**

Compound	Sample ID	
	AV-C-1	H <sub>2</sub> O Blank
Lower Detection Limit	2.0	0.0015
Benzo(a)pyrene	ND	ND
Indeno(1,2,3-cd)pyrene	16.12	ND
Dibenz(a,h)anthracene	ND	ND
Benzo(g,h,i)perylene	ND	ND
Biphenyl	36.63	ND
Quinoline	36.63	ND
Xylenes	ND	ND
Cresols	ND	ND

ND = not detected.

Sample AV-C-2 was furnished to the lab as a duplicate but was not analyzed by the client's request.

## **4. Sampling and Analytical Procedures**

### **4.1. Test Methods**

#### **4.1.1. EPA Test Method 25A**

This method was used in the measurement of total hydrocarbons on the inlet and outlet of the incinerator, to determine the incinerator removal efficiency for organics.

The system consisted of an in-stack filter, heated sample line, a Ratfisch Model RS55 total hydrocarbon analyzer, and a strip chart recorder. The analyzer was calibrated approximately once every hour during testing. Calibration gases were NIST calibrated cylinders of propane with a nitrogen balance. The data were recorded as propane.

In conjunction with the EPA Method 25A test, EPA Methods 1 through 4 were used for the determination of sampling locations, sampling traverse points, stack gas molecular weight, stack moisture, stack flow rate, and emissions rate. Determination of the stack moisture on the outlet of the incinerator was conducted with a Method 4 sampling train with the impinger water and silica gel analyzed gravimetrically. Moisture determination on the inlet was conducted using the wet bulb/dry bulb method (Method 4).

A fyrite analyzer was used to measure O<sub>2</sub> and CO<sub>2</sub> concentrations in the stack for the determination of molecular weight of stack gas (Method 3). A standard or calibrated S-type pitot tube was used to measure the velocity head in the stack for the determination of flow rate (Methods 1 and 2).

Deviation from the Methods: (1) Velocity measurements on the inlet pipe were measured with a hot wire anemometer; (2) One port location was used during the velocity traverse; and, (3) The outlet O<sub>2</sub> and CO<sub>2</sub> measurements were made on the exhaust of the Method 4 sampling system. Figure 4-1 is the sample train schematic.

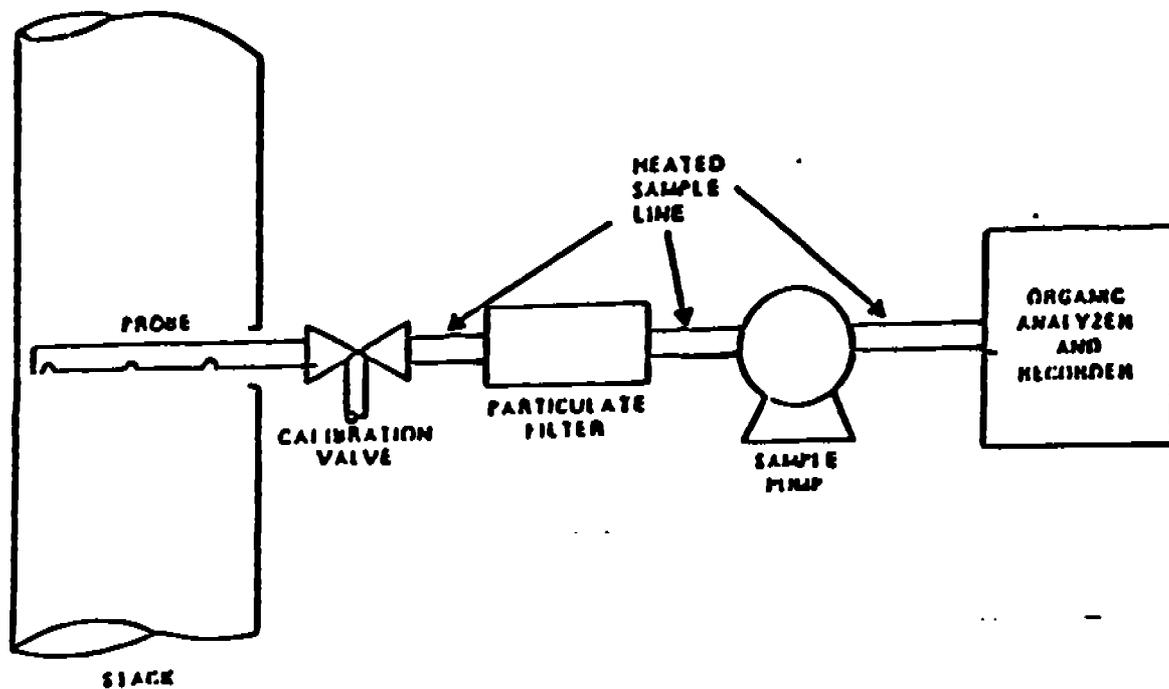


Figure 4-1. Organic concentration measurement system.

#### **4.1.2. EPA Test Method TO-13**

This method was used to analyze the creosote in the creosote storage tank. A sample was taken from the creosote tank during the testing period. The sample was placed in a glass container and then sent to the laboratory where it was analyzed for speciated organic compounds.

### **4.2. Process Data**

Representatives from Research Triangle Institute (RTI) monitored the wood treatment process during testing. The time, temperature, and pressure of each stage of the treatment and conditioning process were monitored and recorded. Additionally, the type of wood to be treated, the product retention time, the total volume of the charge, and the amount of preservative consumed were also recorded for each charge treated during testing.

#### **4.2.1. Operating Conditions**

KMCC test # 1 was conducted on cylinder number 1. During this test, the Boulton cycle lasted for 14 hours, and the final vacuum lasted for 4 hours. Figure 4-2 shows the temperature and pressure in cylinder number 1 throughout the entire wood treatment process.

