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PET
AP-42
Section 5.13.2
Reference Number

6

A 100-year start on tomorrow

November 7, 1980

Mr. Asdakorn Limpiti
Energy and Environmental Analysis, Inc.
2607A Carver Street
Durham, North Carolina 27705

Dear Mr. Limpiti:

Subject: Submission of Completed Questionnaire in Connection with
Development of a New Source Performance Standard for the
Polymers and Resins Industry

Enclosed is the completed questionnaire submitted in response to Mr. Jack Farmer's August 25, 1980 notification letter. This completed questionnaire contains requested information about the polymer manufacturing facility at Tennessee Eastman Company. We understand that EPA will use this information in the development of a New Source Performance Standard for the polymers and resins industry.

The information designated as "Confidential" that we are disclosing to Energy and Environmental Analysis, Inc. (EEA) is considered by Tennessee Eastman Company to be proprietary information and, as such, is entitled to protection from unauthorized disclosure in accordance with the non-disclosure agreement regarding EPA Contract No. 68-02-3061, which was executed by EEA and Tennessee Eastman Company on October 2, 1980.

If you have any questions concerning this information, please contact Mr. J. C. Edwards, (615) 246-2111, Extension 2444.

Yours very truly,



Robert E. Lee
Director, Manufacturing Staff

drp

Enclosure

cc: Mr. Edwin Vincent
U. S. Environmental Protection Agency
Research Triangle Park, North Carolina 27711

ENCLOSURE 4

PLANT SURVEY QUESTIONNAIRE

Polymer Name: Poly(ethylene terephthalate)

Parent Corporation Name: Eastman Kodak Company

Subsidiary Name: Tennessee Eastman Company

Mailing Address: P. O. Box 511

Kingsport, Tennessee 37660

Plant Name: Tennessee Eastman Company

Physical Location: Lincoln Street

(including county and
air quality control region.) Sullivan County; Tennessee 207

Person EPA should contact regarding information supplied in this questionnaire.

Name: J. C. Edwards

Title: Manager, Clean Environment Program

Mailing Address: P. O. Box 511

Kingsport, TN

Telephone Number: ----- (615)246-2111, Ext. 2444

Date Questionnaire Completed: September 24, 1980

DOCKET NO.

Category II-B

The following information is located in the confidential files of the Director, Emission Standards and Engineering Division, Office of Air Quality Planning and Standards, U.S. Environmental Protection Agency, Research Triangle Park, North Carolina. This information is confidential, pending final determination by the Administrator, and is not available for public inspection.

Page 2. Capacity information claimed confidential.

