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PES

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**Source Test for Measurement of
Oxides of Nitrogen, Carbon Monoxide
and VOC from Boiler Exhaust at**

**Candlewick Yarns
711 Cinnamon Drive
Lemoore, California 93245**

Prepared for:

**San Joaquin Valley Unified
Air Pollution Control District
1999 Tuolumne
Suite 200
Fresno, California 93721**

Attention: Mr. Gary Martin

Prepared By:

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*Reviewed by
Gabe*

**PES Job Number: 4778.001
April 21, 1993**

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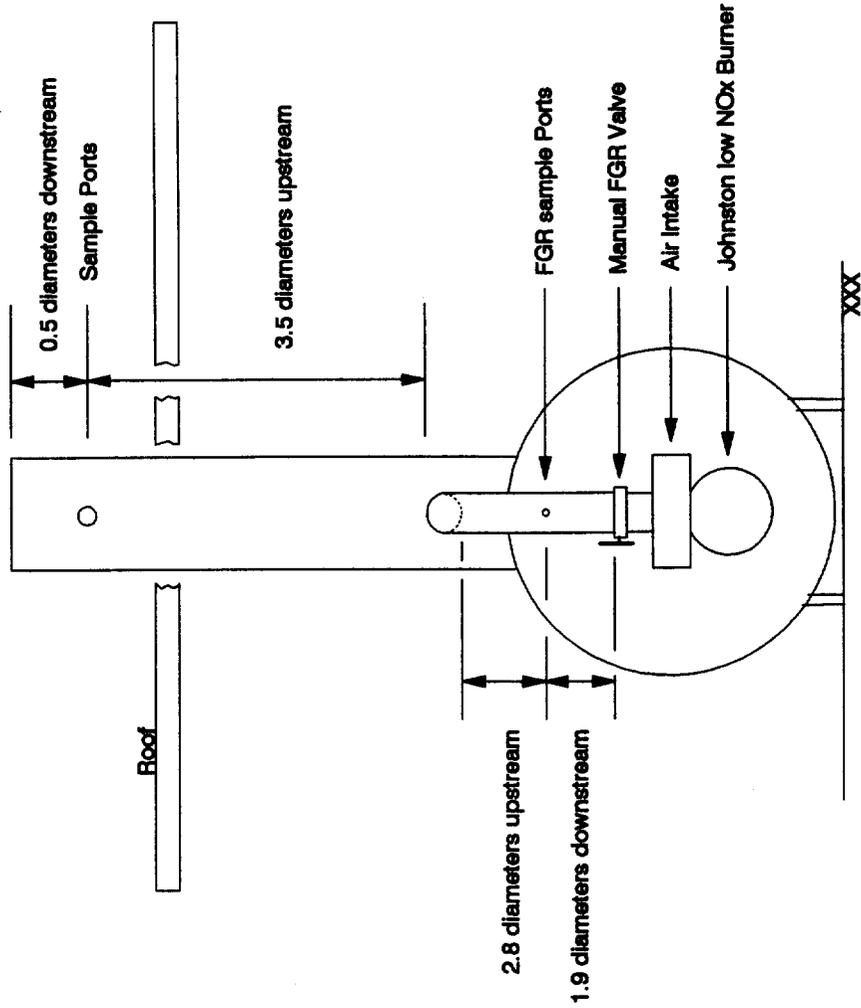
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INTRODUCTION

Pacific Environmental Services, Inc (PES) was retained by Central Boiler Inc, to measure CO, and NOx emissions from a 21 MMBTU/HR boiler located at Candlewick Yarns, 711 Cinnamon Drive, Lemoore, California 93245. The tests were conducted to determine compliance with SJVUAPCD Authority to Construct No. C-233-1-0.

The object of the source test was to determine **NO_x, CO and VOC** emissions under maximum load conditions with the boiler operating on natural gas only. Instruments were used to determine the concentrations of oxides of nitrogen, oxygen, carbon monoxide and ROG on a continuous basis. PES conducted the test on April 21, 1993. The tests were performed by Steve Hernandez and Robert Nguyen of PES. Mr. M. Dean High, Senior Vice President provided guidance for planning and supervision purposes.

The unit tested was a Johnston Model PTFA-500 fire tube boiler with a Johnston Low ~~Ma~~ burner assembly rated at 21.0 MMBTU/HR for natural gas. The boiler was equipped with a manual flue gas recirculation (FGR) system (see Figure 1). The Authority to Construct for this equipment can be found in Appendix A.



Candlewick Yarns 21.0 MMBTU/HR Fire Tube Boiler Layout

Figure 1

Process Description

Candlewick Yarns operated two mills which receive bailed nylon staple which is blended to produce the desired color and texture combinations. It is then directed to a cording process which evens the staple and brings it to a desired weight.

the staple is then spun into yarns and sent to ply twisting where the desired thickness is gained. it is then reeled into skeins and sent to heat setting where the yarn is heated to approximately 270°F to gain a perminant (controlled) twist.

The skeins are then wound onto cones and sent to packaging in preparation for shippment to customers.

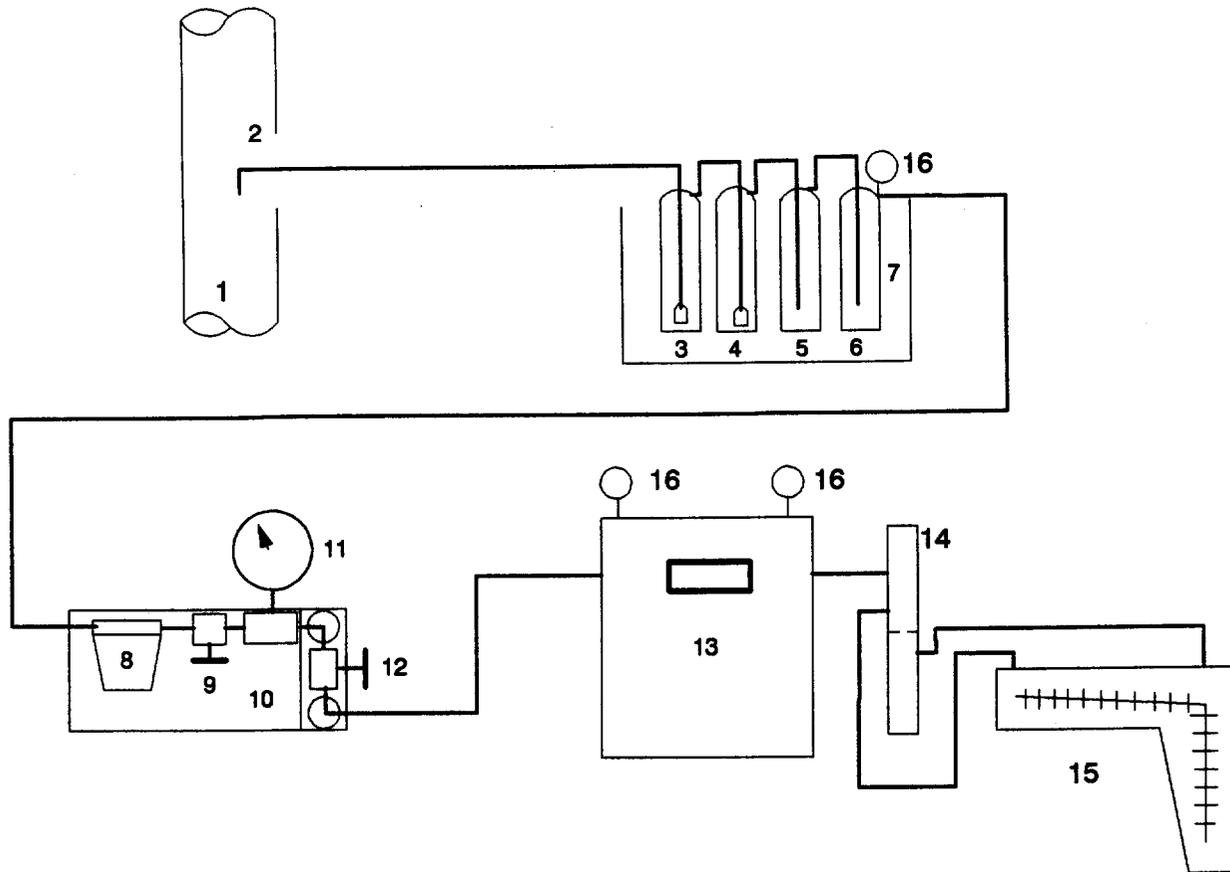
TEST METHODOLOGY

The sampling ports are shown in Figure 1. Since the test location was 0.5 diameters upstream and 0.5 diameters downstream of any disturbance, a total of 16 traverse points were measured (CARB Method 1). At each point the velocity head was measured using an S-type pitot tube attached to an inclined manometer. Temperature was measured with a chromel-alumel (type K) thermocouple (CARB Method 2). Temperature was measured with a Fluke Model 52 Digital thermometer (DTM). Moisture was determined using CARB Method 4 at each load condition tested (Figure 2). A stratification test was performed using the oxygen analyzer prior to the start of testing and all points were found within 0.3% oxygen.

For each test condition, CARB Method 100 was used to determine the oxides of nitrogen (NO_x), carbon monoxide (CO), carbon dioxide (CO₂), oxygen (O₂), and THC levels in the effluent gas from the boiler. The concentrations were measured by using a Continuous Emissions Monitoring System (CEMS) installed in a PES mobil monitoring van. A schematic of this system is shown in Figure 3.

A Rosemount Analytical (formerly Beckman Industrial) model 880 non-dispersive infrared analyzer was used to determine the carbon monoxide concentration. A Rosemount Analytical Model 880 non-dispersive Infrared Analyzer was used to determine the CO₂ concentration. A Rosemount Analytical Model 755R Paramagnetic Analyzer was used to determine the oxygen concentration. A Thermo Electron Model 10 Chemiluminescent Analyzer was used to determine the NO_x concentration. A J.U.M. Engineering Model VE-7 Heated THC Analyzer was used to determine ROG. And CARB Method 25 using a Vacuum Bottle (Figure 4) was used for methane determination which was subtracted from the THC for VOC determination. Specifications for each analyzer are shown in Table 1. The output of the analyzers was linearized by the manufacturers.