KMCC test #2 was conducted on cylinder number 2. During this test, the Boulton cycle lasted 12 hours. Figure 4-3 shows the temperature and pressure in cylinder number 2 throughout the entire wood treatment process.

#### **4.2.2. Occurrences During Normal Operating Conditions**

Under normal conditions, both treatment cylinders are in operation, but only one cylinder

was operating during each of the test runs. The sump tank is a closed tank located underneath cylinder number one. The condensate in the sap tanks, the condensate from the condenser, and the water from the water-sealed vacuum pumps collect in the sump tank. The level in the sump tank is maintained by an automatic level controller. When the level in the sump is too high, the controller activates two pumps that periodically transfer the water in the sump tank to the creosote recovery and wastewater treatment system. When the sump pumps were activated during the two tests, variations were seen in the total hydrocarbon monitoring. The periodic dumping of the sap tank into the sump tank also caused momentary variations in the total hydrocarbon monitoring at the inlet of the incinerator.

#### **4.2.3. Problems and/or Variations During Testing**

During Test #1 (8:45 a.m. to 6:00 a.m., 9/08/93 and 9/09/93), cylinder number 1 ran normally. During Test #2, (8:02 a.m. to 10:42 p.m., 09/10/93), cylinder number 2 ran normally until the start of pressurization. When the cylinder reached approximately 65 psi, the gasket on the cylinder door failed. Normal operations could not continue until the gasket was replaced on the following Monday morning. As a result, testing of the remainder of the process could not be completed because of the gasket failure. Based on creosote work tank levels, approximately 300 gallons of creosote were expelled from the cylinder due to the gasket failure. This creosote was held in secondary containment and was then recycled to the creosote recovery system for reuse.



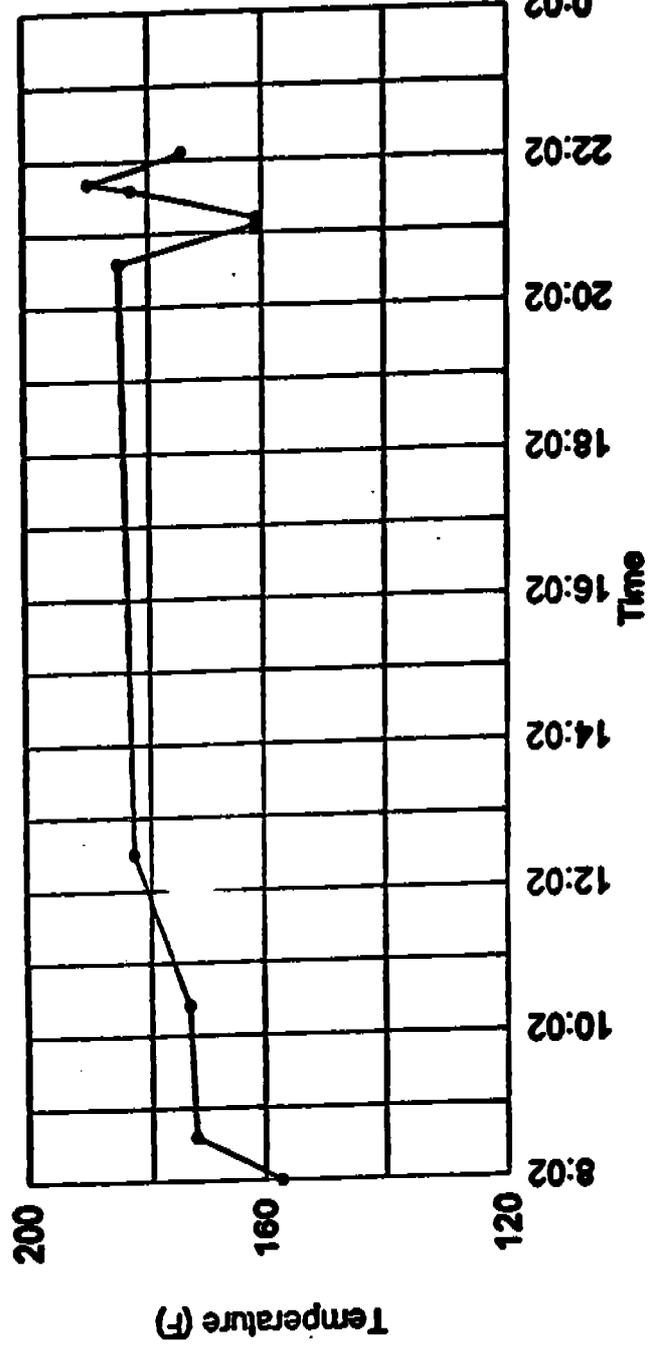
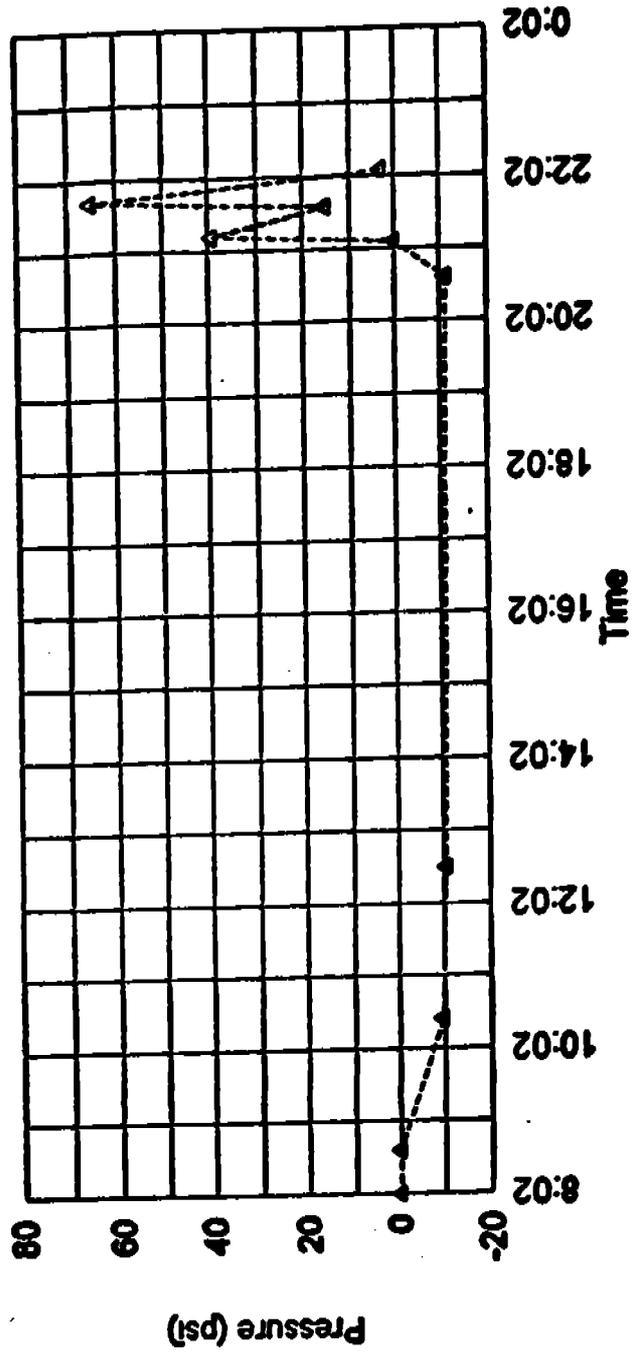


