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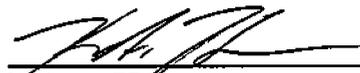
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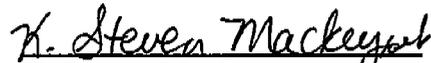
November 15, 1994

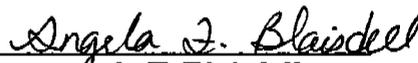
Prepared For:

LOUISIANA-PACIFIC CORPORATION
ORIENTED STRANDBOARD FACILITY
FUEL DRYER E-TUBE STACK
MONTROSE, COLORADO
OCTOBER 25, 1994

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We certify that the information contained herein is accurate and complete to the best of our knowledge.

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1.0

INTRODUCTION

The purpose of this source emission evaluation was to quantify emissions of particulate and condensible matter at the fuel dryer E-tube exhaust stack at Louisiana-Pacific Corporation's oriented strandboard (OSB) facility located in Montrose, Colorado. This testing was conducted to demonstrate compliance with the Colorado Department of Health permit requirements. Louisiana-Pacific Corporation contracted Am Test-Air Quality, Inc. based in Preston, Washington to conduct these source tests.

Testing and analysis procedures used for this project are presented in the July 1, 1993 edition of the Environmental Protection Agency (EPA) document Title 40, Code of Federal Regulations, Part 60 (40 CFR 60), Appendix A, Methods 1, 2, 3A, 4 and 5 and in 40 CFR 51, Appendix M, Method 202. Methods 1 and 2 were performed to measure the stack gas temperature and velocity for calculating volumetric flow rate. Method 3A was performed to determine the molecular weight of the stack gas based on measurements of the oxygen (O₂) and carbon dioxide (CO₂) concentration in the stack gas. Method 4 was performed to measure the moisture content of the stack gas. Method 5 and Method 202 were performed to quantify particulate and condensible matter emissions. The Colorado Department of Health requires that the condensible particulate matter present in the gas stream be quantified by performing an extraction of the back-half portion of the Method 5 sample train. Three (3) Method 1, 2, 3A, 4 and 5/202 samples were collected at the dryer E-tube stack on October 25, 1994.

Mr. K. Steven Mackey of Am Test-Air Quality, Inc. performed the field sampling. Sample recovery and gravimetric laboratory analysis of the Method 5 samples were performed by Ms. Stacy Akin and Ms. Annika M. Woehr of Am Test. Data reduction, quality assurance review and final report preparation were performed by Mr. Kris A. Hansen, Ms. Angela F. Blaisdell, Ms. Cassie B. Heaton and Ms. Annika M. Woehr of Am Test. This testing program was coordinated by Ms. Susan Somers and Mr. Dick Klopschinski of Louisiana-Pacific Corporation who also provided process data. Mr. Charlie Cox of the Colorado Department of Health observed the testing.

2.0

SUMMARY OF RESULTS

The following section of this report presents results from the particulate matter testing performed at the dryer E-tube stack. Refer to the Table of Contents to locate specific information for each test. The summary tables in this section contain information obtained from computer printouts for each individual run which are included in Appendix A of this report. Appendix B of this report contains example calculations of results and copies of the original field data sheets. Appendix C of this report contains process data. Appendix D of this report contains miscellaneous supporting information.

The results of the three (3) 60-minute Method 5/202 tests for quantifying particulate and condensable matter emissions at the dryer E-tube stack are summarized in Table 2.0 below, and on the following computer printout titled "Summary of Results - Methods 1, 2, 3A, 4 and 5".

Table 2.0. Summary of particulate matter emission test results from samples collected on October 25, 1994 at the dryer E-tube stack at Louisiana-Pacific Corporation's facility in Montrose, Colorado.

Sample Run #	Front-half P.M. Emission Conc. (gr/dscf)	Back-half P.M. Emission Conc. (gr/dscf)	Total P.M. Emission Conc. (gr/dscf)	Total P.M. Emission Rate (lb/hr)
1	0.007	0.021	0.028	8.20
2	0.002	0.021	0.022	6.65
3	0.007	0.023	0.030	8.66
Average	0.005	0.022	0.027	7.84

The front-half, back-half and total particulate matter emission concentrations were calculated in units of grains per dry standard cubic foot (gr/dscf) and milligrams per dry standard cubic meter (mg/dscm). The total particulate matter emission rates were calculated in units of pounds per hour (lb/hr). The particulate matter emission standard for this source is 7.44 lb/hr.



SUMMARY OF RESULTS - METHODS 1, 2, 3A, 4, 5 AND 202
AM TEST - AIR QUALITY, INC.

FILE NAME: S721\LPSUM
CLIENT: Louisiana-Pacific Corporation
LOCATION: Montrose, Colorado

DRYER E-TUBE STACK

	RUN #1	RUN #2	RUN #3	AVERAGE
LAB #:	6787	6788	6789	
DATE:	10/25/94	10/25/94	10/25/94	
START TIME:	08:16	10:18	12:13	
STOP TIME:	09:34	11:22	13:15	
SAMPLE LENGTH (minutes):	60.0	60.0	60.0	
VOLUME SAMPLED (cubic feet):	50.236	54.836	55.564	53.545
VOLUME SAMPLED (dry std. cubic feet):	43.329	45.176	44.799	44.435
VOLUME SAMPLED (dry std. cubic meters):	1.227	1.279	1.269	1.258
STACK GAS MOISTURE (percent):	22.01	20.99	23.03	22.01
BAROMETRIC PRESSURE (inches of Hg):	25.22	25.22	25.19	25.21
STATIC PRESSURE (inches of H2O):	0.15	0.15	0.12	0.14
STACK PRESSURE (inches of Hg):	25.23	25.23	25.20	25.22
STACK TEMPERATURE (degrees F.):	150.5	149.8	151.8	150.7
STACK TEMPERATURE (degrees R.):	610.5	609.8	611.8	610.7
CARBON DIOXIDE (percent):	2.2	2.2	2.2	2.2
OXYGEN (percent):	17.3	17.3	17.1	17.2
MOLECULAR WEIGHT (dry, g/g-mole):	29.04	29.04	29.04	29.04
MOLECULAR WEIGHT (wet, g/g-mole):	26.61	26.73	26.49	26.61
AVERAGE VELOCITY HEAD (inches of H2O):	1.348	1.368	1.343	1.353
PITOT TUBE Cp:	0.84	0.84	0.84	
STACK GAS VELOCITY (feet per second):	79.5	79.9	79.7	79.7
STACK DIAMETER (inches):	48.0	48.0	48.0	
STACK AREA (square feet):	12.6	12.6	12.6	
STACK GAS AIRFLOW (dry std. cubic feet per min.):	34099.7	34744.1	33606.0	34149.9
STACK GAS AIRFLOW (actual cubic feet per min.):	59952.1	60222.6	60064.8	60079.8
NOZZLE DIAMETER (inches):	0.230	0.230	0.230	
ISOKINETICS (percent):	92	94	97	
FRONT-HALF PARTICULATE EMISSION CONC. (gr/dscf):	0.007	0.002	0.007	0.005
BACK-HALF PARTICULATE EMISSION CONC. (gr/dscf):	0.021	0.021	0.023	0.022
TOTAL PARTICULATE EMISSION CONC. (gr/dscf):	0.028	0.022	0.030	0.027
TOTAL PARTICULATE EMISSION CONC. (mg/dscm):	64.2	51.1	68.8	61.4
TOTAL PARTICULATE MATTER EMISSION RATE (lb/hr):	8.20	6.65	8.66	7.84

3.0 PROJECT OVERVIEW/EXCEPTIONS

An acceptable leak check of less than 0.02 cfm at the highest vacuum rate (or greater) used during the test preceded and followed each run. The average percentage isokinetics for each run were within the acceptable limits of $100 \pm 10\%$.

During run 1 there was a power failure which delayed the end of the test by 15 minutes.

A nitrogen purge to remove sulfates from the sample is an option of Method 202. It was not performed since no sulfates were expected from this source. The pH of each back-half sample was measured and confirmed to be greater than 4.5 so the Method 202 analysis for sulfates was not performed.

4.0

SOURCE OPERATION

Louisiana-Pacific Corporation's Montrose, Colorado facility is an oriented strand board (OSB) manufacturing plant with a rotary dryer. The MEC 60-foot triple pass rotary dryer is equipped with multiclones and a Geoenergy E-tube wet electrostatic precipitator (ESP) to control emissions. Dry wood fines and waferboard trimmings are used to fire a McConnell burner rated at 40 million British thermal units per hour (MMBtu/hr). The burner heats the triple pass rotary dryer which is used to dry wood chips for OSB production. The process was operated on primarily aspen with 25% pine. Dryer exhaust gases are controlled by multiclones, followed by an ID fan, a quench duct and a Geoenergy E-tube wet ESP. The scrubber water had a 25-28% solids content. A process information sheet, completed in the field, is included in the appendices of this report.

5.0

SAMPLING AND ANALYSIS PROCEDURES

5.1 EPA Methods 1 and 2 - Velocity, Temperature and Airflow

EPA Method 1 procedures were used to assure that representative measurements of volumetric flow rate were obtained by dividing the cross-section of the stack or duct into a number of equal areas, and then locating a traverse point within each of the equal areas. Refer to the "Stack Schematic and Location of Sample Points" data sheet and/or the figure titled, "Location of Sampling Ports and Traverse Points", located in the appendices of this report for a schematic of the stack and the point locations selected for testing. Method 2 was performed to measure the stack gas velocity using a type S or a standard pitot tube, and the gas temperature using a calibrated thermocouple probe connected to a digital thermocouple indicator. The type S pitot tubes were connected with tubing to an oil-filled inclined manometer, a hook gauge manometer or magnehelic gauges to obtain velocity measurements. The pitot tube lines were leak-checked and the pressure measurement device was leveled and zeroed prior to use. Calibration information for each pressure and temperature measurement device used are included in the appendices of this report.

5.2 EPA Method 3A - Molecular Weight

The stack gas composition was evaluated during each test by collecting integrated samples of the stack gas in multilayer bags. The integrated bag samples were analyzed both on-site and in Am Test's laboratory using a paramagnetic analyzer to measure the percent (%) O₂. The samples were analyzed in the laboratory using a non-dispersive infrared (NDIR) analyzer to measure the % carbon dioxide (CO₂) in

the gas stream. The gas composition data were used to calculate the molecular weight of the stack gas.

5.3 EPA Method 4 - Moisture

The percent moisture in the gas stream was quantified by weighing the impingers to 0.1 grams before and after each Method 5 run on a digital top-loading balance. The net weight (final minus initial) was used to calculate the amount of moisture condensed from the known volume of stack gas collected.

5.4 EPA Method 5/202 - Particulate and Condensible Matter

The sample train used for collecting particulate matter samples was an EPA Method 202 design as illustrated in the figure titled "Method 202 Sample Train" in the appendices of this report. The "Sample Train Information Sheet" (also in the appendices) details the type of nozzle, probe, probe liner and filter used along with the contents of the sample train impingers. The gas is drawn from the stack through a quartz nozzle and heated quartz probe liner, through a pre-tared front-half quartz filter, and into a Method 202 impinger train for collecting condensible particulate matter. A stainless steel probe sheath with a heated quartz probe liner was used to draw gas from the filter to the impinger train. The probe was equipped with type S pitot tubes for measuring gas velocity and a thermocouple sensor for measuring stack gas temperature. The thermocouple sensor was connected to a digital thermocouple indicator which was used to measure the stack gas temperature at each sample point. The temperature of the probe liner was monitored to assure that condensation did not occur within the probe liner. The nozzle, probe liner and filter are often referred to as the "front-half" of the sample train. Following the filter was a probe and condenser (impinger) section which, by convention, is referred to as the "back-half". The impinger section was maintained at a

temperature below 68° F by keeping ice on the impingers. The temperature at the outlet of the silica gel bubbler was monitored to verify that it did not exceed 68° F during a test. Note: Some subparts or methods specify alternate temperatures for the probe liner, filter holder box and impinger ice bath. The "Sample Train Information Sheet" details any exceptions.

The sample train was connected to a control box by means of an umbilical cord which contained a vacuum hose, pitot lines, thermocouple wires and a 4-wire electrical cord. The control box (meter box) was used to monitor stack conditions and to facilitate isokinetic sampling. The control box consists of a leak-free pump used to pull the stack gas through the sample train, fine and coarse metering valves to control the sampling rate, a vacuum gauge which measures the pressure drop from the sampling nozzle to the metering valves, and a calibrated dry gas meter readable to 0.001 cubic feet. The dry gas meter inlet and outlet temperatures were monitored by thermocouples which were connected to a multichannel thermocouple indicator. The dry gas meter calibration factor, Y, was determined by calibrating the meter against a wet test meter or calibrated dry gas meter. At the outlet of the dry gas meter was a calibrated orifice which was used to monitor the flow of gas through the metering system to assure that samples were collected isokinetically. The pressure drop across the orifice was monitored at each sample point. The pitot tubes utilized to measure stack gas velocity were connected to the control box via the umbilical cord. The control box contained either low and high range magnehelic gauges or an incline manometer, which were used for the velocity measurement.

Stack condition measurements were made prior to collecting a sample, including measurements of velocity, temperature and a check for cyclonic flow in the stack. A

sample nozzle was selected and isokinetic operating parameters were established utilizing a Hewlett-Packard programmable calculator. The sampling nozzle, probe and prefilter connective glassware were cleaned and rinsed with dichloromethane prior to use. The sample train was assembled and determined to be leak free following the procedures outlined in Method 5. The sample nozzle was positioned in the stack at the first sample point. The sample pump was turned on, and the gas sampling rate was adjusted for isokinetic sampling. At the conclusion of the test, the pump was turned off, the probe was removed from the stack, and a post-test leak check was performed according to Method 5 procedures. Following the leak check, the impingers were removed from the sample train. A nitrogen purge to remove sulfates from the sample is an option of Method 202. It was not performed since no sulfates were expected from this source. The front-half of the sample train was recovered with acetone per EPA Method 5 procedures and the particulate matter was quantified gravimetrically after evaporation of the acetone. The back-half of the sample train was recovered per EPA Method 202.

Following sample collection, the Method 5 sample trains were transferred to an area free from air disturbances and airborne particulate matter. The filters were transferred to petri dishes labeled with the sample date, client name and run number. Am Test refers to the front-half filter portion of the particulate catch as the "A" section. Care was taken to assure that any loose particulate matter and filter mat were quantitatively transferred to the petri dish. In the laboratory, the filters were placed in an oven and baked at 105° C for two (2) hours, then transferred to a constant humidity desiccator containing silicon dioxide (SiO₂) for at least 24 hours of desiccation prior to obtaining weights. The same weighing procedures were followed to obtain the tare weights for the filters. The tare and final weights were made using an electronic balance set to a time integrating mode with a readability

of 0.1 milligrams. The filters were weighed to a constant weight of ± 0.5 milligrams. The interval between weighings was at least 6 hours. These weights were recorded in a bound laboratory notebook.

Following each run, the contents of the nozzle, probe liner and prefilter connective glassware were quantitatively transferred to the "B" section storage container labeled with sample date, client name and run number. Several rinses of acetone, with simultaneous loosening of particulate matter using a clean nylon brush, were used for the front-half clean-up. An iodine flask with a female ball joint end was attached to the male ball joint end of the probe to assure that no particulate matter was lost during the rinsing and brushing of the probe. The contents of the iodine flask were quantitatively transferred to the "B" section storage container. In the laboratory, the "B" section acetone rinse was transferred to a tared, graduated 150 milliliter (mL) beaker. The volume of acetone was recorded and the beakers were placed in an evaporation chamber at a temperature of approximately 75-80° F until dry. Note: Some subparts or methods specify alternate "B" section cleanup solvents. The "Sample Train Information Sheet" details any exceptions.

The bubblers and impingers utilized for the condenser section, or "back-half" of the sample train were weighed with a readability of 0.1 grams before and after sampling using an electronic top loading balance. The difference between the initial and final weights of the condenser section constitute the amount of moisture gain during the run. To perform the Method 202 back-half analysis, the impinger solutions were recovered into a graduated cylinder. The impingers were rinsed with deionized water and the rinses were combined with the contents of the graduated cylinder and the final volume was recorded. The pH of each sample was measured and confirmed to be greater than 4.5 so the Method 202 analysis for sulfates was not

performed. An aliquot of the impinger water was taken for potential future analysis. The condensable matter was quantified in the Am Test laboratory by performing a solvent extraction using a separatory funnel on the remaining impinger solution. The impingers were given a final rinse with 75 mL of CH_2Cl_2 , which was collected in a glass sample container for later use in the first extraction. Two (2) additional 75 mL portions of CH_2Cl_2 were used for a total of three (3) 75 mL extractions. The organic layer was transferred to tared 150 mL beakers labeled as the "Cx" section and these beaker's contents were allowed to evaporate in an evaporation chamber at ambient temperature to dryness, the beakers were desiccated, then weighed to a constant weight. The water layer was transferred to a tared 150 mL beaker with glass boiling beads labeled the "C" section and this beaker's contents were heated on a hot plate to boiling until approximately 50 milliliters remained in the beaker. The remaining 50 milliliters in the beaker were evaporated to dryness in an oven at 105° C. Since no sulfates were present, the ammonium hydroxide titration to remove sulfuric acid interference was not necessary and was omitted. Sample blanks containing acetone, DI water and CH_2Cl_2 were analyzed in an identical fashion as the representative "section". All beakers were desiccated for at least 24 hours and weighed to constant weights of ± 0.5 milligrams after their contents had evaporated. The total particulate matter weight is the sum of the net weights of the particulate matter found on the filter plus the net weights found in the beakers containing sample, minus the acetone, water and methylene chloride blank contributions.

6.0

QUALITY ASSURANCE PLAN

The purpose of the quality assurance plan is to provide guidelines for achieving quality control in air pollution measurements. The detailed procedures which are utilized are included in the Environmental Protection Agency's (EPA's) reference manual titled Quality Assurance Handbook for Air Pollution Measurement Systems, Volume 3, EPA-600/4-77-027b. These procedures are followed throughout equipment preparation, field sampling, sample recovery, analysis and data reduction. Am Test-Air Quality, Inc.'s quality assurance procedures are discussed below.

6.1 Calibration Procedures and Frequency

Field equipment utilized for on-site measurements is calibrated at a frequency recommended by the equipment manufacturer or industry practice. Prior to field use, each instrument is calibrated and the calibration value is recorded. If any measuring or test device requiring calibration cannot immediately be removed from service, the Project Manager may extend the calibration cycle providing a review of the equipment's history warrants the issuance of an extension. No equipment will be extended more than twice a calibration cycle, nor will the extension exceed one-half the prescribed calibration cycle. Test equipment consistently found to be out of calibration will be repaired or replaced.

The sample nozzles used to collect isokinetic samples are calibrated on-site before sampling using digital inside calipers readable to 0.001 inch. Three (3) measurements were taken at varying points around the inside of the nozzle tip and

averaged. The dry gas meters used to accurately measure sample volumes are calibrated using a standard laboratory dry gas meter. The type S pitot tubes utilized for velocity determination are calibrated using Method 2, Section 4.1, and are inspected regularly for wear. The magnehelic gauges used for pressure measurements are checked against an oil-filled manometer. The digital thermocouple indicator used for temperature measurement has a readability of 1 degree Fahrenheit and is periodically re-certified by the manufacturer. Each thermocouple probe used to monitor temperature is checked periodically at three (3) temperature settings. Copies of calibration information for each measurement device used are included in the appendices of this report. A barometer readable to 0.01 inches of mercury is used in the field to obtain barometric pressure readings.

In the laboratory reagent blanks are carried throughout the gravimetric analysis procedures. The samples were weighed to constant weights of ± 0.5 milligrams following desiccation in a cabinet desiccator. This desiccator is an electronic dehumidifier which automatically maintains the humidity inside the desiccator. The dehumidifier automatically recharges the internal desiccant every 5.5 hours. An Airguide humidity indicator accurate to $\pm 1\%$ is used to check the humidity inside the desiccator when obtaining tare and final weights. A small container of indicating silica gel is placed in the desiccators to maintain the desired humidity. The Mettler AE163 electronic balance used to obtain weights is set to a time integrating mode (100,000 readings per minute) with a readability of 0.01 milligrams. The balance is calibrated prior to every weighing session and an audit is performed with class S weights once a week. The calibration of Am Test's Mettler balances is checked by the manufacturer on a yearly basis.

Support equipment is defined as all equipment, not previously discussed, that is required for completing an environmental monitoring or measurement task. This equipment may include storage and transportation containers, sample recovery glassware, and communications gear. Support equipment is periodically inspected to maintain the performance standards necessary for proper and efficient execution of all tasks and responsibilities.

