



CALPUFF Chemistry Improvements

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Reactive Plume Models

- Two major reactive plume models currently available for impact assessments:
 - CALPUFF (recommended by EPA for PSD and BART analyses)
 - SCIPUFF/SCICHEM (alternative model)

CALPUFF vs. SCICHEM

- CALPUFF
 - 1st order closure diffusion
 - Simple chemistry
 - Computationally efficient
 - Regulatory model with prescribed options for typical applications
- SCICHEM
 - 2nd order closure diffusion
 - Comprehensive chemistry
 - Computationally demanding
 - Alternative model

CALPUFF Chemistry Upgrade- Issues

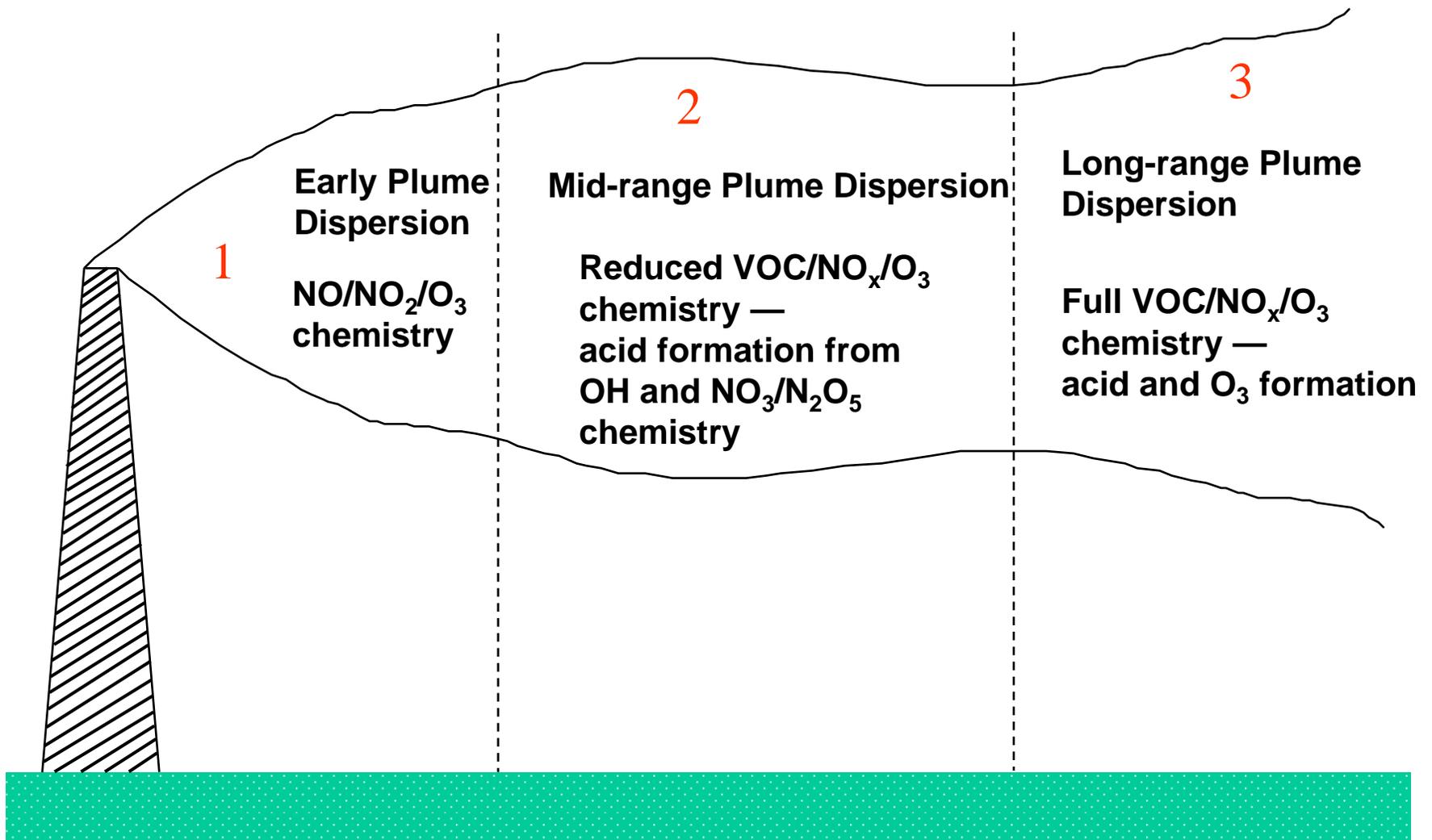
- Gas-phase chemistry highly simplified but difficult to replace with comprehensive chemistry:
 - Significant development effort to implement non-linear chemistry
 - Increases complexity of model
 - Alternative puff models with detailed chemistry already available (e.g., SCICHEM)



CALPUFF Chemistry Upgrade- Approach

- Correct an existing error in the RIVAD gas-phase chemistry option and update RIVAD chemistry rate constants
- Focus on improving treatments for PM formation and cloud chemistry
- All new modules added as new options – none of the previous CALPUFF chemistry options modified