Figure 4-3. Process profile during Test #2 at Kerr-McGee Chemical Corporation in Avoca, Pennsylvania (09/10/93)

## 5. Internal QA/QC Activities

### 5.1. QA/QC Problems

One THC analyzer was damaged during shipment, leaving only one operational THC analyzer available for run 1. An additional THC analyzer was shipped and was on site 48 hours later (approximately 4 hours into run 2). At the direction of the Technical Coordinator, the inlet THC concentrations were monitored for 7 3/4 hours into run 1, at which time the analyzer was switched to monitor the outlet THC concentrations for the remaining 15 hours of run 1.

### 5.2. QA Audits

Calibrations of the THC analyzers were done at approximately two-hour intervals using certified gases of known concentrations of THC (propane). The certifications are contained in Appendix B.

Calibration of the Method 4 control box was done both prior to and following the test program. Table 5-1 presents the pre-test and post-test calibration data. Pre-test and post-test calibration worksheets are contained in Appendix C.

A sample blank was prepared and shipped to the lab for analysis in an identical manner as the preservative sample. De-ionized, distilled water was used as the blank solution.

Table 5-1. Method 4 Calibrations Data.

Control Box ID	Gas Meter Coefficient (Yd)		Deviation
	Pre-Test	Post-Test	
OMST20	0.967	0.967	0%

**Appendix A**

**Preservative Analysis Laboratory Results**

## AIR TOXICS LTD.

SAMPLE NAME: AV-C-1

ID#: 9309158-01A

Semi-Volatiles By EPA METHOD TO-13 GC/MS Full Scan

File Name:	9092408	Date of Collection:	9/2/93
Dil. Factor:	1100	Date of Extraction:	NA
		Date of Analysis:	9/2/93

Compound	Det. Limit (uG)	Amount (uG)
Phenol	550	Not Detected
bis(2-Chloroethyl) Ether	550	Not Detected
2-Chlorophenol	550	Not Detected
1,3-Dichlorobenzene	550	Not Detected
1,4-Dichlorobenzene	550	Not Detected
1,2-Dichlorobenzene	550	Not Detected
bis(2-Chloroisopropyl) Ether	550	Not Detected
N-Nitroso-d-n-Propylamine	550	Not Detected
Hexachloroethane	550	Not Detected
Nitrobenzene	550	Not Detected
Isophorone	550	Not Detected
2-Nitrophenol	550	Not Detected
2,4-Dimethylphenol	550	Not Detected
Benzoic Acid	550	Not Detected
bis(2-Chloroethoxy) Methane	550	Not Detected
2,4-Dichlorophenol	550	Not Detected
1,2,4-Trichlorobenzene	550	Not Detected
Naphthalene	550	61000
4-Chloroaniline	550	Not Detected
Hexachlorobutadiene	550	Not Detected
4-Chloro-3-Methylphenol	550	Not Detected
2-Methylnaphthalene	550	41000
Hexachlorocyclopentadiene	550	Not Detected
2,4,6-Trichlorophenol	550	Not Detected
2,4,5-Trichlorophenol	550	Not Detected
2-Chloronaphthalene	550	Not Detected
2-Nitroaniline	550	Not Detected
Dimethylphthalate	550	Not Detected
Acenaphthylene	550	59000
2,6-Dinitrotoluene	550	Not Detected
3-Nitroaniline	550	Not Detected
Acenaphthene	550	Not Detected
2,4-Dinitrophenol	550	Not Detected
4-Nitrophenol	550	Not Detected
2,4-Dinitrotoluene	550	Not Detected
Dibenzofuran	550	36000
Diethylphthalate	550	Not Detected
Fluorene	550	35000
4-Chlorophenyl-Phenyl Ether	550	Not Detected
4-Nitroaniline	550	Not Detected
4,6-Dinitro-2-Methylphenol	550	Not Detected
N-Nitrosodiphenylamine	550	Not Detected

(cont.)