During a project, a systems audit is performed, consisting of an on-site qualitative inspection and review of the total measurement system. This inspection is conducted on a daily basis by the Project Leader. During the systems audit, the auditor observes the procedures and techniques of the field team in the following general areas:

- Setting up and leak testing the sample train
- Isokinetic sampling check
- Final leak check of the sample train
- Sample recovery

Visual inspections of pitot tubes, glassware, and other equipment are also made. The main purpose of a systems audit is to ensure that the measurement system will generate valid data, if operated properly.

6.2 Sample Recovery and Field Documentation

Data collected during each test, are immediately inspected for completeness and placed under the custody of the Project Leader until custody is transferred when the samples were returned to the Air Quality laboratory. Sample recovery is carried out in a suitable area free from particulate matter contamination. Each sample is assigned an identifying lab number to assist the chemists in tracking the sample.

6.3 Data Reduction, Validation and Reporting

Raw data are handled according to strict guidelines when being transposed into computer files or to other logs. The guidelines include document receipt control procedures, file review, and sign-off by a project assistant. Raw data are entered into the appropriate computer spreadsheet by a "processor", then the entered figures are checked for accuracy by a "checker", different from the "processor". Any mistakes are corrected, and figures are rechecked and signed off by the "checker". In addition, a by-hand calculation check of each spreadsheet is made using a hand-held calculator to validate the computer output. All data generated by each phase of a laboratory or field sampling program are reviewed by the senior reviewer. The data package is signed off by the senior reviewer prior to releasing the data for report preparation.

The test results are calculated in accordance with current EPA 40 CFR 60 criteria. Copies of the pertinent equations used to derive the results for this test are included in the appendices of this report. Standard conditions are 29.92 inches of mercury and 68° F.

7.0**METHODOLOGY REFERENCES**

- EPA. Title 40 Code of Federal Regulations, Part 60 (40 CFR 60), Appendix A, Reference Methods 1, 2, 3A, 4 and 5. July 1, 1993.
- EPA. Title 40 Code of Federal Regulations, Part 51 (40 CFR 51), Appendix M, Reference Method 202. July 1, 1993.
- EPA. EPA 450/2-79-006, APTI Course, "Course 450 - Source Sampling For Particulate Pollutants", December 1979.
- EPA. Quality Assurance Handbook for Air Pollution Measurement Systems, Volume 3, EPA-60/4-77-027b.

APPENDIX A
Computer Printouts of Results

METHODS 1, 2, 3A, 4, AND 5
AM TEST-AIR QUALITY, INC.

FILE NAME: S721\LPCOR1
CLIENT: Louisiana Pacific
LOCATION: Montrose, Colorado
SAMPLE SITE: Dryer E-Tube Stack
SAMPLE DATE: October 25, 1994
RUN #: 1 - Method 5
OPERATORS: Mackey

LAB #: 6787
START TIME: 08:16 o'clock
STOP TIME: 09:34 o'clock
SAMPLE LENGTH: 60.0 minutes

IMPINGER WEIGHTS NET
FINAL INITIAL grams grams
809.9 643.3 166.6
700.5 633.3 67.2
665.4 654.2 11.2
829.8 815.5 14.3
TOTAL H2O GAIN: 259.3
TOTAL VOLUME (scf): 12.23
PERCENT MOISTURE: 22.01
Bws: 0.2201

PITOT TUBE Cp: 0.84
NOZZLE DIAMETER: 0.230 inches
NOZZLE AREA: 0.0003 sq. feet
STACK DIAMETER: 48.0 inches
STACK AREA: 12.6 sq. feet
METER TEMPERATURE: 57.2 degrees F
BAROMETRIC PRES.: 25.22 inches Hg
STATIC PRESSURE: 0.15 inches H2O
STACK PRESSURE: 25.23 inches Hg
ORIFICE PRESSURE: 2.173 inches H2O
METER PRESSURE: 25.38 inches Hg

INIT. METER VOLUME: 923.899
FINAL METER VOLUME: 974.135
VOLUME SAMPLED: 50.236
STD VOLUME (dscf): 43.329
STD VOLUME (dscm): 1.227
Y FACTOR: 0.996

AVERAGE CONC. CO2: 2.2 percent
AVERAGE CONC. O2: 17.3 percent
AVERAGE CONC. CO: NA ppm
MOLECULAR WEIGHT: 29.04 g/g-mole-dry
MOLECULAR WEIGHT: 26.61 g/g-mole-wet

FRONT-HALF PARTICULATE MATTER MASS LOADING
FILTER NUMBER: #110-377
TARE WEIGHT OF FILTER (grams): 0.5905
FINAL WEIGHT OF FILTER (grams): 0.5998
NET WEIGHT OF PARTICULATE MATTER (grams): 0.0093

PROBE RINSE BEAKER NUMBER: #150-1978
TARE WEIGHT OF BEAKER (grams): 70.2140
FINAL WEIGHT OF BEAKER (grams): 70.2241
NET WEIGHT OF PARTICULATE MATTER (grams): 0.0101
VOLUME OF ACETONE (milliliters): 160.0
WT./VOL. OF ACETONE BLANK (milligrams/mL): 0.0000
NET WEIGHT OF PARTIC. DUE TO ACETONE (grams): 0.0000

TOTAL FRONT-HALF PARTICULATE MATTER (grams): 0.0194

BACK-HALF PARTICULATE MATTER MASS LOADING

"C" SECTION - CONDENSER PARTICULATE #150-2011
TARE WEIGHT OF BEAKER (grams): 66.0861
FINAL WEIGHT OF BEAKER (grams): 66.0964
NET WEIGHT OF PARTIC. MATTER (grams): 0.0107
TOTAL VOLUME OF WATER (milliliters): 590.0
VOLUME OF ALIQUOT (milliliters): 20.0
VOLUME OF WATER CONDENSED (milliliters): 245.0
NET VOLUME OF WATER FOR BLANK (milliliters): 345.0
WT./VOL. OF WATER BLANK (milligrams/mL): 0.0000
NET WEIGHT OF PARTIC. DUE TO WATER (grams): 0.0000

"Cx" SECTION - HYDROCARBON EXTRACTION #150-1979 & 1980
TARE WEIGHT OF BEAKER (grams): 140.7433
FINAL WEIGHT OF BEAKER (grams): 140.7905
NET WEIGHT OF PARTIC. MATTER (grams): 0.0489
TOTAL VOLUME OF CH2Cl2 (milliliters): 220.0
WT./VOL. OF CH2Cl2 BLANK (milligrams/mL): 0.0007
NET WEIGHT OF PARTIC. DUE TO CH2Cl2 (grams): 0.0002

TOTAL BACK-HALF PARTICULATE MATTER (grams): 0.0594
TOTAL WEIGHT OF PARTICULATE MATTER (grams): 0.0788

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE °F	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE °F
N 1	1.23	149	W 1	1.33	149
N 2	1.26	150	W 2	1.33	150
N 3	1.26	150	W 3	1.38	151
N 4	1.21	150	W 4	1.36	152
N 5	1.30	149	W 5	1.28	154
N 6	1.23	148	W 6	1.10	150
N 7	1.22	148	W 7	1.38	153
N 8	1.32	149	W 8	1.48	153
N 9	1.39	149	W 9	1.51	152
N 10	1.42	149	W 10	1.54	152
N 11	1.41	151	W 11	1.52	153
N 12	1.42	149	W 12	1.54	153

PERCENT ISOkinetics: 92 %
STACK GAS TEMPERATURE: 150.5 degrees F
AVERAGE VELOCITY HEAD: 1.348 " of H2O
STACK GAS VELOCITY: 5952.1 acf/min
STACK GAS AIR FLOW: 34099.7 dscf/min
PARTICULATE EMISSION CONCENTRATION (FRONT-HALF): 0.007 gr/dscf
PARTICULATE EMISSION CONCENTRATION (BACK-HALF): 0.021 gr/dscf
TOTAL PARTICULATE EMISSION CONC. (FRONT & BACK-HALF): 0.028 gr/dscf
TOTAL PARTICULATE EMISSION CONCENTRATION: 64.2 mg/dscm
TOTAL PARTICULATE MATTER EMISSION RATE: 8.20 lb/hr

METHODS 1, 2, 3A, 4, AND 5
AM TEST-AIR QUALITY, INC.

FILE NAME: SZ21ALPCOR2
CLIENT: Louisiana Pacific
LOCATION: Montrose, Colorado
SAMPLE SITE: Dryer E-Tube Stack
SAMPLE DATE: October 25, 1994
RUN #: 2 - Method 5
OPERATORS: Mackey

LAB #: 6788
START TIME: 10:18 o'clock
STOP TIME: 11:22 o'clock
SAMPLE LENGTH: 60.0 minutes

IMPINGER WEIGHTS

FINAL	INITIAL	NET
grams	grams	grams
900.2	699.2	201.0
676.2	640.0	36.2
669.9	666.0	3.9
838.1	824.7	13.4
TOTAL H2O GAIN:		254.5
TOTAL VOLUME (scf):		12.00
PERCENT MOISTURE:		20.99
BWS:		0.2099

INIT. METER VOLUME: 974.940
FINAL METER VOLUME: 1029.776
VOLUME SAMPLED: 54.836
STD VOLUME (dscf): 45.176
STD VOLUME (dscm): 1.279
Y FACTOR: 0.996

PITOT TUBE Cp: 0.84
NOZZLE DIAMETER: 0.230 inches
NOZZLE AREA: 0.0003 sq. feet
STACK DIAMETER: 48.0 inches
STACK AREA: 12.6 sq. feet
METER TEMPERATURE: 81.9 degrees F
BAROMETRIC PRES.: 25.22 inches Hg
STATIC PRESSURE: 0.15 inches H2O
STACK PRESSURE: 25.23 inches Hg
ORIFICE PRESSURE: 2.444 inches H2O
METER PRESSURE: 25.40 inches Hg

AVERAGE CONC. CO2: 2.2 percent
AVERAGE CONC. O2: 17.3 percent
AVERAGE CONC. CO: NA ppm
MOLECULAR WEIGHT: 29.04 g/g-mole-dry
MOLECULAR WEIGHT: 26.73 g/g-mole-wet

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE °F	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE °F
S 1	1.22	150	W 1	1.38	149
S 2	1.23	150	W 2	1.39	148
S 3	1.24	150	W 3	1.38	149
S 4	1.23	149	W 4	1.42	149
S 5	1.23	150	W 5	1.26	153
S 6	1.22	149	W 6	1.12	153
S 7	1.29	150	W 7	1.41	153
S 8	1.40	149	W 8	1.51	148
S 9	1.39	148	W 9	1.53	150
S 10	1.39	149	W 10	1.60	153
S 11	1.42	149	W 11	1.61	149
S 12	1.44	149	W 12	1.60	149

PERCENT ISOKINETICS: 94 %

STACK GAS TEMPERATURE: 149.8 degrees F
AVERAGE VELOCITY HEAD: 1.368 " of H2O
STACK GAS VELOCITY: 6022.6 acf/min
STACK GAS AIR FLOW: 34744.1 dscf/min
PARTICULATE EMISSION CONCENTRATION (FRONT-HALF): 0.002 gr/dscf
PARTICULATE EMISSION CONCENTRATION (BACK-HALF): 0.021 gr/dscf
TOTAL PARTICULATE EMISSION CONC. (FRONT & BACK-HALF): 0.022 gr/dscf
TOTAL PARTICULATE EMISSION CONCENTRATION: 51.1 mg/dscm
TOTAL PARTICULATE MATTER EMISSION RATE: 6.65 lb/hr

FRONT-HALF PARTICULATE MATTER MASS LOADING
FILTER NUMBER: #110-703
TARE WEIGHT OF FILTER (grams): 0.5935
FINAL WEIGHT OF FILTER (grams): 0.5935
NET WEIGHT OF PARTICULATE MATTER (grams): 0.0000

PROBE RINSE BEAKER NUMBER: #150-1981
TARE WEIGHT OF BEAKER (grams): 78.3910
FINAL WEIGHT OF BEAKER (grams): 78.3962
NET WEIGHT OF PARTICULATE MATTER (grams): 0.0052
VOLUME OF ACETONE (milliliters): 160.0
WT./VOL. OF ACETONE BLANK (milligrams/mL): 0.0000
NET WEIGHT OF PARTIC. DUE TO ACETONE (grams): 0.0000
TOTAL FRONT-HALF PARTICULATE MATTER (grams): 0.0052

BACK-HALF PARTICULATE MATTER MASS LOADING

"C" SECTION - CONDENSER PARTICULATE
TARE WEIGHT OF BEAKER (grams): #150-2012
67.2238
FINAL WEIGHT OF BEAKER (grams): 67.2238
NET WEIGHT OF PARTIC. MATTER (grams): 0.0141
TOTAL VOLUME OF WATER (milliliters): 650.0
VOLUME OF ALIQUOT (milliliters): 20.0
VOLUME OF WATER CONDENSED (milliliters): 241.1
NET VOLUME OF WATER FOR BLANK (milliliters): 408.9
WT./VOL. OF WATER BLANK (milligrams/mL): 0.0000
NET WEIGHT OF PARTIC. DUE TO WATER (grams): 0.0000

"CX" SECTION - HYDROCARBON EXTRACTION #150-1982 & 1983
TARE WEIGHT OF BEAKER (grams): 144.9391
FINAL WEIGHT OF BEAKER (grams): 144.9859
NET WEIGHT OF PARTIC. MATTER (grams): 0.0462
TOTAL VOLUME OF CH2CL2 (milliliters): 218.0
WT./VOL. OF CH2CL2 BLANK (milligrams/mL): 0.0007
NET WEIGHT OF PARTIC. DUE TO CH2CL2 (grams): 0.0002

TOTAL BACK-HALF PARTICULATE MATTER (grams): 0.0602
TOTAL WEIGHT OF PARTICULATE MATTER (grams): 0.0654

METHODS 1, 2, 3A, 4, AND 5
AM TEST-AIR QUALITY, INC.

FILE NAME: S721\1PCOR3
CLIENT: Louisiana Pacific
LOCATION: Montrose, Colorado
SAMPLE SITE: Dryer E-Tube Stack
SAMPLE DATE: October 25, 1994
RUN #: 3 - Method 5
OPERATORS: Mackey

LAB #: 6789
START TIME: 12:13 o'clock
STOP TIME: 13:15 o'clock
SAMPLE LENGTH: 60.0 minutes

FRONT-HALF PARTICULATE MATTER MASS LOADING

FILTER NUMBER: #110-379
TARE WEIGHT OF FILTER (grams): 0.5956
FINAL WEIGHT OF FILTER (grams): 0.6033
NET WEIGHT OF PARTICULATE MATTER (grams): 0.0077

PROBE RINSE BEAKER NUMBER: #150-1984
TARE WEIGHT OF BEAKER (grams): 77.9157
FINAL WEIGHT OF BEAKER (grams): 77.9271
NET WEIGHT OF PARTICULATE MATTER (grams): 0.0114
VOLUME OF ACETONE (milliliters): 110.0
WT./VOL. OF ACETONE BLANK (milligrams/mL): 0.0000
NET WEIGHT OF PARTIC. DUE TO ACETONE (grams): 0.0000

TOTAL FRONT-HALF PARTICULATE MATTER (grams): 0.0191

BACK-HALF PARTICULATE MATTER MASS LOADING

"C" SECTION - CONDENSER PARTICULATE #150-2013
TARE WEIGHT OF BEAKER (grams): 66.0167
FINAL WEIGHT OF BEAKER (grams): 66.0503
NET WEIGHT OF PARTIC. MATTER (grams): 0.0140
TOTAL VOLUME OF WATER (milliliters): 720.0
VOLUME OF ALIQUOT (milliliters): 20.0
VOLUME OF WATER CONDENSED (milliliters): 268.7
NET VOLUME OF WATER FOR BLANK (milliliters): 451.3
WT./VOL. OF WATER BLANK (milligrams/mL): 0.0000
NET WEIGHT OF PARTIC. DUE TO WATER (grams): 0.0000

"CX" SECTION - HYDROCARBON EXTRACTION #150-1985 & 1986
TARE WEIGHT OF BEAKER (grams): 129.8080
FINAL WEIGHT OF BEAKER (grams): 129.8608
NET WEIGHT OF PARTIC. MATTER (grams): 0.0543
TOTAL VOLUME OF CH2CL2 (milliliters): 208.0
WT./VOL. OF CH2CL2 BLANK (milligrams/mL): 0.0007
NET WEIGHT OF PARTIC. DUE TO CH2CL2 (grams): 0.0001

TOTAL BACK-HALF PARTICULATE MATTER (grams): 0.0682
TOTAL WEIGHT OF PARTICULATE MATTER (grams): 0.0875

IMPINGER WEIGHTS
FINAL INITIAL NET
grams grams grams
846.0 642.5 203.5
693.7 633.7 60.0
601.9 596.7 5.2
826.3 810.7 15.6
TOTAL H2O GAIN: 284.3
TOTAL VOLUME (scf): 13.40
PERCENT MOISTURE: 23.03
BWS: 0.2303

PITOT TUBE Cp: 0.84
NOZZLE DIAMETER: 0.230 inches
NOZZLE AREA: 0.0003 sq. feet
STACK DIAMETER: 48.0 inches
STACK AREA: 12.6 sq. feet
METER TEMPERATURE: 93.1 degrees F
BAROMETRIC PRES.: 25.19 inches Hg
STATIC PRESSURE: 0.12 inches H2O
STACK PRESSURE: 25.20 inches Hg
ORIFICE PRESSURE: 2.471 inches H2O
METER PRESSURE: 25.37 inches Hg

AVERAGE CONC. CO2: 2.2 percent
AVERAGE CONC. O2: 17.1 percent
AVERAGE CONC. CO: NA ppm
MOLECULAR WEIGHT: 29.04 g/g-mole-dry
MOLECULAR WEIGHT: 26.49 g/g-mole-wet

SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE °F	SAMPLE POINT	VELOCITY " OF H2O	TEMPERATURE °F
S 1	1.28	150	W 1	1.33	151
2	1.23	151	2	1.39	153
3	1.31	151	3	1.41	153
4	1.23	150	4	1.38	151
5	1.32	150	5	1.28	151
6	1.29	153	6	1.22	149
7	1.25	155	7	1.28	149
8	1.22	156	8	1.43	150
9	1.32	155	9	1.39	150
10	1.43	156	10	1.39	149
11	1.43	154	11	1.39	150
12	1.46	155	12	1.61	150

PERCENT ISOKINETICS: 97 %
STACK GAS TEMPERATURE: 151.8 degrees F
AVERAGE VELOCITY HEAD: 1.343 " of H2O
STACK GAS AIR FLOW: 60064.8 acf/min
PARTICULATE EMISSION CONCENTRATION (FRONT-HALF): 79.7 ft/second
PARTICULATE EMISSION CONCENTRATION (BACK-HALF): 33606.0 dscf/min
TOTAL PARTICULATE EMISSION CONC. (FRONT & BACK-HALF): 0.007 gr/dscf
TOTAL PARTICULATE EMISSION CONC. (FRONT & BACK-HALF): 0.023 gr/dscf
TOTAL PARTICULATE EMISSION CONCENTRATION: 68.8 mg/dscm
TOTAL PARTICULATE MATTER EMISSION RATE: 8.66 lb/hr

APPENDIX B

Example Calculations and Field Data Sheets

EXAMPLE CALCULATION SHEET (continued)
EPA METHODS 1, 2, 3A, 4 and 5

$$\begin{aligned} \text{Total } C_s &= (0.001 \text{ g/mg}) * (15.43 \text{ grains/gram}) * \underline{65.4} \text{ mg} / \underline{45.176} \text{ dscf} \\ &= \underline{0.022} \text{ gr/dscf (Equation 5-6)} \end{aligned}$$

$$\begin{aligned} \text{gr/dscf @ } \underline{\quad} \% \text{ O}_2 &= \underline{\quad} \text{ gr/dscf} * (20.9\% - \underline{\quad} \% \text{ O}_2) / (20.9\% - \underline{\quad} \% \text{ O}_2) \\ &= \underline{NA} \text{ gr/dscf @ } \underline{\quad} \% \text{ O}_2 \end{aligned}$$

$$\begin{aligned} \text{gr/dscf @ } \underline{\quad} \% \text{ CO}_2 &= \underline{\quad} \text{ gr/dscf} * \underline{\quad} \% / \underline{\quad} \% \text{ CO}_2 \\ &= \underline{NA} \text{ gr/dscf @ } \underline{\quad} \% \text{ CO}_2 \end{aligned}$$

$$\begin{aligned} \text{mg/dscm} &= \underline{65.4} \text{ mg} / \underline{1.279} \text{ dscm} \\ &= \underline{51.1} \text{ mg/dscm} \end{aligned}$$