Chemistry of NO_x Plumes

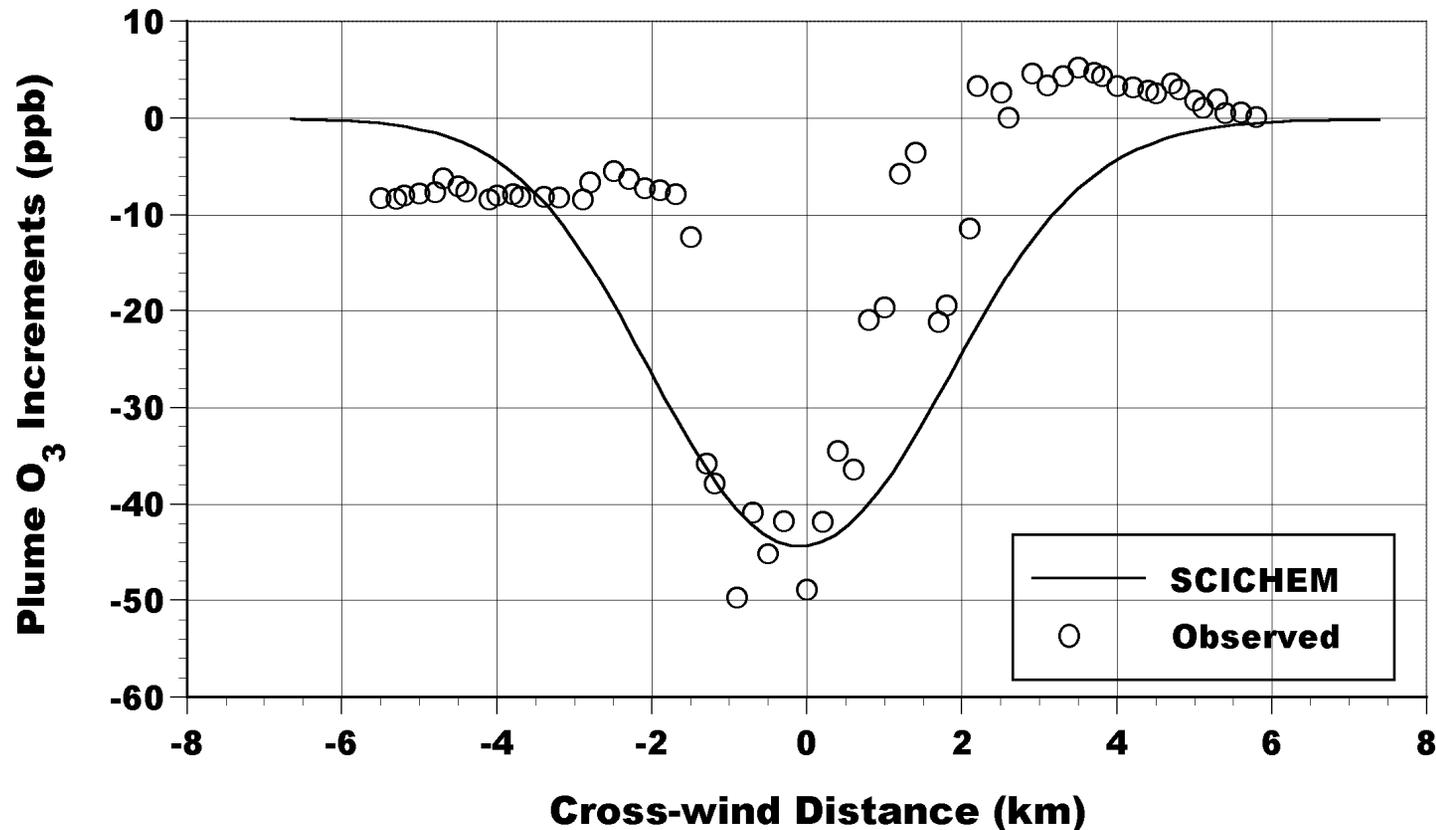


RIVAD Chemistry Option in CALPUFF

- The RIVAD chemistry option in CALPUFF uses Stage 1 chemistry for NO/NO₂/O₃ and part of Stage 2 chemistry for OH and the formation of H₂SO₄ and HNO₃
- O₃ puff concentration in CALPUFF is replenished to its background value at each time step; the code must be corrected to account for the O₃ depletion in the puff in the early stages of plume dispersion
- Correction: Store puff O₃ history and calculate new puff O₃ concentration at each time step

Ozone Depletion in a Fresh Plume

Helicopter measurements of Cumberland power plant plume, July 6, 1999;
downwind distance = 11 km