## AIR TOXICS LTD.

SAMPLE NAME: AV-C-1

ID#: 9309158-01A

Semi-Volatiles By EPA METHOD TO-13 GC/MS Full Scan

File Name:	9092408	Date of Collection:	9/18/93
Dil. Factor:	1100	Date of Extraction:	NA
		Date of Analysis:	9/24/93

Compound	Det. Limit (uG)	Amount (uG)
4-Bromophenyl-Phenyl Ether	550	Not Detected
Hexachlorobenzene	550	Not Detected
Pentachlorophenol	550	Not Detected
Phenanthrene	550	82000
Anthracene	550	11000
di-n-Butylphthalate	550	Not Detected
Fluoranthene	550	62000
Pyrene	550	46000
Butylbenzylphthalate	550	Not Detected
3,3'-Dichlorobenzidine	550	Not Detected
Chrysene	550	14000
Benzo(a)anthracene	550	11000
bis(2-Ethylhexyl)phthalate	550	Not Detected
Di-n-Octylphthalate	550	Not Detected
Benzo(b)fluoranthene	550	6000
Benzo(k)fluoranthene	550	3700
Benzo(a)pyrene	550	Not Detected
Indeno(1,2,3-cd)pyrene	550	4400
Dibenz(a,h)anthracene	550	Not Detected
Benzo(g,h,i)perylene	550	Not Detected
Biphenyl	550	10000
Quinoline	550	10000
Xylenes	550	Not Detected
Cresols	550	Not Detected

\*Surrogates added after dilution.

Surrogates	% Recovery	Method Limits
2-Fluorophenol	118	25-121
Phenol-d5	81	24-113
Nitrobenzene-d5	117	23-120
2-Fluorobiphenyl	107	30-115
2,4,6-Tribromophenol	100	19-122
Terphenyl-d14	108	18-137

## AIR TOXICS LTD.

SAMPLE NAME: AV-C-2\*

ID#: 9309158-01B

Semi-Volatiles By EPA METHOD TO-13 GC/MS Full Scan

File Name:	NA	Date of Collection:	9/18/93
Dil. Factor:	1.0	Date of Extraction:	NA
		Date of Analysis:	NA

Compound	Det. Limit (uG)	Amount (uG)
Phenol	0.50	Not Analyzed
bis(2-Chloroethyl) Ether	0.50	Not Analyzed
2-Chlorophenol	0.50	Not Analyzed
1,3-Dichlorobenzene	0.50	Not Analyzed
1,4-Dichlorobenzene	0.50	Not Analyzed
1,2-Dichlorobenzene	0.50	Not Analyzed
bis(2-Chloroisopropyl) Ether	0.50	Not Analyzed
N-Nitroso-dl-n-Propylamine	0.50	Not Analyzed
Hexachloroethane	0.50	Not Analyzed
Nitrobenzene	0.50	Not Analyzed
Isophorone	0.50	Not Analyzed
2-Nitrophenol	0.50	Not Analyzed
2,4-Dimethylphenol	0.50	Not Analyzed
Benzoic Acid	0.50	Not Analyzed
bis(2-Chloroethoxy) Methane	0.50	Not Analyzed
2,4-Dichlorophenol	0.50	Not Analyzed
1,2,4-Trichlorobenzene	0.50	Not Analyzed
Naphthalene	0.50	Not Analyzed
4-Chloroaniline	0.50	Not Analyzed
Hexachlorobutadiene	0.50	Not Analyzed
4-Chloro-3-Methylphenol	0.50	Not Analyzed
2-Methylnaphthalene	0.50	Not Analyzed
Hexachlorocyclopentadiene	0.50	Not Analyzed
2,4,6-Trichlorophenol	0.50	Not Analyzed
2,4,5-Trichlorophenol	0.50	Not Analyzed
2-Chloronaphthalene	0.50	Not Analyzed
2-Nitroaniline	0.50	Not Analyzed
Dimethylphthalate	0.50	Not Analyzed
Acenaphthylene	0.50	Not Analyzed
2,6-Dinitrotoluene	0.50	Not Analyzed
3-Nitroaniline	0.50	Not Analyzed
Acenaphthene	0.50	Not Analyzed
2,4-Dinitrophenol	0.50	Not Analyzed
4-Nitrophenol	0.50	Not Analyzed
2,4-Dinitrotoluene	0.50	Not Analyzed
Dibenzofuran	0.50	Not Analyzed
Diethylphthalate	0.50	Not Analyzed
Fluorene	0.50	Not Analyzed
4-Chlorophenyl-Phenyl Ether	0.50	Not Analyzed
4-Nitroaniline	0.50	Not Analyzed
4,6-Dinitro-2-Methylphenol	0.50	Not Analyzed
N-Nitrosodiphenylamine	0.50	Not Analyzed

(cont.)

## AIR TOXICS LTD.

SAMPLE NAME: AV-C-2\*

ID#: 9309158-01B

Semi-Volatiles By EPA METHOD TO-13 GC/MS Full Scan

File Name:	NA	Date of Collection:	9/16/93
DIL Factor:	1.0	Date of Extractions:	NA
		Date of Analysis:	NA

Compound	Det. Limit (uG)	Amount (uG)
4-Bromophenyl-Phenyl Ether	0.50	Not Analyzed
Hexachlorobenzene	0.50	Not Analyzed
Pentachlorophenol	0.50	Not Analyzed
Phenanthrene	0.50	Not Analyzed
Anthracene	0.50	Not Analyzed
di-n-Butylphthalate	0.50	Not Analyzed
Fluoranthene	0.50	Not Analyzed
Pyrene	0.50	Not Analyzed
Butylbenzylphthalate	0.50	Not Analyzed
3,3'-Dichlorobenzidine	0.50	Not Analyzed
Chrysene	0.50	Not Analyzed
Benzo(a)anthracene	0.50	Not Analyzed
bis(2-Ethylhexyl)phthalate	0.50	Not Analyzed
Di-n-Octylphthalate	0.50	Not Analyzed
Benzo(b)fluoranthene	0.50	Not Analyzed
Benzo(k)fluoranthene	0.50	Not Analyzed
Benzo(a)pyrene	0.50	Not Analyzed
Indeno(1,2,3-cd)pyrene	0.50	Not Analyzed
Dibenz(a,h)anthracene	0.50	Not Analyzed
Benzo(g,h,i)perylene	0.50	Not Analyzed
Biphenyl	0.50	Not Analyzed
Quinoline	0.50	Not Analyzed
Xylenes	0.50	Not Analyzed
Cresols	0.50	Not Analyzed

\*Not analyzed per client's request.