Particulate Matter Emission Rate

$$\begin{aligned} \text{pounds/hour} &= C_s * \text{dscf/min} * 60 \text{ min/hr} * 1 \text{ lb/7000 grains} \\ &= \underline{0.022} \text{ gr/dscf} * \underline{34744.1} \text{ dscf/min} * 60 \text{ min/hr} * 1 \text{ lb/7000 grains} \\ &= \underline{6.65} \text{ lb/hr} \end{aligned}$$

$$\begin{aligned} \text{tons/year} &= \underline{\quad} \text{ lb/hr} * 24 \text{ hr/day} * 365 \text{ days/yr} * 1 \text{ ton/2000 lb} \\ &= \underline{NA} \text{ tons/yr} \end{aligned}$$

Moisture - Equation 5-2 and 5-3

$$\begin{aligned} V_{w\text{std}} &= 0.04715 \text{ ft}^3/\text{g} * \underline{254.5} \text{ grams of H}_2\text{O collected in impingers} \\ &= \underline{12.00} \text{ scf} \end{aligned}$$

$$\begin{aligned} B_{ws} &= (\underline{12.00} \text{ scf}) / (\underline{12.00} \text{ scf} + \underline{45.176} \text{ dscf}) \\ &= \underline{0.2099} \end{aligned}$$

$$\% \text{ Moisture} = \underline{20.99} \% = B_{ws} * 100$$

Molecular weight - Equation 3-2

$$\begin{aligned} M_d &= 0.440 * (\underline{2.2} \% \text{ CO}_2) + 0.320 * (\underline{17.3} \% \text{ O}_2) + 0.280 * (100\% - \underline{2.2} \% \text{ CO}_2 - \underline{17.3} \% \text{ O}_2 (\% \text{ CO} + \% \text{ N}_2)) \\ &= \underline{29.04} \text{ g/g-mole (dry)} \end{aligned}$$

$$\begin{aligned} M_s &= M_d * (1 - B_{ws}) + 18.0 * B_{ws} = \underline{29.04} \text{ g/g-mole} * (1 - \underline{0.2099}) + 18.0 \text{ g/g-mole} * \underline{0.2099} \\ &= \underline{26.73} \text{ g/g-mole (wet)} \end{aligned}$$

$$F_o = (20.9 - \underline{\quad}) \% \text{ O}_2 / \underline{\quad} \% \text{ CO}_2 = \underline{NA}$$

EXAMPLE CALCULATION SHEET (continued)
EPA METHODS 1, 2, 3A, 4 and 5

Stack gas velocity and volumetric flow rate - Equation 2-9 and 2-10

$$V_s = 85.49 * C_p * \sqrt{\Delta P * T_s^{\circ} R / (M_s \text{ g/g-mole} * P_s \text{ "Hg})}$$

$$= 85.49 * 0.84 * \sqrt{\frac{1.368 * 609.8^{\circ} R}{(49.8^{\circ} F + 460)^{\circ} R} / \frac{25.23 \text{ "Hg}}{(23.22 P_B + 2.15 P_s) / 13.6}}$$

$$= 79.9 \text{ ft/sec (std)}$$

$$Q_{sd} = 3600 * (1 - B_{ws}) * V_s \text{ ft/sec} * A_s \text{ ft}^2 * (T_{std}^{\circ} R / T_s^{\circ} R) * (P_s \text{ "Hg} / P_{std} \text{ "Hg}) / 60 \text{ min/hr}$$

$$= 3600 * (1 - 0.2099) * 79.9 \text{ ft/sec} * 12.6 \text{ ft}^2 * (528^{\circ} R / 609.8^{\circ} R) * (25.23 \text{ "Hg} / 29.92 \text{ "Hg}) / 60$$

$$= 34744.1 \text{ dscf/min (dry standard cubic feet per minute)}$$

$$\text{acfm} = 79.9 \text{ ft/sec} * 12.6 \text{ ft}^2 * 60 \text{ sec/min}$$

$$= 60528.6 \text{ acfm (actual cubic feet per minute)}$$

Isokinetic variation - Equation 5-8

$$I = 0.09450 * V_{m_{std}} \text{ dscf} * T_s^{\circ} R / [P_s \text{ "Hg} * V_s \text{ ft/sec} * \text{minutes} * A_n \text{ ft}^2 * (1 - B_{ws})]$$

$$= 0.09450 * 45.176 \text{ dscf} * 609.8^{\circ} R / [25.23 \text{ "Hg} * 79.9 \text{ ft/sec} * 60 \text{ min} * \frac{0.0003 \text{ ft}^2 * (1 - 0.2099)}{(.230 \text{ N}_{dia} / 12 / 2)^2 * \text{Pi}}]$$

$$= 94 \%$$

All of the above numbered equations are from the 40 CFR 60 and assume English units.

[cbh\c:\word\b-plate\M5.doc]

EXAMPLE CALCULATION SHEET (continued)
EPA METHODS 1, 2, 3A, 4 and 5
BACK-HALF PARTICULATE

"C" Section

(write in final and initial wts for all appropriate impingers for c below)

- | | | | |
|--|--|-------------------------|-------|
| | | 100.2 | 617.2 |
| | | 676.2 | 540.0 |
| | | 619.9 | 666.0 |
| | | 2246.3 - 2005.2 = 241.1 | |
- a 13.7 mg particulate in "C" Section Beaker
- b 650 ml of water in condensers, including rinses
- c 841.1 ml condensation in 1st, 2nd and 3rd bubblers (final wt - initial wt, assumes 1g/1ml water density)
- d 408.9 ml DI water used in bubblers including rinses = $b - c$
- e 0.0 mg/ml blank particulate = $\frac{0.0 \text{ mg blank}}{500 \text{ ml blank}}$
- f 0.0 mg blank particulate = $e * d$
- g 20 ml aliquot
- h 1.032 correction factor = $b / (b - g)$
- 14.1 mg of "C" particulate = $a * h - f = (13.7 \text{ mg} * 1.032) - 0.0 \text{ mg}$

"Cx" Section

- a 46.2 mg particulate in "Cx" Section Beaker = 44.8 mg * 1.032 correction factor (h)
- b 218 ml CH₂Cl₂ used for sample
- c 0.0007 mg/ml blank particulate = $\frac{0.1 \text{ mg blank}}{150 \text{ ml blank}}$
- d 0.2 mg blank particulate = $b * c$
- 46.1 mg of "Cx" particulate = $a - d = 46.2 \text{ mg} - 0.2 \text{ mg}$

"D" Section

- a _____ mg particulate in "D" Section Beaker
- b _____ ml Acetone used for sample
- c _____ mg/ml blank particulate (Same as "B" Section)
- d _____ mg blank particulate = $b * c$
- NA mg of "D" particulate = $a - d = \text{_____ mg} - \text{_____ mg}$

TOTAL BACK-HALF PARTICULATE

- + 14.1 mg "C" Section Particulate
- + 46.1 mg "Cx" Section Particulate
- + — mg "D" Section Particulate
- + — mg Back-half Filter Particulate (If applicable)

= 60.2 mg Back-half Particulate

STACK SCHEMATIC AND LOCATION OF SAMPLE POINTS

Client L.P. Montrose

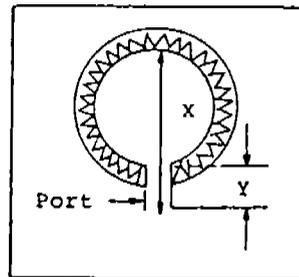
Location Montrose, CO

Sampling Location Dryer E-tube stack

Inside of far wall to outside of port (distance, X) 54"

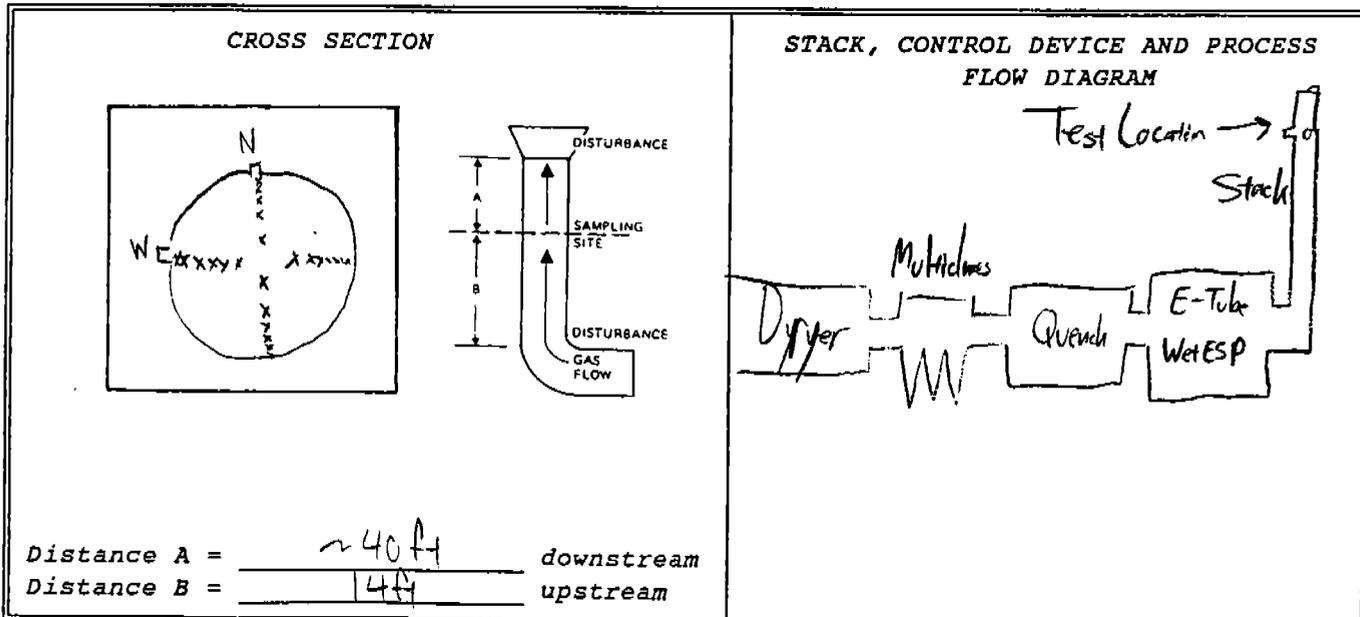
Inside of near wall to outside of port (distance, Y) 6"

Stack I.D. (distance X // distance Y) 48" ϕ



Schematic of Sampling Location

1	2	3	4	5	6
Traverse Point #	Fractional Percent of Stack I.D.	Stack I.D. inches	Column 2 x 3	Distance Y	Traverse Point Location from Outside of Port columns 4 + 5
1	.021	48.0"	1.0"	6.0"	7.0"
2	.067		3.2		9.2
3	.118		5.7		11.7
4	.177		8.5		14.5
5	.250		12.0		18.0
6	.356		17.1		23.1
7	.644		30.9		36.9
8	.750		36.0		42.0
9	.823		39.5		45.5
10	.882		42.3		48.3
11	.933		44.8		50.8
12	.979		48.0		53.0



6787

TRAVERSE SAMPLING DATA SHEET

Client Los Alamitos Pacific - Mantro
 Location Mantro, CO
 Sample Site Dryer E-1/6 Stalk
 Stack Diameter 48"
 Date 10/25/94
 Operators MSH
 Run I.D. 1 - Meth: 5/200

QA FORMS COMPLETED

Stack Schematic
 Sample Train
 Pitot Tube Insp.
 Magnehelic Cal.
 Temp. Probe Cal.
 Gas Meter Calib.

Start Time 8:16
 Stop Time 9:34
 Barometric 25.27
 Pressure "Hg
 Static Pres "H₂O 70.15
 Production Rate

EQUIPMENT CHECKS

Initial/Final
 Leak Rate cfm .009 / .004
 Leak Test Vacuum 15" / 15"
 Pitots, Pre Leak Ck
 Pitots, Post Leak Ck
 Gas Sampling System
 Integrated Bag
 M5 Rinse Acetone/H₂O/Other

Filter # 10 Box # 12.1

SAMPLING PARAMETERS

	Final	Initial	Net
	Wt.	Wt.	Wt.
	gram	gram	gram
#1 Imp.	809.9	643.3	166.6
#2 Imp.	700.5	633.3	67.2
#3 Imp.	665.4	654.2	11.2
#4 Imp.			
#5 Imp.			
#6 S.G.	829.8	815.5	14.3
Total H ₂ O Volume			259.3 g

% Moisture 22/27/45.27
 Meter Temp. 45/55
 Stack Temp. 139/157/148/153
 Δ H@ 1.868 Y 0.996
 Pitot # 54 (side # A)
 Cp .34
 Nozzle Diameter .237 inch
 D₁ .230 D₂ .230 D₃ .231
 K Factor 1.52/1.05/1.55

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading Δ P " H ₂ O	Orifice Setting (Δ H) " H ₂ O		Gas Meter Temp °F		Pump Vac. Gauge " Hg	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F	O ₂ %
				Ideal	Actual	In	Out					
1	0	923.899	1.23	2.14	2.14	48	48		263	50	149	
2	2.5	926.510	1.26	2.19	2.19	49	48	1	263	50	150	
3	5	928.710	1.26	1.91	1.92	50	44	1	264	52	150	
4	7.5	930.819	1.21	1.84	1.84	50	44	1	264	54	150	
5	10	932.741	1.30	1.97	1.97	52	50	1	262	55	149	
6	12.5	934.820	1.23	1.87	1.87	53	51	1	265	58	148	
7	15	936.800	1.22	1.85	1.85	55	52	1	265	59	148	
8	17.5	938.755	1.32	2.00	2.00	56	52	1	262	60	149	
9	20	940.805	1.39	2.29	2.29	57	53	1	265	61	149	
10	22.5	942.869	1.42	2.34	2.34	58	54	1	265	61	149	
11	25	944.889	1.41	2.33	2.33	59	55	1	264	62	151	
12	27.5	946.928	1.42	2.34	2.34	59	55	1	263	62	149	
13	30	949.147	1.33	2.20	2.20	60	57	1	265	59	149	
14	32.5	951.102	1.33	2.20	2.20	60	57	1	265	60	150	
15	35	953.209	1.38	2.25	2.25	62	58	1	266	59	151	
16	37.5	955.320	1.36	2.25	2.25	62	58	1	265	58	152	
17	40	957.314	1.28	2.11	2.11	63	60	1	266	59	154	
18	42.5	959.310	1.10	1.91	1.91	64	60	1	269	59	150	
19	45	961.350	1.38	2.25	2.25	65	61	1	265	60	153	
20	47.5	963.505	1.48	2.44	2.44	65	62	1	264	60	153	
21	50	965.525	1.51	2.35	2.35	65	62	1	264	59	152	
22	52.5	967.620	1.54	2.39	2.39	66	62	1	265	60	152	
23	55	969.825	1.52	2.36	2.36	66	62	1	265	60	153	
24	57.5	972.080	1.54	2.39	2.39	66	62	1	264	59	153	
25	60	974.135										

← Power Filter (15 minute delay)

METHOD 5
LABORATORY ANALYSIS

Client LOUISIANA PACIFIC Run Number 1
Sample Location DRYER E-TUBE STACK
Date 10/25/94

"A" Section (Filter)

Filter # 110-377 Tare Weight 0.5905 grams
Final Weight 0.5998 grams
Net Weight 0.0093 grams

"B" Section (Probe Wash)

Beaker # 150-1978 Tare Weight 70.2190 grams
Volume Acetone 160 mls Final Weight 70.2241 grams
Net Weight 0.0101 grams

"C" Section (Condenser Particulate - Inorganic Catch)

Beaker # 150-2011 Tare Weight 66.0861 grams
Volume Water 570/590 mls Final Weight 66.0964 grams
Net Weight 0.0103 grams

"Cx" Section (Condenser Particulate - Organic Catch)

Beaker # 150-1979/150-1980 Tare Weight 140.7433 grams
Volume CH₂Cl₂ 145/75 mls Final Weight 140.7905 grams
Net Weight _____ grams

"D" Section (Final Acetone Rinse of Impingers)

Beaker # _____ Tare Weight _____ grams
Volume Acetone _____ mls Final Weight _____ grams
Net Weight _____ grams

73 # F11024
7/11 16/14 Acetone Catch

106-1648
200
2036
278.5
20150K
6787
George
Chuck Cox
Steve Cox

Run 1 Gas Bag (InField Analysis) = 17.3% O₂

6788

TRAVERSE SAMPLING DATA SHEET

<p>Client <u>Montrose</u> Location <u>Montrose, CO</u> Sample Site <u>Duper E-Tube Stn</u> Stack Diameter <u>48.0" φ</u> Date <u>10/25/04</u> Operators <u>Ken</u> Run I.D. <u>2-11/5/2002</u></p>	<p>QA FORMS COMPLETED Stack Schematic <input checked="" type="checkbox"/> Sample Train <input checked="" type="checkbox"/> Pitot Tube Insp. <input checked="" type="checkbox"/> Magnehelic Cal. <input checked="" type="checkbox"/> Temp. Probe Cal. <input checked="" type="checkbox"/> Gas Meter Calib. <input checked="" type="checkbox"/></p>	<p>Start Time <u>10:18</u> Stop Time <u>11:22</u> Barometric Pressure "Hg <u>25.27</u> Static Pres "H₂O <u>+0.15</u> Production Rate _____</p>																																
<p>EQUIPMENT CHECKS</p> <p>Initial/Final</p> <p>Leak Rate cfm <u>.006 / .007</u> Leak Test Vacuum <u>15" / 15"</u> <input checked="" type="checkbox"/> Pitots, Pre Leak Ck <input checked="" type="checkbox"/> Pitots, Post Leak Ck <input checked="" type="checkbox"/> Gas Sampling System <input checked="" type="checkbox"/> Integrated Bag M5 Rinse Acetone/H₂O/Other _____</p>	<p>Filter # <u>110</u> Box # <u>B2</u></p> <table border="1" style="width:100%; border-collapse: collapse;"> <tr> <th></th> <th>Final Wt. gram</th> <th>Initial Wt. gram</th> <th>Net Wt. gram</th> </tr> <tr> <td>#1 Imp.</td> <td>900.2</td> <td>699.2</td> <td>201.0</td> </tr> <tr> <td>#2 Imp.</td> <td>676.2</td> <td>640.0</td> <td>36.2</td> </tr> <tr> <td>#3 Imp.</td> <td>669.9</td> <td>666.0</td> <td>3.9</td> </tr> <tr> <td>#4 Imp.</td> <td>669.9</td> <td></td> <td></td> </tr> <tr> <td>#5 Imp.</td> <td></td> <td></td> <td></td> </tr> <tr> <td>#6 S.G.</td> <td>838.1</td> <td>824.7</td> <td>13.4</td> </tr> <tr> <td>Total H₂O Volume</td> <td colspan="2"></td> <td>254.5</td> </tr> </table>		Final Wt. gram	Initial Wt. gram	Net Wt. gram	#1 Imp.	900.2	699.2	201.0	#2 Imp.	676.2	640.0	36.2	#3 Imp.	669.9	666.0	3.9	#4 Imp.	669.9			#5 Imp.				#6 S.G.	838.1	824.7	13.4	Total H ₂ O Volume			254.5	<p style="text-align: center;">SAMPLING PARAMETERS</p> <p>% Moisture <u>28%</u> Meter Temp. <u>76</u> Stack Temp. <u>150</u> Δ H@ <u>1.868</u> Y <u>0.006</u> <input checked="" type="checkbox"/> Pitot # <u>54</u> (54") Side # <u>1</u> <input checked="" type="checkbox"/> Cp <u>84</u> Nozzle Diameter <u>.230</u> inch D₁ <u>.230</u> D₂ <u>.230</u> D₃ <u>.231</u> <input checked="" type="checkbox"/> K Factor <u>1.79</u></p>
	Final Wt. gram	Initial Wt. gram	Net Wt. gram																															
#1 Imp.	900.2	699.2	201.0																															
#2 Imp.	676.2	640.0	36.2																															
#3 Imp.	669.9	666.0	3.9																															
#4 Imp.	669.9																																	
#5 Imp.																																		
#6 S.G.	838.1	824.7	13.4																															
Total H ₂ O Volume			254.5																															