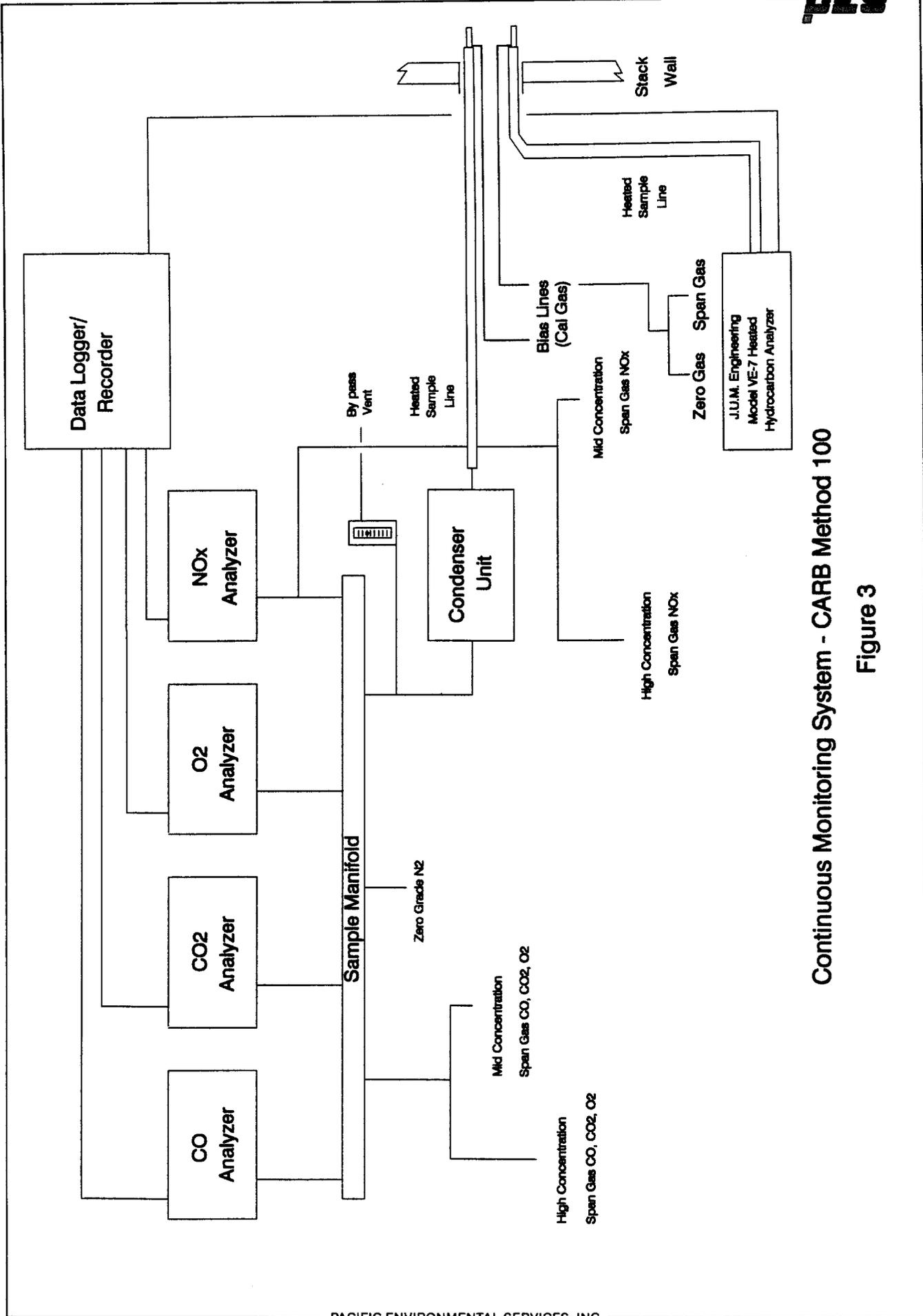
The monitoring train consisted of a 3/8-inch stainless steel sampling probe, a 3/8-inch heated teflon sampling line, a sample refrigeration system (operated at 40°F), a glass fiber filter in a 47 mm Teflon holder, a diaphragm vacuum pump, and a sample distribution manifold. The distribution manifold was equipped with a series of 5-way valves with flow meters (rotometer style). One flow meter acted as a bypass, and the others were connected to the individual analyzers. The output of each analyzer was logged 60 times per minute with a Yokogawa Model 2400 multi-channel strip chart recorder. The recorder monitored the output of the CO, CO₂, O₂, NO_x and stack temperature on channels scaled specifically for each component.



- | | |
|----------------------------------|-----------------------------|
| 1. Stack | 9. Metering Valve |
| 2. Probe | 10. Sealed Pump (Leak Free) |
| 3. Impinger with 100 mL DI water | 11. Vacuum Gage |
| 4. Impinger with 100 mL DI water | 12. Bypass Valve |
| 5. Empty Impinger | 13. Dry Gas Meter |
| 6. Impinger With Silica Gel | 14. Orifice Plate |
| 7. Ice Bath | 15. Inclined Manometer |
| 8. Filter for Pump | 16. Temperature Indicator |

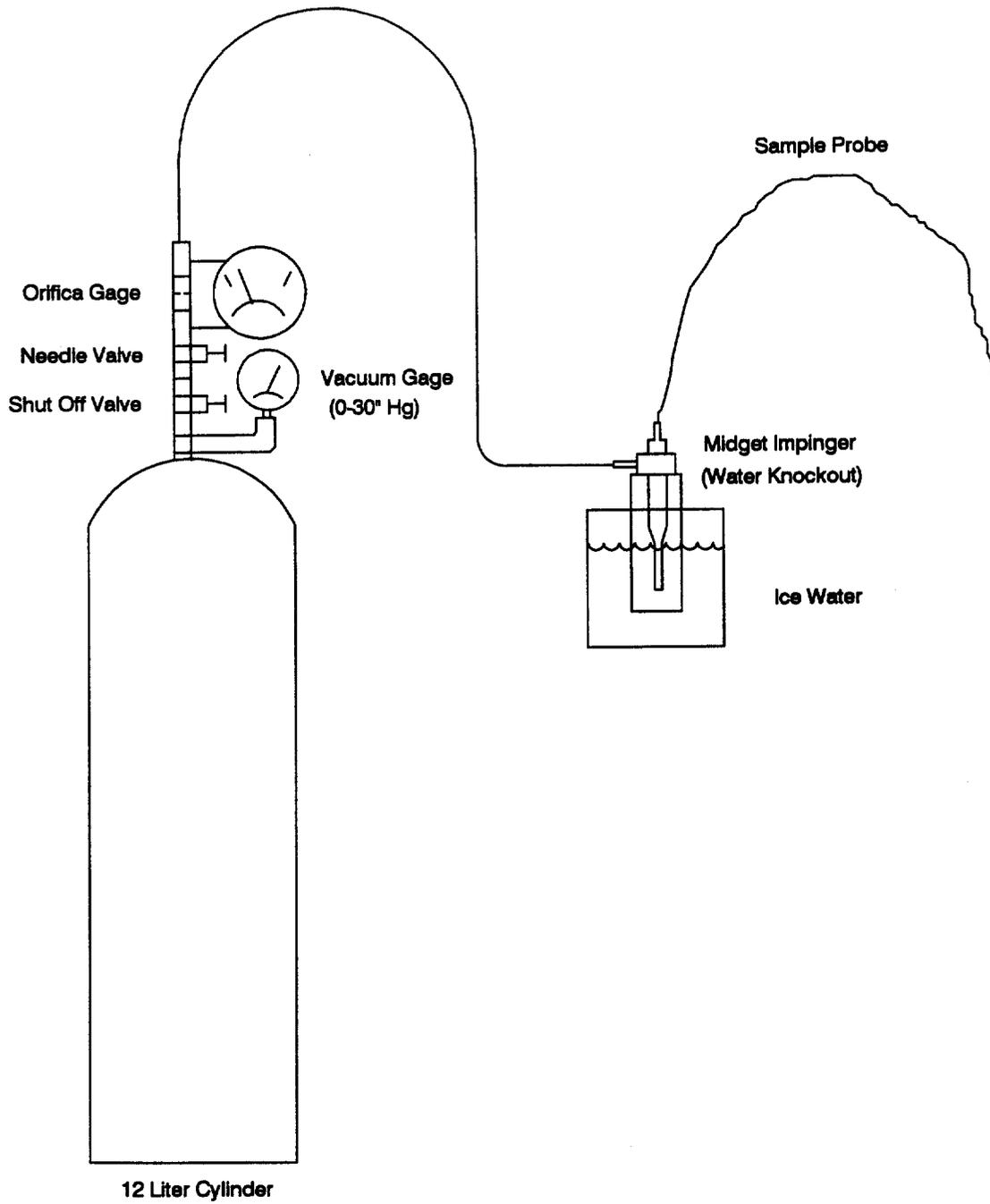
CARB Method 4 Sampling train

Figure 2



Continuous Monitoring System - CARB Method 100

Figure 3



CARB Modified Method 25 Sample Train

Figure 4

Table No. 1 Continuous Monitor Specifications**NO_x Chemiluminescent Analyzer - Thermo-Electron Model 10A**

Response Time	1.5 sec - NO, 1.7 sec - NO _x
Zero Drift	± 0.5% after warm up (30 min)
Span Drift	± 1% of full scale
Linearity	± 1% of full scale
Accuracy	Derived from the calibration NO/NO _x ± 1% gas was used.
Output	NO 0-5.0 Vdc (scaled 0-50 ppm) NO _x 0-0.5 Vdc (scaled 0-50 ppm)

O₂ Paramagnetic Analyzer - Beckman Model 755R

Response Time	2 Sec
Zero Drift	± 1% of full scale
Span Drift	± 1% of full scale
Linearity	± 1% of full scale
Accuracy	Derived from the calibration O ₂ ± 1% gas was used.
Output	0-1.0 Vdc (scaled 0-25%)

CO/CO₂ Infrared Analyzers - Beckman Model 880

Response Time	2 sec.
Zero Drift	± 1% of full scale
Span Drift	± 1% of full scale
Linearity	± 1% of full scale
Accuracy	Derived from the calibration CO ± 1% gas was used.

Table 1 Continuous Monitor Specifications Cont'd**CO/CO2 Infrared Analyzers - Beckman Model 880 Cont'd**

Output 0-1.0 Vdc (scaled 0-100 ppm CO)
0-1.0 Vdc (scaled 0-20 % CO₂)

ROG Heated Hydrocarbon Analyzer - J.U.M. Engineering Model VE-7 FID Analyzer

Response Time 0-95% in less than 1.2 Seconds

Zero Drift ± 1% of Full Scale in 24 Hrs

Span Drift ± 1% of Full scale in 24 Hrs

Linearity ± 1% of Full Scale

Accuracy Derived from the Calibration Gas
± 1% gas was used

Sensitivity 1 ppb

Range Change Consistency Less than 1% Full Scale

Oxygen Synergism Less than 2%

Output 0-10.0 Vdc Scaled:

R ₁	0-10	ppm as C ₁
R ₂	0-100	ppm as C ₁
R ₃	0-1,000	ppm as C ₁
R ₄	0-10,000	ppm as C ₁
R ₅	0-100,000	ppm as C ₁

Sample Flow Rate 3 Liters/Minute

Prior to the source tests, the suction side of the monitoring system was leak-checked at 20.5" Hg vacuum with no loss in 5 minutes, and the sampling bias of the system was determined by introducing CO, CO₂, O₂, and NOx span gas blends at the tip of the sampling line. A comparison of the analyzer responses was made between the span gas introduced at the sample line tip and the span gas introduced directly to the analyzers to ensure a differential of less than 5%. Since all analyzers were left on line during this procedure, a cross interference check was accomplished at the same time. The analyzers were spanned at a point between 20% and 90% of full scale before and after each test run NBS traceable calibration gases, and with a zero grade nitrogen. Table 2 lists all the gases used and Appendix C contains copies of the certifications.

Readings from the continuous monitors were permanently recorded on the multichannel recorder and a back up data logger (Rustrak Ranger II) was used. Average gas concentrations were obtained by numeric integration of the area under the curves on the stripchart using time intervals of two minutes. The average strip chart readings were then corrected using the averages of the initial and the final zero and span calibrations. The average zero and span calibrations were determined using the following equations:

$$Z_a = \frac{Z_f - Z_i}{2} + Z_i$$

$$S_a = \frac{S_f - S_i}{2} + S_i$$

Where:

Z_a, S_a = average zero or span strip chart divisions.

Z_f, S_f = final zero or span strip chart divisions.

Z_i, S_i = initial zero or span strip chart divisions.

Table No. 2 Calibration Gases

Gas Composition	Use:	Cylinder Serial No.	Certified Accuracy	Analysis Date
Nitrogen	Zero Gas	ALMO 27878	Zero Grade	N/A
11.0% O ₂ 50.0 ppm CO 11.0% CO ₂ Bal N ₂	Span Gas	IL 2572	± 1%	4-08-93
19.1% O ₂ 75.5 ppm CO 18.0% CO ₂ Bal N ₂	Span Gas	AAL 13428	± 1%	6-02-92 5-17-93*
22.5 ppm NO 22.8 ppm NOx Bal N ₂	Span Gas	ALMO 27851	± 1%	3-19-93 3-94 exp
40.1 ppm NO 40.7 ppm NOx Bal N ₂	Span Gas	AAL 18076	± 1%	4-24-93
80.0 ppm CH ₄ Bal Air	Span Gas	ALMO 14380	± 1%	2-14-92
7.99 ppm CH ₄ Bal Air	Span Gas	AAL 7535	± 1%	9-11-92

* Recertified Gas

The following equation was then used to determine corrected gas concentrations:

$$C_{gas} = (C_m - Z_a) \times \frac{C_s}{S_a - Z_a}$$

Where:

C_m = measured gas concentration from strip chart.

C_{gas} = effluent gas concentration.

C_s = span concentration.