Surrogates	% Recovery	Method Limits
2-Fluorophenol	NA	25-121
Phenol-d5	NA	24-113
Nitrobenzene-d5	NA	23-120
2-Fluorobiphenyl	NA	30-115
2,4,6-Tribromophenol	NA	19-122
Terphenyl-d14	NA	18-137

## AIR TOXICS LTD.

SAMPLE NAME: H2O BLANK

ID#: 9309158-03A

Semi-Volatiles By EPA METHOD TO-13 GC/MS Full Scan

File Name:	9092507	Date of Collection:	9/16/93
DIL Factor:	1.0	Date of Extraction:	9/24/93
		Date of Analysis:	9/25/93

Compound	Det. Limit (uG)	Amount (uG)
Phenol	0.50	Not Detected
bis(2-Chloroethyl) Ether	0.50	Not Detected
2-Chlorophenol	0.50	Not Detected
1,3-Dichlorobenzene	0.50	Not Detected
1,4-Dichlorobenzene	0.50	Not Detected
1,2-Dichlorobenzene	0.50	Not Detected
bis(2-Chloroisopropyl) Ether	0.50	Not Detected
N-Nitroso-di-n-Propylamine	0.50	Not Detected
Hexachloroethane	0.50	Not Detected
Nitrobenzene	0.50	Not Detected
Isophorone	0.50	Not Detected
2-Nitrophenol	0.50	Not Detected
2,4-Dimethylphenol	0.50	Not Detected
Benzoic Acid	0.50	Not Detected
bis(2-Chloroethoxy) Methane	0.50	Not Detected
2,4-Dichlorophenol	0.50	Not Detected
1,2,4-Trichlorobenzene	0.50	Not Detected
Naphthalene	0.50	Not Detected
4-Chloroaniline	0.50	Not Detected
Hexachlorobutadiene	0.50	Not Detected
4-Chloro-3-Methylphenol	0.50	Not Detected
2-Methylnaphthalene	0.50	Not Detected
Hexachlorocyclopentadiene	0.50	Not Detected
2,4,6-Trichlorophenol	0.50	Not Detected
2,4,5-Trichlorophenol	0.50	Not Detected
2-Chloronaphthalene	0.50	Not Detected
2-Nitroaniline	0.50	Not Detected
Dimethylphthalate	0.50	Not Detected
Acenaphthylene	0.50	Not Detected
2,6-Dinitrotoluene	0.50	Not Detected
3-Nitroaniline	0.50	Not Detected
Acenaphthene	0.50	Not Detected
2,4-Dinitrophenol	0.50	Not Detected
4-Nitrophenol	0.50	Not Detected
2,4-Dinitrotoluene	0.50	Not Detected
Dibenzofuran	0.50	Not Detected
Diethylphthalate	0.50	Not Detected
Fluorene	0.50	Not Detected
4-Chlorophenyl-Phenyl Ether	0.50	Not Detected
4-Nitroaniline	0.50	Not Detected
4,6-Dinitro-2-Methylphenol	0.50	Not Detected
N-Nitrosodiphenylamine	0.50	Not Detected

(cont.)

**AIR TOXICS LTD.**

SAMPLE NAME: H2O BLANK

ID#: 9309158-03A

Semi-Volatiles By EPA METHOD TO-13 GC/MS Full Scan

File Name:	9092507	Date of Collection:	9/16/93
Dil. Factor:	1.0	Date of Extraction:	9/24/93
		Date of Analysis:	9/25/93

Compound	Det. Limit (uG)	Amount (uG)
4-Bromophenyl-Phenyl Ether	0.50	Not Detected
Hexachlorobenzene	0.50	Not Detected
Pentachlorophenol	0.50	Not Detected
Phenanthrene	0.50	Not Detected
Anthracene	0.50	Not Detected
di-n-Butylphthalate	0.50	Not Detected
Fluoranthene	0.50	Not Detected
Pyrene	0.50	Not Detected
Butylbenzylphthalate	0.50	Not Detected
3,3'-Dichlorobenzidine	0.50	Not Detected
Chrysene	0.50	Not Detected
Benzo(a)anthracene	0.50	Not Detected
bis(2-Ethylhexyl)phthalate	0.50	Not Detected
Di-n-Octylphthalate	0.50	Not Detected
Benzo(b)fluoranthene	0.50	Not Detected
Benzo(k)fluoranthene	0.50	Not Detected
Benzo(a)pyrene	0.50	Not Detected
Indeno(1,2,3-cd)pyrene	0.50	Not Detected
Dibenz(a,h)anthracene	0.50	Not Detected
Benzo(g,h,i)perylene	0.50	Not Detected
Biphenyl	0.50	Not Detected
Quinoline	0.50	Not Detected
Xylenes	0.50	Not Detected
Cresols	0.50	Not Detected

Surrogates	% Recovery	Method Limits
2-Fluorophenol	74	25-121
Phenol-d5	51	24-113
Nitrobenzene-d5	83	22-120
2-Fluorobiphenyl	58	30-115
2,4,6-Tribromophenol	56	19-122
Terphenyl-d14	94	18-137



**AIR TOXICS LTD.**  
AN ENVIRONMENTAL ANALYTICAL LABORATORY

100 BLUE RAVINE ROAD, SUITE B  
FOLSOM, CA 95630  
(916) 985-1000 • FAX (916) 985-1020

## CHAIN OF CUSTODY RECORD

Page 3 of     

PROJECT # 01-1179-03-4790 PO #       
REMARKS     

COLLECTED BY (Signature) [Signature]