Sample Point	Elap Time Min.	Dry Gas Meter Reading Cu. Ft.	Pitot Reading Δ P " H ₂ O	Orifice Setting (Δ H) " H ₂ O		Gas Meter Temp °F		Pump Vac. Gauge " Hg	Filter Box Temp °F	Imp. Exit Temp °F	Stack Temp °F	O ₂ %
				Ideal	Actual	In	Out					
1	0	974.740	1.22	2.18	2.18	75	75	1	268	59	150	
2	2.5	977.101	1.23	2.20	2.20	76	76	1	263	59	150	
3	5	979.290	1.24	2.21	2.21	76	77	1	263	59	150	
4	7.5	981.281	1.23	2.20	2.20	76	77	1	263	59	149	
5	10	983.484	1.23	2.20	2.20	78	79	1	269	55	150	
6	12.5	985.616	1.22	2.18	2.18	78	79	1	269	57	149	
7	15	987.801	1.29	2.30	2.30	79	80	1	270	56	150	
8	17.5	989.848	1.40	2.50	2.50	80	81	1	270	57	149	
9	20	992.150	1.39	2.43	2.43	80	81	1	263	57	149	
10	22.5	994.499	1.39	2.43	2.43	81	82	1	272	58	149	
11	25	996.683	1.42	2.53	2.53	82	82	1.5	270	58	149	
12	27.5	999.015	1.44	2.57	2.57	82	82	1.5	269	58	149	
West 1	30	1001.495	1.38	2.46	2.46	83	83	1.5	261	59	149	
2	32.5	1003.651	1.39	2.48	2.48	83	84	1.5	262	59	148	
3	35	1005.857	1.38	2.46	2.46	83	84	1.5	261	58	149	
4	37.5	1008.290	1.42	2.53	2.53	83	85	1.5	259	58	149	
5	40	1010.500	1.26	2.25	2.25	84	85	1.0	264	59	153	
6	42.5	1012.688	1.12	2.00	2.00	84	86	1	263	59	153	
7	45	1014.743	1.41	2.52	2.52	84	86	1.5	261	60	153	
8	47.5	1017.061	1.51	2.70	2.70	85	86	2	261	60	148	
9	50	1019.453	1.53	2.73	2.73	85	87	2	261	60	150	
10	52.5	1021.733	1.60	2.86	2.86	85	87	2	261	61	153	
11	55	1024.314	1.61	2.87	2.87	85	87	2	252	62	149	
12	57.5	1026.977	1.60	2.86	2.86	85	87	2	258	63	149	
	60	1029.776										
				2.444 / 2.19 /						149.2 /		

METHOD 5
LABORATORY ANALYSIS

client LOUISIANA PACIFIC Run Number 2
Sample Location DRYER E-TUBE STACK
Date 10/25/94

"A" Section (Filter)

Filter # 110-203 Tare Weight 0.5935 grams F
Final Weight 0.5875 grams 2.00
Net Weight -0.0060 grams

"B" Section (Probe Wash)

Beaker # 150-1981 Tare Weight 78.3710 grams
Volume Acetone 160 mls Final Weight 78.3962 grams F
Net Weight 0.0052 grams

"C" Section (Condenser Particulate - Inorganic Catch)

Beaker # 150-2012 Tare Weight 67.2101 grams
Volume Water 630/650 mls Final Weight 67.2236 grams F
Net Weight 0.0137 grams

"Cx" Section (Condenser Particulate - Organic Catch)

Beaker # 150-1982/150-1983 Tare Weight 144.9291 grams
Volume CH₂Cl₂ 143/75 mls Final Weight 144.9839 grams
63 150 Net Weight 0.0448 grams

"D" Section (Final Acetone Rinse of Impingers)

Beaker # _____ Tare Weight _____ grams
Volume Acetone _____ mls Final Weight _____ grams
Net Weight _____ grams

$(\sqrt{AP})_{avg} = 1.170$
 $AP_{DP} = 1.368$
 $V_s = 80.24/ac$

6788

94590150-11

Run 2 Gas Bag (Initial Analysis) : 17.39% O₂

METHOD 5
LABORATORY ANALYSIS

Client LOUISIANA PACIFIC Run Number 3
Sample Location DRYER E-TUBE STACK
Date 10/25/94

"A" Section (Filter)

Filter # 110-379 Tare Weight 0.5956 grams
Final Weight 2.2083 grams F
Net Weight 1.6127 grams

"B" Section (Probe Wash)

Beaker # 150-1984 Tare Weight 77.9157 grams
Volume Acetone 110 mls Final Weight 77.9241 grams F
Net Weight 2.0114 grams

"C" Section (Condenser Particulate - Inorganic Catch)

Beaker # 150-2013 Tare Weight 66.0167 grams
Volume Water 700/720 mls Final Weight 66.0303 grams F
Net Weight 0.0136 grams

"Cx" Section (Condenser Particulate - Organic Catch)

Beaker # 150-1985/150-1986 Tare Weight 123.7000 grams
Volume CH₂Cl₂ 133/75 mls Final Weight 129.8608 grams F
Net Weight 6.1608 grams

"D" Section (Final Acetone Rinse of Impingers)

Beaker # _____ Tare Weight _____ grams
Volume Acetone _____ mls Final Weight _____ grams
Net Weight _____ grams

Gas Bag Analysis In Field 17.190 O₂

METHOD 5
LABORATORY ANALYSISAcetone Blank

Beaker # 150-1987 Tare Weight 64.9525 grams
Volume Acetone 140 mls Final Weight 64.9523 grams
Net Weight grams

 $0.0 \frac{mg}{ml}$ Distilled, Deionized Water Blank

Beaker # 150-2014 Tare Weight 68.3090 grams
Volume Water 300 mls Final Weight 68.3082 grams
Net Weight grams

 $0.0 \frac{mg}{ml}$ Methylene Chloride (CH₂Cl₂) Blank

Beaker # 150-1988 Tare Weight 65.0619 grams
Volume CH₂Cl₂ 150 mls Final Weight 65.0620 grams
Net Weight 0.0001 grams

 $\frac{0.1mg}{150ml} = 0.0007 \frac{mg}{ml}$

SAMPLE TRAIN INFORMATION

Fill out one sheet per site and per test type.

CLIENT: L.P. Montrose
 LOCATION: Montrose CO
 SITE: Dryer E-Tube Stack
 TEST TEAM: KSM DATE(S): 10/25/94
 RUN #S: 1-3 TYPE: Method 5/202
 Probe/Filter Temperature: 248+25 F 320 F Other
 Impinger Temperature: <68 F Other

THIMBLE: yes no NOZZLE TYPE: quartz steel none

PROBE LINER: quartz glass steel teflon

PROBE TYPE: regular water-cooled

FRONT-HALF FILTER: yes no SIZE (mm): 90 110 125 other

FRONT-HALF FILTER MEDIA: quartz fiber glass fiber teflon

SUPPORT: steel glass frit teflon GASKET: silicon teflon

BACK-HALF FILTER: yes no

BACK-HALF FILTER MEDIA: quartz fiber glass fiber teflon tared untared

NOTE: Show the back-half filter location with an arrow on the table below.

CONTENTS	Initial Volume (mL)	Clean-up Solution Used	Bottle Type	Comments
Nozzle/Probe Rinse		Acetone	500 ml	
Filter		—	Petri Dish	
#1 100 mls DI H ₂ O	100	DCM	1000 ml	
#2 100 mls DI H ₂ O	100	DCM	1000 ml	
#3 100 mls DI H ₂ O	100	DCM	1000 ml	
#4 Silica Gel				
#5				
#6				
#7				

IF THIS INFORMATION IS NOT ACCURATE FOR ALL RUNS, NOTE ALL EXCEPTIONS.

The optional post test N₂ purge was not performed on these tests

<u>DATE</u>	<u>TIME</u>	<u>TEMP</u>	<u>RH Room</u>	<u>RH Des</u>	<u>Actual Wgt</u>	<u>Recorded Wgt</u>	<u>CAL.</u>	<u>I</u>
7/31A	0840	58.8	56	21	100	100.0000	Y	Am
↓	↓	↓	↓	↓	50	50.0002	↓	↓
↓	↓	↓	↓	↓	0.5	0.5000	↓	↓
7/31P	1501	66.9	49	26	100	99.9998	Y	SP
11/1A	0825	57.9	50	24	100	100.0000	Y	Am
11/1P	1307	67.8	44	28	100	99.9997	Y	S
12A	0858	59.6	48	22	100	99.9997	Y	S
11/2P	1610	67.5	44	25	100	99.9987	Y	SP
3A	0825	57.0	43	21	100	99.9999	Y	S
11/3P	1521	70.1	38	22	100	100.0000	Y	S
14A	0820	57.0	47	21	100	100.0000	Y	Am
11/4P	1615	68.0	43	24	100	100.0000	Y	Am
11/7A	0836	56.0	51	21	100	100.0003	Y	S
↓	↓	↓	↓	↓	50	50.0002	↓	↓
↓	↓	↓	↓	↓	0.5	0.9999	↓	↓
11/7P	1620	68.7	44	25	100	100.0000	Y	Am
11/8A	0759	59.8	49	22	100	100.0000	Y	S
11/8P	1517	67.9	44	30	100	100.0001	Y	S
11/9A	0820	62.9	46	21	100	100.0001	Y	Am

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Date

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Date

PROJECT

Continued From Page

ATE	TIME	TEMP	RH Room	RH Des	Audit Weight	Recorded Weight	CAL	In
1/2 P	1622	72.7	46	26	100	100.0000	Y	Ann
↓	↓	↓	↓	↓	50	50.0003	↓	↓
↓	↓	↓	↓	↓	0.5	0.4999	↓	↓
1/11 A	0838	62.4	50	24	100	99.9998	Y	SI
1/11 P	1500	67.2	45	25	100	99.9998	Y	SI
1/12 A	0832	59.9	45	19	100	99.9999	Y	SI
1/2 P	1542	66.9	44	22	100	100.0000	Y	SI
1/13 A	0823	59.5	46	20	100	100.0000	Y	SI
1/13 P	1532	68.4	44	24	100	100.0001	Y	SI
1/14 A	0922	62.0	49	21	100	99.9999	Y	SI
1/14 P	1630	66.7	45	20-25	100	100.0001	Y	CB
1/17 A	0844	57.6	53	21	100	100.0001	Y	Ann
↓	↓	↓	↓	↓	50	50.0001	↓	↓
↓	↓	↓	↓	↓	0.5	0.4999	↓	↓
1/17 P	1645	68.2	49	29	100	100.0000	Y	Ann
1/18 A	0810	60.2	54	24	100	100.0000	Y	Ann
1/18 P	1446	68.7	49	21	100	100.0000	Y	SI
1/19 A	0844	61.7	51	23	100	100.0000	Y	Ann
1/19 P	15:28	68.5	48	24	100	100.0000	Y	SI
1/19 A	08:26	62.5	52	23	100	100.0000	Y	Ann
1/20 P	1552	69.5	48	20	100	99.9998	Y	SI
1/21 A	0820	61.3	53	26	100	100.0000	Y	Ann
1/21 P	1654	66.3	47	21	100	100.0000	Y	Ann
1/21 A	0828	57.7	50	20	100	100.0000	Y	Ann
↓	↓	↓	↓	↓	50	50.0001	Y	↓
↓	↓	↓	↓	↓	0.5	0.4997	Y	↓
1/21 P	16:03	69.3	48	29	100	99.9998	Y	SI
1/25 A	0814	60.8	47	21	100	100.0001	Y	Ann
1/25 P	15:41	69.7	47	22	100	100.0000	Y	SI
1/26 A	08:57	65.5	57	24	100	100.0001	Y	SI
1/26 P	1600	71.5	52	24	100	100.0000	Y	Ann
1/27 A	0747	62.1	55	35	100	100.0000	Y	SI
1/27 P	1506	70.1	50	33	100	100.0000	Y	SI
1/28 A	0832	60.9	52	26	100	100.0000	Y	Ann
1/28 P	1517	69.0	45	31	100	99.9999	Y	SI

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Date

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Date

BEAKER	L	Run	Sec	mLs	TARE	WEIGHT
198	1	1	A	160	78.3924	109A
198	2	2	C4	67	65.3685	109A
198	3	2	C4	190	79.5702	109A
198	4	3	B	110	77.9157	109A
198	5	3	C4	133	65.9169	109A
198	6	3	C4	75	63.8286	109A
198	7	8	B	140	64.9520	109A
198	8	13	C4	150	65.0416	109A
198	9	13	B	270	65.0721	109A
199	0	14	B	240	65.1251	109A
199	1	1	B	66.0644	109A	66.0644
199	2	1	D	80	65.9337	109A
199	3	1	C4	150	65.9169	109A
199	4	2	B	67.2804	109A	67.2804
199	5	2	D	90	66.1119	109A
199	6	2	C4	81.2954	109A	81.2954
199	7	3	B	79.7559	109A	79.7559
199	8	3	D	100	80.4757	109A
199	9	3	C4	150	79.8165	109A
200	0	3	B	50	66.7439	109A
200	1	3	C4	150	65.6537	109A
200	2	3	B	200	66.0768	109A
200	3	1	B	180	67.1091	109A
200	4	1	D	190	68.6634	109A
200	5	1	C	250	68.6318	109A
200	6	1	C	430	66.1780	109A
200	7	2	C	400	64.0589	109A
200	8	2	C	300	68.9885	109A
200	9	2	C	400	66.5455	109A
201	0	3	C	400	68.1169	109A
201	1	3	C	500	64.0800	109A
201	2	3	C	600	67.2493	109A
201	3	3	C	700	66.0166	109A
201	4	3	C	300	68.3085	109A
201	5	3	C	400	67.3803	109A
201	6	3	C	400	67.2320	109A

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Date

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Date

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Continued on Page

BEAKER	F	IN A L	WEIGHT
198	1	78.3963	109A
198	2	65.4165	109A
198	3	79.5705	109A
198	4	77.9278	109A
198	5	64.0219	109A
198	6	63.8345	109A
198	7	64.9527	109A
198	8	65.0624	109A
198	9	65.1235	109A
199	0	65.1958	109A
199	1	65.9357	109A
199	2	65.9693	109A
199	3	66.1730	109A
199	4	66.1724	109A
199	5	66.1730	109A
199	6	66.1724	109A
199	7	66.1724	109A
199	8	66.1724	109A
199	9	66.1724	109A
200	0	65.6541	109A
200	1	66.0784	109A
200	2	66.1119	109A
200	3	65.6391	109A
200	4	66.1119	109A
200	5	66.1119	109A
200	6	66.1119	109A
200	7	66.1119	109A
200	8	66.1119	109A
200	9	66.1119	109A
201	0	68.1193	109A
201	1	66.0768	109A
201	2	67.2493	109A
201	3	66.0166	109A
201	4	68.3085	109A
201	5	67.3803	109A
201	6	67.2320	109A

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Date

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FILTER	TARE	FINAL WEIGHT				RUN	JOB
347	0.5972	1.3092 ^{3/2} A	1.3115 ^{3/2} P	1.3114 ^{3/2} A	2	5297	
348	0.5912	1.2087 ^{3/2} A	1.2086 ^{3/2} P		3	5298	
349	0.5883	0.5728 ^{3/2} A	0.5701 ^{3/2} P	0.5666 ^{3/30} A 0.5645 ^{3/20} P	3	5351	
350	0.6587	0.6216 ^{3/2} A	0.6221 ^{3/2} P		2	5350	
351	0.5931						
352	0.5947	0.6468 ^{3/25} A	0.6535 ^{3/28} A	0.6506 ^{3/28} P 0.6493 ^{3/24} A	2	5324	
353	0.5950	0.7514 ^{3/25} A	0.7027 ^{3/28} P	0.6812 ^{3/28} P 0.6716 ^{3/29} A	3	5330	
354	0.5996	0.6755 ^{3/25} A	0.6614 ^{3/28} A	0.6598 ^{3/28} P 0.6587 ^{3/25} A	1	5328	
355	0.6516	0.6604 ^{3/29} A	0.6436 ^{3/29} P	0.6481 ^{3/30} A 0.6477 ^{3/29} P	1	5349	
356	0.6101						
357	0.6098						
358	0.6156	0.7157 ^{3/29} A	0.7162 ^{3/29} P		2	5353	
359	0.6144	0.6993 ^{3/29} A	0.6994 ^{3/29} P		3	5354	
360	0.6576						
361	0.8188						
362	0.8231	1.1618 ^{5/2} A	1.1616 ^{5/2} P		2	5578	
363	0.8129	0.8817 ^{4/11} A	0.8811 ^{4/12} A	0.8812 ^{4/12} P	2	5407	
364	0.8050	0.8671 ^{4/11} P	0.8666 ^{4/12} A		1	5406	
365	0.8197						
366	0.8141	0.8875 ^{4/11} P	0.8868 ^{4/12} A	0.8874 ^{4/12} P	3	5408	
367	0.8018	0.8730 ^{4/11} P	0.8726 ^{4/12} A		1	5409	
368	0.8143	0.8691 ^{4/11} P	0.8680 ^{4/12} A	0.8675 ^{4/12} P	3	5411	
369	0.8265	0.8827 ^{4/11} P	0.8826 ^{4/12} A		2	5410	
370	0.6537	0.7091 ^{3/29} A	0.7103 ^{3/29} P	0.7099 ^{3/30} A	3	5357	
371	0.5898	0.6358 ^{3/29} A	0.6340 ^{3/29} P	0.6336 ^{3/30} A	1	5355	
372	0.5925	0.6518 ^{3/29} A	0.6499 ^{3/29} P	0.6487 ^{3/30} A 0.6486 ^{3/30} P	2	5356	
373	0.5933						
374	0.5963	0.6601 ^{9/23} A	0.6499 ^{9/23} P	0.6474 ^{9/23} P 0.6382 ^{9/26} A 0.6304 ^{9/26} P	1	6612	
375	0.6500						
376	0.5956						
377	0.5905	0.5998 ^{10/27} A	0.6009 ^{10/27} P	0.5991 ^{10/28} A 0.5999 ^{10/28} A	1	6787	
378	0.5900	0.8268 ^{10/11} A	0.8269 ^{10/11} P		2	6788	
379	0.5956	0.6035 ^{10/27} A	0.6047 ^{10/27} P	0.6033 ^{10/28} A	3	6789	
380	0.6525						
381	0.5890	0.6284 ^{4/7} A	0.6285 ^{4/7} P		3	5396	
382	0.5939	0.6295 ^{4/7} A	0.6290 ^{4/7} P		1	5397	

Continued on page 38

354 0.6614 ^{3/29}P 349 0.5662 ^{3/31}A Read and Understood By _____

353 0.6714 ^{3/29}P 374 0.6114 ^{9/27}A 0.6116 ^{9/27}P

ITEM	RUN	LAB	TARE	WEIGHT
672	1	6632	0.81312	0.81003
673	1	6634	0.81003	0.81002
674	1	6635	0.80911	0.80913
675	1	6636	0.80950	0.80955
676	1	6637	0.81176	
677	1	6638	0.81517	0.81512
678	1	6639	0.81571	
679	1	6640	0.81206	
680	1	6641	0.81228	
681	1	6642	0.81527	
682	1	6643	0.81226	0.81524
683	1	6644	0.80969	0.81225
684	1	6645	0.81545	
685	1	6646	0.81099	
686	1	6647	0.81076	
687	1	6648	0.80926	
688	1	6649	0.82227	
689	1	6650	0.82259	
690	1	6651	0.82267	
691	1	6652	0.81053	0.81654
692	1	6653	0.81571	
693	1	6654	0.81954	0.81573
694	1	6655	0.81671	0.81966
695	1	6656	0.81180	
696	1	6657	0.81578	
697	1	6658	0.81622	0.81654
698	1	6659	0.81573	
699	1	6660	0.81954	0.81573
700	1	6661	0.81671	0.81966
701	1	6662	0.81180	
702	1	6663	0.81578	
703	1	6664	0.81622	0.81654
704	1	6665	0.81573	
705	1	6666	0.81954	0.81573
706	1	6667	0.81671	0.81966
707	1	6668	0.81180	

Continued on Page _____

Read and Understood By _____

Signed _____ Date _____

ITEM	RUN	LAB	TARE	WEIGHT
697	1	6669	0.81550	0.81547
698	1	6670	0.81634	0.81634
699	1	6671	0.81634	0.81634
700	1	6672	0.81634	0.81634
701	1	6673	0.81634	0.81634
702	1	6674	0.81634	0.81634
703	1	6675	0.81634	0.81634
704	1	6676	0.81634	0.81634
705	1	6677	0.81634	0.81634
706	1	6678	0.81634	0.81634
707	1	6679	0.81634	0.81634
708	1	6680	0.81634	0.81634
709	1	6681	0.81634	0.81634
710	1	6682	0.81634	0.81634
711	1	6683	0.81634	0.81634
712	1	6684	0.81634	0.81634
713	1	6685	0.81634	0.81634
714	1	6686	0.81634	0.81634
715	1	6687	0.81634	0.81634
716	1	6688	0.81634	0.81634
717	1	6689	0.81634	0.81634
718	1	6690	0.81634	0.81634
719	1	6691	0.81634	0.81634
720	1	6692	0.81634	0.81634
721	1	6693	0.81634	0.81634
722	1	6694	0.81634	0.81634
723	1	6695	0.81634	0.81634
724	1	6696	0.81634	0.81634
725	1	6697	0.81634	0.81634
726	1	6698	0.81634	0.81634
727	1	6699	0.81634	0.81634
728	1	6700	0.81634	0.81634
729	1	6701	0.81634	0.81634
730	1	6702	0.81634	0.81634
731	1	6703	0.81634	0.81634
732	1	6704	0.81634	0.81634
733	1	6705	0.81634	0.81634
734	1	6706	0.81634	0.81634
735	1	6707	0.81634	0.81634