The following calculation was used to correct to 3% O_2 :

$$C_{3\%O_2} = \frac{20.95 - 3.0}{20.95 - C_{gasO_2}}$$

Where:

$C_{3\%O_2}$ = gas concentration corrected to 3% O_2 .

C_{gasO_2} = measured O_2 concentration.

TEST RESULTS

The 3 thirty minute tests were conducted with the boiler using natural gas and running at high fire.

The results of each test are presented with all relative data with the monitored and corrected concentrations. Table 3 details the results for natural gas.

Table No. 3 natural gas

FGR:	50% open	541 DSCFM (FGR)
FGR _{Return} :	16%	3,380 DSCFM (EXH.)
Condition:	95% of load capacity	19.8 MMBTU/HR
Fuel:	natural gas	18,334 cfh

RUN 1:

Pollutant	Units	Measured	Corrected (to 3% O ₂)	Lbs/Hr	Allowed (lbs/hr)
NOx	ppm	27.5	25.9	0.66	0.92
CO	ppm	6.9	6.4	0.10	0.73
VOC	ppm	0.03	0.001	0.0002	0.06
O ₂	%	1.8			

RUN 2:

Pollutant	Units	Measured	Corrected (to 3% O ₂)	Lbs/Hr	Allowed (lbs/hr)
NOx	ppm	28.5	26.8	0.69	0.92
CO	ppm	11.8	11.1	0.17	0.73
VOC	ppm	0.4	0.4	0.0025	0.06
O ₂	%	1.8			

RUN 3:

pollutant	Units	Measured	Corrected (to 3% O ₂)	Lbs/Hr	Allowed (lbs/hr)
NOx	ppm	28.9	27.2	0.70	0.92
CO	ppm	4.1	3.9	0.06	0.73
VOC	ppm	1.5	1.4	0.0095	0.06
O ₂	%	1.9			

Averages

pollutant	Units	Lbs/Hr	Allowed (lbs/hr)
NOx	ppm	0.68	0.92
CO	ppm	0.11	0.73
VOC	ppm	0.0041	0.06
O ₂	%	1.9	

QUALITY ASSURANCE/QUALITY CONTROL

Source tests are performed to determine the types and amounts of pollutants emitted by a source. Information from this source test program may be used for obtaining permits, evaluating control equipment performance, updating emission inventories, and determining compliance with present and future emission regulations. For these purposes, reliable data are required. PES provides this reliability by using the following work practices:

Sample Handling and Conditioning

Quality control procedures used during this test included the use of non-reactive 316 stainless steel or teflon tubing and fittings throughout the system. Pre-test and post-test leak checks were made pulling a vacuum of 19.5" Hg at the beginning and 20" at the end of the testing day. Bias checks were made and found to be within tolerance. Zero and span drift errors were minimized by using average zero and span values to correct the data. A refrigeration unit was used with the pump down stream of the conditioned sample gas. All instrumentation was continuously monitored and checked to insure data reliability during all sample runs.

Calibration Gases

All calibration gases used were $\pm 1\%$ accuracy and provided by Scott Specialty Gases in San Bernardino, Ca. All instruments were run for 4 hours prior to testing to insure proper operating order. Certifications for the gases are located in Appendix C.

Use Of Standard Test Procedures

CARB Methods 1 and 2 were utilized to determine the flow rate and CARB Method 100 was used to determine the NO_x , CO, O_2 , and VOC emission rates. CARB Modified Method 25 was used for methane determination and subtracted from the THC concentration measured. And CARB Method 4 was used to determine moisture.

A procedure must be thoroughly studied under various conditions in order to be designated as a CARB Method. Results of many executions of the procedure are compared to demonstrate accuracy and repeatability before adoption of the procedure as a source testing method.

Use Of Trained Test Personnel

Because of the complexity of typical source testing methods, testers should be trained and experienced with the test procedures in order to assure reliable results. PES personnel have had professional training and routinely conduct source tests.

Knowledge Of Source's Operation

The source testing team should have sufficient knowledge of the process to be tested in order to properly document the process parameters during the tests. Without documentation of the process parameters used, results are much less meaningful. PES has experience with boiler operations and is familiar with the processes.

Equipment Maintenance and Calibration

Use of properly maintained and calibrated test equipment is essential for minimizing systematic errors in results. All sampling devices are constructed, maintained, and calibrated as suggested in the CARB Source Testing Manual.

Thorough Record Keeping

All data relating to the operation of the sampling train was immediately recorded to ensure that it was not lost or misinterpreted. Any unusual occurrences in the process operation, unusual test instrument readings, or any other items that could affect the test results were also noted.

Use Of Standardized Data Reduction Techniques

Data reduction was accomplished by the use of step by step calculation sheets which are included in Appendix B.

Appendix A



San Joaquin Valley
Unified Air Pollution Control District

AUTHORITY TO CONSTRUCT

PERMIT NO: C-233-1-0

ISSUANCE DATE: 10/29/92

LEGAL OWNER OR OPERATOR: CANDLEWICK YARNS
MAILING ADDRESS: P.O. BOX 70
LEMOORE, CA 93245

*Pay #
233-2057*

LOCATION: 711 CINNAMON DRIVE, LEMOORE

EQUIPMENT DESCRIPTION:

JOHNSON MODEL PTFA-500 BOILER WITH ONE INDUSTRIAL COMBUSTION ZIMMERTU MODEL LDNG-252P LOW NO
NATURAL GAS FIRED BURNER WITH FLUE GAS RECIRCULATION.

CONDITIONS

- 1 - No air contaminant shall be discharged into the atmosphere for a period or periods aggregating more than 3 minutes in any one hour which is as dark as or darker than Ringelmann 1 or equivalent 20% opacity.
- 2 - The boiler shall be fired on natural gas only.
- 3 - Natural Gas consumption shall not exceed 504,000 SCF per day.
- 4 - The flue gas recirculation system shall be operated in accordance with the manufacturer's directions whenever the boiler is operating.
- 5 - The concentration of nitrogen oxides in the exhaust shall not exceed 30 ppmv corrected to 3% oxygen averaged over any one hour period.

(CONDITIONS CONTINUE ON THE NEXT PAGE)

This is NOT a PERMIT TO OPERATE. Approval or denial of a PERMIT TO OPERATE will be made after an inspection to verify that the equipment has been constructed in accordance with the approved plans, specifications and conditions of this Authority to Construct, and to determine if the equipment can be operated in compliance with all Rules and Regulations of the San Joaquin Valley Unified Air Pollution Control District. PLEASE NOTIFY THE DISTRICT COMPLIANCE DIVISION (916) 427-1040 WHEN CONSTRUCTION OF THE EQUIPMENT IS COMPLETED. Unless construction has commenced pursuant to Rule 2250, this Authority to Construct shall expire and application shall be cancelled two years from the date of issuance. The applicant is responsible for complying for all laws, ordinances and regulations of any other governmental agencies which may pertain to the above equipment.

DAVID L. CROW, EXECUTIVE DIRECTOR/APCO

PHILIP MERSDEN, DIRECTOR OF PERMIT SERVICES

- 6 - A source test shall be performed to document actual emissions reductions and demonstrate compliance within 90 days of the startup of the boiler. 1-25-93 START-UP DATE
- 7 - The source test shall be performed by a qualified contractor according to a test plan submitted to the District for approval at least 30 days prior to the test date.
- 8 - Records of the fuel usage for the boiler shall be maintained for a period of at least two years and made available to the District upon request.
- 9 - Records of the monthly fuel usage shall be submitted to the District by March 1, of the following year. FEBRUARY 1993 TO FEBRUARY 1994
- 10 - The emissions of Nitrogen Oxides from the boiler shall not exceed 0.92 lb/Hr or 22.2 lb/day.
- 11 - The emissions of PM10 from the boiler shall not exceed 0.03 lb/Hr. or 0.8 lb/day.
- 12 - The emissions of VOC from this boiler shall not exceed 0.06 lb/Hr or 1.4 lb/day.
- 13 - The emissions of carbon monoxide from the boiler shall not exceed 0.73 lb/Hr. or 17.6 lb/day.
- 14 - All air pollution control equipment shall be maintained in good operating condition.
- 15 - Permit No. 7954D and 7954E for the two boilers being replaced shall be surrendered prior to startup of the boiler.

} Source
TEST

State of California
Air Resources Board

Approved Independent Contractor

Pacific Environmental Services, Inc.

This is to certify that the ~~services~~ ~~contract~~ has been approved by the Air Resources Board for compliance testing, pursuant to Section 19147, Title 17, California Code of Regulations, until June 30, 1987 for the test methods listed below:



James J. Morgester
James J. Morgester, Chief
Compliance Division

H. Roye Jackson
H. Roye Jackson, Manager
Field Evaluation Section

Appendix B

EMISSION TEST CALCULATIONS

Definiton of Symbols:

A	: cross-sectional area of stack, ft ²
A _n	: cross-sectional area of nozzle, ft ²
B _w	: water vapor in gas stream, proportional by volume
C _p	: pitot tube coeficient
I	: percent of isokinetic sampling
M _d	: dry molecular weight, lb/lb-mole
M _s	: molecular weight of stack gas, lb/lb-mole
P _{bar}	: barometric pressure at the sampling site, in. of Hg
P _s	: stack gas pressure, in. of Hg
Q _s	: stack volumetric flow rate , actual condition acfm
Q _{std}	: stack volumetric flow rate, standard conditions dscfm
T _m	: average dry gas meter temperature , R = 460 + °F
T _s	: average stack gas temperature, R = 460 + °F
V _{lc}	: total volume of liquid collected in impingers & silica gel, in ml
V _m	: Volume of gas sample as measured by dry gas meter, dscf
V _{mstd}	: Vm corrected to standard conditions, dscf
V _{wstd}	: volume of water vapor in gas sample corrected to standard conditions, scf
V _s	: stack gas velocity, ft/sec
γ	: dry gas meter calibration factor
ΔH	: average pressure differential across the orifice meter, in. of H ₂ O
θ	: total sampling time, in minutes
13.6	: specific gravity of mercury



SAMPLE RECOVERY DATA

Plant: CANDLEWICK YARNS
 Date: 4-21-93
 Sampling Location: BOILER EXHAUST
 Sample Type: SCAQMD 4.1
 Run Number: ONE - HIGH FINE
 Sample Box Number: 1B
 Clean-up Man: NGUYEN
 Job Number: _____
 Comments: _____

FRONT HALF

Filter Number: _____
 Description of Filter: _____

MOISTURE

	#1		#2		#3
Impingers					
Final Volume:	<u>728.3</u> ml		<u>585.8</u> ml		<u>496.6</u> ml
Initial Volume:	<u>580.2</u> ml		<u>569.6</u> ml		<u>494.5</u> ml
Net Volume:	<u>148.1</u> ml		<u>16.2</u> ml		<u>2.1</u> ml
Total H ₂ O:	_____		_____		_____

	#4			
Silica Gel				
Final Volume:	<u>757.4</u> g		_____ g	_____ g
Initial Volume:	<u>746.5</u> g		_____ g	_____ g
Net Volume:	<u>10.9</u> g		_____ g	_____ g
Total Moisture:	_____		_____	<u>188.2 ml</u>

Description of Impinger Catch: CLEAN

Plant: CANDLEWICK YARNS

Date: 4-21-93

Source/Sample Number: ONE - HIGH FIRE

$$1. Vm(std) = (17.64)(Vm)(Y) \left[\frac{P_{bar} + (\Delta H/13.6)}{Tm} \right]$$

$$Vm(std) = (17.64)(39.40)(0.988) \left[\frac{(29.95) + (1.5/13.6)}{(548)} \right]$$

$$Vm(std) = \underline{37.67} \text{ dscf.}$$

2. Volume water vapor collected (standard conditions).

$$V(1o) = \underline{188.2} \text{ condensate from impingers and silica gel.}$$

$$Vw(std) = (0.04707) V(1o) = (0.04707)(188.2)$$

$$Vw(std) = \underline{8.86} \text{ scf.}$$

3. Percent moisture, by volume.

$$Bw_s = \frac{Vw(std)}{Vw(std) + Vm(std)} = \frac{(8.86)}{(8.86) + (37.67)} = \underline{0.190}$$

$$Bw_s = \underline{19.0\%} .$$

4. Molecular weight, stack gas.

Dry molecular weight.