Original PM Chemistry in CALPUFF

- PM chemistry includes formation of inorganic species (sulfate, nitrate and ammonium) and organic species (secondary organic aerosols, SOA)
- H_2SO_4 and HNO_3 lead to the formation of ammonium sulfate and ammonium nitrate according to a simple gas/particle algorithm that uses a constant NH_3 concentration
- Includes a treatment for the formation of SOA from anthropogenic and biogenic VOCs (developed for Wyoming DEQ)
 - simplified treatment
 - only includes toluene and xylene as anthropogenic SOA precursors



New PM Chemistry in CALPUFF

- Formation of ammonium sulfate and ammonium nitrate is treated with the thermodynamic equilibrium model ISORROPIA
- Inorganic PM formation that is now consistent with that of other operational models (e.g., CMAQ) while retaining computational efficiency
- Formation of SOA includes oxidation of anthropogenic VOCs (aromatics, long-chain alkanes and PAH) by OH to form condensable products, which are partitioned according to Pankow's absorption algorithm (based on MADRID formulation)

Original CALPUFF Cloud Chemistry

- No explicit treatment of aqueous-phase chemistry
- MESOPUFF-II chemistry option uses a simple parameterization to approximate the increased oxidation of SO_2 in presence of clouds or fog:
 - function of relative humidity (RH)
 - may significantly underestimate SO_2 oxidation rates when clouds are present
 - may overestimate SO_2 oxidation when clouds are not present but RH is high



New Cloud Chemistry in CALPUFF

- Based on CMAQ treatment
- Includes SO_2 oxidation by hydrogen peroxide and ozone as well as iron and manganese catalyzed oxidation by oxygen
- Includes gas-aqueous equilibria to calculate liquid-phase concentrations and cloud pH



CALPUFF Upgrade & Testing

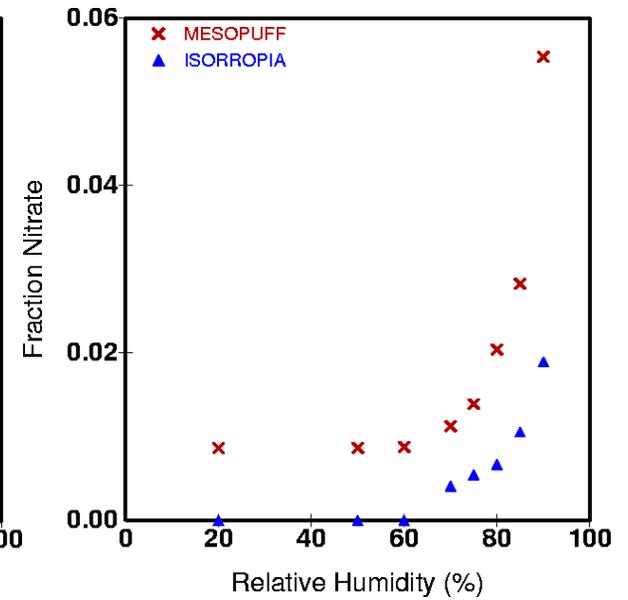
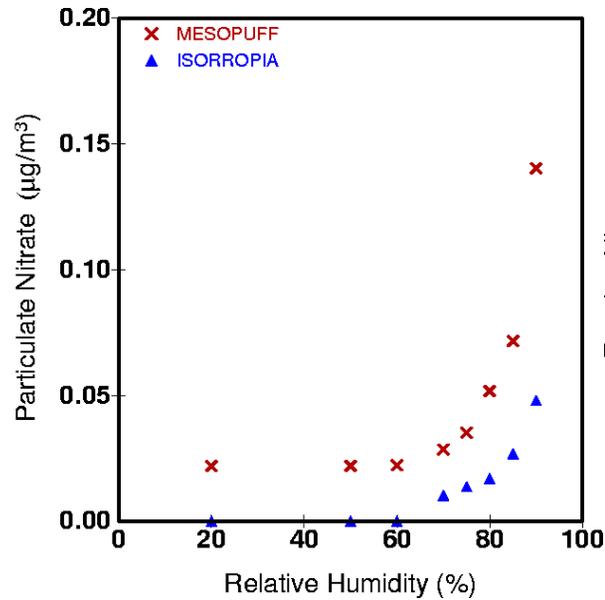
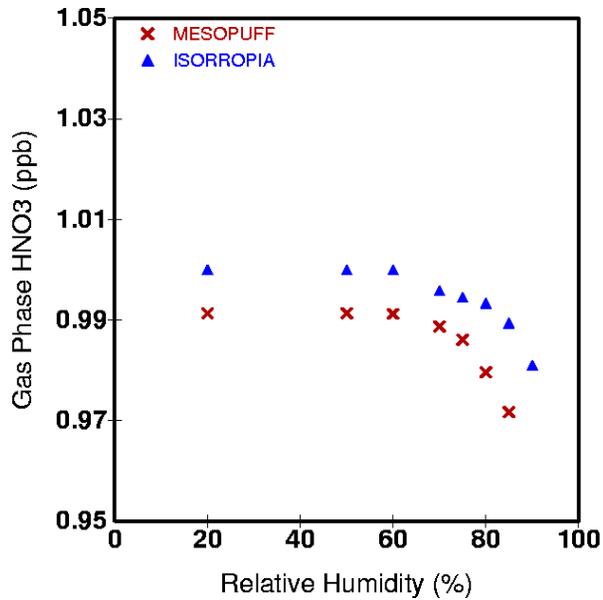
- Gas-phase chemistry corrections and new PM and aqueous-phase chemistry modules incorporated into 2 versions of CALPUFF
 - Unofficial Version 6 (April 2006 release)
 - EPA-approved Version 5.8 (June 2007 release)
- Box-model sensitivity studies with old and new inorganic PM modules
- Original (MCHEM options = 3,4) and new (MCHEM options = 5,6) CALPUFF codes tested using a power plant plume database used in previous SCICHEM/CALPUFF studies