Continued on Page _____

Read and Understood By _____

Signed _____ Date _____

L.P. Montrose Gas Bag Analysis Results

10/25/94 Run 1

2.2% CO₂
17.3% O₂

10/25/94 Run 2

2.2% CO₂
17.3% O₂

10/25/94 Run 3

2.2% CO₂
17.1% O₂

Analyzed 10/28/94 on Servomex 1410B CO₂ analyzer by KSM
Servomex 1420B O₂ analyzer

APPENDIX C
Source and Process Information



SOURCE AND PROCESS INFORMATION

CLIENT: L.P. Montrose

LOCATION: Montrose CO

SAMPLE SITE: Dryer E-TU

TEST DATE: 11/2/94

CLIENT CONTACT: Sue Somers - LP Wisconsin Dick Klopschinski - LP Montrose

OPERATION PERSONNEL: LP Operation Personnel

AGENCY REPRESENTATIVE: Charlie Cox State of Colorado

AM TEST-AIR QUALITY PERSONNEL: Steve Mackel

TYPE OF PROCESS: Oriental Strand Board (OSB) Manufacture Plant Rotary Dryer

EQUIPMENT MANUFACTURER: MEC Rotary Kiln with Geonergy E-Tube wet electrostatic precipitator

MODEL: MEC 60ft Rotary Kiln CONSTRUCTED/INSTALLED: _____

IDENTIFICATION NO.: _____ DATE LAST TESTED: August 1994

PROCESS RATE: See L.P. Process Data to be provided to Amtest for inclusion in report

FUEL FIRING RATE: _____ TYPE OF FUEL: Sawdust, Sawdust, Waterboard Flies

TYPE OF EMISSION CONTROL DEVICE: Multiclones & Geonergy E-tube wet electrostatic precipitator

EQUIPMENT MANUFACTURER: Geonergy Jarrard Mackel

PURPOSE OF TESTING: Compliance Determining (Geonergy Representative onsite)

EMISSION STANDARD (in units): 7.44 lbs/hr in Perm

PROCESS DESCRIPTION: (.030 gr/lsc was Geonergy's guarantee)

This OSB manufacturing plant uses a MEC 60ft triple pass rotary dryer to dry wood chips for OSB production. The wood furnish is primarily aspen with 25% pine. The dryer is fired with a suspension burner rated at 40 MMBTU/hr. Dryer exhaust gases are controlled by multiclones, followed by an ID fan, a quench duct, and a Geonergy E-Tube wet e.s.p. (25-28% Solids Content in Scrubber Water)

PROCESS SHUTDOWNS/PROBLEMS: None Noted

AUTHORIZED SIGNATURE _____

MONTROSE TESTING 10-25-94
PROCESS DATA

<u>CONTENTS</u>	<u>PAGE</u>
Test schedule and process data summary	2
MONTROSE DRYER TESTING 10-25-94, Pm	
BOARD WEIGHTS / PRODUCTION	3
DRYER DATA SHEETS	4-5
DRYER CHARTS	6
PRESS CHARTS	7
PRESS REPORTS	8
E-TUBE READINGS	9-10

MONTROSE TESTING 10-25-94

TEST SCHEDULE

MONTROSE TESTING 10-25-94

<u>DATE</u>	<u>POLLUTANT</u>	<u>RUN #1</u>	<u>RUN #2</u>	<u>RUN #3</u>
10-25	PM	0816-0934	1018-1122	1213-1315

PROCESS DATA SUMMARY

MONTROSE TESTING 10-25-94

- 11.5 =Plant production rate in TONS per HOUR
- 1.96 =estimated TONS of dry fuel burned based on fuel measurement
- 27392 =LB per HOUR of furnish produced by the dryer
- 1168 =average dryer inlet temperature in deg. f.
- 43.3% =moisture content of incoming wood
- 5.5% =moisture content of wood after drying

MONTROSE TESTING 10-25-94 PM testing

DATA TIME: START= 08:15 END= 13:15 HOURS= 5.00
TOTAL HOURS = 5.00

BOARD WEIGHTS - LBS

weight of approximately every 25th untrimmed board

199	183	197
185	193	189
185	173	196
188	179	192
187	200	
178	191	
181	187	
188	191	
182	200	
182	194	

188.33 lb = average
untrimmed
mat weight

173.26 lb = average
finished board
weight
(untrimmed mat
weight-weight of trim)

8.0% = TRIM

PLANT PRODUCTION RATE

5.00 =hours during testing
83 =pressloads
664 =No. of (8' X 16') boards produced (no. of pressloads x 8 boards per load)
115045 =Lbs. finished product (no. of boards x weight of finished board)
23009 =Lbs. finished product produced per hour (lbs. of finished product / testing hours)
11.50 =Tons finished product per hour (lbs. of finished product per hour / 2000 lb)
note: the press was down for the last 12 minutes of run no. 3, however the
dryer continued to operate at a constant rate, therefore production has been
estimated using the rate from 08:00-13:00

DRYER PRODUCTION RATE:

27392 = lbs of dryer production per hour (lb. of finished product / (1-%trim + % fines))
8.0% =%FINES
8.0% =%TRIM
1168 =average dryer inlet temperature
43.3% =moisture content of incoming wood
5.5% =moisture content of wood after drying

DRYER FUEL BURNING RATE

5.147 =fuel calibration on lbs. per count
3802 =total counts during testing
5.00 =hours during testing
19569 =total lbs. of fuel burned during testing (calibration x total counts)
3914 =lbs of fuel burned per hour (total lbs fuel burned / testing hours)
1.96 = Tons of fuel burned per hour (lbs of fuel burned per hour / 2000 lbs.)
8600 =estimated BTU content per lb. of wood fuel
33.7 =estimated mmbtu input per hour (BTU content per lb. x lb. per hour)

DRYER DATA SHEET

DATE 25 OCTOBER 1994

BY Cony

PLANT MONTROSE, COLORADO

READINGS EVERY 10 MIN.

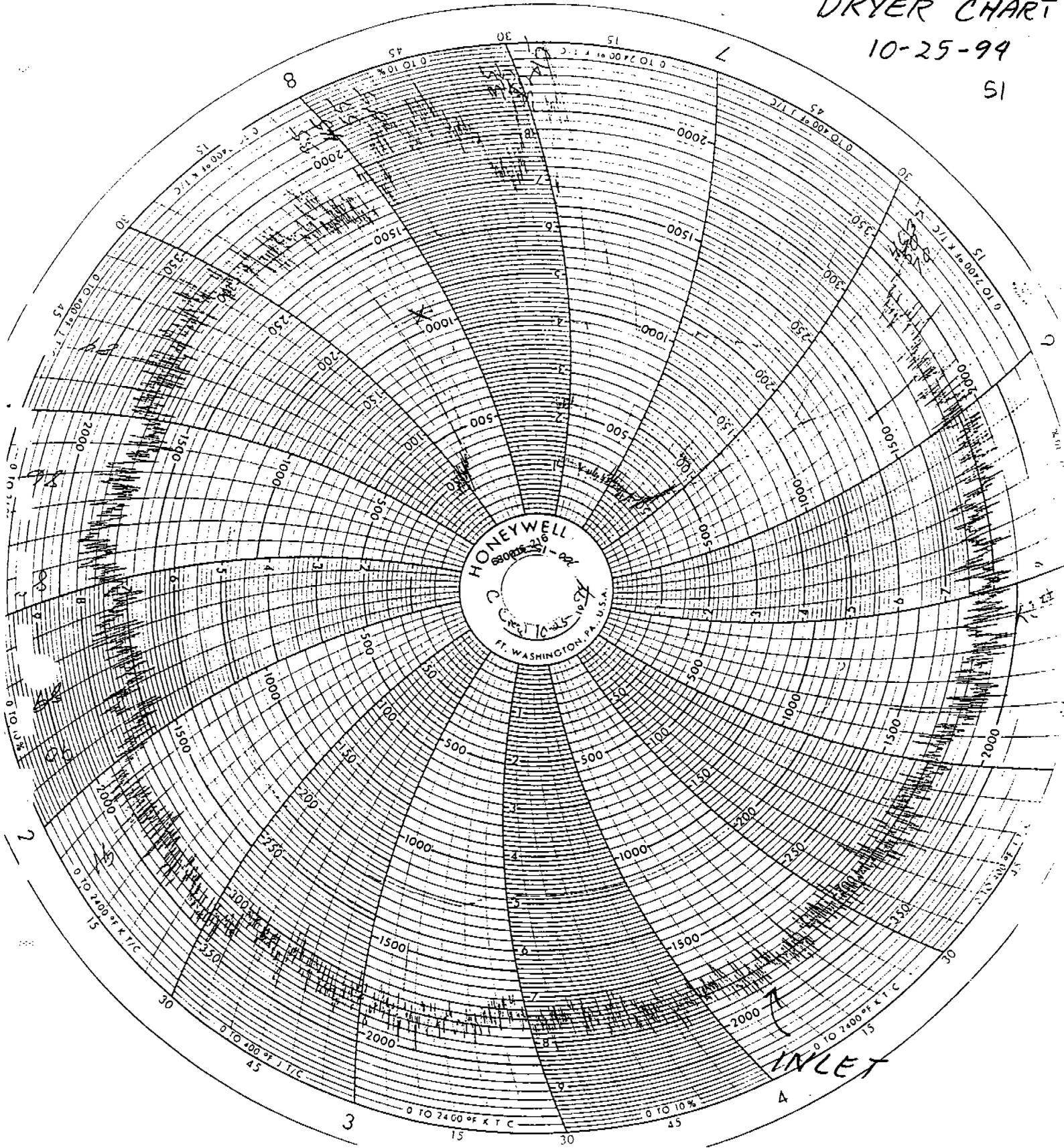
FUEL CALIBRATION:

TIME	OUTLET SET POINT	FEED RATE	DRYER INLET TEMP	DRYER OUTLET TEMP	FUEL COUNT	WET BIN LEVEL	DRY BIN LEVEL		EVERY HOUR FLAKE MOISTURE	
							SUR.	CORE	IN	OUT
800	220	84	1148	219	477464	3/4	3/4	3/4		5.2
810	219	86	1177	219	477601	3/4	1/2	1/2		
820	219	86	1194	217	477732	3/4	1/2	1/2		
830	217	89	1106	215	477856	Full	1/2	1/2	39.6	
840	217	89	1145	217	477984	Full	1/2	1/2		
850	217	90	1129	215	478106	Full	3/4	3/4		
900	216	90	1228	213	478232	Full	3/4	3/4		5.7
910	215	91	1244	214	478376	Full	3/4	3/4		
920	215	91	1291	215	478497	Full	3/4	3/4		
930	215	91	1259	216	478668	Full	3/4	3/4	45.8	
940	215	91	1235	216	478782	3/4	3/4	3/4		
950	215	91	1199	216	478906	Full	3/4	3/4		
1000	215	91	1172	215	479034	Full	3/4	3/4		5.7
1010	215	91	1179	216	479164	3/4	3/4	3/4		
1020	215	91	1120	217	479289	3/4	3/4	3/4		
1030	215	91	1079	215	479409	Full	3/4	3/4	46.5	
1040	215	91	1080	214	479539	Full	3/4	3/4		
1050	215	91	1114	215	479646	Full	3/4	3/4		
1100	215	91	1092	215	479754	Full	3/4	3/4		5.6
1110	215	91	1106	215	479886	Full	3/4	3/4		
1120	215	91	1109	215	480021	Full	3/4	3/4		
1130	215	91	1114	216	480140	Full	3/4	3/4	41.2	
1140	215	91	1122	214	480257	Full	3/4	3/4		
1150	215	91	1125	215	480395	Full	3/4	3/4		
1200	215	91	1160	214	480502	Full	3/4	3/4		5.2
1210	215	91	1192	215	480643	Full	3/4	3/4		
1220	215	91	1206	214	480756	Full	3/4	3/4		
1230	215	91	1183	215	480887	3/4	3/4	3/4	43.5	
1240	215	91	1195	215	481015	-1/2	3/4	3/4		
1250	215	91	1235	215	481147	-1/2	3/4	3/4		

DRYER CHART

10-25-99

51



PRESS CHART
10-25-94

12/0700

0900

52

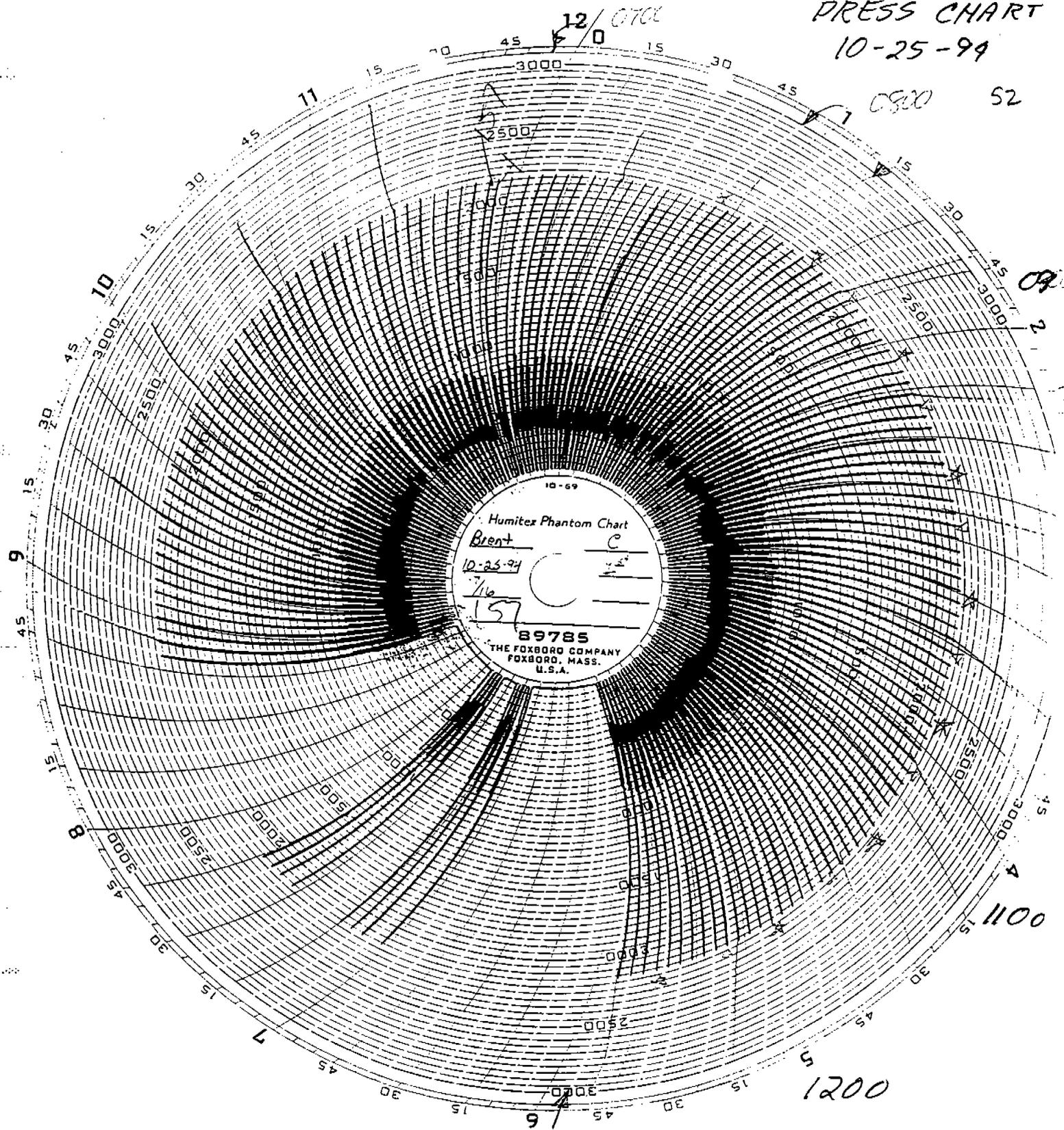
09

2

1100

1200

1300



10-69
Humitex Phantom Chart
Brent C
10-25-94
7/6
157
89785
THE FOXBORO COMPANY
FOXBORO, MASS.
U.S.A.

READINGS EVERY 10 MIN.

E-TUBE DATA SHEET

DATE 25 OCTOBER 1994

PLANT: MONTROSE, COLORADO

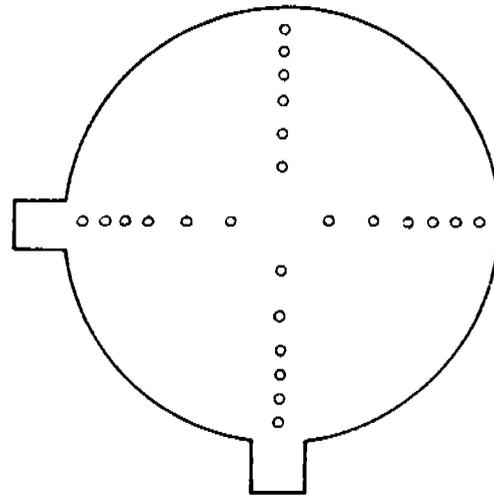
BY *Mike Kennedy*

TIME	QUENCH CHAMBER TEMP	SOUTH TR SET			NORTH TR SET			INDICATE TIME FLUSH CYCLE
		KV	MA	SPARK RATE	KV	MA	SPARK RATE	
0800	128.1	45	160	27.7	49	150	26.8	
0810	128.7	46	190	27.7	48	150	27.9	
0820	129.1	44	190	28.0	48	160	27.6	
0830	126.2	45	200	27.8	48	150	28.0	
0840	127.8	44	190	28.0	48	150	27.8	
0850	127.8	44	200	28.0	48	150	28.0	
0900	130.5	44	200	27.9	48	150	27.7	0901 South
0910	130.4	48	110	27.8	47	150	27.8	
0920	132.1	47	130	27.8	46	180	27.9	
0930	133.3	47	190	27.7	47	180	27.9	
0940	130.9	46	180	27.8	46	180	27.9	
0950	130.5	46	180	27.7	47	180	27.8	
1000	129.6	46	180	27.8	47	180	27.8	
1010	130.9	45	170	27.7	46	180	27.8	1016 North
1020	127.9	45	170	27.9	48	110	27.9	
1030	127.2	44	170	27.8	48	140	27.6	
1040	126.8	44	170	27.9	48	170	27.8	
1050	127.3	45	210	28.0	49	160	28.0	
1110	128.8	45	230	27.9	48	180	27.7	
1120	128.3	45	210	27.7	49	170	27.7	
1130	128.7	44	210	27.9	48	180	27.9	1131 South
1140	128.4	47	130	27.9	47	170	27.8	
1150	129.0	47	130	27.8	48	170	27.7	
1200	129.2	47	160	27.9	48	170	27.9	
1210	130.8	46	170	27.9	46	180	27.9	
1220	131.1	46	180	27.7	47	170	27.8	
1230	130.2	46	180	27.8	47	170	27.8	
1240	130.7	45	200	28.0	47	170	27.8	1246 North
1250	132.7	45	160	27.9	49	90	28.0	
1300	132.5	45	170	27.7	49	140	27.4	

APPENDIX D
Supporting Information

CROSS SECTIONAL AREA

<u>Traverse Point</u>	<u>Distance (inches)</u>
1	1.0
2	3.2
3	5.7
4	8.5
5	12.0
6	17.1
7	30.9
8	36.0
9	39.5
10	42.3
11	44.8
12	47.0