II. Process

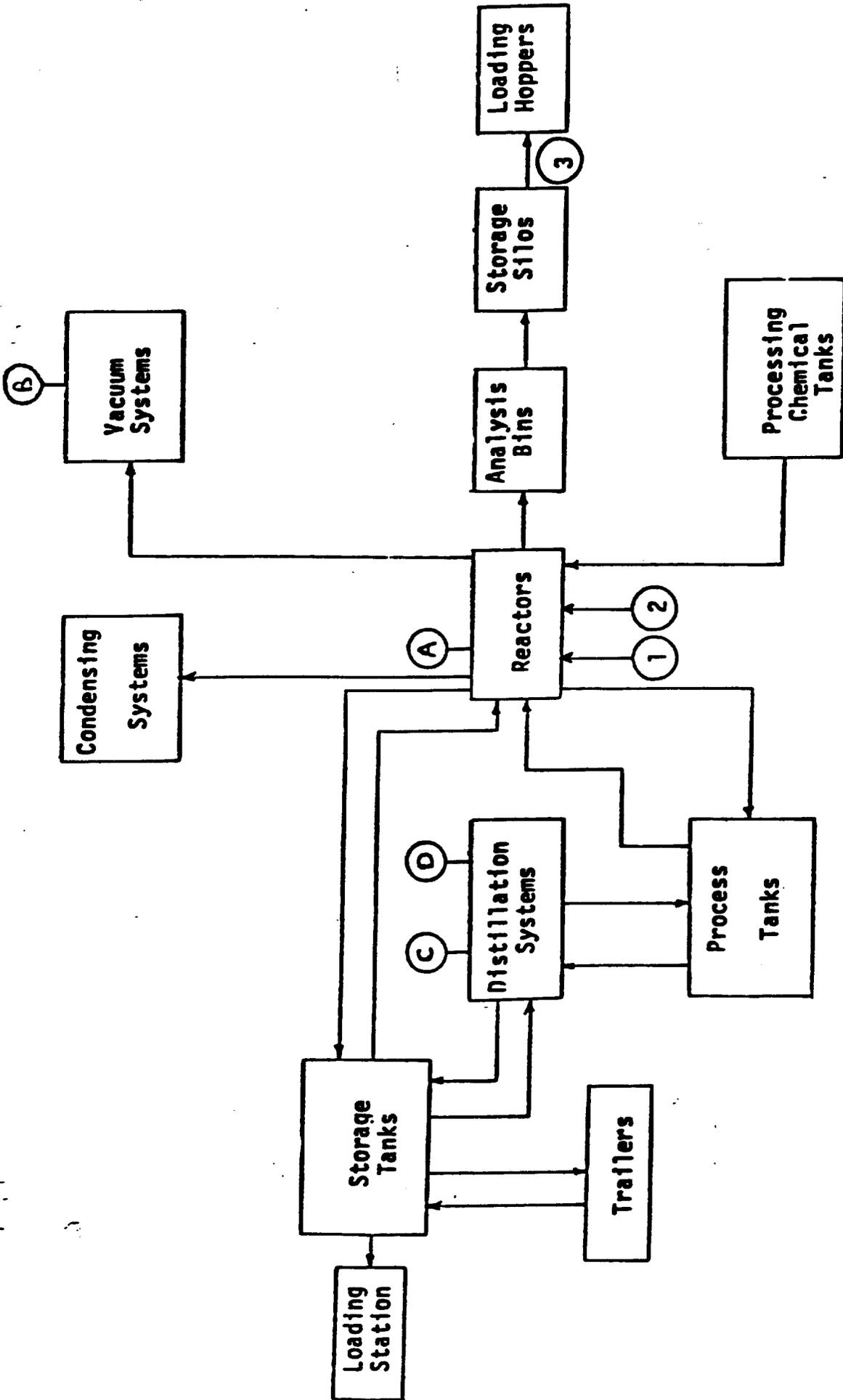
1. Process Name and Type:

Continuous Melt Phase Polymerization

2. Flow Sheet: (overall process block diagram)

(See Attachment A) Page 4

ATTACHMENT A



II. Process: (continued)

3. Raw Materials and Products.

Raw Materials

<u>Name</u>	<u>Quantity</u>	<u>Composition</u>	wt percent <u>X</u> or vol percent <u> </u>
Dimethyl Terephthalate	379.6M lbs.	99 + %	
Ethylene Glycol	245.0M lbs.	99 + %	

By-product

<u>Name</u>	<u>Quantity</u>	<u>Composition</u>	wt percent <u>X</u> or vol percent <u> </u>
Methanol	125.3	MeOH 91.0 Misc. Low Boilers 5.0 Water 4.0	

4. Number of:

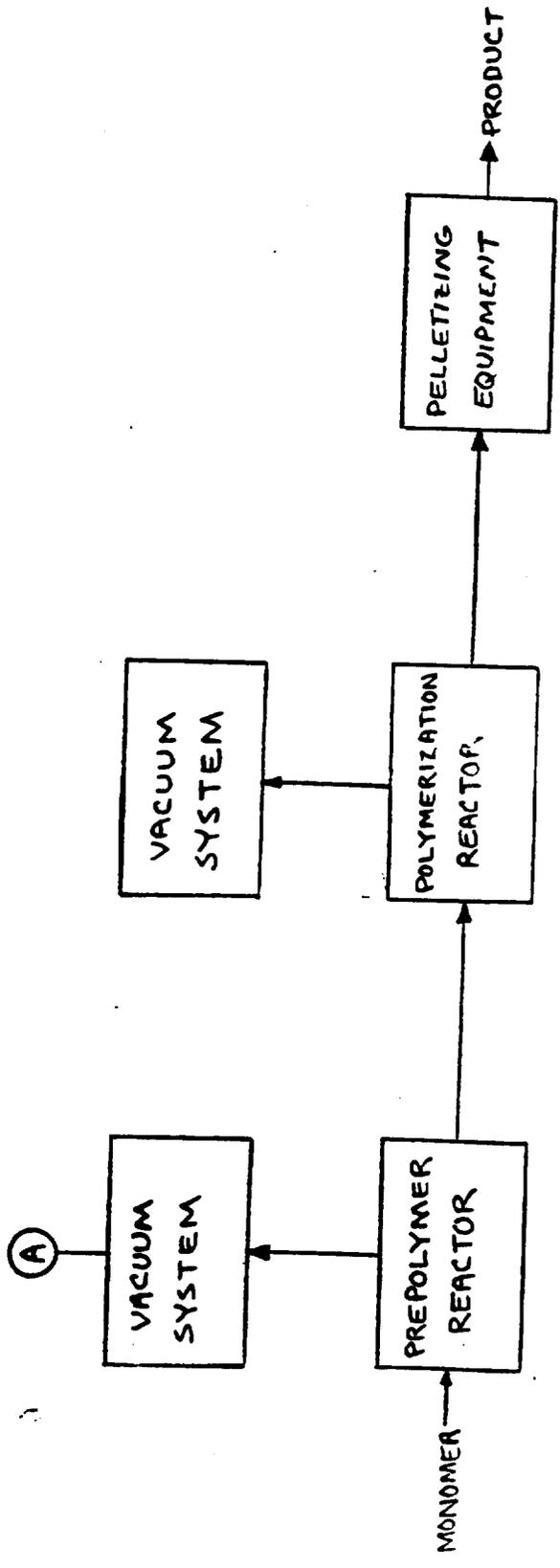
Pumps 100
Compressors 0
Valves 1,500
Pressure Relief Devices 270
Cooling Tower 1
Agitators 32
Open-Ended Valves or Line 250
Sampling Connections 100
Flanges 4,000