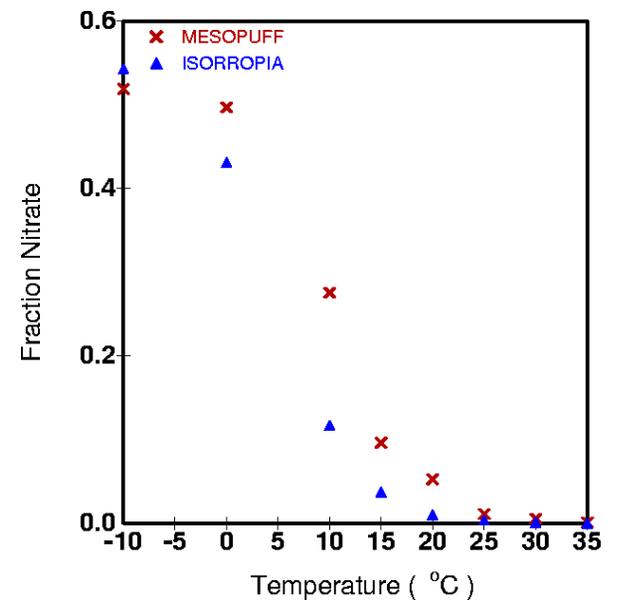
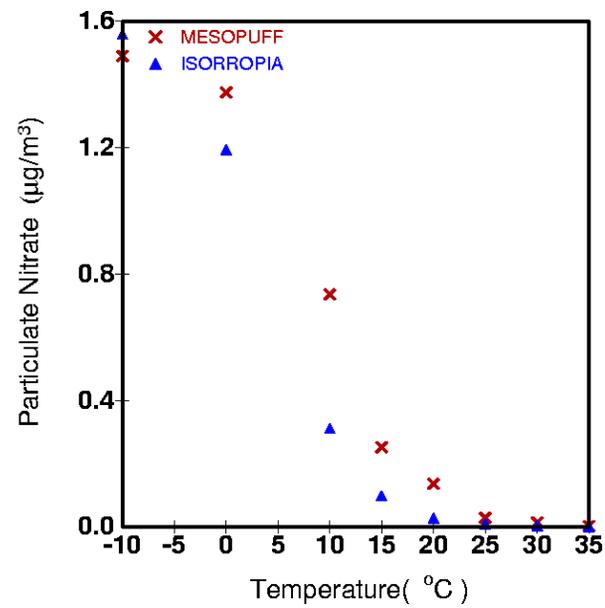
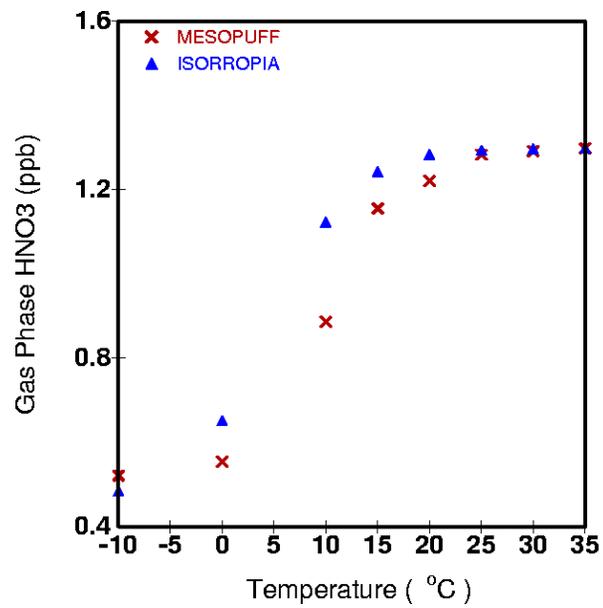
Box-Model Sensitivity Studies with Inorganic PM Modules

- Sensitivity of original CALPUFF module (MESOPUFF) and new CALPUFF module (ISORROPIA) to
 - Relative humidity
 - Temperature
 - Background ammonia
 - Background sulfate
 - Total nitrate