**STACK DIMENSIONS**

48-inch diameter circular stack

2 ports

A = 40 feet downstream

B = 14 feet upstream

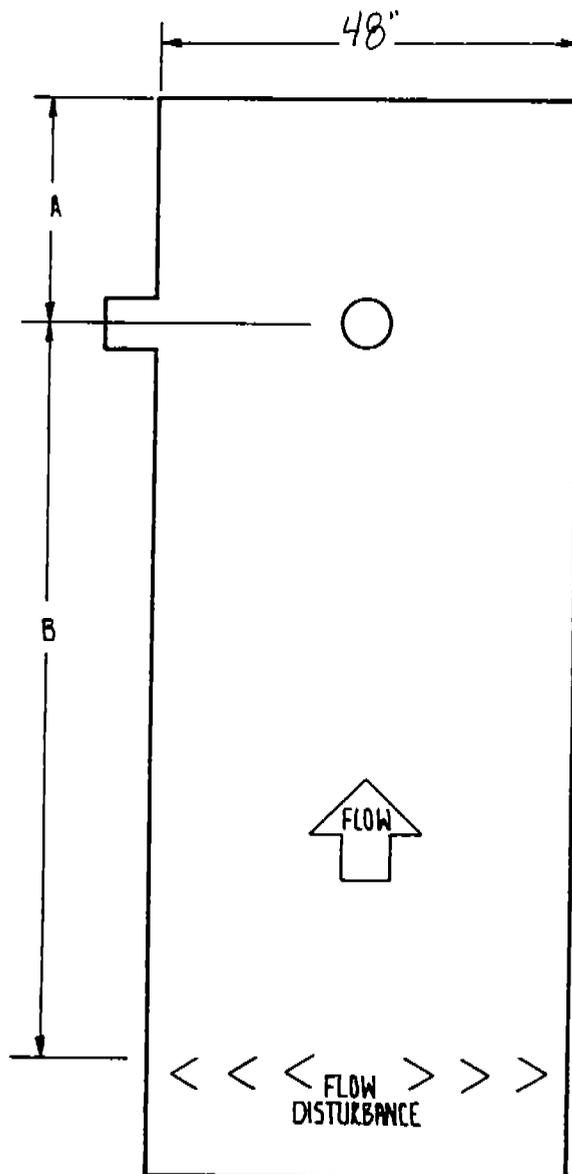


Figure 1. Location of Sampling Ports and Traverse Points

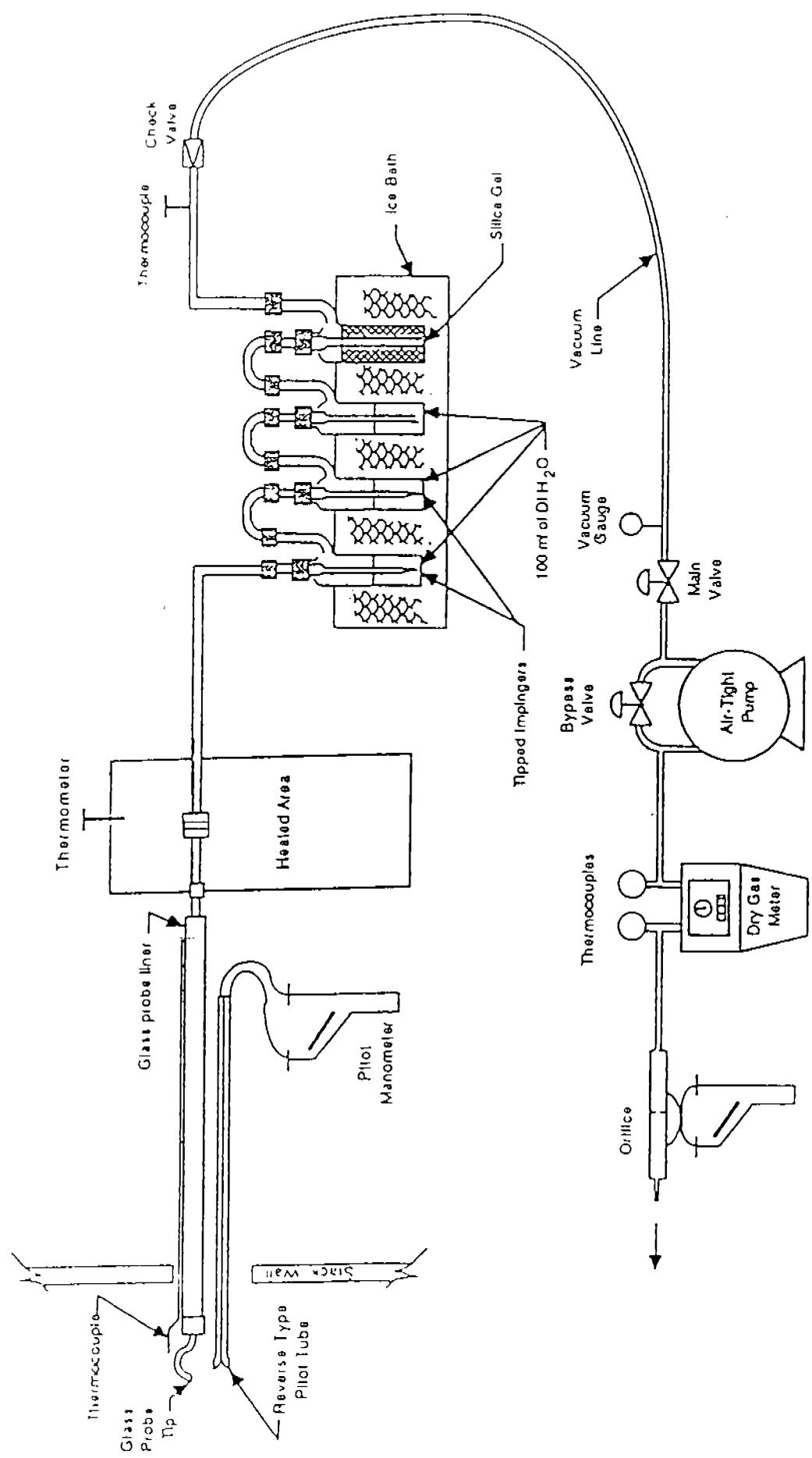


Figure: 2 EPA Method 202 Sample Train

METHOD 1 - LOCATION OF TRAVERSE POINTS

Circular Stacks

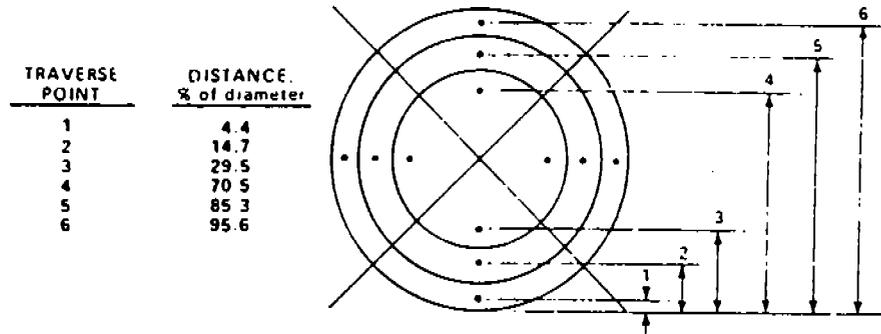


Figure 1-3. Example showing circular stack cross section divided into 12 equal areas, with location of traverse points indicated.

TABLE 1-2. LOCATION OF TRAVERSE POINTS IN CIRCULAR STACKS

(Percent of stack diameter from inside wall to traverse point)

Traverse point number on a diameter	Number of traverse points on a diameter—																								
	2	4	6	8	10	12	14	16	18	20	22	24													
1	14.6	6.7	4.4	3.2	2.6	2.1	1.8	1.6	1.4	1.3	1.1	1.1													
2	85.4	25.0	14.6	10.5	8.2	6.7	5.7	4.9	4.4	3.9	3.5	3.2													
3		75.0	29.6	19.4	14.6	11.8	9.9	8.5	7.5	6.7	6.0	5.5													
4			93.3	70.4	52.3	42.6	37.7	33.5	30.9	28.7	26.7	25.0													
5				85.4	67.7	54.2	45.0	39.1	35.9	33.2	30.9	29.0													
6					95.6	80.6	65.8	55.6	49.0	45.5	42.6	40.6													
7						89.5	77.4	64.4	54.0	47.4	44.4	42.4													
8							96.8	85.4	75.0	63.4	55.0	51.8													
9								91.8	82.3	73.1	62.5	58.2													
10									97.4	88.2	79.9	71.7													
11										93.3	85.4	78.0													
12											90.1	83.1													
13												94.3													
14													87.5												
15														98.2											
16															91.5										
17																95.1									
18																	98.4								
																		92.5							
																			87.1						
																				90.3					
																					95.6				
																						90.3			
																							93.3		
																								88.4	
																									83.9

Rectangular Stacks

For a rectangular cross section, an equivalent diameter (D_e) shall be calculated from the following equation, to determine the upstream and downstream distances:

$$D_e = \frac{2LW}{L+W}$$

where L =length and W =width.

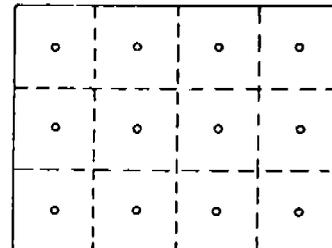


Figure 1-4. Example showing rectangular stack cross section divided into 12 equal areas, with a traverse point at centroid of each area.

METHOD 1 - MINIMUM NUMBER OF TRAVERSE POINTS

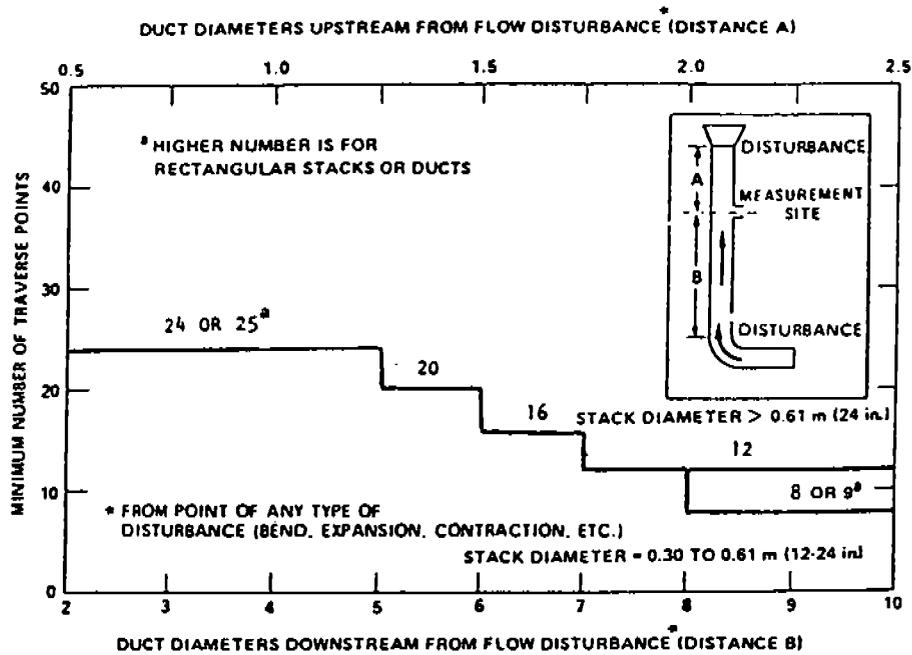


Figure 1-1. Minimum number of traverse points for particulate traverses.

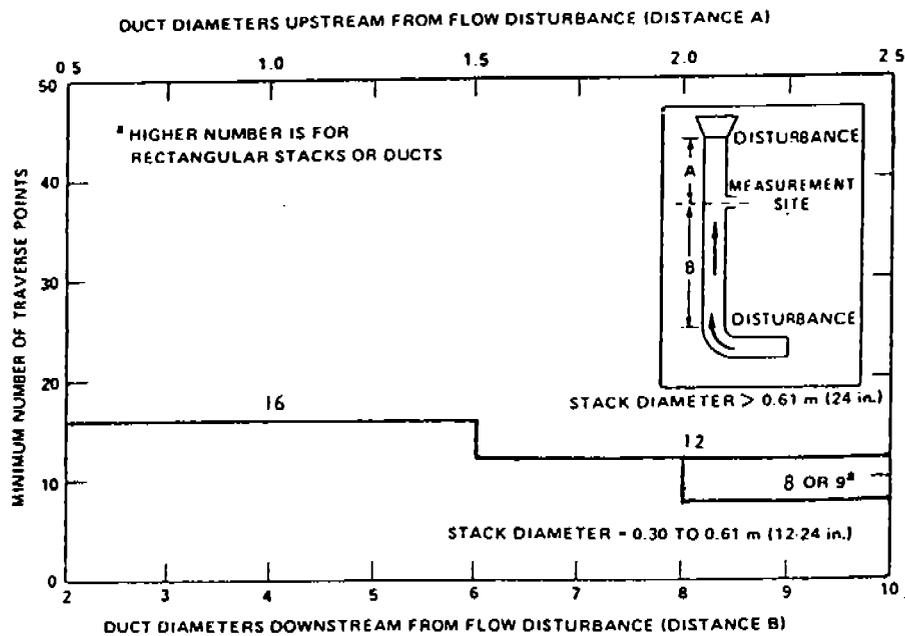


Figure 1-2. Minimum number of traverse points for velocity (nonparticulate) traverses.

METHOD 2 - STACK GAS VELOCITY AND VOLUMETRIC FLOW CALCULATIONS

5.1 Nomenclature.

A = Cross-sectional area of stack, m^2 (ft^2).
 B_w = Water vapor in the gas stream (from Method 5 or Reference Method 4), proportion by volume.
 C_p = Pitot tube coefficient, dimensionless.
 K_p = Pitot tube constant.

$$34.97 \frac{m}{sec} \left[\frac{(g/g\text{-mole})(mm\ Hg)}{(^{\circ}K)(mm\ H_2O)} \right]^{1/2}$$

for the metric system and

$$85.49 \frac{ft}{sec} \left[\frac{(lb/lb\text{-mole})(in.\ Hg)}{(^{\circ}R)(in.\ H_2O)} \right]^{1/2}$$

for the English system.

M_d = Molecular weight of stack gas, dry basis (see Section 3.6) $g/g\text{-mole}$ ($lb/lb\text{-mole}$).
 M_w = Molecular weight of stack gas, wet basis, $g/g\text{-mole}$ ($lb/lb\text{-mole}$).
 $= M_d(1 - B_w) + 18.0 B_w$

Eq. 2-5

P_{bar} = Barometric pressure at measurement site, $mm\ Hg$ ($in.\ Hg$).

P_s = Stack static pressure, $mm\ Hg$ ($in.\ Hg$).

P_t = Absolute stack gas pressure, $mm\ Hg$ ($in.\ Hg$).

$= P_{bar} + P_s$

Eq. 2-6

Eq. 2-6

P_{std} = Standard absolute pressure, $760\ mm\ Hg$ ($29.92\ in.\ Hg$).

Q_{std} = Dry volumetric stack gas flow rate corrected to standard conditions, $dscm/hr$ ($dscf/hr$).

t = Stack temperature, $^{\circ}C$ ($^{\circ}F$).

T = Absolute stack temperature, $^{\circ}K$ ($^{\circ}R$).
 $= 273 + t$ for metric.

Eq. 2-7

$= 460 + t$ for English.

Eq. 2-8

T_{std} = Standard absolute temperature, $293\ ^{\circ}K$ ($528\ ^{\circ}R$).

v_s = Average stack gas velocity, m/sec (ft/sec).

Δp = Velocity head of stack gas, $mm\ H_2O$ ($in.\ H_2O$).

3.600 = Conversion factor, sec/hr .

18.0 = Molecular weight of water, $g/g\text{-mole}$ ($lb/lb\text{-mole}$).

5.2 Average Stack Gas Velocity.

$$v_s = K_p C_p (\sqrt{\Delta p})_{0.75} \sqrt{\frac{T_{std}}{P_s M_d}}$$

Equation 2-9

5.3 Average Stack Gas Dry Volumetric Flow Rate.

$$Q_{std} = 3.600(1 - B_w) v_s A \left(\frac{T_{std}}{T_s} \right) \left(\frac{P_t}{P_{std}} \right)$$

Eq. 2-10

METHOD 3 - MOLECULAR WEIGHT AND EXCESS AIR CALCULATIONS

6.1 Nomenclature.

M_d = Dry molecular weight, $g/g\text{-mole}$ ($lb/lb\text{-mole}$).

%EA = Percent excess air.

%CO₂ = Percent CO₂ by volume (dry basis).

%O₂ = Percent O₂ by volume (dry basis).

%CO = Percent CO by volume (dry basis).

%N₂ = Percent N₂ by volume (dry basis).

0.264 = Ratio of O₂ to N₂ in air, v/v .

0.280 = Molecular weight of N₂ or CO, divided by 100.

0.320 = Molecular weight of O₂, divided by 100.

0.440 = Molecular weight of CO₂, divided by 100.

6.2 Percent Excess Air. Calculate the percent excess air (if applicable), by substituting the appropriate values of percent O₂, CO, and N₂ (obtained from Section 4.1.3 or 4.2.4) into Equation 3-1.

% EA =

$$\frac{\%O_2 - 0.5\% CO}{0.264\% N_2 (\%O_2 - 0.5\% CO)} \times 100$$

Eq. 3-1

Note: The equation above assumes that ambient air is used as the source of O₂ and that the fuel does not contain appreciable amounts of N₂ (as do coke oven or blast furnace gases). For those cases when appreciable amounts of N₂ are present (coal, oil, and natural gas do not contain appreciable amounts of N₂) or when oxygen enrichment is used, alternate methods, subject to approval of the Administrator, are required.

6.3 Dry Molecular Weight. Use Equation 3-2 to calculate the dry molecular weight of the stack gas

$$M_d = 0.440(\%CO_2) + 0.320(\%O_2) + 0.280(\%N_2 + \%CO)$$

Eq. 3-2

METHOD 4 - STACK GAS MOISTURE CALCULATIONS

2.3.1 Nomenclature.

B_{wt} = Proportion of water vapor, by volume, in the gas stream.

M_w = Molecular weight of water, 18.0 g/g-mole (18.0 lb/lb-mole).

P_m = Absolute pressure (for this method, same as barometric pressure) at the dry gas meter, mm Hg (in. Hg).

P_{std} = Standard absolute pressure, 760 mm Hg (29.92 in. Hg).

R = Ideal gas constant, 0.06236 (mm Hg) (m³)/(g-mole) (°K) for metric units and 21.85 (in. Hg) (ft³)/(lb-mole) (°R) for English units.

T_m = Absolute temperature at meter, °K (°R).

T_{std} = Standard absolute temperature, 293°K (528°R).

V_m = Dry gas volume measured by dry gas meter, dcm (dcf).

ΔV_m = Incremental dry gas volume measured by dry gas meter at each traverse point, dcm (dcf).

$V_{m(std)}$ = Dry gas volume measured by the dry gas meter, corrected to standard conditions, dscm (dscf).

$V_{wv(std)}$ = Volume of water vapor condensed corrected to standard conditions, scm (scf).

$V_{wv(std)}$ = Volume of water vapor collected in silica gel corrected to standard conditions, scm (scf).

V_f = Final volume of condenser water, ml.

V_i = Initial volume, if any, of condenser water, ml.

W_f = Final weight of silica gel or silica gel plus impinger, g.

W_i = Initial weight of silica gel or silica gel plus impinger, g.

Y = Dry gas meter calibration factor.

ρ_w = Density of water, 0.9982 g/ml (0.002201 lb/ml).

2.3.2 Volume of Water Vapor Condensed.

$$V_{wv(std)} = \frac{(V_f - V_i)\rho_w RT_{std}}{P_{std}M_w}$$

$$= K_1(V_f - V_i)$$

Eq. 4-1

$K_1 = 0.001333$ m³/ml for metric units
 $= 0.04707$ ft³/ml for English units

2.3.3 Volume of Water Vapor Collected in Silica Gel.

$$V_{wv(std)} = \frac{(W_f - W_i)RT_{std}}{P_{std}M_w}$$

$$= K_2(W_f - W_i)$$

Eq. 4-2

Where:

$K_2 = 0.001335$ m³/g for metric units
 $= 0.04715$ ft³/g for English units

2.3.4 Sample Gas Volume.

$$V_{m(std)} = V_m Y \frac{(P_m)(T_{std})}{(P_{std})(T_m)}$$

$$= K_3 Y \frac{V_m P_m}{T_m}$$

Eq. 4-3

Where:

$K_3 = 0.3858$ °K/mm Hg for metric units
 $= 17.64$ °R/in. Hg for English units

NOTE: If the post-test leak rate (Section 2.2.6) exceeds the allowable rate, correct the value of V_m in Equation 4-3, as described in Section 6.3 of Method 5.

2.3.5 Moisture Content.

$$B_{wt} = \frac{V_{wv(std)} + V_{wv(std)}}{V_{wv(std)} + V_{wv(std)} + V_{m(std)}}$$

Eq. 4-4

NOTE: In saturated or moisture droplet-laden gas streams, two calculations of the moisture content of the stack gas shall be made, one using a value based upon the saturated conditions (see Section 1.2), and another based upon the results of the impinger analysis. The lower of these two values of B_{wt} shall be considered correct.