$$Md = 0.440(\% CO_2) + 0.320(\% O_2) + 0.280(\% N_2 + \% CO)$$

$$Md = 0.440(9.5) + 0.320(1.9) + 0.280(79)$$

$$Md = \underline{26.91} \text{ lb/lb-mole.}$$

$$Ms = Md + Bw_s (18 - Md) = (26.91) + (0.190)(18 - 26.91)$$

$$Ms = \underline{25.22} \text{ lb/lb-mole.}$$

Plant: CANDLEWICK YARNS

Date: 4-21-93

Source/Sample Number: ONE - HIGH FIRE

5. Stack gas velocity average.

$$V_s(\text{avg}) = (85.49)(C_p)(\sqrt{\Delta P}) \left[\text{avg} \sqrt{\frac{(T_s)}{(P_s)(M_s)}} \right]$$
$$V_s(\text{avg}) = (85.49)(0.84)(0.55) \left[\sqrt{\frac{(831)}{(29.95)(25.22)}} \right]$$

$V_s(\text{avg}) = \underline{41.43} \text{ ft/sec.}$

6. Stack volumetric flow rate, actual conditions (stack temperature and pressure).

$Q_s = (60)(V_s)(A) = (60)(41.43)(2.64)$

$Q_s = \underline{6,563} \text{ acfm.}$

7. Stack volumetric flow rate, standard conditions (68 degrees F, 29.92 Hg).

$Q(\text{std}) = (17.64)(Q_s)(1 - Bw_s) \left[\frac{(P_s)}{(T_s)} \right]$

$Q(\text{std}) = (17.64)(6,563)(1 - 0.19) \left[\frac{(29.95)}{(831)} \right]$

$Q(\text{std}) = \underline{3,380} \text{ dscfm.}$

8. Isokinetic variation.

$\%I = (K) \left[\frac{(T_s)(V_m(\text{std}))}{(P_s)(V_s)(A_n)(\theta)(1 - Bw_s)} \right]$

$\%I = (0.0945) \left[\frac{(\quad)(\quad)(\quad)}{(\quad)(\quad)(\quad)(\quad)(1 - \quad)} \right]$

$\%I = \underline{N/A} \%$



FGR Flow data

Operators: NGUYEN / HERNANDEZ Date: 4-21-93

Test Site: CANDLEWICK YARNS Client: CENTRAL BOILER
Pitot readings using CARB Method 1 (modified):

Run 1: Load Tested: HIGH FIRE Fuel: NAT. GAS %FGR_{open} 50

1. <u>0.06</u> Temp. <u>346</u>	5. <u>0.17</u> Temp. <u>352</u>
2. <u>0.05</u> Temp. <u>349</u>	6. <u>0.20</u> Temp. <u>355</u>
3. <u>0.07</u> Temp. <u>357</u>	7. <u>0.21</u> Temp. <u>356</u>
4. <u>0.12</u> Temp. <u>358</u>	8. <u>0.22</u> Temp. <u>356</u>

A_{SqRt} 0.358 DSCFM 540.7 %FGR_{Return} 16.0 T_{s+460} 814 R

Run 2: Load Tested: _____ Fuel: _____ %FGR_{open} _____

1. _____ Temp. _____	5. _____ Temp. _____
2. _____ Temp. _____	6. _____ Temp. _____
3. _____ Temp. _____	7. _____ Temp. _____
4. _____ Temp. _____	8. _____ Temp. _____

A_{SqRt} _____ DSCFM _____ %FGR_{Return} _____ T_{s+460} _____

Run 3: Load Tested: _____ Fuel: _____ %FGR_{open} _____

1. _____ Temp. _____	5. _____ Temp. _____
2. _____ Temp. _____	6. _____ Temp. _____
3. _____ Temp. _____	7. _____ Temp. _____
4. _____ Temp. _____	8. _____ Temp. _____

A_{SqRt} _____ DSCFM _____ %FGR_{Return} _____ T_{s+460} _____

FGR Pipe (I.D.) = 10" / 2 = radius / 12 = rad'⁽²⁾ X Pi = Area 0.545ft²

Diameters Upstream (A) 19" 1.9ø Diameters Downstream (B) 28" 2.8ø

Pitot Tube type and S/N STANDARD Temperature Readout NUTECH 2A

Where:

- A_{SqRt} = Average square root of the "delta p"
- FGR_{Return} = percentage of flue gas returned to burner.
- %FGR_{Open} = FGR valve opening.
- T_{s+460} = Average Stack Temperature in °F → R
- DSCFM = Volume of gas measured in FGR line corrected to dry gas standards.



Plant: CANDLEWICK YARNS

Date: 4-21-93

Source/Sample Number: ONE - HIGH FIRE (FGR)

5. Stack gas velocity average.

$$V_s(\text{avg}) = (85.49)(C_p)(\sqrt{\Delta P}) \left[\text{avg} \sqrt{\frac{(T_s)}{(P_s)(M_s)}} \right]$$

$$V_s(\text{avg}) = (85.49)(0.99)(0.358) \left[\sqrt{\frac{(814)}{(29.95)(25.22)}} \right]$$

$$V_s(\text{avg}) = \underline{31.5} \text{ ft/sec.}$$

6. Stack volumetric flow rate, actual conditions (stack temperature and pressure).

$$Q_s = (60)(V_s)(A) = (60)(31.5)(0.545)$$

$$Q_s = \underline{1028.5} \text{ acfm.}$$

7. Stack volumetric flow rate, standard conditions (68 degrees F, 29.92 Hg).

$$Q(\text{std}) = (17.64)(Q_s)(1 - Bw_s) \left[\frac{(P_s)}{(T_s)} \right]$$

$$Q(\text{std}) = (17.64)(1028.5)(1 - 0.19) \left[\frac{(29.95)}{(814)} \right]$$

$$Q(\text{std}) = \underline{540.7} \text{ dscfm.}$$

8. Isokinetic variation.

$$\frac{541}{3380} \times 100 =$$

$$\%I = (K) \left[\frac{(T_s)(V_m(\text{std}))}{(P_s)(V_s)(A_n)(\theta)(1 - Bw_s)} \right]$$

$$\%I = (0.0945) \left[\frac{(\quad)(\quad)(\quad)(\quad)}{(\quad)(\quad)(\quad)(\quad)(1 - \quad)} \right]$$

$$\%I = \underline{N/A} \%$$



HYDROCARBON SAMPLING FIELD DATA

Project No. _____

CLIENT: CANDLEWICK YARNS / CENTRAL BOILER

Date: 4-21-93

Sampling Location: BOILER EXHAUST

Tank # Flowmeter #	101 MR19	102 6F79	105 MR19
Time (min)	Sample A 24 HR CLOCK " Hg "H ₂ O	Sample B 24 HR CLOCK " Hg "H ₂ O	Sample C 24 HR CLOCK " Hg "H ₂ O
<u>0</u>	12:56 <u>30.0 1.0</u>	13:30 <u>29.7 1.0</u>	14:21 <u>30.0 1.0</u>
<u>5</u>	13:01 <u>27.6 1.0</u>	13:35 <u>27.7 1.0</u>	14:26 <u>27.5 1.0</u>
<u>10</u>	13:06 <u>25.1 1.0</u>	13:40 <u>25.7 1.0</u>	14:31 <u>24.9 1.0</u>
<u>15</u>	13:11 <u>21.9 1.0</u>	13:45 <u>23.7 1.0</u>	14:36 <u>22.0 1.0</u>
<u>20</u>	13:16 <u>18.8 1.0</u>	13:50 <u>21.5 1.0</u>	14:41 <u>19.0 1.0</u>
<u>25</u>	13:21 <u>15.8 1.0</u>	13:55 <u>19.4 1.0</u>	14:46 <u>15.9 1.0</u>
<u>30</u>	13:26 <u>12.7 1.0</u>	14:00 <u>16.5 1.0</u>	14:51 <u>12.7 1.0</u>

TANK 101
HAS A STICKY
VACUUM GATE

LEAK CHECK

<u>0</u>	13:28 <u>12.7 0.0</u>	14:10 <u>16.4 0.0</u>	14:53 <u>12.7 0.0</u>
<u>10</u>	13:38 <u>12.7 0.0</u>	14:20 <u>16.4 0.0</u>	15:03 <u>12.7 0.0</u>

Pre Leak-check ek

Post Leak-check _____

METHANE ANALYSES

Project No.: 4778.002

Client: Central Boiler / Candlewick Yarns

Cylinder#	Sample Id.	Total Area Of Sample	C _f	CH ₄ (ppm)	R _{1(v)}	P _{1(N2)}	D _{1(f)}	R _{2(v)}	P _{2(N2)}	D _{2(f)}	CH ₄ (ppm) corrected
101	A	0	0.00109	0	-9.35	1.95	1.56	-	-	-	0
102	B	179	0.00109	0.195	-13.80	0.65	1.92	0.55	5.30	1.16	0.4
105	C	0	0.00109	0	-10.90	0.40	1.61	0.30	5.00	1.16	0

$$C_f = \frac{\text{concentration of calibration gas (ppm)}}{\text{average total area of calibration gas}}$$

$$D_{n(f)} = \frac{P_{\text{bar}} + P_{n(N_2)}}{P_{\text{bar}} - R_{n(v)}}$$

$$\text{CH}_4 \text{ (ppm)} = C_f \times \text{average total area of sample}$$

$$\text{Concentration of CH}_4 \text{ corrected (ppm)} = \text{CH}_4 \times D_{n(f)}$$



PACIFIC ENVIRONMENTAL SERVICES, INC.

Project No. 4778.002	Page of
Client CENTRAL BOILER / CANDLEWICK YARNS	
Location BOILER EXHAUST	

Prepared By RTN	Date 4-26-93	Checked By	Date	Sheet Title
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4-26-93

BAROMETRIC PRESSURE: 29.46 in. Hg

TEMPERATURE: 74°F

5-7-93

BAROMETRIC PRESSURE: 29.51 in

TEMPERATURE: 72°F

Cylinder #	Sample Id.	$R_1(V)$ (in. Hg)	$P_1(N_2)$ (in. Hg)	$R_2(V)$ (in. Hg)	$P_2(N_2)$ (in. Hg)
101	A	-9.35	1.95		
102	B	-13.80	0.65	0.55	5.30
105	C	-10.90	0.40	0.30	5.00



CHANNEL A INJECT 05/10/93 12:16:25



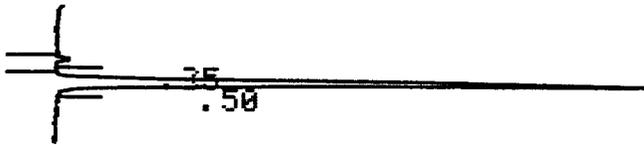
05/10/93 12:16:25 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 16 INDEX 16

PEAK#	AREA%	RT	AREA	BC
1	1.258	0.35	92	01
2	98.742	0.5	7221	01
TOTAL	100.		7313	

Calibrated w/ 8 ppm CH4

CHANNEL A INJECT 05/10/93 12:18:44



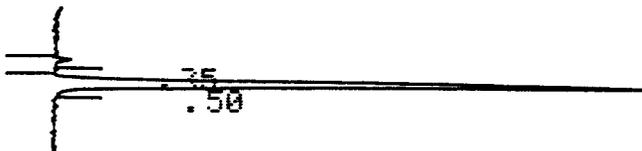
05/10/93 12:18:44 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 17 INDEX 17

PEAK#	AREA%	RT	AREA	BC
1	1.291	0.35	95	01
2	98.709	0.5	7261	01
TOTAL	100.		7356	

Calibrated w/ 8 ppm CH4

CHANNEL A INJECT 05/10/93 12:21:28



05/10/93 12:21:28 CH= "A" PS= 1.