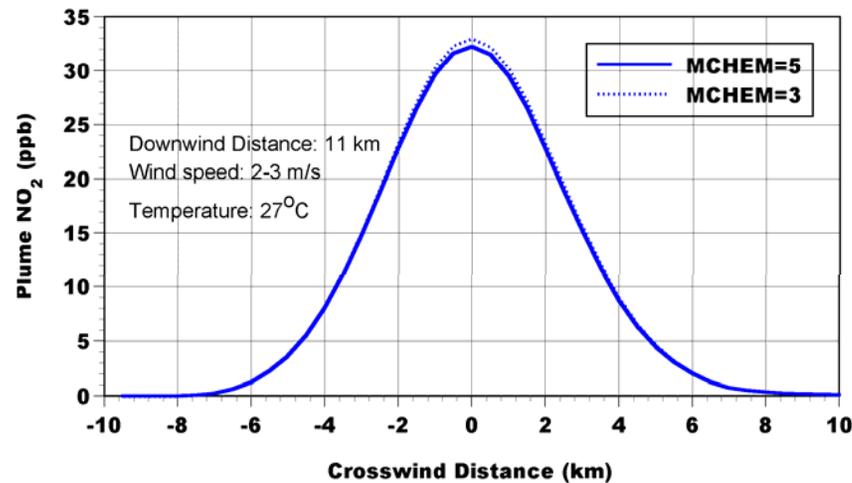
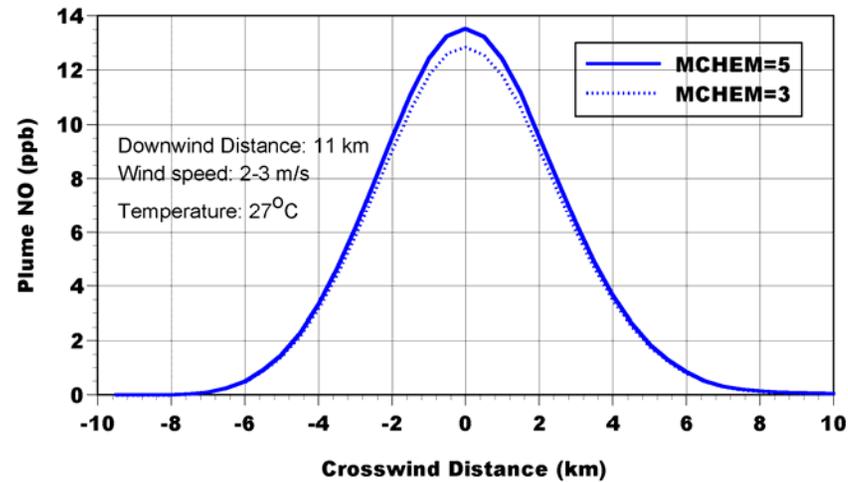
Sensitivity to Relative Humidity



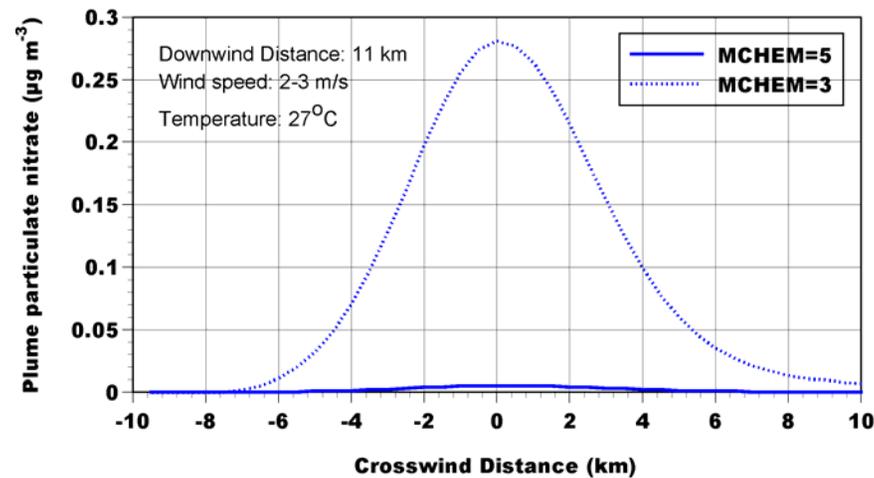
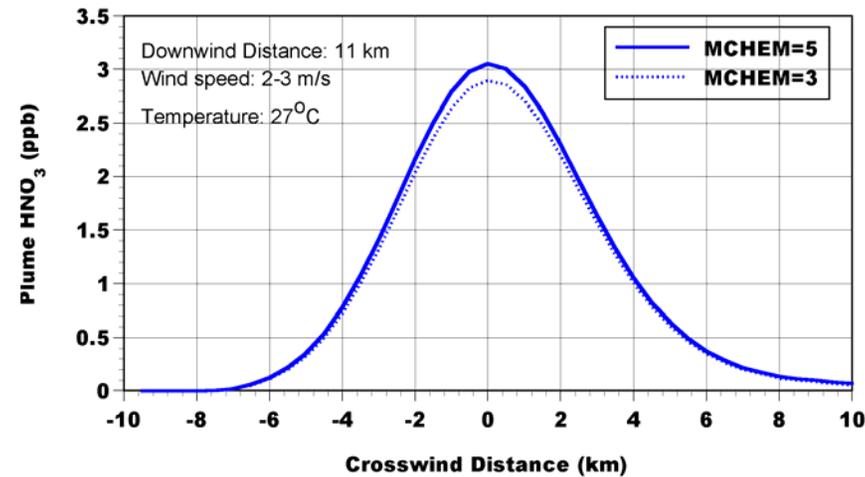
Sensitivity to Temperature