FIELD SAMPLE I.D.#	SAMPLING MEDIA (Nasal, Canister etc.)	DATE/TIME	ANALYSIS	VAC./PRESSURE	LAB I.D. #
01A AV-C-1	250 ml bottle	9-16-93	sample GC/MS		
01B AV-C-2			duplicate sample		
02A DB-C-4a			sample GC/MS		
02B DB-C-4b			duplicate sample		
03A H <sub>2</sub> O Blank	300 ml bottle		GC/MS		

RELINQUISHED BY: DATE/TIME

RECEIVED BY: DATE/TIME

RELINQUISHED BY: DATE/TIME

RECEIVED BY: DATE/TIME

[Signature] 9/20/93

LAB USE ONLY

SHIPPER NAME

AIR BILL #

OPENED BY: DATE/TIME

TEMP (°C)

CONDITION

REMARKS

**Appendix B**

**Calibration Gas Certifications**

**(included in Final Report)**



# Scott Specialty Gases, Inc.

Shipped  
From:

1290 COMBERMERE STREET  
TROY MI 48083  
Phone: 313-589-2950

Fax: 313-589-2134

## CERTIFICATE OF ANALYSIS

C A E INSTRUMENT RENTAL

246 WOODWORK LANE

PALATINE

IL 60067

PROJECT #: 05-52393-002  
PO#: 8673-71500  
ITEM #: 05023451 2AL  
DATE: 6/18/93

CYLINDER #: AAL9218

ANALYTICAL ACCURACY: +-1%

BLEND TYPE : ACUBLEND MASTER GAS

COMPONENT  
PROPANE  
NITROGEN

REQUESTED GAS  
CONC MOLES  
850. PPM  
BAL

ANALYSIS  
(MOLES)  
851.6 PPM  
BAL

ACUBLEND MASTER GAS  
ALM030711 ALM008587

ANALYTICAL METHOD: AMG

ANALYST: Robert Lane

APPROVED BY: J. C. Shapiro

ANALYSTS: PHILADELPHIA, PENNSYLVANIA / TROY, MICHIGAN / HOUSTON, TEXAS / DUNSMITH, ILLINOIS

1992



# Scott Specialty Gases, Inc.

1200 COMBERMERE STREET TROY, MI 48063 PHONE: (313) 589-2050 FAX NO: (313) 589-2134

C A E INSTRUMENT RENTAL  
246 WOODWORK LANE  
PALATINE, IL. 60067

Date: 02/27/92

Our Project No.: 0534012

Your P.O. No.: 4559-71500

Gentlemen:

Thank you for choosing Scott for your Specialty gas needs. The analyses for the gases ordered, as reported by our laboratory, are listed below. Results are in volume percent, unless otherwise indicated.

## ANALYTICAL REPORT

Cyl. No. _____	Analytical Accuracy $\pm 1\% \text{CMG}$
Component	Concentration
PROPANE	54.50PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	
ALM-004703, ALM-004944, AAL-9952	

Cyl. No. _____	Analytical Accuracy $\pm 1\% \text{CMG}$
Component	Concentration
PROPANE	2495PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	
ALM-008679, ALM-010619, ALM-010140	

Cyl. No. _____	Analytical Accuracy $\pm 1\% \text{CMG}$
Component	Concentration
PROPANE	5496PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	
ALM-010293, ALM-010029, ALM-011352	

Cyl. No. _____	Analytical Accuracy $\pm 1\% \text{CMG}$
Component	Concentration
PROPANE	8423PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	
ALM-003391, ALM-004786, ALM-004952	

Analyst David Bano

Approved J Shapers

The only liability of this Company for gas which fails to comply with this analysis shall be replacement thereof by the Company without extra cost.

CERTIFIED REFERENCE MATERIALS    EPA PROTOCOL GASES  
 ACUBLEND®    CALIBRATION & SPECIALTY GAS MIXTURES    PURE GAS  
 ACCESSORY PRODUCTS    CUSTOM ANALYTICAL SERVICES

PLUMBSTEADVILLE, PENNSYLVANIA / SAN BERNARDINO, CALIFORNIA / HOUSTON, TEXAS / WHEELING, ILLINOIS



# Scott Specialty Gases, Inc.

Shipped  
From:

1290 COMBERMERE STREET  
TROY MI 48083  
Phone: 313-589-2950

Fax: 313-589-2134

## CERTIFICATE OF ANALYSIS

C A E INSTRUMENT RENTAL

246 WOODWORK LANE

PALATINE

IL 60067

PROJECT #: 05-50366-003

PO#: 8266-71500

ITEM #: 05023451 2K

DATE: 4/27/93

CYLINDER #: K000379

ANALYTICAL ACCURACY: +-1%

BLEND TYPE : ACUBLEND MASTER GAS

COMPONENT	REQUESTED GAS		ANALYSIS	
	CONC MOLES		(MOLES)	
PROPANE	500.	PPM	504.9	PPM
NITROGEN		BAL		BAL

ACUBLEND MASTER GAS  
K011786 K000735

ANALYTICAL METHOD: AMG

ANALYST: PLT APPROVED BY: J. C. Shapiro

**ANALYTICAL REPORT - cont'd**

C A E INSTRUMENT RENTAL

Date: 11/22/91

Our Project No.: 0530468

Your P.O. No.: 3750-71500

**COPIES**

Cyl. No. _____	Analytical Accuracy $\pm 1\% \text{MG}$
Component	Concentration
PROPANE	851.3PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	
ALM-008898, ALM-004772, ALM-015402	

Cyl. No. <u>K-006450</u>	Analytical Accuracy $\pm 1\% \text{MG}$
Component	Concentration
PROPANE	2479PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	

Cyl. No. <u>K-001820</u>	Analytical Accuracy $\pm 1\% \text{MG}$
Component	Concentration
PROPANE	8490PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	

Cyl. No. _____	Analytical Accuracy $\pm 1\% \text{MG}$
Component	Concentration
SULFUR DIOXIDE	25.40PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	
AAL-8382, AAL-3001, AAL-989	