FILE 1. METHOD 0. RUN 18 INDEX 18

PEAK#	AREA%	RT	AREA	BC
1	1.05	0.35	77	01
2	98.95	0.5	7259	01
TOTAL	100.		7336	

Calibrated w/ 8 ppm CH4



CHANNEL A INJECT 05/10/93 13:40:14

|

#101

NO DATA, CHANNEL A

CHANNEL A INJECT 05/10/93 13:43:17

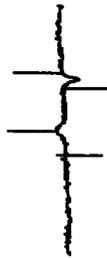
|

#101

NO DATA, CHANNEL A



CHANNEL A INJECT 05/10/93 12:58:02



.49

#102

05/10/93 12:58:02 CH= "A" PS= 1.

FILE	1.	METHOD	0.	RUN	23	INDEX	23
PEAK#		AREA%	RT	AREA	BC		
1		100.	0.49	185	01		
TOTAL		100.		185			

CHANNEL A INJECT 05/10/93 13:01:43



.49

#102

05/10/93 13:01:43 CH= "A" PS= 1.

FILE	1.	METHOD	0.	RUN	24	INDEX	24
PEAK#		AREA%	RT	AREA	BC		
1		100.	0.49	173	01		
TOTAL		100.		173			



EL A INJECT 05/10/93 13:47:13

|

#105

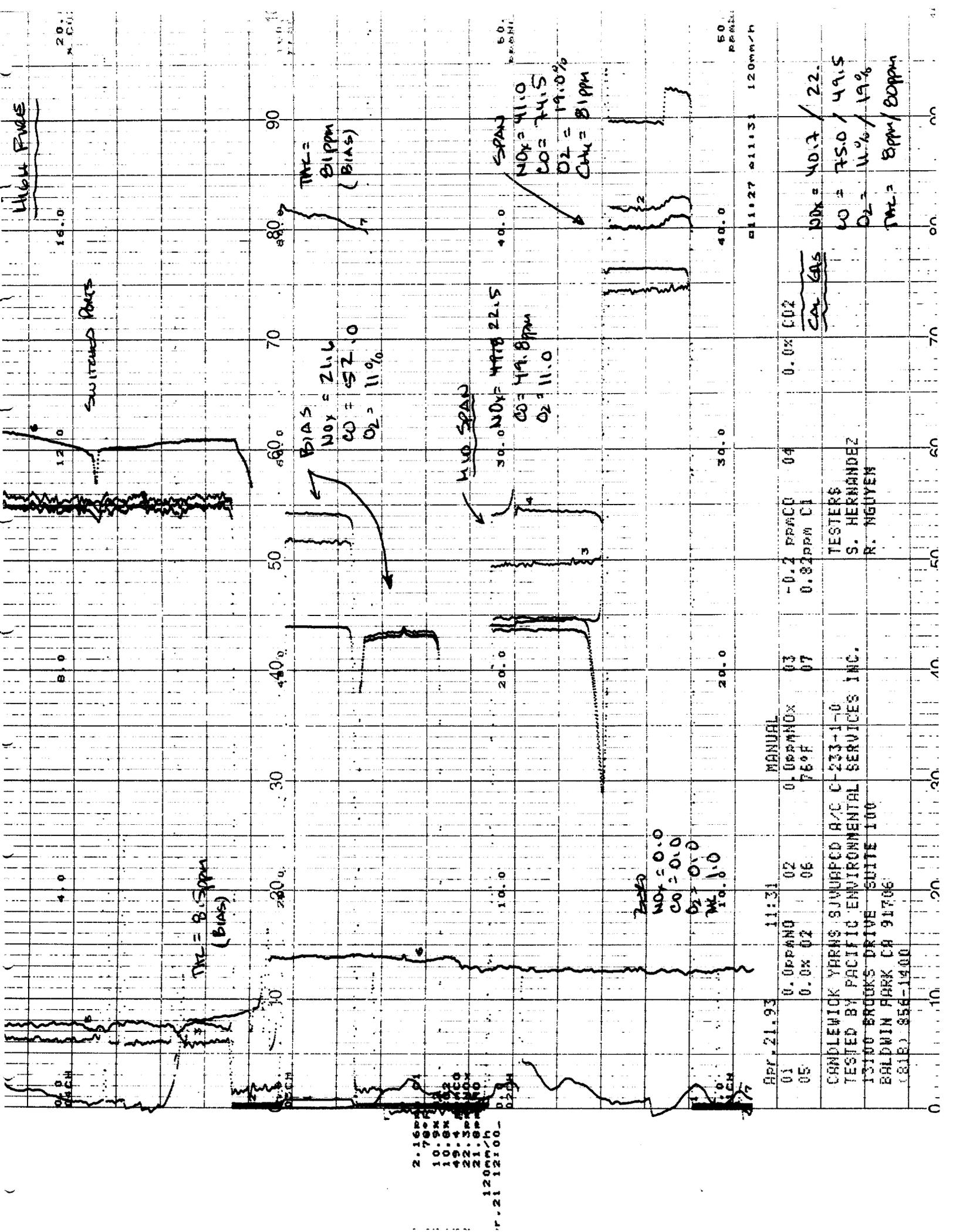
NO DATA, CHANNEL A

CHANNEL A INJECT 05/10/93 13:50:15

|

#105

NO DATA, CHANNEL A



HIGH PULSE

SWITCHED PULSE

TMC = 81ppm
(BIAS)

TMC =
81ppm
(BIAS)

BIAS
NOx = 21.6
CO = 52.0
O2 = 11.0

SPAN
NOx = 41.0
CO = 74.5
O2 = 19.0
CH4 = 81ppm

BIAS
NOx = 0.0
CO = 0.0
O2 = 0.0
CH4 = 0.0

2.16ppm
10.9X
10.8X
49.4
22.3
21.8

r.21 12:00-

APR. 21.93	11:31	MANUAL	0.0ppmNOx	03	-0.2ppmCO	04	0.0% CO2
01	0.0ppmNOx	02	76°F	07	0.82ppm CI		
05	0.0% O2	06					
CANDLEWICK YARDS SUPERFCD A/C C-233-1-0 TESTED BY PACIFIC ENVIRONMENTAL SERVICES INC. 13100 BROOKS DRIVE SUITE 100 BALDWIN PARK CA 91706 (818) 856-1400							
CAL GAS NOx = 40.7 / 22. CO = 75.0 / 49.5 O2 = 11.0 / 19.0 TMC = 81ppm / 80ppm							

TESTERS
S. HERNANDEZ
R. NGUYEN

CANDLEWICK YARNS SUMMIT RD C-211-1-0
TESTED BY PACIFIC ENVIRONMENTAL SERVICES INC.
13100 BROOKS DRIVE SUITE 100
BALDWIN PARK CA 91706
CIB 856-1400

Apr 21 93	15:43	MANUAL
01	0.0ppm O2	0.0ppm NOx
05	0.0% CO2	0.0% CO2
		1.73ppm C1

MID PMS
9.0 EXP 8.9

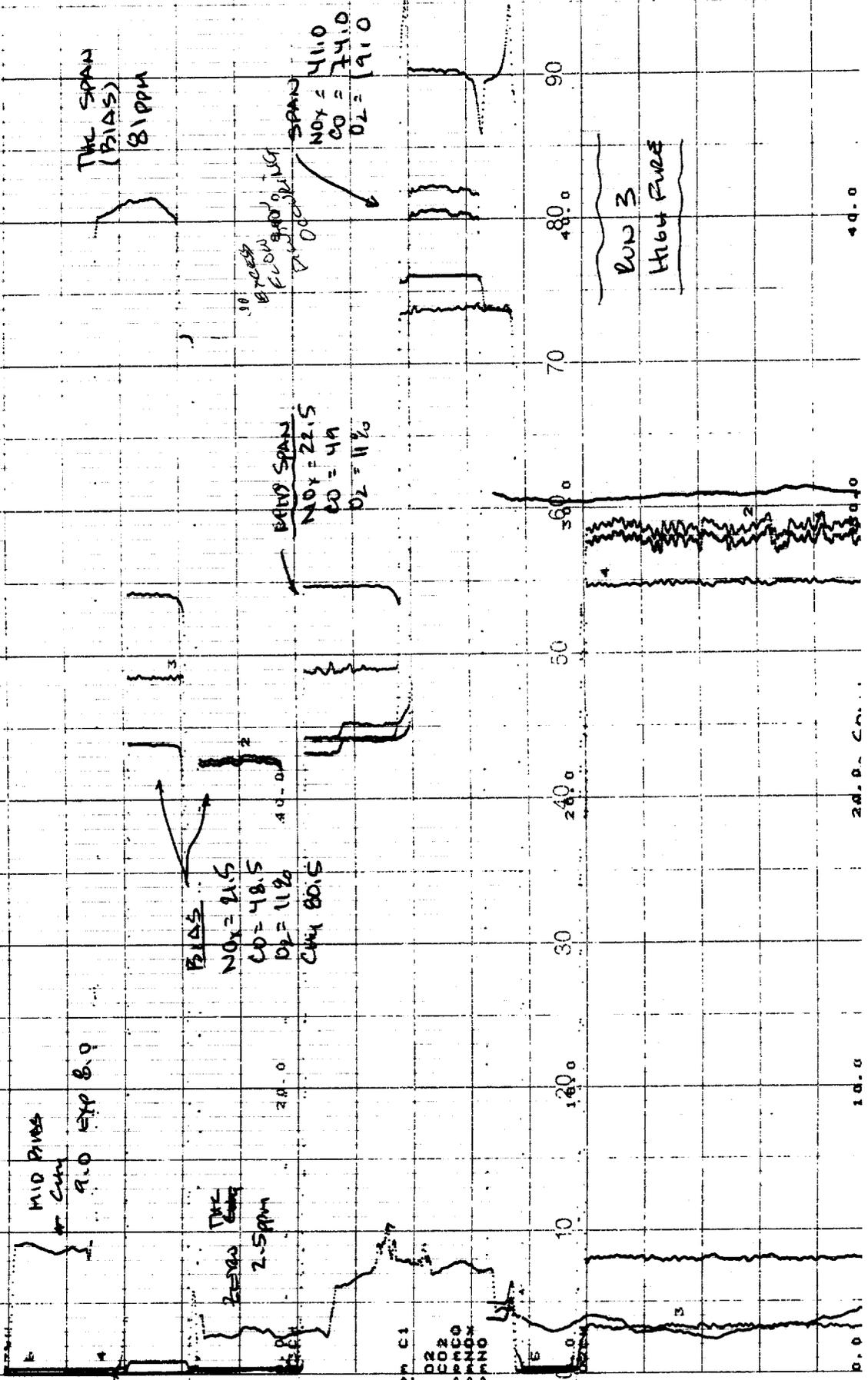
BIAS
NOx = 21.5
CO = 48.5
O2 = 11.2
C1 = 80.5

BIAS SPAN
NOx = 22.5
CO = 44
O2 = 11%

TAL SPAN (BIAS)
81ppm
NOx = 41.0
CO = 74.0
O2 = 19.0

EXCESS FLOW SPAN
200ppm

4.47ppm C1
362ppf O2
18.3% CO2
73.7 ppm CO
3.1 ppm NOx
2.5 ppm NO
120ppm H
Apr. 21 15:00



1000

Location: CANDEWICK YARNS Client: CENTRAL BOILER
 Prepared By: SMIT Checked By: - Date: 4-21-93
 Run # 1 Load Tested: 100% Fuel: NAT GAS