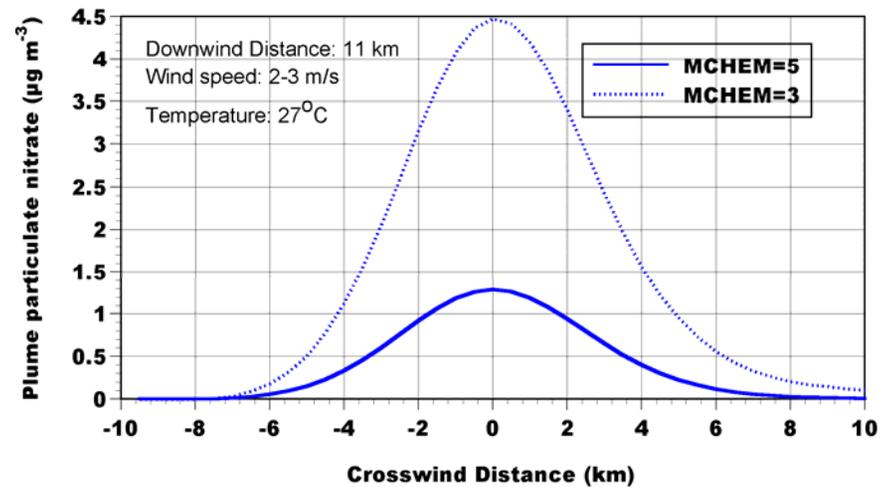
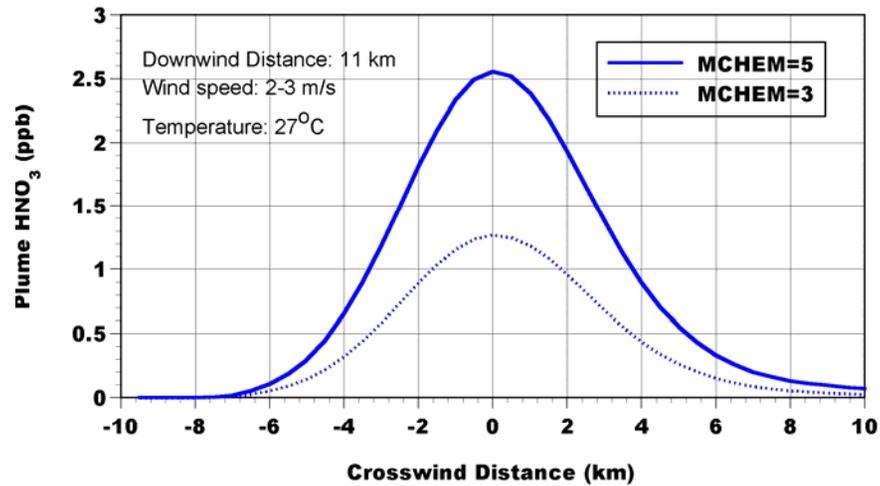
CALPUFF Results Effect of O₃ Correction