Cyl. No. _____	Analytical Accuracy $\pm 1\% \text{MG}$
Component	Concentration
SULFUR DIOXIDE	224.7PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	
AAL-13200, AAL-5250, AAL-2882	

Cyl. No. _____	Analytical Accuracy $\pm 1\% \text{MG}$
Component	Concentration
CARBON MONOXIDE	10.08PPM
NITROGEN	BALANCE
ACUBLEND MASTER GAS	
ALM-013733, ALM-010002, ALM-016772	

Analyst *[Signature]*

Approved By *[Signature]*

**Scott Specialty Gases**

**CERTIFICATE OF ANALYSIS**

SCIENCE APPLICATIONS PD No. 5953735

CPS Reg. Component No.	Component	Certified Analysis
74-98-6	PROPANE	5.00 PCT/V
7727-37-9	NITROGEN	BRL

Analysis Date 06/31/93 Project No 01-58270  
 Analytical Accuracy  $\pm 2\%$  Analyst WALTER SPITUS Cylinder No. A1003  
 Item No 01023453 4R

Grade CERTIFIED MASTER GAS

Reorder/Service Contact (215)766-8861 PLUMSTEDVILLE PA 18949

**Scott Specialty Gases**

**CERTIFICATE OF ANALYSIS**

SCIENCE APPLICATIONS PD No. 5953735

CPS Reg. Component No.	Component	Certified Analysis
74-98-6	PROPANE	9.00 PCT/V
7727-37-9	NITROGEN	BRL

Analysis Date 06/31/93 Project No 01-58270  
 Analytical Accuracy  $\pm 2\%$  Analyst WALTER SPITUS Cylinder No. X96549  
 Item No 01023453 4R

Grade CERTIFIED MASTER GAS

Reorder/Service Contact (215)766-8861 PLUMSTEDVILLE PA 18949

## **Appendix C**

### **Method 4 Calibration Data**

## Calibration Data Sheet

### Dry Gas Meter Against a Calibration ~~Test~~ Test Gas Meter

Dry Gas Meter Id.: DMST 20 Calibration Meter Id.: DMSE 19 Date: SEPT 28 '73  
 Barometric Pressure: 30.01 " Hg Cal. meter coefficient ( $Y_d$ ) 1.001  
 Orifice leak-check: Yes        No        Calibrated by: AS

		Run #1	Run #2	Run #3	Run #4
Cal. Meter Vol., Initial	ft <sup>3</sup>	178.398	187.238	196.098	
Cal. Meter Vol. Final	ft <sup>3</sup>	187.238	196.098	204.779	
Cal. Meter Vol. Diff. ( $V_d$ )	ft <sup>3</sup>	8.840	8.860	8.677	
Dry Gas Meter Vol. Initial	ft <sup>3</sup>	74.169	723.343	732.600	
Dry Gas Meter Vol. Final	ft <sup>3</sup>	723.343	732.600	741.734	
Dry Gas Meter Vol. Diff. ( $V_d$ )	ft <sup>3</sup>	9.174	9.257	9.134	
Cal. Meter Temp. ( $T_d$ )	°F	69	69	69	
Dry Gas Meter Temp. ( $T_d$ )	°F	72 <sub>1.000</sub>	78 <sub>1.007</sub>	81 <sub>1.023</sub>	
Time (t)	min	10	10	10	
<del><math>\Delta H</math> Wet</del>	<del>in. Hg</del>	<del>      </del>	<del>      </del>	<del>      </del>	<del>      </del>
$\Delta H$ Dry	in. Hg	2.5" H <sub>2</sub> O	2.5" H <sub>2</sub> O	2.5" H <sub>2</sub> O	
Dry Gas Meter Coeff. ( $Y_d$ )	N/A	.965	.969	.966	
Flow rate	SCFM	1.30 <sup>85</sup>	.887	.869	
$Y_{ds}$ from Graph	N/A	<del>      </del>	<del>      </del>	<del>      </del>	<del>      </del>
Average Dry Gas Meter Coefficient					.967

Comments: POST-TEST CAL - EPA-EMB WA-4 01-1129-03-4790-006  
 $Y = .967$  7% DEVIATION @ 0.97

Where:  $\Delta H_{@}$  = Orifice pressure differential that gives 0.75 cfm of air at 70°F and 29.92 inches of mercury, in. H<sub>2</sub>O. Acceptable maximum dry gas meter calibration tolerance  $\pm 0.15$ .

$$Y_d = Y_{ds} \times \frac{V_{ds}}{V_d} \times \frac{T_d + 460}{T_{ds} + 460} \times \frac{P_b + \frac{\Delta H}{13.6} \text{ dry}}{P_b + \frac{\Delta H}{13.6} \text{ wet}} \times$$

1.001 . . . . . 0.974

$$Q = 17.65 \times \frac{30.01}{P_b} \times \frac{V_{ds}}{T_{ds} + 460} \times \frac{1}{t}$$

$$\Delta H_{@} = \frac{0.0317 \Delta H}{P_b (T_d + 460)} \times \frac{(T_{ds} + 460) \theta}{V_{ds}}$$

## Calibration Data Sheet

### Meter Box Gas Meter Against a Calibration Dry Gas Meter

Box Meter Id.: OMST20      Calibration Meter Id.: OMSE19      Date: 6.4.93  
 Barometric Pressure: 29.63      ( $Y_{ds}$ ) Cal. meter coefficient 1.001  
 Orifice leak-check: cross      Calibrated by: JT