One Minute Integration in % of Chart

	<u>Time</u>	<u>NO_x</u>	<u>CO</u>	<u>THC</u>	<u>O₂</u>	
1	<u>12:40</u>	<u>27.5</u>	<u>5.8</u>	<u>-0.1</u>	<u>1.9</u>	<u>361</u>
2	<u>12:41</u>	<u>27.8</u>	<u>6.1</u>	<u>0.16</u>	<u>1.9</u>	<u>361</u>
3	<u>12:42</u>	<u>27.6</u>	<u>6.3</u>	<u>0.65</u>	<u>1.9</u>	<u>360</u>
4	<u>12:43</u>	<u>27.7</u>	<u>6.1</u>	<u>0.47</u>	<u>1.9</u>	<u>360</u>
5	<u>12:44</u>	<u>27.7</u>	<u>5.8</u>	<u>0.51</u>	<u>1.9</u>	<u>360</u>
6	<u>12:45</u>	<u>27.8</u>	<u>7.1</u>	<u>0.50</u>	<u>1.8</u>	<u>359</u>
7	<u>12:46</u>	<u>27.4</u>	<u>6.2</u>	<u>0.15</u>	<u>1.8</u>	<u>356</u>
8	<u>12:47</u>	<u>27.6</u>	<u>6.1</u>	<u>0.48</u>	<u>1.8</u>	<u>359</u>
9	<u>12:48</u>	<u>27.7</u>	<u>6.0</u>	<u>0.61</u>	<u>1.9</u>	<u>361</u>
10	<u>12:49</u>	<u>27.6</u>	<u>6.0</u>	<u>0.78</u>	<u>1.9</u>	<u>363</u>
11	<u>12:50</u>	<u>27.5</u>	<u>6.2</u>	<u>0.90</u>	<u>1.8</u>	<u>363</u>
12	<u>12:51</u>	<u>27.6</u>	<u>6.1</u>	<u>1.28</u>	<u>1.8</u>	<u>361</u>
13	<u>12:52</u>	<u>27.7</u>	<u>6.2</u>	<u>1.51</u>	<u>1.8</u>	<u>365</u>
14	<u>12:53</u>	<u>27.9</u>	<u>6.1</u>	<u>1.55</u>	<u>1.7</u>	<u>366</u>
15	<u>12:54</u>	<u>27.6</u>	<u>6.1</u>	<u>1.47</u>	<u>1.9</u>	<u>367</u>

OBSERVED JUMPS IN CO WHEN STACK VIBRATES. (OSCILLATION IN PER)

Multiplication Factors:

NO_x 0 - 100% Full Scale = 0 - 50 ppm NO_x % Chart X 0.5
 CO 0 - 100% Full Scale = 0 - 100 ppm CO % Chart X 1.0
 O₂ 0 - 100% Full Scale = 0 - 25% O₂ % Chart X 0.25
 THC 0 - 100% Full Scale = 0 - 100 C₁ % Chart X 1.0



Location: CANREMIC YARNS Client: CANTON BOILER

Prepared By: SMH Checked By: - Date: 4-21-93

Run # 1 Load Tested: 100% Fuel: NAT GAS

One Minute Integration in % of Chart

	<u>Time</u>	<u>NO_x</u>	<u>CO</u>	<u>THC</u>	<u>O₂</u>	
1	<u>12:55</u>	<u>27.7</u>	<u>6.2</u>	<u>1.57</u>	<u>1.8</u>	368
2	<u>12:56</u>	<u>28.0</u>	<u>6.1</u>	<u>2.29</u>	<u>1.8</u>	369
3	<u>12:57</u>	<u>27.8</u>	<u>6.5</u>	<u>2.3</u>	<u>1.8</u>	369
4	<u>12:58</u>	<u>27.8</u>	<u>6.6</u>	<u>2.26</u>	<u>1.8</u>	370
5	<u>12:59</u>	<u>27.9</u>	<u>6.4</u>	<u>2.14</u>	<u>1.8</u>	371
6	<u>13:00</u>	<u>27.9</u>	<u>6.4</u>	<u>2.12</u>	<u>1.8</u>	371
7	<u>13:01</u>	<u>28.2</u>	<u>6.3</u>	<u>2.92</u>	<u>1.8</u>	372
8	<u>13:02</u>	<u>28.0</u>	<u>6.8</u>	<u>3.34</u>	<u>1.7</u>	373
9	<u>13:03</u>	<u>28.0</u>	<u>6.6</u>	<u>3.68</u>	<u>1.7</u>	373
10	<u>13:04</u>	<u>28.0</u>	<u>6.7</u>	<u>3.62</u>	<u>1.7</u>	374
11	<u>13:05</u>	<u>27.9</u>	<u>6.7</u>	<u>3.46</u>	<u>1.8</u>	374
12	<u>13:06</u>	<u>28.0</u>	<u>6.8</u>	<u>3.43</u>	<u>1.7</u>	375
13	<u>13:07</u>	<u>28.2</u>	<u>6.8</u>	<u>3.77</u>	<u>1.7</u>	375
14	<u>13:08</u>	<u>27.9</u>	<u>6.7</u>	<u>3.72</u>	<u>1.8</u>	374
15	<u>13:09</u>	<u>28.1</u>	<u>29.0</u>	<u>3.25</u>	<u>1.7</u>	375
		27.8 ppm NO _x	7.1 ppm CO	1.83 ppm C ₁	1.8% O ₂	

Multiplication Factors:

- NO_x 0 - 100% Full Scale = 0 - 50 ppm NO_x * Chart X 0.5
- CO 0 - 100% Full Scale = 0 - 100 ppm CO * Chart X 1.0
- O₂ 0 - 100% Full Scale = 0 - 25% O₂ * Chart X 0.25
- THC 0 - 100% Full Scale = 0 - 100 ppm C₁ * Chart X 1.0

Location: CANDSWICK YARDS Client: CENTRAL POWER

Prepared By: SMH Checked By: — Date:

Run # 2 Load Tested: 100% Fuel: HOT GAS

One Minute Integration in % of Chart Conc.

	<u>Time</u>	<u>NO_x</u>	<u>CO</u>	<u>THC</u>	<u>O₂</u>	
1	<u>13:30</u>	<u>28.5</u>	<u>8.1</u>	<u>2.2</u>	<u>1.7</u>	371
2	<u>13:31</u>	<u>28.6</u>	<u>10.7</u>	<u>2.5</u>	<u>1.7</u>	370
3	<u>13:32</u>	<u>28.7</u>	<u>11.0</u>	<u>2.1</u>	<u>1.8</u>	370
4	<u>13:33</u>	<u>28.7</u>	<u>6.5</u>	<u>2.1</u>	<u>1.8</u>	370
5	<u>13:34</u>	<u>28.4</u>	<u>6.8</u>	<u>2.1</u>	<u>1.8</u>	370
6	<u>13:35</u>	<u>28.5</u>	<u>7.0</u>	<u>2.0</u>	<u>1.8</u>	369
7	<u>13:36</u>	<u>28.4</u>	<u>6.8</u>	<u>1.9</u>	<u>1.8</u>	369
8	<u>13:37</u>	<u>28.5</u>	<u>6.6</u>	<u>1.9</u>	<u>1.8</u>	369
9	<u>13:38</u>	<u>28.6</u>	<u>6.9</u>	<u>2.0</u>	<u>1.8</u>	369
10	<u>13:39</u>	<u>28.8</u>	<u>7.6</u>	<u>3.1</u>	<u>1.7</u>	369
11	<u>13:40</u>	<u>28.8</u>	<u>40.6</u>	<u>3.6</u>	<u>1.7</u>	368
12	<u>13:41</u>	<u>28.4</u>	<u>7.4</u>	<u>2.6</u>	<u>1.7</u>	368
13	<u>13:42</u>	<u>28.5</u>	<u>7.5</u>	<u>2.4</u>	<u>1.7</u>	368
14	<u>13:43</u>	<u>28.2</u>	<u>96.1</u>	<u>4.7</u>	<u>1.5</u>	367
15	<u>13:44</u>	<u>28.5</u>	<u>7.1</u>	<u>2.0</u>	<u>1.7</u>	368

Multiplication Factors:

NO_x 0 - 100% Full Scale = 0 - 50 ppm NO_x % Chart X 0.5
 CO 0 - 100% Full Scale = 0 - 100 ppm CO % Chart X 1.0
 O₂ 0 - 100% Full Scale = 0 - 25% O₂ % Chart X 0.25
 THC 0 - 100% Full Scale = 0 - 100 ppm C₁ % Chart X 1.0



Location: Candlewick Yarns Client: Central Boiler

Prepared By: SMH Checked By: — Date: 7-21-93

Run # 2 Load Tested: 100% Fuel: NAT GAS

One Minute Integration in ~~1/2~~ of Chart Record display

	<u>Time</u>	<u>NO_x</u>	<u>CO</u>	<u>THC</u>	<u>O₂</u>	
1	<u>13:45</u>	<u>28.5</u>	<u>7.1</u>	<u>2.1</u>	<u>1.7</u>	369
2	<u>13:46</u>	<u>28.5</u>	<u>7.0</u>	<u>2.1</u>	<u>1.8</u>	369
3	<u>13:47</u>	<u>28.5</u>	<u>7.2</u>	<u>2.1</u>	<u>1.8</u>	368
4	<u>13:48</u>	<u>28.1</u>	<u>22.6</u>	<u>2.3</u>	<u>1.7</u>	367
5	<u>13:49</u>	<u>28.4</u>	<u>9.4</u>	<u>2.2</u>	<u>1.7</u>	367
6	<u>13:50</u>	<u>28.3</u>	<u>7.6</u>	<u>2.1</u>	<u>1.7</u>	367
7	<u>13:51</u>	<u>28.8</u>	<u>6.7</u>	<u>1.9</u>	<u>1.8</u>	366
8	<u>13:52</u>	<u>29.2</u>	<u>6.6</u>	<u>1.9</u>	<u>1.9</u>	366
9	<u>13:53</u>	<u>29.4</u>	<u>6.6</u>	<u>1.8</u>	<u>2.0</u>	365
10	<u>13:54</u>	<u>29.3</u>	<u>6.5</u>	<u>1.6</u>	<u>1.9</u>	365
11	<u>13:55</u>	<u>29.5</u>	<u>6.7</u>	<u>1.7</u>	<u>1.9</u>	365
12	<u>13:56</u>	<u>29.3</u>	<u>6.8</u>	<u>1.7</u>	<u>1.9</u>	365
13	<u>13:57</u>	<u>29.6</u>	<u>6.7</u>	<u>1.8</u>	<u>2.0</u>	365
14	<u>13:58</u>	<u>29.6</u>	<u>6.3</u>	<u>1.7</u>	<u>2.0</u>	365
15	<u>13:59</u>	<u>29.3</u>	<u>6.8</u>	<u>1.7</u>	<u>1.9</u>	366

13:52 Fuel moved from SS - 50% TO STOP OSCILLATING NO_x
 Moves up 0.5ppm
 28.8 ppm NO_x 11.9 ppm CO 2.2 ppm C₁ 1.8% O₂

Multiplication Factors:

- NO_x 0 - 100% Full Scale = 0 - 50 ppm NO_x * Chart X 0.5
- CO 0 - 100% Full Scale = 0 - 100 ppm CO * Chart X 1.0
- O₂ 0 - 100% Full Scale = 0 - 25% O₂ * Chart X 0.25
- THC 0 - 100% Full Scale = 0 - 100 ppm C₁ * Chart X 1.0



Location: CANDLERVILLE YARDS Client: Carroll Police

Prepared By: [Signature] Checked By: - Date: 4-21

Run # 3 Load Tested: 100% Fuel: LAT GAS

One Minute Integration in % of Chart

	<u>Time</u>	<u>NO_x</u>	<u>CO</u>	<u>THC</u>	<u>O₂</u>	
1	<u>14:23</u>	<u>29.1</u>	<u>3.1</u>	<u>3.3</u>	<u>1.9</u>	<u>362</u>
2	<u>14:24</u>	<u>29.2</u>	<u>3.1</u>	<u>3.5</u>	<u>1.9</u>	<u>363</u>
3	<u>14:25</u>	<u>29.4</u>	<u>3.1</u>	<u>3.4</u>	<u>1.9</u>	<u>363</u>
4	<u>14:26</u>	<u>29.4</u>	<u>3.1</u>	<u>4.0</u>	<u>1.9</u>	<u>363</u>
5	<u>14:27</u>	<u>29.3</u>	<u>3.1</u>	<u>4.4</u>	<u>1.9</u>	<u>363</u>
6	<u>14:28</u>	<u>29.4</u>	<u>3.1</u>	<u>4.7</u>	<u>1.9</u>	<u>364</u>
7	<u>14:29</u>	<u>29.2</u>	<u>3.1</u>	<u>4.4</u>	<u>1.9</u>	<u>363</u>
8	<u>14:30</u>	<u>29.0</u>	<u>3.0</u>	<u>4.1</u>	<u>1.9</u>	<u>364</u>
9	<u>14:31</u>	<u>29.2</u>	<u>3.1</u>	<u>3.9</u>	<u>1.9</u>	<u>364</u>
10	<u>14:32</u>	<u>29.0</u>	<u>3.2</u>	<u>3.6</u>	<u>1.9</u>	<u>365</u>
11	<u>14:33</u>	<u>29.2</u>	<u>3.2</u>	<u>3.6</u>	<u>1.9</u>	<u>366</u>
12	<u>14:34</u>	<u>29.2</u>	<u>3.0</u>	<u>3.3</u>	<u>1.9</u>	<u>366</u>
13	<u>14:35</u>	<u>29.2</u>	<u>3.3</u>	<u>3.2</u>	<u>1.9</u>	<u>366</u>
14	<u>14:36</u>	<u>28.9</u>	<u>3.2</u>	<u>3.2</u>	<u>1.9</u>	<u>366</u>
15	<u>14:37</u>	<u>29.6</u>	<u>3.0</u>	<u>3.1</u>	<u>2.0</u>	<u>364</u>