CALPUFF Results - Inorganic PM Module Comparison

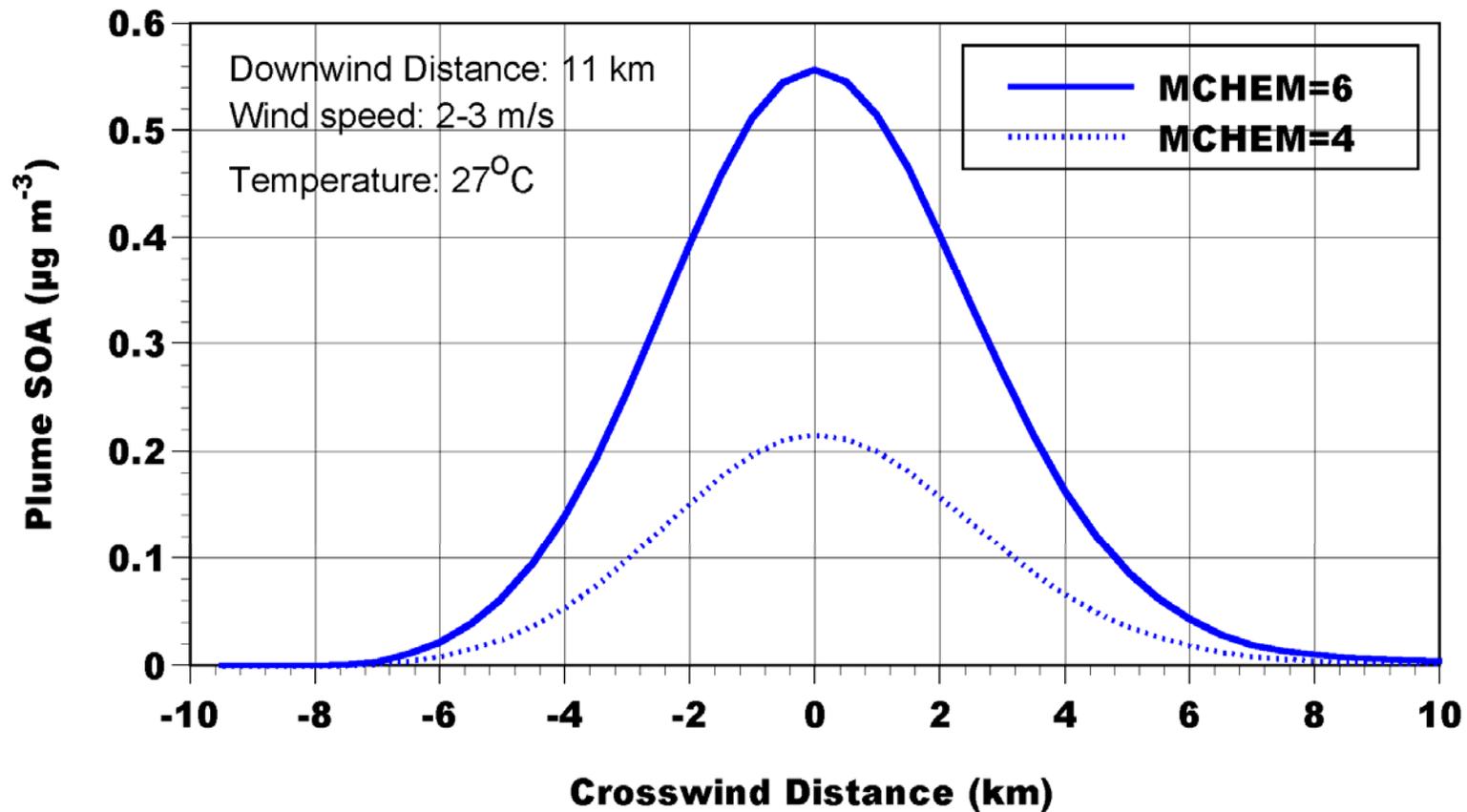


CALPUFF Results - Inorganic PM Module Comparison at High RH (95%)

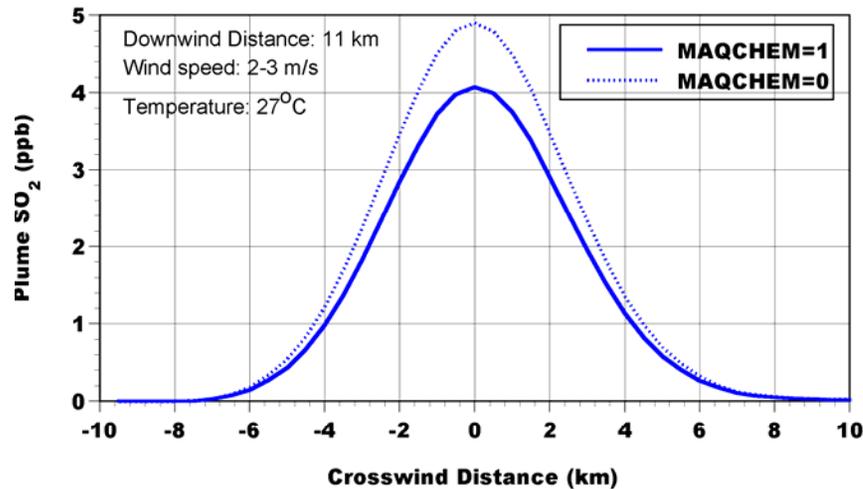




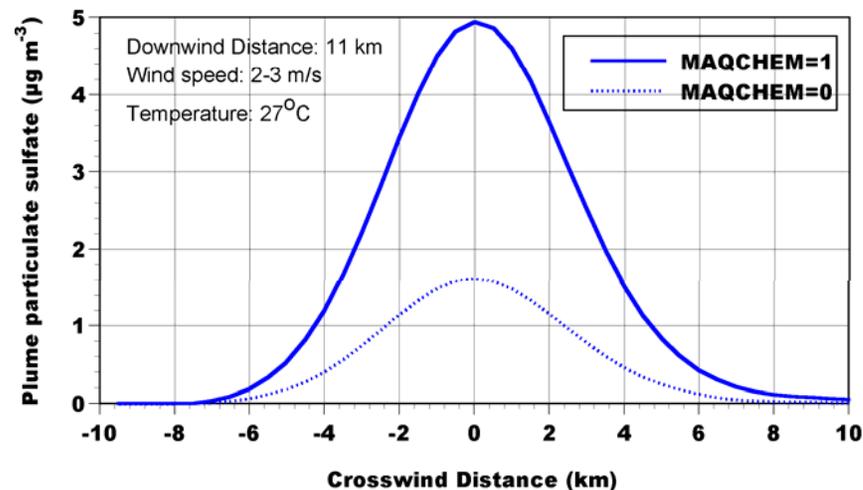
CALPUFF Results - Organic PM (SOA) Module Comparison