Which Handle Volatile Organic Compound (VOC)

II. Process (continued)

- 5. Polymerization Section Process Flow Sheet: (For example; polymerization reactor, vent gas compressors, and any other polymerization unit operations.)**

(See Attachment B) Page 7

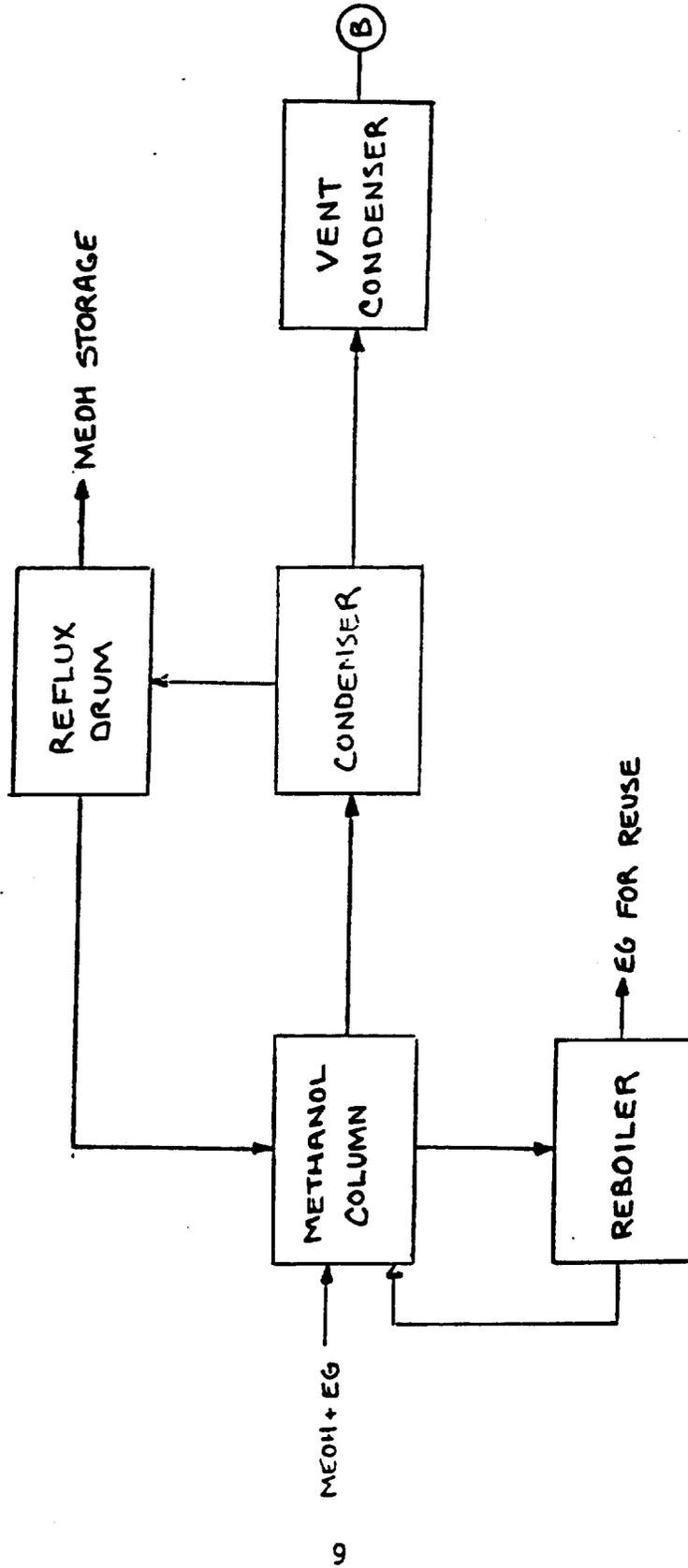


ATTACHMENT B
POLYMERIZATION REACTOR SECTION

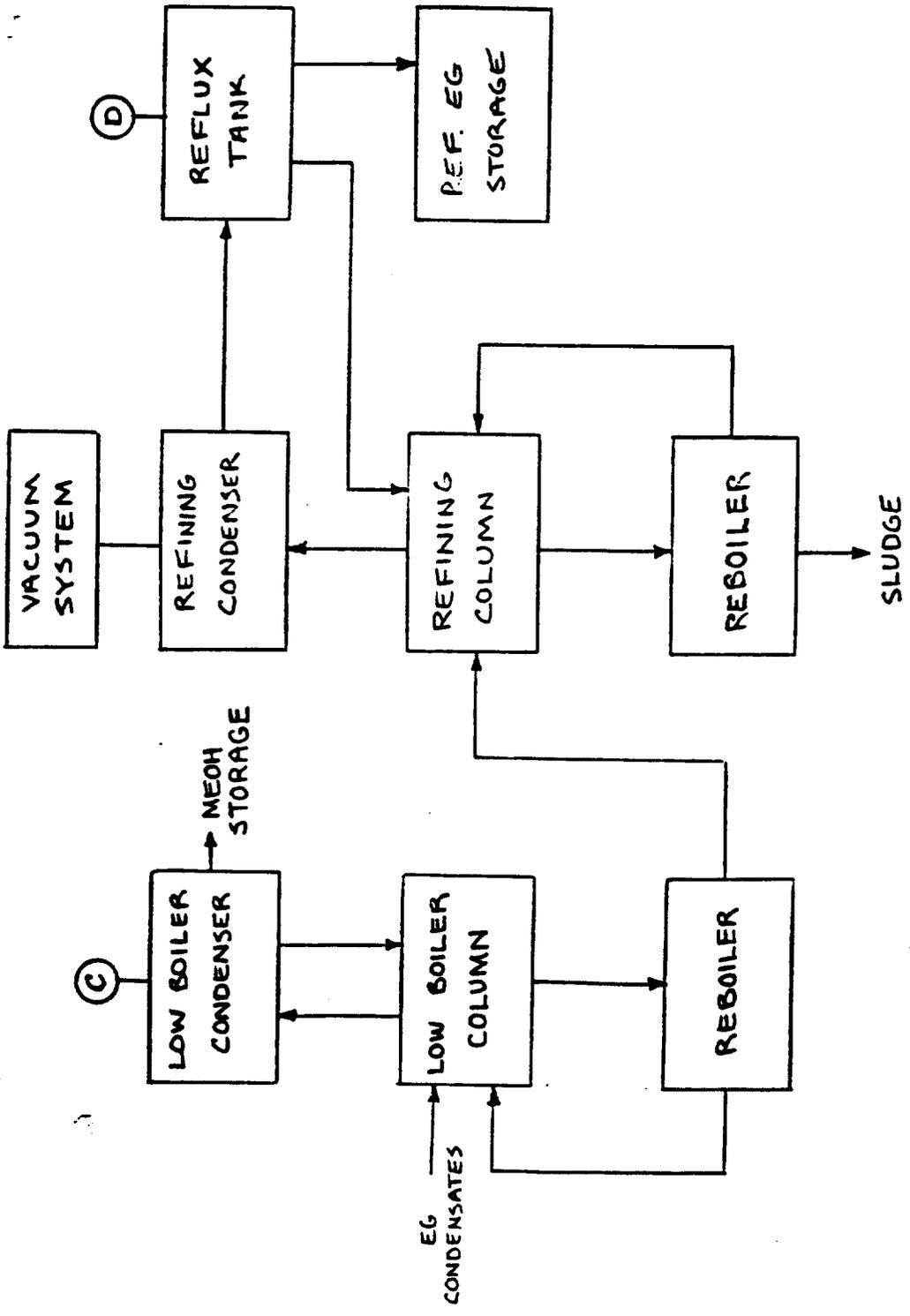
II. Process (continued)