High Water Alarm at ^{2-inch} 14:34 →

Multiplication Factors:

- NO_x 0 - 100% Full Scale = 0 - 50 ppm NO_x % Chart X 0.5
- CO 0 - 100% Full Scale = 0 - 100 ppm CO % Chart X 1.0
- O₂ 0 - 100% Full Scale = 0 - 25% O₂ % Chart X 0.25
- THC 0 - 100% Full Scale = 0 - 100 ppm C₁ % Chart X 1.0



Location: CANTLEWICK YARNS Client: CANTON BOILER

Prepared By: SMU Checked By: - Date: 4-21-93

Run # 3 Load Tested: 100% Fuel: NAT GAS

One Minute Integration in % of Chart

	<u>Time</u>	<u>NO_x</u>	<u>CO</u>	<u>THC</u>	<u>O₂</u>	
1	<u>14:38</u>	<u>29.1</u>	<u>3.2</u>	<u>2.9</u>	<u>2.0</u>	<u>364</u>
2	<u>14:39</u>	<u>29.0</u>	<u>3.2</u>	<u>2.8</u>	<u>1.9</u>	<u>363</u>
3	<u>14:40</u>	<u>28.8</u>	<u>3.1</u>	<u>2.6</u>	<u>1.9</u>	<u>364</u>
4	<u>14:41</u>	<u>28.9</u>	<u>3.1</u>	<u>2.5</u>	<u>2.0</u>	<u>364</u>
5	<u>14:42</u>	<u>29.4</u>	<u>3.2</u>	<u>2.3</u>	<u>2.0</u>	<u>364</u>
6	<u>14:43</u>	<u>29.5</u>	<u>3.1</u>	<u>2.3</u>	<u>2.0</u>	<u>364</u>
7	<u>14:44</u>	<u>29.2</u>	<u>3.1</u>	<u>2.5</u>	<u>1.9</u>	<u>363</u>
8	<u>14:45</u>	<u>29.1</u>	<u>3.2</u>	<u>2.8</u>	<u>1.9</u>	<u>363</u>
9	<u>14:46</u>	<u>29.3</u>	<u>3.1</u>	<u>2.9</u>	<u>1.9</u>	<u>363</u>
10	<u>14:47</u>	<u>29.0</u>	<u>3.1</u>	<u>2.9</u>	<u>1.9</u>	<u>363</u>
11	<u>14:48</u>	<u>29.1</u>	<u>3.0</u>	<u>2.8</u>	<u>2.0</u>	<u>363</u>
12	<u>14:49</u>	<u>29.4</u>	<u>3.1</u>	<u>2.7</u>	<u>2.0</u>	<u>362</u>
13	<u>14:50</u>	<u>29.3</u>	<u>3.1</u>	<u>3.1</u>	<u>2.0</u>	<u>362</u>
14	<u>14:51</u>	<u>29.2</u>	<u>3.0</u>	<u>3.6</u>	<u>2.0</u>	<u>362</u>
15	<u>14:52</u>	<u>29.1</u>	<u>3.1</u>	<u>3.8</u>	<u>1.9</u>	<u>362</u>

29.2 ppm NO_x 3.1 ppm CO 3.3 ppm C₁ 1.9% O₂

Multiplication Factors:

- NO_x 0 - 100% Full Scale = 0 - 50 ppm NO_x † Chart X 0.5
- CO 0 - 100% Full Scale = 0 - 100 ppm CO † Chart X 1.0
- O₂ 0 - 100% Full Scale = 0 - 25% O₂ † Chart X 0.25
- THC 0 - 100% Full Scale = 0 - 100 C₁ † Chart X 1.0



Calibration Summaries

Location: Candlewick Yarns Client: Central Boiler Date: 4-21-93

Prepared By: _____

Checked by: _____

Run #	1	NO _x	CO	O ₂	THC
Z _i	=	0.0	0.0	0.0	1.0
Z _f	=	0.0	2.0	0.0	2.5
S _i	=	41.0	74.5	19.0	81.0
S _f	=	41.3	76.8	19.0	81.0
Z _s /S _s	=	0.0/41.2	0.1/75.7	0.0/19.0	1.8/81.0

Run #	2	NO _x	CO	O ₂	
Z _i	=	0.0	2.0	0.0	---
Z _f	=	0.0	-1.0	0.0	---
S _i	=	41.3	76.8	19.0	---
S _f	=	41.1	73.5	19.0	---
Z _s /S _s	=	0.0/41.2	0.1/75.2	0.0/19.0	---/---

Run #	3	NO _x	CO	O ₂	
Z _i	=	0.0	-1.0	0.0	---
Z _f	=	0.0	-1.0	0.0	---
S _i	=	41.1	73.5	19.0	---
S _f	=	41.0	74.0	19.0	---
Z _s /S _s	=	0.0/41.1	-1.0/73.8	0.0/19.0	---/---

Run #	-	NO _x	CO	O ₂	
Z _i	=	---	---	---	---
Z _f	=	---	---	---	---
S _i	=	---	---	---	---
S _f	=	---	---	---	---
Z _s /S _s	=	---/---	---/---	---/---	---/---

Run #	-	NO _x	CO	O ₂	
Z _i	=	---	---	---	---
Z _f	=	---	---	---	---
S _i	=	---	---	---	---
S _f	=	---	---	---	---
Z _s /S _s	=	---/---	---/---	---/---	---/---

Run #	-	NO _x	CO	O ₂	
Z _i	=	---	---	---	---
Z _f	=	---	---	---	---
S _i	=	---	---	---	---
S _f	=	---	---	---	---
Z _s /S _s	=	---/---	---/---	---/---	---/---

$$Z_s = \frac{Z_f - Z_i}{2} + Z_i$$

Where:

Z_s, S_s = Average zero or span strip chart divisions.

Z_f, S_f = Final zero or span strip chart divisions.

Z_i, S_i = Initial zero or span strip chart divisions.

$$S_s = \frac{S_f - S_i}{2} + S_i$$



Data Drift Correction Worksheet

$$C_{gas (corr)} = (C_m - Z_m) (C_s + (S_m - Z_m)) \times C_{3\%O_2}$$

Run 1 $C_{3\%O_2}$ = (20.95 - 3.0) + (20.95 - 1.8) = 0.94

Start 12:40 NO_x (27.8 - 0.0) (40.7 + (41.2 - 0.0)) = 25.9
(27.5)

End 13:09 CO (7.1 - 0.1) (75.0 + (75.7 - 0.1)) = 6.4
(6.7)

O_2 (1.8 - 0.0) (19.0 + (19.0 - 0.0)) = 1.8

THC (1.83 - 1.8) (80.0 + (81.0 - 1.8)) = 0.0
(0.03 - 0.4 = -0.37)

Run 2 $C_{3\%O_2}$ = (20.95 - 3.0) + (20.95 - 1.8) = 0.94

Start 13:30 NO_x (28.8 - 0.0) (40.7 + (41.2 - 0.0)) = 26.8
(28.5)

End 13:59 CO (11.9 - 0.1) (75.0 + (75.2 - 0.1)) = 11.1
(11.8)

O_2 (1.8 - 0.0) (19.0 + (19.0 - 0.0)) = 1.8

THC (2.2 - 1.8) (80.0 + (81.0 - 1.8)) = 0.0
(0.4 - 0.4 = 0.0)

Run 3 $C_{3\%O_2}$ = (20.95 - 3.0) + (20.95 - 1.9) = 0.94

Start 14:23 NO_x (29.2 - 0.0) (40.7 + (41.1 - 0.0)) = 27.2
(28.9)

End 14:52 CO (3.1 - -1.0) (75.0 + (73.8 - -1.0)) = 3.9
(4.1)

O_2 (1.9 - 0.0) (19.0 + (19.0 - 0.0)) = 1.9

THC (3.3 - 1.8) (80.0 + (81.0 - 1.8)) = 1.03
(1.5 - 0.4 = 1.1)

Run $C_{3\%O_2}$ = (20.95 - 3.0) + (20.95 -) =

Start : NO_x (-) (+ (-)) =

End : CO (-) (+ (-)) =

O_2 (-) (+ (-)) =

THC (-) (+ (-)) =
(- =)

CO emissions calculated to lbs/hr

Calculation:

$$\frac{\text{ppmV} \times 28 \text{ lb/lb-mole} \times 60 \text{ min/hr} \times \text{DSCFM}}{385 \text{ SCF/lb-mole} \times 1,000,000} = \text{lbs/hr CO}$$

Run 1 $\frac{6.9 \times 28 \times 60 \times 3380 \text{ DSCFM}}{385 \times 10^6} = 0.10 \text{ lbs/hr CO}$

Run 2 $\frac{11.8 \times 28 \times 60 \times 3380 \text{ DSCFM}}{385 \times 10^6} = 0.17 \text{ lbs/hr CO}$

Run 3 $\frac{4.1 \times 28 \times 60 \times 3380 \text{ DSCFM}}{385 \times 10^6} = 0.06 \text{ lbs/hr CO}$

Run _____ $\frac{\text{X } 28 \times 60 \times \text{X DSCFM}}{385 \times 10^6} = \text{_____ lbs/hr CO}$

Run _____ $\frac{\text{X } 28 \times 60 \times \text{X DSCFM}}{385 \times 10^6} = \text{_____ lbs/hr CO}$

Run _____ $\frac{\text{X } 28 \times 60 \times \text{X DSCFM}}{385 \times 10^6} = \text{_____ lbs/hr CO}$

WHERE:

ppmV = ppm average of one minute integrations for NO_x corrected to 3% O₂.

28 lb/lb mole = Molecular weight of CO

DSCFM = Flows corrected to dry standards. Taken from calculations based on measured flow rates.

Appendix C



Scott Specialty Gases, Inc.

Shipped From: 2600 CAJON BLVD.
 SAN BERNARDINO CA 92411
 Phone: 714-887-2571

Fax: 714-887-0549

CERTIFICATE OF ANALYSIS

PACIFIC ENVIRONMENTAL SER
 PO# 0640-102
 13100 BROOKS DRIVE

PROJECT #: 02-25048-002
 PO#: PO # 0640-102
 ITEM #: 02024520 2AL
 DATE: 4/08/93

BALDWIN PARK CA 91706

CYLINDER #: 1L2572
 FILL PRESSURE: 2000 PSIG
 BLEND TYPE : ACUBLEND MASTER GAS

ANALYTICAL ACCURACY: +/-1%

COMPONENT	REQUESTED GAS		ANALYSIS	
	CONC	MOLES	(MOLES)	
CARBON DIOXIDE	11.	PCT	11.00	PCT
CARBON MONOXIDE	50.	PPM	50.00	PPM
OXYGEN	11.	PCT	11.00	PCT
NITROGEN		BAL		BAL

Post-It™ brand fax transmittal memo 7671 # of pages 1

To: <i>Steve</i>	From: <i>Janie</i>
Co.	Co.
Dept.	Phone#
Fax #	Fax #

4/9 2000 PSIG BIN#1

ANALYST: *[Signature]*

MAYNARD JOHNSON

APPROVED BY: *[Signature]*

DR. ARMANDO VARGAS, PhD

PLUMSTEADVILLE, PENNSYLVANIA / TROY, MICHIGAN / HOUSTON, TEXAS / DURHAM, NORTH CAROLINA
 SOUTH PLAINFIELD, NEW JERSEY / FREMONT, CALIFORNIA / WAKEFIELD, MASSACHUSETTS / LONGMONT, COLORADO
 BATON ROUGE, LOUISIANA



Scott Specialty Gases, Inc.