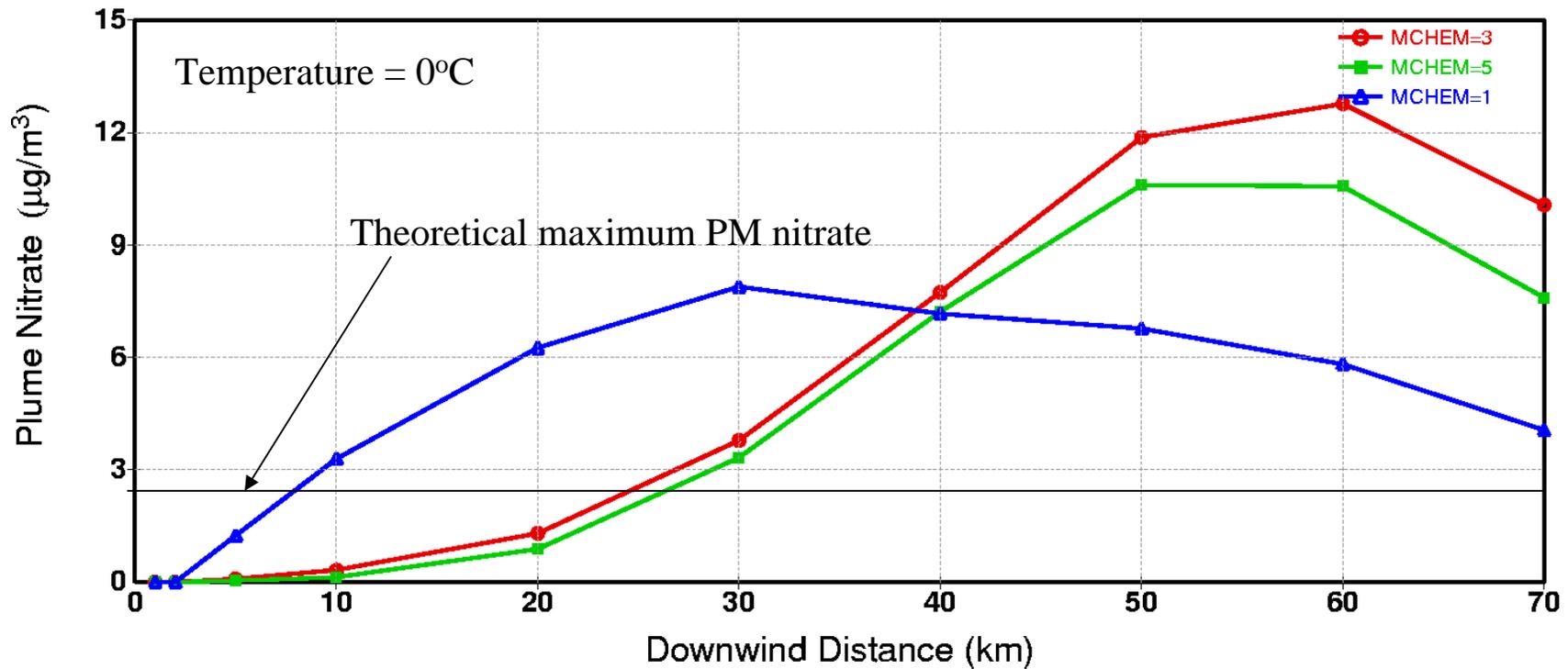
CALPUFF Results Aqueous-Phase Chemistry



Cloud cover and liquid water content hard-coded for testing aqueous-phase chemistry option



Shortcoming in CALPUFF Ammonia Treatment



Ammonia limitation handled in POSTUTIL
(optional CALPUFF post-processing program)



Ongoing Work

- Model is currently being evaluated with the Southwest Wyoming Technical Air Forum (SWWYTAF) data base
- Additional model updates:
 - Update ammonia limitation method in POSTUTIL to use ISORROPIA algorithm
 - Allow vertical profiles in input ammonia concentrations

Recommendations

- Modify CALPUFF to read 3-D grid model outputs
 - More realistic specification of oxidant (OH, O₃, H₂O₂) and ammonia concentrations
 - Temporal and spatial variability
- Include cloud fields in CALMET and CALPUFF for using the aqueous-phase chemistry option
- Additional testing and evaluation of new chemistry options with other data bases

Acknowledgments

- API:
 - Provided funding for this study and ongoing CALMET/CALPUFF evaluation study with SWWYTAF data base
- Wyoming Department of Environmental Quality:
 - Provided SWWYTAF data base for model application and evaluation for ongoing evaluation study