- 6. Material Recovery/Separation Section Flow Sheet: (i.e., Process deactivation/decanting, diluent recovery, solvent recovery, polymer drying, catalyst drying, devolatilizing and other similar unit operations.)**

(See Attachments C - Page 9, D - Page 10)



ATTACHMENT C
METHANOL RECOVERY



III. Emissions (Composition and Flow) for Polymerization Section:

Stream Name: Prepolymer Vacuum Discharge - Attachment B

The stream flow shown on Polymerization section flow sheet by number/letter: A

Check One: basis on 100% capacity X; basis on production rate ; (specified)

1) flow 50 SCFM; N₂ temperature 77 °F; pressure Atm. psig
 ACFM Check One: 1b/hr X; wt% ; vol% (For composition and composition range)
 lb/hr

<u>Component Name</u>	<u>Formula^{a)}</u>	<u>State</u>	<u>Average Amount or Composition</u>	<u>Composition Range</u>
Nitrogen	N ₂	Gas	220 lbs/hr.	0-440 lbs/hr.
Methanol	CH ₃ OH	Vapor	0.0091 lbs/hr.	0-0.1 lbs/hr.

a) Give approximate molecular weight if the exact formula is not specified.

*Volatile organic compound (VOC) should be described as fully as possible

III. (continued)

For stream flow shown on Polymerization section flow sheet by letter/
number A stream name Prepolymer Vacuum Discharge.

2. Composition Variation

Nitrogen

The nitrogen discharged from this vent is used to control pressure in the Prepolymer Reactor. Usage will vary with vacuum operation.

Methanol

The amount of methanol discharged from this vent will depend on the operation of the Prepolymer Reactor and vacuum system. Normally this variation will not be more than $\pm 10\%$ of the average amount shown.

3. Method Used to Determine Composition and Flow

 X ENGINEERING ESTIMATES
 DESIGN CALCUALTIONS
 ANALYTICAL MEASUREMENTS

III. (continued)

For stream flow shown on Polymerization section flow sheet by letter/number A stream name Prepolymer Vacuum Discharge .

4. Sample Procedure

The emissions from this vent have been determined by Engineering calculations and material balances. This vent discharges to the atmosphere through a Barometric Seal leg and has not been sampled in the past.

5. Analytical Procedure

Not applicable.

6. Sampling Frequency

Not applicable.

7. Confidence Level

Not applicable.

8. Ease of Sampling

A step ladder would be required to obtain access to the top of the barometric leg.

111. Emissions (Composition and Flow) for Material Recovery/Separation Section:

Stream Name: Methanol Column Condenser - Attachment C

The stream flow shown on Material Recovery/Separation flow sheet by number/letter B

Check One: basis on 100% capacity X; basis on production rate _____; (specified _____)

1) flow 1.2 SCFM; N₂ temperature 100 °F; pressure _____ Atm. _____ psig
_____ ACFM Check One: 1b/hr X; wt% _____; vol% _____ (for composition and composition range)
_____ lb/hr

Component Name	Formula	a)	State	Average Amount or Composition	Composition Range
Nitrogen	N ₂		Gas	8.7 lb/hr.	0-8.7 lb/hr.
Methanol	CH ₃ OH		Vapor	8.2 lb/hr.	0-82.0 lb/hr.

a) Give approximate molecular weight if the exact formula is not specified.

*Volatile organic compound (VOC) should be described as fully as possible _____

III. (continued)

For stream flow shown on Material Recovery/Separation section flow sheet by letter/number B stream name Methanol Column Condenser .

2. Composition Variation:

The Nitrogen emission from this vent is from a rotameter and should remain constant.

The amount of Methanol discharged from this vent will remain constant except during upset conditions at which time the maximum value listed could be reached.

3. Method Used to Determine Composition and Flow:

- X ENGINEERING ESTIMATES
- DESIGN CALCULATIONS
- ANALYTICAL MEASUREMENTS

III. (continued)

For stream flow shown on Material Recovery/Separation reaction flow sheet by letter/number B stream name Methanol Column Condensate

4. Sampling Procedure:

This vent has not been sampled. Emissions have been determined by Engineering calculation.

5. Analytical Procedure:

Not applicable.

6. Sampling Frequency:

Not applicable.

7. Confidence Level:

Not applicable.

8. Ease of Sampling:

Readily accessible.

111. Emissions (Composition and Flow) for Material Recovery/Separation Section:

Stream Name: Ethylene Glycol Recovery - Low Boiler Column - Attachment D

The stream flow shown on Material Recovery/Separation flow sheet by number/letter C.

Check One: basis on 100% capacity X; basis on production rate _____; (specified _____)

1) flow _____ SCFM; temperature 77 °F; pressure 0.2 psig
 _____ ACFM Check One: 1b/hr X; wt% _____; vol% _____ (for composition and composition range)
X _____ lb/hr (see below)

Component Name	Formula ^{a)}	State	Average Amount or Composition	Composition Range
Nitrogen	N ₂	Gas	5.62 lb/hr.	0-5.62 lb/hr.
2-Methyl-1,3 Dioxolane	CH ₃ CH(OCH ₂) ₂	Vapor	0.66 lb/hr.	0-.73 lb/hr.
Methanol	CH ₃ OH	Vapor	0.36 lb/hr.	0-.40 lb/hr.
Water	H ₂ O	Vapor	0.04 lb/hr.	0-.05 lb/hr.

AVG MW =

a) Give approximate molecular weight if the exact formula is not specified.

*Volatile organic compound (VOC) should be described as fully as possible _____

III. (continued)

For stream flow shown on Material Recovery/Separation section flow sheet by letter/number C stream name Low Boiler Column Condenser.

2. Composition Variation:

Nitrogen

Nitrogen is purged to the column vent through a rotameter. Flow variations should be very small.

Low Boilers

The low boiler composition is dependent upon the operating condition of the column. The composition should not vary more than 10% from the average values given.

3. Method Used to Determine Composition and Flow:

X ENGINEERING ESTIMATES

DESIGN CALCULATIONS

ANALYTICAL MEASUREMENTS

III. (continued)

For stream flow shown on Material Recovery/Separation reaction
flow sheet by letter/number C stream name Low Boiler Condenser

4. Sampling Procedure:

This vent is not routinely sampled. Emissions are based on
Engineering calculations.

5. Analytical Procedure:

Not applicable.

6. Sampling Frequency:

Not applicable.

7. Confidence Level:

Not applicable.

8. Ease of Sampling:

Step ladder required.

111. Emissions (Composition and Flow) for Material Recovery/Separation Section:

Stream Name: Ethylene Glycol Recovery - Refining Column - Attachment D

The stream flow shown on Material Recovery/Separation flow sheet by number/letter D.

Check One: basis on 100% capacity X; basis on production rate _____; (specified _____)

1) flow _____ SCFM; temperature 77 °F; pressure Atm. psig

_____ ACFM Check One: lb/hr X; wt% _____; vol% _____ (for composition and composition range)
X lb/hr (See below)

<u>Component Name</u>	<u>Formula</u>	<u>a)</u>	<u>State</u>	<u>Average Amount or Composition</u>	<u>Composition Range</u>
Nitrogen	N ₂		Gas	306 lb/hr.	0-306 lb/hr.

a) Give approximate molecular weight if the exact formula is not specified.

*Volatile organic compound (VOC) should be described as fully as possible _____

III. (continued)

For stream flow shown on Material Recovery/Separation reaction
flow sheet by letter/number D stream name Refining Co]umn

4. Sampling Procedure:

This vent is not routinely sampled.

5. Analytical Procedure:

Not applicable.

6. Sampling Frequency:

Not applicable.

7. Confidence Level:

Not applicable.

8. Ease of Sampling:

Step ladder required.

IV. Emission control device

for device shown on block diagram by number _____

Device name _____

1. Engineering description.

No emission control devices used; hence, none registered for this process.

IV. (continued)

For device shown on block diagram by number _____

device name _____

2. Capital Cost of Emission Control System:

a) Capital Cost

Major equipment cost: \$ _____ each

Total installed cost: \$ _____ each

Year:

Cost:

IV. (continued)

For device shown on block diagram by number _____

b) Check List. Mark whether items listed are included in total cost included in IV.2.a Do not give dollar value.

Yes	No	Cost	Explanation
			Site development
			Buildings
			Laboratory equipment
			Stack
			Rigging etc.
			Piping
			Insulation
			Instruments
			Instrument panels
			Electrical
			Facilities outside battery limits*
			Storage tanks, spheres drums, bins, silos
			Catalyst
			Spare parts and non-installed parts

*Such as - process pipe lines such as steam, condensate, water, gas, fuel, air, fire, instrument and electric lines.