Shipped
From:

2600 CAJON BLVD.
SAN BERNARDINO CA 92411
Phone: 714-887-2571

Fax: 714-887-0549

CERTIFICATE OF ANALYSIS

PACIFIC ENVIRONMENTAL SER
ATTN: STEVE HERNANDEZ
13100 BROOKS DRIVE

PROJECT #: 02-25328-002
PON: 0640-102
ITEM #: 0202RC00 AL
DATE: 5/17/93

BALDWIN PARK

CA 91706

CYLINDER #: AAL13428
FILL PRESSURE: 700 PSIG

COMPONENT	ANALYSIS
CARBON DIOXIDE	17.99 PCT
CARBON MONOXIDE	75.21 PPM
OXYGEN	19.07 PCT
NITROGEN	BALANCE

5/17 700 PSIG

ANALYST:

jd
JOSEPH DE LA TORRE

APPROVED BY:

Robert Shealy
ROBERT SHEALY



Scott Specialty Gases, Inc.

2600 CAJON BOULEVARD, SAN BERNARDINO, CA 92411

(909) 887-2571 FAX: (909) 887-0549

CERTIFICATE OF ANALYSIS: EPA PROTOCOL GAS

Customer:
Pacific Environmental Services
13100 Brooks Drive
Baldwin Park, CA 91706-0740

Assay Laboratory
Scott Specialty Gases
2600 Cajon Boulevard
San Bernardino, CA 92411

Purchase Order 0640-102
Scott Project # 24700

ANALYTICAL INFORMATION

Certified to exceed the minimum specifications of EPA Protocol 1 Procedure #G1, Section Number 3.0.4

Cylinder Number ALM027851
Cylinder Pressure 1950 psig

Certification Date 03-17-93
Previous Certification Dates NONE

Expiration Date 09-13-94

ANALYZED CYLINDER

Components
Nitric Oxide

Certified Concentration
22.51 ppm

Analytical Uncertainty*
± 1 % NIST Traceable

NOX
Balance Gas: Nitrogen

22.82 ppm

*Analytical uncertainty is inclusive of usual known error sources which at least includes reference standard error & precision of the measurement processes.

REFERENCE STANDARD

Type SRM 2629A
Expiration Date 09-93

Cylinder Number
FF28519

Concentration
19.3 ppm

INSTRUMENTATION

Instrument/Model/Serial #
TECO / 10AR / 14853-150

Last Date Calibrated
1-20-93

Analytical Principle
Chemi-Luminescent

ANALYZER READINGS (Z=Zero Gas R=Reference Gas T=Test Gas r=Correlation Coefficient)

Components

First Triad Analysis

Second Triad Analysis

Calibration Curve

Nitric Oxide

Date: 03-01-93 Response Units: mv					
Z1= 0.000	R1= 78.33	T1= 90.95	Z2= 0.000	R2= 78.30	T2= 90.92
Z3= 0.000	T3= 90.94	R3= 78.28	Avg. Conc. of Cust Cyl. 22.41 ppm		

Date: 03-17-93 Response Units: mv					
Z1= 0.000	R1= 77.12	T1= 90.20	Z2= 0.000	R2= 77.24	T2= 90.54
Z3= 0.000	T3= 90.50	R3= 77.21	Avg. Conc. of Cust Cyl. 22.60 ppm		

Concentration=	Ax^2+Bx+C
A=	-0.000504296
B=	1.0489680
C=	-0.292506

Date: Response Units: mv					
Z1=	R1=	T1=	Z2=	R2=	T2=
Z3=	T3=	R3=	Avg. Conc. of Cust Cyl.		

Date: Response Units: mv					
Z1=	R1=	T1=	Z2=	R2=	T2=
Z3=	T3=	R3=	Avg. Conc. of Cust Cyl.		

Concentration=	
----------------	--

Date: Response Units:					
Z1=	R1=	T1=	Z2=	R2=	T2=
Z3=	T3=	R3=	Avg. Conc. of Cust Cyl.		

Date: Response Units:					
Z1=	R1=	T1=	Z2=	R2=	T2=
Z3=	T3=	R3=	Avg. Conc. of Cust Cyl.		

Concentration=	
----------------	--

Special Notes

Joseph De La Torre
Analyst Joseph De La Torre

Scott Specialty Gases, Inc.

Shipped From: 2600 CAJON BLVD.
SAN BERNARDINO CA 92411
Phone: 714-887-2571 Fax: 714-887-0549

CERTIFICATE OF ANALYSIS

PACIFIC ENVIRONMENTAL SER
ATTN: STEVE HERNANDEZ
13100 BROOKS DRIVE

BALDWIN PARK

CA 91706

PROJECT #: 02-19360
PO#: 0640-102
ITEM #: 02022914 4AL
DATE: 4/24/92

CYLINDER #: AAL18076

ANALYTICAL ACCURACY: +/-1%

COMPONENT

NITRIC OXIDE
NITROGEN - OXYGEN FREE

REQUESTED GAS

CONC MOLES
40. PPM
BAL

ANALYSIS

(MOLES)
40.01 PPM
BAL

CERTIFIED

PACIFIC ENVIRONMENTAL

CYLINDER PRESSURE 2000 PSI

NIST TRACEABLE TO SRM1683B

CYL#CAL12736 47.9 PPM NO

NOX = 40.66 PPM

ANALYST:

[Signature]
JOSEPH DE LA TORRE

APPROVED BY:

[Signature]
DR. ARMAND LEBLANC PHD

PLUMSTEADVILLE, PENNSYLVANIA / TROY, MICHIGAN / HOUSTON, TEXAS / DURHAM, NORTH CAROLINA
SOUTH PLAINFIELD, NEW JERSEY / FREMONT, CALIFORNIA / WAKEFIELD, MASSACHUSETTS / LONGMONT, COLORADO
BATON ROUGE, LOUISIANA



Scott Specialty Gases, Inc.

Shipped From:

2600 CAJON BLVD.
SAN BERNARDINO CA 92411
Phone: 714-887-2571

Fax: 714-887-0549

CERTIFICATE OF ANALYSIS

PACIFIC ENVIRONMENTAL SER
13100 BROOKS DRIVE
BALDWIN PARK CA 91706

PROJECT #: 02-21555
PO#: 4450.039
ITEM #: 02022711 4AL
DATE: 9/11/92

CYLINDER #: AAL7535
FILL PRESSURE: 2000 PSIG

ANALYTICAL ACCURACY: +/-1%

COMPONENT
METHANE
AIR

REQUESTED GAS
CONC MOLES
8. PPM
BAL

ANALYSIS
(MOLES)
7.998 PPM
BAL

PACIFIC ENVIRONMENTAL 9/11

ANALYST:

MJ
MAYNARD JOHNSON

APPROVED BY:

Armand Lange
DR ARMAND LANGE PH D

Scott Specialty Gases, Inc.

Shipped From:

2600 CAJON BLVD.
Phone: 714-887-2571

SAN BERNARDINO
Fax: 714-887-0549

CA 92411-0000

CERTIFICATE OF ANALYSIS

PACIFIC ENVIRONMENTAL SER
ATTN: STEVE HERNANDEZ
13100 BROOKS DRIVE

PROJECT #: 02-18169
PON: 0640-102
ITEM #: 02022712 4AL
DATE: 2/14/92

BALDWIN PARK

CA 91706

CYLINDER #: ALM014380

ANALYTICAL ACCURACY: +/-1%

COMPONENT
METHANE
AIR

REQUESTED GAS
CONC MOLES
80. PPM
BAL

ANALYSIS
(MOLES)
80.00 PPM
BAL

CYLINDER PRESSURE 2000 PSI

ANALYST:

MJ
MAYNARD JOHNSON

APPROVED BY:

ALJ
DR ARMAND LANGE PhD

PLUMSTEADVILLE, PENNSYLVANIA / TROY, MICHIGAN / HOUSTON, TEXAS / DURHAM, NORTH CAROLINA
SOUTH PLAINFIELD, NEW JERSEY / FREMONT, CALIFORNIA / WAKEFIELD, MASSACHUSETTS / LONGMONT, COLORADO
BATON ROUGE, LOUISIANA

Appendix D

C A L I B R A T I O N S H E E T
 - - - - -

Customer : PES
 Date : 3/27/93

Serial : 80669

CALCULATION DATA FOR RUN :	1	2	3
1. Barometric Pressure, P(B) :	30.36	30.36	30.36
2. Orifice Setting, Delta H :	2.00	0.75	6.00
3. Final Reading (Test) :	374.406	385.184	396.448
4. Initial Reading (Test) :	363.568	374.655	385.768
5. Volume, V(T) Cubic Feet :	10.838	10.529	10.680
6. Temp Initial T(T(I)) F :	70	70	69
7. Temp Final T(T(F)) F :	70	69	69
8. Final Reading (Box) :	102.946	113.914	125.252
9. Initial Reading (Box) :	92.000	103.200	114.500
10. Volume, V(B) Cubic Feet :	10.946	10.714	10.752
11. Temp Initial T(B(I)) F :	75	74	76
12. Temp Final T(B(F)) F :	76	75	81
13. Elapsed Time, Minutes :	14.0	22.0	8.0

Delta H(a)	:	1.8278	1.7934	1.8267
Gamma	:	0.9956	0.9902	0.9967

$\overline{\Delta H} = 1.8160$
 $\overline{\gamma} = 0.9942$

Calibration Performed By : Alexander J. Poston



THEMOCOUPLER CALIBRATION

CALIBRATED BY: *Jay Miller*

DATE: 12-31-92

Thermocouple number	Thermocouple reading (°C)	Thermometer reading (°C)
TC-1	0.0	0.0
	23.0	24.0
	98.0	99.0
TC-2	OUT OF SERVICE	
TC-3	0.0	0.0
	23.0	24.0
	97.0	98.0
TC-4	0.0	0.0
	25.5	25.0
	100.5	100.0
TC-5	NOT AVAILABLE	
TC-6	NOT AVAILABLE	
TC-7	NOT AVAILABLE	
TC-8	NOT AVAILABLE	



TC-9

NOT AVAILABLE

TC-10

0.0
25.5
100.5

0.0
25.0
100.0

TC-11

NOT AVAILABLE

C-1

NOT AVAILABLE

C-2

NOT AVAILABLE

S-1A

1.0
22.0
97.0

0.0
23.0
99.0

S-2A

0.0
25.0
98.5

0.0
25.0
100.0

S-16A

1.0
22.0
97.0

0.0
24.0
99.0

S-17A

1.0
22.0
98.0

0.0
24.0
99.0

Thermometer Standard Serial Number: 128239
Thermometer NIST I.D. Number: 88024



PITOT TUBE CALIBRATION DATA SHEET

Calibrated By: Aiza Mah

Date: 2-1-93

Pitot I.D. No.: D-3

Effective Length: 47"

Pitot Tube Assembly Level? Yes No *

Pitot Tube Openings Damaged? Yes (explain below) _____ No _____

$\alpha_1 = \underline{0}^\circ (<10^\circ)$ $\alpha_2 = \underline{0}^\circ (<10^\circ)$

$\beta_1 = \underline{0}^\circ (<5^\circ)$ $\beta_2 = \underline{0}^\circ (<5^\circ)$

$\gamma = \underline{0}^\circ$ $\theta = \underline{0}^\circ$ $A = \underline{0.821}^\circ$

$Z = A \sin \gamma = \underline{\hspace{2cm}} \text{ cm (in.)}$ 0.32 cm (<1/8 in.)

$W = A \sin \theta = \underline{\hspace{2cm}} \text{ cm (in.)}$ 0.08 cm (<1/32 in.)

$P_A = \underline{0.317} \text{ cm (in.)}$

$P_B = \underline{0.316} \text{ cm (in.)}$

$D_c = \underline{0.417} \text{ cm (in.)}$

Comments: * A LITTLE ROUGH AROUND THE EDGES.

Calibration Required? Yes _____ No