METHOD 5 - PARTICULATE EMISSION CALCULATIONS - (1)

6.1 Nomenclature.

A_n = Cross-sectional area of nozzle, m^2 (ft^2).
 B_m = Water vapor in the gas stream, proportion by volume.
 C_o = Acetone blank residue concentration, mg/g .
 C_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, $g/dscm$ ($g/dscf$).
 l = Percent of isokinetic sampling.
 L_o = Maximum acceptable leakage rate for either a pretest leak check or for a leak check following a component change; equal to $0.0057 m^3/min$ ($0.02 cfm$) or 4 percent of the average sampling rate, whichever is less.
 L_i = Individual leakage rate observed during the leak check conducted prior to the " i^{th} " component change ($i=1, 2, 3, \dots, n$), m^3/min (cfm).
 L_p = Leakage rate observed during the post-test leak check, m^3/min (cfm).
 m_o = Mass of residue of acetone after evaporation, mg .
 m_m = Total amount of particulate matter collected, mg .
 M_w = Molecular weight of water, $18.0 g/g\text{-mole}$ ($18.0 lb/lb\text{-mole}$).
 P_{bar} = Barometric pressure at the sampling site, $mm\ Hg$ ($in. Hg$).
 P_s = Absolute stack gas pressure, $mm\ Hg$ ($in. Hg$).
 P_{std} = Standard absolute pressure, $760 mm\ Hg$ ($29.92 in. Hg$).
 R = Ideal gas constant, $0.06236 mm\ Hg\text{-}m^3/K\text{-}g\text{-mole}$ ($21.85 in. Hg\text{-}ft^3/R\text{-}lb\text{-mole}$).
 T_m = Absolute average dry gas meter temperature (see Figure 5-2), $^{\circ}K$ ($^{\circ}R$).
 T_s = Absolute average stack gas temperature (see Figure 5-2), $^{\circ}K$ ($^{\circ}R$).
 T_{std} = Standard absolute temperature, $293^{\circ}K$ ($528^{\circ}R$).
 V_o = Volume of acetone blank, ml .
 V_{wa} = Volume of acetone used in wash, ml .
 V_{lc} = Total volume of liquid collected in impingers and silica gel (see Figure 5-3), ml .
 V_m = Volume of gas sample as measured by dry gas meter, dcm ($dscf$).
 V_{mstd} = Volume of gas sample measured by the dry gas meter, corrected to standard conditions, $dscm$ ($dscf$).
 V_{wstd} = Volume of water vapor in the gas sample, corrected to standard conditions, scm (scf).
 v_s = Stack gas velocity, calculated by Method 2, Equation 2-9, using data obtained from Method 5, m/sec (ft/sec).
 W_o = Weight of residue in acetone wash, mg .
 Y = Dry gas meter calibration factor.
 ΔH = Average pressure differential across the orifice meter (see Figure 5-2), $mm\ H_2O$ ($in. H_2O$).
 ρ_w = Density of acetone, mg/ml (see label on bottle).

ρ_w = Density of water, $0.9982 g/ml$ ($0.002201 lb/ml$).

θ = Total sampling time, min .

θ_1 = Sampling time interval, from the beginning of a run until the first component change, min .

θ_i = Sampling time interval, between two successive component changes, beginning with the interval between the first and second changes, min .

θ_p = Sampling time interval, from the final (n^{th}) component change until the end of the sampling run, min .

13.6 = Specific gravity of mercury.

60 = Sec/min.

100 = Conversion to percent.

6.2 Average Dry Gas Meter Temperature and Average Orifice Pressure Drop. See data sheet (Figure 5-2).

6.3 Dry Gas Volume. Correct the sample volume measured by the dry gas meter to standard conditions ($20^{\circ}C$, $760 mm\ Hg$ or $68^{\circ}F$, $29.92 in. Hg$) by using Equation 5-1.

$$V_{m(std)} = V_m Y \left(\frac{T_{std}}{T_m} \right) \left[\frac{P_{bar} + \frac{\Delta H}{13.6}}{P_{std}} \right]$$

$$= K_1 V_m Y \frac{P_{bar} + (\Delta H/13.6)}{T_m}$$

Equation 5-1

Where:

$K_1 = 0.3858^{\circ}K/mm\ Hg$ for metric units

$= 17.64^{\circ}R/in. Hg$ for English units

NOTE: Equation 5-1 can be used as written unless the leakage rate observed during any of the mandatory leak checks (i.e., the post-test leak check or leak checks conducted prior to component changes) exceeds L_o . If L_p or l_i exceeds L_o , Equation 5-1 must be modified as follows:

(a) Case I. No component changes made during sampling run. In this case, replace V_m in Equation 5-1 with the expression:

$$[V_m - (L_p - L_o)\theta]$$

(b) Case II. One or more component changes made during the sampling run. In this case, replace V_m in Equation 5-1 by the expression:

$$\left[V_m - (L_i - L_o)\theta_i - \sum_{i=2}^n (L_i - L_o)\theta_i - (L_p - L_o)\theta_p \right]$$

and substitute only for those leakage rates (L_i or L_p) which exceed L_o .

6.4 Volume of Water Vapor.

METHOD 5 - PARTICULATE EMISSION CALCULATIONS - (2)

$$V_{w(s,d)} = V_{1c} \left(\frac{p_w}{M_w} \right) \left(\frac{RT_{s,d}}{P_{s,d}} \right) = K_2 V_{1c} \quad \text{Equation 5-2}$$

Where:
 $K_2 = 0.001333 \text{ m}^3/\text{ml}$ for metric units
 $= 0.04707 \text{ ft}^3/\text{ml}$ for English units.

6.5 Moisture Content.

$$B_{w,s} = \frac{V_{w(s,d)}}{V_{m(s,d)} + V_{w(s,d)}}$$

Eq. 5-3

NOTE: In saturated or water droplet-laden gas streams, two calculations of the moisture content of the stack gas shall be made, one from the impinger analysis (Equation 5-3), and a second from the assumption of saturated conditions. The lower of the two values of B_w shall be considered correct. The procedure for determining the moisture content based upon assumption of saturated conditions is given in the Note of Section 1.2 of Method 4. For the purposes of this method, the average stack gas temperature from Figure 5-2 may be used to make this determination, provided that the accuracy of the in-stack temperature sensor is $\pm 1^\circ \text{C}$ (2°F).

6.6 Acetone Blank Concentration.

$$C_a = \frac{m_a}{V_a \rho_a} \quad \text{Eq. 5-4}$$

6.7 Acetone Wash Blank.

$$W_a = C_a V_w \rho_w \quad \text{Eq. 5-5}$$

6.8 Total Particulate Weight. Determine the total particulate catch from the sum of the weights obtained from Containers 1 and 2 less the acetone blank (see Figure 5-3).

NOTE: Refer to Section 4.1.5 to assist in calculation of results involving two or more filter assemblies or two or more sampling trains.

6.9 Particulate Concentration.

$$C_p = (0.001 \text{ g/mg}) (m_p / V_m (s,d))$$

Eq. 5-6

6.10 Conversion Factors:

From	To	Multiply by
scf	m ³	0.02832
g/ft ³	gr/ft ³	15.43
g/ft ³	lb/ft ³	2.205×10^{-3}
g/ft ³	g/m ³	35.31

6.11 Isokinetic Variation.

6.11.1 Calculation From Raw Data.

$$I = \frac{100 T_s [K_2 V_r + (P_m / T_m) (P_{bar} + \Delta H / 13.6)]}{600 v_r P_s A_s} \quad \text{Eq. 5-7}$$

Where:

$K_2 = 0.003454 \text{ mm Hg} \cdot \text{m}^3/\text{ml} \cdot ^\circ\text{K}$ for metric units.
 $= 0.002669 \cdot \text{in. Hg} \cdot \text{ft}^3/\text{ml} \cdot ^\circ\text{R}$ for English units.

6.11.2 Calculation From Intermediate Values.

$$I = \frac{T_s V_m (s,d) P_{s,d} 100}{T_{s,d} v_r \theta A_s P_s 60 (1 - B_{w,s})}$$

$$= K_3 \frac{T_s V_m (s,d)}{P_s v_r A_s \theta (1 - B_{w,s})}$$

Equation 5-8

where:

$K_3 = 4.320$ for metric units
 $= 0.09450$ for English units.

6.12 Acceptable Results. If 90 percent < I < 110 percent, the results are acceptable. If the particulate results are low in comparison to the standard, and I is over 110 percent or less than 90 percent, the Administrator may accept the results.

NOMENCLATURE
METHOD 5 CALCULATIONS

- $V_{m_{std}}$ = Volume of gas sample measured by the dry gas meter, corrected to standard conditions, dscm (dscf).
- Y = Dry gas meter calibration factor
- P_b = Barometric pressure at the sampling site, mm Hg (in. Hg)
- H = Average pressure differential across the orifice meter, mm H₂O (in. H₂O)
- T_m = Absolute average dry gas meter temperature, ° K (° R)
- dscm = Dry standard cubic meters
- dscf = Dry standard cubic feet
- W_a = Weight of residue in acetone wash
- M_a = Mass of residue of acetone after evaporation, mg
- C_a = Acetone blank residue concentration, mg/g
- V_a = Volume of acetone blank
- V_{aw} = Volume of acetone used in wash, ml
- M_n = Total amount of particulate matter collected, mg
- C_s = Concentration of particulate matter in stack gas, dry basis, corrected to standard conditions, mg/dscm (gr/dscf)
- gr/dscf = grains per dry standard cubic foot
- $V_{w_{std}}$ = Volume of water vapor in the gas sample, corrected to standard conditions, scm (scf)
- B_{ws} = Water vapor in the gas stream, proportion by volume
- M_d = Molecular weight of stack gas, g/g-mole on dry basis
- M_s = Molecular weight of stack gas, g/g-mole on wet basis
- V_s = Stack gas velocity, calculated by Method 2, Equation 2-9, using data obtained from Method 5, m/sec (ft/sec)
- C_p = Pitot tube coefficient, dimensionless
- Δ_p = Velocity head of stack gas, mm H₂O (in. H₂O)
- P_s = Absolute stack gas pressure, mm Hg (in. Hg)

NOMENCLATURE (continued)
METHOD 5 CALCULATIONS

- Q_{std} = Dry volumetric stack gas flow rate corrected to standard conditions, dscm/hr (dscf/hr)
- dscf/min = dry standard cubic feet per minute (also identified as dcfm or scfm)
- acfm = actual cubic feet per minute
- I = Percent of isokinetic sampling
- A_n = Cross-sectional area of nozzle, m^2 (ft^2)



POST TEST METER CALIBRATION AM TEST - AIR QUALITY, INC.

FILE NAME: POSTLPCO
 METER BOX #: BLUE MAG
 CALIBRATION DATE: 11/7/94
 METHOD OF CALIB.: STANDARD DRY GAS METER (Method 5 Section 7.1)

TOTAL TIME min	DELTA H "H2O	METER VOL V1 cf	METER VOL V2 cf	TEMP IN deg F	TEMP OUT deg F	BARO. PRES. "Hg	STD DGM V1	STD DGM V2	ST.DGM TEMP IN deg F	ST.DGM TEMP OUT deg F	ST.DGM Yds	Y FACTOR	DELTA HD
7.0	2.0	364.646	369.940	82.0	82.0	29.71	852.027	857.352	82.0	82.0	1.001	1.0019	1.995
8.0	2.0	369.940	376.007	82.0	81.0	29.71	857.352	863.473	82.0	82.0	1.001	1.0040	1.975
7.0	2.0	376.007	381.300	83.0	81.0	29.71	863.473	868.800	82.0	81.0	1.001	1.0034	1.993
7.0	2.0	381.300	386.634	83.0	81.0	29.71	868.800	874.180	82.0	82.0	1.001	1.0047	1.958
AVERAGE												1.003	1.980



DRY GAS METER CALIBRATION
AM TEST - AIR QUALITY, INC.

FILE NAME: BLUE494
 METER BOX #: BLUE MAGNEHELIC
 CALIBRATION DATE: 4-19-94
 METHOD OF CALIB.: STANDARD DRY GAS METER (Method 5 Section 7.1)

TOTAL TIME min	DELTA H "H2O	METER VOL V1 cf	METER VOL V2 cf	TEMP IN deg F	TEMP OUT deg F	BARO. PRES. "Hg	STD DGM V1	STD DGM V2	ST.DGM TEMP IN deg F	ST.DGM TEMP OUT deg F	ST.DGM Yds	Y FACTOR	DELTA HQ
11.21	0.5	548.549	552.892	64.0	62.0	29.74	772.500	776.800	61.0	63.0	1.001	0.9918	1.887
10.56	1.0	553.406	559.268	67.0	63.0	29.74	777.400	783.200	61.0	63.0	1.001	0.9936	1.837
12.78	1.5	560.072	568.666	70.0	65.0	29.74	783.900	792.400	62.0	63.0	1.001	0.9958	1.876
13.08	2.0	569.776	579.891	73.0	66.0	29.74	793.500	803.500	63.0	63.0	1.001	0.9970	1.893
12.26	2.5	580.807	591.527	76.0	68.0	29.74	804.400	815.000	66.0	63.0	1.001	0.9978	1.854
12.27	3.0	592.242	604.074	78.0	70.0	29.74	815.700	827.350	66.0	64.0	1.001	0.9951	1.841
11.90	4.0	604.689	617.714	79.0	71.0	29.74	828.000	840.900	68.0	64.0	1.001	0.9985	1.887
AVERAGE													1.868

PRESSURE SENSOR CALIBRATION DATA FORM

Date 1-4-94 Control Box # Blue Manometer
 Ambient Temperature 70 °F Barometric Pressure 29.25 in Hg

	MAGNEHELIC GAUGE #	REFERENCE MANOMETER READING Inches H ₂ O	MAGNEHELIC GAUGE READING Inches H ₂ O	PRESSURE DIFFERENCE	
				Inches H ₂ O	%
	Blue Mag	M5 Box			
High	0-1	0.20	0.21	0.01	5.0
		0.41	0.42	0.01	2.4
		0.61	0.63	0.02	3.3
		0.80	0.81	0.01	1.3
Low	0-1	0.20	0.20	0.00	0.0
		0.40	0.42	0.02	5.0
		0.61	0.62	0.01	1.6
		0.80	0.80	0.00	0.0
High	0-4	0.80	0.82	0.02	2.5
		1.60	1.63	0.03	1.9
		2.40	2.43	0.03	1.3
		3.22	3.24	0.02	0.6
Low	0-4	0.80	0.83	0.03	3.7
		1.60	1.62	0.02	1.3
		2.40	2.42	0.02	0.8
		3.20	3.23	0.03	0.9

$(\text{ref pres. } ^\circ \text{H}_2\text{O} - \text{test pres. } ^\circ \text{H}_2\text{O}) \cdot 100 \leq 5\%$
 $(\text{ref pres. } ^\circ \text{H}_2\text{O})$

PRESSURE SENSOR CALIBRATION DATA FORM

Date 1-4-94 Control Box # Blue ManometerAmbient Temperature 70 °F Barometric Pressure 29.25 in Hg

	MAGNEHELIC GUAGE #	REFERENCE MANOMETER READING Inches H ₂ O	MAGNEHELIC GUAGE READING Inches H ₂ O	PRESSURE DIFFERENCE	
				Inches H ₂ O	%
	Blue Mag	MS Box			
High	0-5	1.00	0.99	0.01	1.0
		2.00	1.98 1.92	0.08	4.0
		3.00	2.87	0.13	4.3
		4.00	3.87	0.13	3.2
Low	0-5	1.00	0.99	0.01	1.0
		2.01	1.98	0.03	1.5
		2.95	2.89	0.06	2.0
		4.00	3.85	0.15	3.7
High	0-2	0.40	0.40	0.00	0.0
		0.81	0.82	0.01	1.2
		1.22	1.20	0.02	1.6
		1.60	1.58	0.02	1.3
Low	0-2	0.41	0.41	0.00	0.0
		0.80	0.77	0.03	3.8
		1.20	1.15	0.05	4.2
		1.60	1.54	0.06	3.8

$$\frac{(\text{ref. pres. } ^\circ \text{H}_2\text{O} - \text{test pres. } ^\circ \text{H}_2\text{O})}{(\text{ref. pres. } ^\circ \text{H}_2\text{O})} \cdot 100 \leq 5\%$$

~~TEMPERATURE~~ TEMPERATURE SENSOR CALIBRATION DATA FORM

Date 6-21-93 Thermocouple Indicator # Blue KAH

Ambient Temperature 64 °F Barometric Pressure 29.4 in Hg

THERMOCOUPLE # OR REFERENCE	REFERENCE THERMOMETER TEMPERATURE °F	THERMOCOUPLE TEMPERATURE °F	TEMPERATURE DIFFERENCE	
			°F	%
Blue	KAH			
Meter in	34	33	1	0.29%
	212	214	2	0.39%
Meter out	34	34	0	0-0
	212	216	4	0.6%

$$\frac{(\text{ref temp. } ^\circ\text{F} + 460) - (\text{test therm. temp. } ^\circ\text{F} + 460)}{(\text{ref temp. } ^\circ\text{F} + 460)} * 100 \leq 1.5\%$$

STANDARD DRY GAS METER CALIBRATION DATA SUMMARY
AMTEST AIR QUALITY, INC.

PERFORMED BY HOMER R. DULIN CO.

Cal. DGM Ind, SCFH	Primary Standard SCFH	Yds
13.5	13.8	1.022
25.7	25.8	1.004
35.9	35.7	0.994
46.5	46.2	0.994
55.9	55.9	1.000
66.8	66.6	0.997
85.5	84.8	0.992
88.8	88.5	0.997
99.9	100.1	1.002
108.6	108.9	1.003
113.3	113.5	1.002
AVG. Yds =		1.001

HOMER R. DULIN CO.

729 EAST WILLOW STREET
LONG BEACH, CALIFORNIA 90806

(310) 424-8533 (213) 636-4096 FAX (310) 426-7707

CERT. NO. 4-304-4

CALIBRATION CERTIFICATION

SUBMITTED BY: KRIS A. HANSEN CO.

FLOWMETER SERIAL No. 0302

MANUFACTURER EQUIMETER MFG. SERIAL No. 27688

TUBE No. _____ FLOAT No. _____
MODEL NO.: T-110

REMARKS: CALIBRATED IN CFH AIR @ 14.7 PSIA & 70 DEG F.

ACCURACY: SEE DATA

INDICATED		ACTUAL	
CFH		CFH	
113.3		113.5	
108.6		108.9	
99.9		100.1	
88.8		88.5	
85.5		84.8	
66.8		66.6	
55.9		55.9	
46.5		46.2	
35.9		35.7	
25.7		25.8	
13.5		13.8	

Flowmeter Certified with HOMER R. DULIN CO.

Equipment No. 12400 Accuracy 0.2% Calib. Due 5-28-94
Procedure No. 101G
NIST Cert. No. 821/249576-92

Our standards are certified by or are traceable to the National Institute of Standards and Technology and comply with MIL-STD-45662A.

P.O. No. 06330

Shipper No. _____

4-18-94

4-18-95

CALIBRATION DATE

RECALIBRATION DATE

B. Richardson

CALIBRATION TECHNICIAN
B. RICHARDSON

TYPE S PITOT TUBE INSPECTION DATA FORM

Date 12-28-93 Pitot Tube # P5HClient Louisiana-PacificLocation Montrose COSite(s) Deyre E-tube stackTest
Date(s) 10-25-94Pitot tube assembly level? yes noPitot tube openings damaged? yes (explain below) no $\alpha_1 = \underline{1.5}^\circ (<10^\circ)$, $\alpha_2 = \underline{2.5}^\circ (<10^\circ)$, $\beta_1 = \underline{2.5}^\circ (<5^\circ)$, $\beta_2 = \underline{2.0}^\circ (<5^\circ)$ $\gamma = \underline{1.5}^\circ$; $\theta = \underline{1.0}^\circ$; $A = \underline{1.18}$ cm (in.) $z = A \sin \gamma = \underline{0.0292}$ cm (in.); <0.32 cm ($<1/8$ in.), $w = A \sin \theta = \underline{0.0195}$ cm (in.); <0.08 cm ($<1/32$ in.) $P_A = \underline{0.559}$ cm (in.) $P_b = \underline{0.559}$ cm (in.) $D_t = \underline{0.375}$ cm (in.)Comments:

Calibration required? yes* no

*If yes, tag and take out of service until repaired.

VP

STACK TEMPERATURE SENSOR CALIBRATION DATA FORM

Date 8/23/94 Thermocouple Indicator: BLUE MAG BOX

Ambient Temperature 66 °F Barometric Pressure 30.79 in Hg

THERMOCOUPLE # OR REFERENCE	REFERENCE THERMOMETER TEMPERATURE °F	THERMOCOUPLE TEMPERATURE °F	TEMPERATURE DIFFERENCE	
			°F	%
T5F	<u>33</u>	<u>31</u>		
ICE BATH	<u>33</u>	<u>31</u>	<u>2.0</u>	<u>0.41</u>
↙ ↘	<u>34</u>	<u>33</u>	<u>1.0</u>	<u>0.20</u>
	<u>34</u>	<u>33</u>	<u>1.0</u>	<u>0.20</u>
BOILING WATER	<u>208</u>	<u>209</u>	<u>1.0</u>	<u>0.15</u>
↙ ↘	<u>208</u>	<u>209</u>	<u>1.0</u>	<u>0.15</u>
	<u>208</u>	<u>209</u>	<u>1.0</u>	<u>0.15</u>
BOILING OIL	<u>362</u>	<u>362</u>	<u>0.0</u>	<u>0.0</u>
	<u>368</u>	<u>368</u>	<u>0.0</u>	<u>0.0</u>
	<u>374</u>	<u>374</u>	<u>0.0</u>	<u>0.0</u>

$$\frac{(\text{ref temp. } ^\circ\text{F} + 460) - (\text{test therm. temp. } ^\circ\text{F} + 460)}{(\text{ref temp. } ^\circ\text{F} + 460)} \cdot 100 \leq 1.5\%$$



**RESUME OF
KRIS A. HANSEN**

**PRESIDENT
AM TEST-AIR QUALITY, INC.
AM TEST ALASKA**

EDUCATION

- B.S., Chemistry, Central Washington University, 1973
- Coursework and 2.5 years research completed towards M.S., Chemistry, Central Washington University
- Several workshops, courses and conferences annually, including a Continuous Emission Monitoring System (CEMS) Workshop, taught by Dr. James Jahnke in October 1992, an Advanced Emission Measurement Workshop in July 1993, a Title III MACT Workshop in 1994, a Title V Operating Permits Workshop in 1994 and Enhanced Monitoring Workshops in 1993 and 1994.

PROFESSIONAL MEMBERSHIPS

- Air and Waste Management Association (AWMA)
- Pacific Northwest International Section of AWMA (PNWIS)
- Source Evaluation Society (SES)

PROFESSIONAL EXPERIENCE

Am Test-Air Quality, Inc. was formed under the direction of Mr. Hansen in 1982. This company conducts full-service source testing activities. Mr. Hansen keeps current on recent developments in EPA methodology and has developed sampling and analysis techniques for many sources for which EPA guidelines are not available. He has worked at numerous types of industrial facilities, including oil and gas refineries, gas, oil and coal-fired power plants, nuclear plants, aluminum plants, wood products industries, smelters, incinerators, and other industrial sources throughout the United States and internationally. Mr. Hansen manages an experienced field testing and laboratory analysis staff. Mr. Hansen manages all phases of project development, including cost estimation, scheduling, sample collection, analysis and report preparation. He also manages all other aspects of the business, including business development and personnel issues.

Prior to joining Am Test, Inc., Mr. Hansen's professional experience included 4.5 years as the Manager of Laboratory services and a Project Leader for an environmental consulting firm which specialized in air quality studies, and 2 years as a laboratory instructor while attending graduate school at Central Washington University, where his research emphasis was in gas chemistry. Mr. Hansen has 18 years of professional experience.

Mr. Hansen has assisted in the instruction of the EPA 450 "Source Sampling for Particulate Pollutants" and the EPA 468 "Source Sampling and Analysis of Gaseous Pollutants" courses offered yearly by the EPA in cooperation with the University of Washington. Students are represented by industry and governmental agency personnel from across the U.S. and other nations. Mr. Hansen was the recipient of the 1987 PNWIS/APCA "Hardhat Award" which was presented in recognition of his contributions to the advancement of source sampling technology in the Pacific Northwest.

**RESUME OF
ANGELA F. BLAISDELL**

VICE PRESIDENT/SR. TECHNICAL WRITER

EDUCATION

- B.S., Marine Resources, Western Washington University, Bellingham, Washington, 1980
- Minors in Chemistry and Biology
- Several workshops, courses and conferences annually, including a Continuous Emission Monitoring System (CEMS) Workshop, taught by Dr. James Jahnke in October 1992 and April 1993, a Title V Operating Permits Workshop in May 1993, an Advanced Emission Measurement Workshop in July 1993, a Title III MACT Workshop in 1994 and Enhanced Monitoring Workshops in 1993 and 1994.

PROFESSIONAL MEMBERSHIPS

- Air and Waste Management Association (A&WMA)
- Pacific Northwest International Section of A&WMA (PNWIS)
- Source Evaluation Society (SES)

PROFESSIONAL EXPERIENCE

Ms. Blaisdell has worked with Am Test-Air Quality, Inc. for the past 10 years and has had 14 years of professional experience in the field of air quality. She helps manage all aspects of source test projects, including initial client contact, scope of work preparation, scheduling, pre-test coordination, implementation of quality assurance programs for field sampling, analysis, data reduction, final data review and report preparation and review. She reviews current literature for each test method and incorporates the methodology into our testing and reporting protocol. Ms. Blaisdell recently assisted in the instruction of the EPA-APTI 464L "Continuous Emission Monitoring" course, the EPA 450 "Source Sampling for Particulate Pollutants" course and the EPA 502 Course "Hazardous Waste Incineration" offered by the EPA's Air Pollution Training Institute in cooperation with the University of Washington. She was the General Chair of the 1994 PNWIS Spring Specialty Conference on Enhanced Monitoring, and was a speaker at the 1990 and 1991 PNWIS annual meetings on the subject of continuous emission monitoring systems (CEMS). Angela is the current Treasurer and past secretary/treasurer for the Pacific Northwest International Section (PNWIS) of the Air and Waste Management Association (A&WMA) and is the past Chair for the Puget Sound Chapter of PNWIS. Ms. Blaisdell was the recipient of the 1991 PNWIS/A&WMA "Labcoat Award" which is presented each year to an individual to recognize his contribution to the advancement of source sampling methodology in the Pacific Northwest.

Prior to joining Am Test, Ms. Blaisdell acted as a Project Leader and Office Manager for an environmental engineering consulting firm in the Seattle area for 2 years. Experience with that firm involved sample collection, analysis and report preparation for source and ambient air, water and industrial hygiene studies. Ms. Blaisdell also worked on various research projects in the Chemistry department while attending Western Washington University.

**RESUME OF
K. STEVEN MACKEY**

SENIOR AIR QUALITY SPECIALIST

EDUCATION

- A.S., Mathematics, North Idaho College, Coeur d'Alene, Idaho, 1978
- Continuous Emission Monitoring System (CEMS) Workshop, taught by Dr. James Jahnke, April 1993.
- Professional training courses and specialty conferences

PROFESSIONAL MEMBERSHIPS

- Source Evaluation Society (SES)

PROFESSIONAL EXPERIENCE

Mr. Mackey has worked for Am Test-Air Quality Inc. for 5 years as a Senior Air Quality Specialist, and has 14 years experience in the field of air quality and industrial source emissions testing. Mr. Mackey conducts source testing and activities related to source emission testing, including field sampling, test equipment maintenance and calibration, test planning and preparation, and data reduction and evaluation. He has performed sampling for particulates, particle size distribution, sulfur and nitrogen oxides, reduced sulfur compounds, semi-volatile and volatile organic compounds, toxic air pollutants, metals, opacity, and other EPA methodology. Mr. Mackey is certified to observe opacity.

Prior to joining Am Test-Air Quality, Inc., Mr. Mackey was the Source Testing Coordinator for the State of Oregon Department of Environmental Quality (ODEQ) for 1.5 years. This position was responsible for coordinating a statewide source sampling program and source quality assurance program for industrial air pollution sources. His work with ODEQ's Air Quality Division involved evaluating source tests, plant site source test observations, approving source test methods, and evaluating source self-monitoring practices.

Prior to his position with the State of Oregon, Mr. Mackey was a Field Supervisor for Omni Environmental Services in Beaverton, Oregon for 4 years. Experience with this firm was primarily in industrial source emission testing and field and laboratory woodstove emissions testing. In addition to source testing at a wide variety of industrial facilities, Mr. Mackey ran an extensive field study on woodstove performance in Vermont and New York involving specially designed sampling equipment. Prior to working with Omni, Mr. Mackey worked 4.5 years as a Project Leader and Field Technician for an environmental consulting firm which specialized in air quality studies. Experience with this firm involved extensive source testing and other emissions monitoring projects.



**RESUME OF
STACY AKIN**

PROJECT ASSISTANT/TECHNICAL WRITER

EDUCATION

- B.S., Marine Biology, Roger Williams University, Bristol, Rhode Island, 1992

PROFESSIONAL EXPERIENCE

Ms. Akin began her employment with Am Test-Air Quality, Inc. in September 1993. She is responsible for pre-field preparation for air quality projects and preparation of sample trains, glassware, labware and sampling hardware to be used on each specific project. In the field, Ms. Akin has acted as a Project Assistant, assisting the Project Manager, Project Engineer, or Project Leader in setting up the equipment at each sample site, performing tests and recording data, recovering the samples after each test, performing any in-field analysis, and in demobilizing. When working in the field, she is responsible for properly labeling and identifying each sample, and initiating chain-of-custody procedures.

In the laboratory, Ms. Akin is experienced in gravimetric analysis of particulate matter samples and particle size distribution samples. She assists in preparation and analysis of samples for quantifying sulfur oxides, reduced sulfur compounds, toxic metals (including hexavalent chromium), semi-volatile compounds (including dioxins and furans), and volatile organic compounds.

In the office, Ms. Akin reduces the field data and inputs values into data reduction programs and performs by-hand calculations to verify computer program integrity.

Prior to her employment with Am Test, Ms. Akin worked in a large ophthalmology clinic as a receptionist and assistant bookkeeper. Her responsibilities included daily record keeping for the optical shop as well as monthly patient and insurance billings.

RESUME OF
CASSIE B. HEATON

PROJECT ENGINEER/CHEMIST

EDUCATION

- B.A. in Chemistry, BSc in Chemical Engineering
University of Colorado, Boulder, Colorado, 1990
- Engineer-in-Training; State of Colorado
- EPA 450 "Source Sampling for Particulate Pollutants Course",
August 1992.

PROFESSIONAL EXPERIENCE

Ms. Heaton began her employment with Am Test-Air Quality, Inc. in January 1992. She performs technical writing of reports and word processing of proposals, test plans, and reports for the Senior Technical Writers to review. She performs data processing of field and laboratory analysis data using computerized data reduction programs, and performs by-hand calculations to verify computer program integrity. She also develops new computerized data reduction programs for field and laboratory analysis data and reviews other staff members data for accuracy.

Ms. Heaton is also experienced preparing and packing sampling equipment for mobilization to the test sites, and in assisting the Project Leaders to collect samples. She has assisted in handling and setting up the equipment at each sample site, running the control box and recording data, sample recovery, in-field analysis, data reduction, and demobilization.

In the laboratory, Ms. Heaton is experienced with gravimetric analysis of particulate matter samples and particle size distribution samples, wet chemical titrametric analysis and glassware preparation for organics and metals testing. She assists in preparation and analysis of samples for quantifying sulfur oxides, nitrogen oxides, hydrogen sulfide, toxic metals (including hexavalent chromium), semi-volatile compounds (including dioxins and furans), and volatile organic compounds.

**RESUME OF
ANNIKA M. WOHR**

PROJECT ASSISTANT/TECHNICAL WRITER

EDUCATION

- B.S., Biology, Florida Atlantic University, Boca Raton, Florida, 1992, emphasis in Marine Biology

PROFESSIONAL EXPERIENCE

Ms. Woehr began her employment with Am Test-Air Quality, Inc. in June 1994. She has been responsible for pre-field preparation for air quality projects and preparation of sample trains, glassware, labware and sampling hardware to be used on each specific project. In the field, Ms. Woehr has acted as a Project Assistant, assisting the Project Manager, Project Engineer, or Project Leader in setting up the equipment at each sample site, performing tests and recording data, recovering the samples after each test, performing any in-field analysis, and in demobilizing. When working in the field, she is responsible for properly labeling and identifying each sample, and initiating chain-of-custody procedures.

In the laboratory, Ms. Woehr is experienced in gravimetric analysis of particulate matter samples and particle size distribution samples. She assists in preparation and analysis of samples for quantifying sulfur oxides, reduced sulfur compounds, toxic metals (including hexavalent chromium), semi-volatile compounds (including dioxins and furans), and volatile organic compounds.

In the office, Ms. Woehr reduces the field data and inputs values into data reduction programs and performs by-hand calculations to verify computer program integrity.

Prior to joining Am Test, Annika worked through an employment agency for a coastal management firm restoring and building up sand dunes on beaches in Florida using native and endangered plants. She worked as a teacher assistant, office assistant, research assistant while attending Florida Atlantic University. One project involved maintaining sea turtle tanks in a laboratory. She also spent a summer working as a secretary in Geneva, Switzerland.

**AM TEST-AIR QUALITY, INC.
CAPABILITIES**

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Am Test-Air Quality, Inc. is an independent company providing comprehensive air pollution testing services to industry and government. Am Test, Inc.'s Air Quality Division was developed in 1982 by Mr. Kris A. Hansen. On January 1, 1991, the Air Quality Division incorporated as a separate company named Am Test-Air Quality, Inc. Am Test-Air Quality's main office and laboratory facility is located in Preston, Washington. The Preston office employs a staff of 20 qualified, experienced engineers, chemists and scientists. In January 1993, Am Test-Air Quality, Inc. opened a branch office, Am Test Alaska in Anchorage, Alaska in order to better serve our many Alaskan clients. One aspect of the testing services provided by Am Test-Air Quality, Inc., which we feel is unsurpassed by other testing firms, is the quality and experience of our personnel. We utilize highly experienced, motivated personnel on all projects. We believe that the success of a project is dependent on the use of state-of-the-art equipment and experienced, knowledgeable personnel. We have performed source testing projects of all different sizes and difficulty levels. Typically, we dispatch test teams of one to ten individuals. On all projects, our attention to detail remains extremely high.

Am Test-Air Quality, Inc. performs source testing projects at all types of facilities throughout the United States and Canada. Some of the types of facilities tested include:

- Oil and Gas Refineries
- Pulp and Paper Mills
- Gas, Oil, Wood, Coal and Nuclear-Fired Power Plants
- Chemical Plants
- Aluminum Reduction Facilities
- Wood Products Industries
- Smelters
- Cement Kilns
- Hazardous Waste Incinerators
- Municipal and Medical Waste Combustors
- Landfill Gas Flares
- Wastewater and Sewage Treatment Facilities
- Coating and Finishing Facilities
- Manufacturers
- Superfund Clean-Up Sites
- Asphalt Plants

Examples of the types of projects Am Test-Air Quality, Inc. becomes involved in include:

- Regulatory compliance with NSPS, MACT, BIF requirements
- Continuous emission monitoring system (CEMS) certifications
- Enhanced monitor certifications (CEMS, CRMS, CPMS)
- Emission inventories for MACT, NESHAP, SARA Title III and SIP determinations
- Air pollution control equipment warranty/guarantee/design testing
- Combustion or control device optimization
- Air toxics studies for hazardous air pollutants (HAPs)
- Particle size distribution studies (PM₁₀)
- Soil remediation unit evaluations
- Vapor recovery system efficiency evaluations
- Destruction/capture efficiency studies
- Ammonia slip evaluations
- Pilot plant or engineering-scale trial burns
- Hands-on stack sampling training workshops
- Industry-agency liaison

MANUAL EMISSIONS TESTING SERVICES

Am Test-Air Quality, Inc. performs all recognized EPA source test methods for many types of industries. We also perform NIOSH methods, BIF Regulation methods, SW-846 methods, toxic organic (TO) methods and are experienced at developing test methods for specialized applications where published methods do not exist. Am Test personnel keep current on new method development by attending training courses and utilizing the EMTIC bulletin board system (BBS). A partial list of pollutants measured using manual test methods follows:

- Particulate Matter
- Particle Size Distribution
- Sulfur Oxides
- Nitrogen Oxides
- Opacity Measurement
- Carbon Monoxide and Fixed Gases
- Hydrogen Sulfide
- Multi-Metals
 - Hexavalent Chromium
 - ICP Metals
 - Low-Level Mercury
- Particulate and Gaseous Fluoride
- Total Reduced Sulfur Compounds
- Volatile Organic Compounds
 - 8240 list of VOCs
 - Alcohols and Acetates
 - BTEX
 - Formaldehyde/Aldehydes
- Semi-Volatile Organic Compounds
 - Dioxin and Furan Isomers
 - PAHs
 - BNAs
 - PCBs
 - Pesticides and Herbicides
- Acid Gases
 - Hydrochloric Acid
 - Hydrofluoric Acid
 - Nitric Acid
 - Sulfuric Acid
- Chlorine, Chlorine Dioxide and Chloroform
- Ammonia and Urea
- Radionucleides
- Hazardous Air Pollutants (HAPs)

CONTINUOUS EMISSION MONITORING SYSTEMS AND INSTRUMENTAL TESTING SERVICES

Am Test-Air Quality, Inc. has two (2) complete continuous emission monitor (CEM) instrument vans and a separate set of instruments which can be shipped to remote locations. We maintain an inventory of approximately 80 EPA Protocol 1 (or best available grade) certified calibration gas mixtures to tailor calibrations gases to each source. Each mobile laboratory is equipped with:

- O₂, CO₂, SO₂, NO_x, and CO analyzers
- Total hydrocarbon (THC) and non-methane hydrocarbon (NMHC) analyzers
- PC-based data acquisition systems for collecting and reducing data to provide on-site results
- Cylinder racks for calibration gases
- Heated or unheated Teflon sample transport lines
- Sample extraction systems, including sample probes, fine particulate filters, moisture removal systems, and Teflon lined sample pumps
- Laboratory bench space which can be equipped with analytical balances, wet chemical analysis equipment and instrumental methods, including gas chromatography

Am Test-Air Quality, Inc.'s mobile laboratories are used to perform New Source Performance Standards (NSPS) testing and to conduct continuous emission monitoring system (CEMS) performance specification testing and audits. We provide a variety of CEMS services, including:

- Performance Specification Tests (PST)
- Relative Accuracy Test Audits (RATA)
- Relative Accuracy Audits (RAA)
- Cylinder Gas Audits (CGA)
- Quality Assurance Plans (QAP)

In addition, Am Test-Air Quality, Inc.'s senior level staff are extremely knowledgeable in the field of CEMS and can assist in monitor selection, calibration gas selection, stratification testing to locate CEMS probes, sampling system troubleshooting, and act as a liaison with the regulatory agency.

AM TEST LABORATORY

AmTest, Inc. is a full service analytical testing laboratory located in Redmond, Washington. It was organized to provide the highest caliber laboratory testing of environmental and industrial samples. Experienced environmental chemists and microbiologists have the best available equipment at their disposal. Experienced personnel, top-notch instrumentation, and a personal interest in your testing requirements assures:

- Quick turnaround time for sample analysis
- Maximum quality control on projects
- Direct dialogue with the analytical staff
- Reasonable rates

Laboratory is departmentalized into the following disciplines:

**AIR QUALITY
ENVIRONMENTAL
INDUSTRIAL**

OIL AND FUELS

TRACE ORGANICS

MICROBIOLOGY

**Source and Ambient Testing
Water, Wastes, Tissue, Vegetation
Food, Materials Testing, Industrial
Hygiene, Special Projects
Lubricating Oil, Fuel Analysis,
Hydraulics, Contamination Analysis
PCB's, Pesticides, Hazardous Wastes,
Priority Pollutants
Water, Wells, Food, Product Evaluation**

AmTest's laboratory utilizes the following instrumentation:

Finnigan Inco 50 GC/MS
Jarrell Ash ICP Plasma Emission Spectrometer
Gas Chromatographs: FID, EC, TC, N₂, P
Jarrell Ash Arc, Spark Emission Spectrometer
Xertex TOX Analyzer
Source Test Equipment-EPA Approved

Atomic Absorption Spectrometer-Flame
Atomic Absorption Spectrometer-Graphite Furnace
Atomic Absorption Spectrometer-Hydride Generation
UV/Visible Spectrometer
Infrared Spectrometer
High Performance Liquid Chromatograph

AmTest is one laboratory in a group of laboratories serving the West. Other facilities include:

**AmTest, Inc., Portland, Oregon
AmTest of Arizona, Phoenix, Arizona
Can Test Ltd., Vancouver, B.C.
Loring Laboratories, Calgary, Alberta
Metropolitan Clinical Laboratories, Ltd., Vancouver, B.C.**

Phone Mr. Shawn Moore, General Manager or Mr. Mark Fugiel, Technical Director, at (206)885-1664 for more details or a price quotation.