	Run #1	Run #2	Run #3	Run #4	*Run #5	+Run #6
Cal. Meter Vol. Beg.	987.767	989.926	996.958	1007.440	1017.779	1027.1
Cal. Meter Vol. End	987.767	995.588	1004.784	1017.105	1025.758	1034.9
Cal. Meter Vol. Dif. ( $V_{ds}$ )	8.030	5.662	7.926	9.665	7.979	7.8
Box Meter Vol. Beg.	485.887	496.740	504.108	515.280	526.171	536.11
Box Meter Vol. End	494.443	502.765	512.490	525.447	534.708	544.41
Box Meter Vol. Dif. ( $V_d$ )	8.556	6.025	8.382	10.167	8.537	8.30
Cal. Meter Temp. (F) ( $T_{ds}$ )	67	67	67	67	67	68
Box Meter Temp. (F) ( $T_d$ )	79	81	82	84	81	82
Time (mins.) (t)	20:00	10:00	10:00	10:00	20:00	10:
$\Delta H$	.5	1.0	2.0	3.0	.5	2.0
$\Delta H@$	1.706	1.710	1.742	1.751	1.722	1.78
Box Meter Coeff. ( $Y_d$ )	.956	.964	.969	.975	.971	.960
CFM	.398	.562	.787	.959	.396	.77

Comments: 6 MONTH CALIBRATION - OMST20  
AVERAGE  $Y = .967$       AVERAGE  $\Delta H@ = 1.736$       \*2<sup>nd</sup> RUN @ 5" TO VERIFY 1"  
 Where:  $\Delta H@$  = Orifice pressure differential that gives 0.75 cfm of air at 70°F and 29.92 inches  
 of mercury, in. H<sub>2</sub>O. Tolerance  $\pm 0.15$ .  
+2<sup>nd</sup> RUN @ 20" TO VERIFY  
AND ALL OTHER RESULTS

$$Y_d = Y_{ds} \times \frac{V_{ds}}{V_d} \times \frac{T_d + 460}{T_{ds} + 460} \times \frac{P_b}{P_b + \frac{\Delta H}{13.6}}$$

$$Q = 17.65 \times \frac{P_b}{T_{ds} + 460} \times \frac{V_{ds}}{t}$$

$$\Delta H@ = \frac{0.0317 \Delta H}{P_b (T_d + 460)} \times \frac{(T_{ds} + 460)^3}{V_{ds}}$$



Environmental

Services, Inc.

OMST20M.CAL

## Temperature Calibration Data Sheet Method 5 Sample Train

Temperature Monitor Type: Digital Readout  
 Identification Number: OMSTR0

Reference Temperature Monitor Type: Mercury-in-glass thermometer  
 Identification Number: OMSTRNBS  
 Readout Device: Mercury in glass.

Umbilical Cord ID# 13 Control Box ID# OMSTR0  
 Calibration Performed by JS Ambient Temperature 66  
 Date 6-4-73 Barometric Pressure 30.00

Reference Point Source	Mercury Temp. of Ref. Point (°F)	Temperature Monitor (°F)						
		Hot Box	Gas Meter		Umbilical (Imp. Exit)	Sample Probe		
			In	Out		Stack	Probe	ID#
Ice Bath	32°	A	30	30	1 31	1 30	1 30	50
					2	2	2	
					3	3	3	
						4	4	
Boiling Water	212°	A	210	210	1 213	1 212	1 210	50
					2	2	2	
					3	3	3	
						4	4	
500 °F NBS traceable thermometer in tube furnace	549°	A	-	-	1 -	1 538°	1 -	50
					2	2	2	
					3	3	3	
						4	4	

**Appendix D**

**Operator Log Sheets and Process Data Sheets**



DATE: 9-8-93  
 CHARGE NO.: 349  
 CYLINDER NO.: 1  
 LOCATION: AVOCA, PA.  
 HERR Mc GEE

**SUMMARY OF TREATING OPERATIONS**

No. of Pieces	Material & Species	Size	Seasoning	Bd. Ft.	Cu. Ft.
1020	X-TIES OAK	6" x 8'6"	GREEN		2887
<b>TOTALS</b> 1020					2887

Retention Req.: 7 LBS.      Retention Actual: 7.03      Water Removal: 5.19 Lbs/Cu. Ft.  
 Total Time: 2 1/2 hrs.      Treating Time: 2 hrs.      Boulton Time: 14 hrs.

**Pretreatment Conditioning**

**Boulton Cycle**

Tank Reading Before Filling:	35220	Temp:	187	Time:	0839	AM
Tank Reading After Filling:	15910	Temp:	176	Time:	0909	AM
Amount Creosote For Boulton, Actual:	16310	Required:				
Tank Reading After Pump back:	33450	Temp:	186	Time:	2322	AM
Creosote Used In Boulton:	1170	Retention:	369			
Water Removed Last Hour, Gals.:	53	Lbs/Cu. Ft.:	.152			
Total Water Removed, Gals.:	1802	Lbs.:	15011	Lbs/Cu. Ft.	519	
<b>Boulton Cycle Readings</b>	<b>Maximum</b>	<b>Average</b>	<b>Minimum</b>			
Temperature, °F	187	185	154			
Vacuum, In. Hg.	24.4	21.0	19.9			

**Treating Cycle**

Tank Reading Before Filling:	33450	Temp:	172	Time:	2320	AM
Initial Air, P.S.I.:	30	Start Time:	2320	Finish Time:	2322	AM
Final Vacuum, In. Hg.:	23	Start Time:	0205	Finish Time:	0605	AM
Final Tank Reading:	32500	Temp:	165	Time:	0205	AM
Creosote Used In Treating:	1070	Retention:	3.37			
Total Creosote Used:	2240	Retention:	7.03			

**Attach Treatment Log, Charts, Check Lists & Borings**

Treating Cycle Readings	Maximum	Average	Minimum
Temperature, °F	195	190	185
Pressure, P.S.I.	200	200	30
Vacuum, In. Hg.	24.0	21.6	-

**Post-Treatment Conditioning**

Toluene Extraction Yes  No  Results:

Treating Operator Comments:  
 Signature: M. G.      Date: 9-8-93

Plant O.C. Comments:  
 Signature: Barry K. ...      Date: 9-9-93

Plant Manager Comments:

Conrail O.C. Comments:

