07/16/2004 UNITED STATES ENVIRONMENTAL PROTECTION AGENCY

July 16, 2004

Roger E. Etherington Vulcan Performance Chemicals 1200 Urban Center Drive Birmingham, AL 35242

Subject:

Technical Sodium Chlorite Solution 50

EPA Registration No. 5382-41 Submission Dated : April 20, 2004 Receipt Date: April 27, 2004

Dear Mr. Etherington:

This submission was provided in response to the August 6, 2003, Agency letter.

The following amendment, submitted in connection with registration under the Federal Insecticide, Fungicide, and Rodenticide Act (FIFRA), as amended, is acceptable with the conditions listed below:

Labeling

The Agency letter dated May 21, 2003 for Technical Sodium Chlorite Solution 31.25, mimicked the language required for the Hazards to Humans and Domestic Animals section for this product on August 6, 2003.

Revise the "Hazards to Humans and Domestic Animals" as follows:

Hazards to Humans and Domestic Animals: **Highly** Corrosive. Causes irreversible eye damage and skin burns. Harmful if swallowed. Irritating to nose and throat. Do not get in eyes, on skin or clothing. Wear protective eyewear (goggles or safety glasses). Wear protective clothing and rubber gloves when handling this product. Avoid breathing dust and fumes. Wash thoroughly with soap and water after handling. Remove contaminated clothing and wash clothing before reuse.

Product Chemistry

The Agency requested in the 8/6/03 letter that the declaration of available chlorine be changed from 58% to 29%. According to the **Handbook of Chlorination and Alternative Disinfectants, Fourth Ed.**, the calculation of the available chlorine is as follows:

| CONCURRENCES | | | | |
|-------------------------|-----------------------|--------------------|--|--|
| SYMBOL 7.57CC | | | | |
| SURNAME MITORI | | | | |
| DATE 7-16-04 | | | | |
| EPA Form 1320-1A (1/90) | Drietor no Populari D | OFFICIAL FILE COPY | | |

 $2*(atomic\ mass\ of\ Cl)(percentage\ of\ NaClO_2\ in\ formulation)/molecular\ weight\ of\ NaClO_2=2*(35.5)(0.37)/90.4=0.29$

Therefore, the original calculation for available chlorine identified in the Agency letter dated 8/6/03, is correct and the available chlorine which 29% has been annotated on the label.

General Comments

The basic CSF dated April 20, 2004, is acceptable and supercedes all previously accepted basic formulas. This CSF has been made a part of the file for this product.

Should you have any questions or comments concerning this letter, please contact Wanda Mitchell at (703) 308-6345.

Sincerely,

Emily H. Mitchel

Product Manager - Team 32

Regulatory Management Branch II Antimicrobials Division (7510C)

TECHNICAL SODIUM CHLORITE SOLUTION 50

PRECAUTIONARY STATEMENTS A GYO'N HAZARDS TO HUMANS & DOMESTIC ANIMALS

DANGER. Corrosive. Causes irreversible eye damage and skin burns. Harmful if swallowed. Irritating to nose and throat. Do not get in eyes, on skin or on clothing. Wear protective eyewear (splashproof goggles). Wear protective clothing and rubber gloves when handling this product. Avoid breathing mists or furnes. Wash thoroughly with soap and water after handling. Remove contaminated clothing and wash before reuse to avoid fire.

ENVIRONMENTAL HAZARDS

This product is toxic to fish and aquatic organisms. Do not discharge effluent containing this product into lakes, streams, ponds, estuaries, oceans or other waters unless in cordance with the requirements of a National Pollutant charge Elimination System (NPDES) permit and the permitting authority has been notified in writing prior to the discharge. Do not discharge effluent containing this product to sewer systems without previously notifying the local sewage treatment plant authority. For guidance contact your State Water Board or Regional Office of the EPA.

CHEMICAL HAZARDS

Technical Sodium Chlorite Solution 50 may partially solidify (crystallize) when exposed to temperatures of 70°F (20°C) or below. Dry sodium chlorite is a strong oxidizing agent. This product becomes a fire or explosive hazard if allowed to dry. Mix only into water. Contamination may start a chemical reaction with generation of heat, liberation of hazardous gases (chlorine dioxide a poisonous, explosive gas), and possible fire and explosion. Do not contaminate with garbage, dirt, organic matter, household products, chemicals, soap products, paint products, solvents, acids, vinegar, beverages, oils, pine oil, dirty rags, or any other foreign matter.

DIRECTIONS FOR USE

's a violation of Federal law to use the product in a manner ansistent with its labeling.

Directions for Controlling the Growth of Algae in **Recirculating Cooling Water Towers**

1. Clean badly fouled systems before starting treatment, 2. When algae are visible, add an initial dosage of 4.6 fluid ounces of Sodium Chlorite per 1,000 gals. of water in the system. Repeat if necessary until control is evident. 3. Where algae control is evident, use a subsequent dose of 2.3 fluid ounces of Sodium Chlorite solution per 1,000 gals. of water in the system twice a week or as needed to maintain control. 4. Add Sodium Chlorite directly to the cooling tower drip pan (cold water basin) near the inlet to the recirculating pump.

Directions for Use in the Mechanical Generation of Chlorine Dioxide as a Disinfectant, or for Microorganism or Mollusk Control and as a Chemical Oxidant in Aquatic Systems.

Feed requirements: Feed rates of Technical Sodium Chlorite Solution 50 will depend on the severity of contamination and the degree of control desired. The exact dosage will depend on the size of the system and residual necessary for effective control. Technical Sodium Chlorite Solution 50 is typically diluted at the point of use to prepare a 3% to 25% active aqueous solution for use in chlorine dioxide generators.

| ACTIVE INGREDIENT: Sodium Chlorite* . | | 37% |
|---------------------------------------|--------|----------|
| OTHER INGREDIENTS: | | 63% |
| | Total: | 100% |
| *AVAILABLE CHLORINE | | 4 |

CONTAINS 4.3 LBS. OF SODIUM CHLORITE PER GALLON AT 70°F

KEEP OUT OF REACH OF CHILDREN **DANGER**

| FIRST AID | | |
|----------------------------|--|--|
| If in eyes: | Hold eye open and rinse slowly and gently with water for 15-20 minutes. Remove contact lenses, if present, after the first 5 minutes, then continue rinsing eye. Call a poison control center or doctor immediately for treatment advice. | |
| If on skin or clothing: | Take off contaminated clothing. Rinse skin immediately with plenty of water for 15-20 minutes. Call a poison control center or doctor for treatment advice if burning or irritation of the skin persists. | |
| If swallowed: | Have person drink a glass of water immediately if able to swallow. Call a poison control center or doctor immediately for treatment advice. Do not induce vomiting unless told to do so by the poison control center or doctor. Do not give anything by mouth to an unconscious person. | |
| If inhaled: | Move person to fresh air and monitor for respiratory distress. If cough or difficulty in breathing develops, consult a | |

physician immediately.
If person is not breathing, call 911 or an ambulance, then give artificial respiration.

Call a poison control center or doctor for further treatment advice

For emergency information call: 800-835-2030 (24 hours)
Have the product container or label with you when calling a poison control center or doctor or going to treatment.

NOTE TO PHYSICIAN:

Probable mucosal damage may contraindicate the use of gastric lavage.

Manufactured By:



PO Box 385015, Birmingham, Alabama 35238-5015 amended, for the pesticide,

ACCEPTED with COMMENTS **EPA** Letter Dated:

JUL 16 2004

Under the Federal Insecticide, A BUSINESS UNIT OF VULCAN MATERIALS COMPANY Fungicide, and Rodenticide Act as registered under EPA Reg. No. 5382-41

| PA Reg. No. 5382-41 | | EPA Est. 5382-KS-1 |
|---------------------|-------------|--------------------|
| | Gals. Net (| 1) |

ome examples of industrial applications of chlorine oxide include:

- · Potable water disinfection and removal of sulfide.
- Control of bacterial slime and algae and mollusks in industrial recirculating and one-pass cooling systems.
- Biocontrol in food processing flumes, water-using equipment, cooling water, and recycled waters.
- · Disinfection of sewage and plant wastes.
- Destruction of phenolics, simple cyanides and sulfides by chemical oxidation.
- Bacterial slime control in white water paper mill systems.
- · Bacterial control in oil well and petroleum systems.

ee product bulletins (or Technical Data Sheets) for peci**- application instructions. Your Vulcan pro ative can guide you in the application chilques.

ethod of feed: Large amounts of chlorine dioxide can be enerated by two common methods, including:

- The chlorine method which utilizes a Sodium Chlorite solution and chlorine gas, or
- The hypochlorite method which utilizes a Sodium Chlorite solution, a hypochlorite solution, and an acid.

our Vulcan representative can guide you in the selection, stallation and operation for feed systems. Consult oduct bulletin and also the instructions on the chlorine oxide generation system before using Technical Sodium hlorite Solution 50.

ser is responsible for compliance with applicable Federal, ate and local laws regarding proper use and disposal of le chlorine dioxide generated.

/ater Treatment

hlonne dioxide (ClO₂) is used as both an oxidant and a isinfectant in drinking water treatment. The required osages will vary with source water conditions and the egree of contamination present. For most municipal and ublic potable water systems, a chlorine dioxide residual oncentration of up to 2 ppm is sufficient to provide dequate disinfection. Residual disinfectant and isinfection byproducts must be monitored as required by the National Primary Drinking Water Regulations (40 CFR art 141) and state drinking water standards.

idustrial Cooling Water Treatment

or control of bacterial same and algae in industrial edirculating and one-pass cooling systems, the required osages will mary depending on the exact application and be degree of contamination present. The required chlorine lioxide residual concentrations range between 0.1 and 5.0 pm. Chlorine dioxide may be applied either continuously intermittently. The typical chlorine dioxide residual concentration range is 0.1 - 1.0 ppm for continuous doses, and 0.1 - 3.0 ppm for intermittent doses. The minimum occeptable residual concentration of chlorine dioxide is 0.1 pm for a minimum one minute contact time.

Mollusk Control in Water Systems

Chlorine dioxide generated from sodium chlorite may be used for mollusk control in commercial and industrial recirculating and one-pass cooling water systems. The required dosages will vary with the system type, system conditions, the degree of water contamination present and the desired level of control. Depending on the extent of the infestation, sodium chlorite may be applied either continuously or intermittently through a chlorine dioxide generating system to achieve the necessary chlorine dioxide residual concentration.

<u>Veliger Control</u>: Maintain a continuous chlorine dioxide residual of 0.1 - 0.5 ppm.

Intermittent Dose: Apply chlorine dioxide to obtain a chlorine dioxide residual concentration of 0.2 - 25 ppm. Repeat as necessary to maintain control.

Continuous Dose: Maintain a chlorine dioxide residual concentration of up to 2 ppm.

Food Plant Process Water Treatment

Chlorine dioxide generated from sodium chlorite is effective for use in controlling microbiological growth in flume water and other food processing water systems such as chill water systems and hydrocoolers. The required dosages will vary with process conditions and the degree of contamination present. Depending on the requirements of the specific water system, sodium chlorite should be applied continuously or intermittently through a chlorine dioxide generating system to achieve a chlorine dioxide residual concentration between 0.25 and 5.0 ppm.

Water, containing up to 3 ppm residual chlorine dioxide may be used for washing fruits and vegetables that are not raw agricultural commodities in accordance with 21CFR§173.300. Treatment of the fruits and vegetables with chlorine dioxide must be followed by a potable water rinse, or by blanching, cooking or canning.

Wastewater Treatment

Chlorine dioxide (CIO₂) is effective as both a disinfectant and an oxidant in wastewater treatment. The required dosages will vary with water conditions and the degree of contamination present. For most municipal and other wastewater systems, a chlorine dioxide residual concentration of up to 5 ppm is sufficient to provide adequate disinfection.

For sulfide odor control, between pH 5-9, a minimum of 5.2 ppm (wt) of chlorine dioxide should be applied to oxidize 1 ppm of sulfide (measured as sulfide ion). For phenol destruction, at pH less than 8, 1.5 ppm chlorine dioxide will oxidize 1 ppm phenol; at pH greater than 10, 3.3 ppm chlorine dioxide will oxidize 1 ppm phenol.

Bacterial Slime Control in Paper Mills

Chlorine dioxide generated from sodium chlorite is effective for use in controlling microbiological growth in white water paper mill systems. The required dosages will vary with the degree of microbiological and process contamination present. Depending on the specific requirements of the system, sodium chlorite should be applied continuously or intermittently through a chlorine dioxide generating system to achieve a chlorine dioxide residual concentration between 0.1 and 5.0 ppm. Intermittent treatments should be repeated as often as necessary to maintain control.

Bacterial Control in Oil Wells And Petroleum Systems

Chlorine dioxide is effective in the remediation of bacterial and sulfide contamination commonly found in oiffield production, injection and disposal fisids. The required dosages will vary with process conditions. Scdium chlorite may be applied either continuously or intermittently through a chlorine dioxide generating system to oil well production water as it is separated from the oil, and before it is re-injected into the well.

For continuous feeds, chlorine dioxide may be applied at dosages slightly higher than sulfide's oxidative demand as determined by a demand study. For intermittent treatment, chlorine dioxide should be applied at a shock dosage of 200 - 3000 ppm.

STORAGE AND DISPOSAL

STORAGE: Do not contaminate water, food or feed by storage or disposal. Keep product in tightly closed container when not in use. Don't drop, roll or skid drum. Keep upright. Always replace cover. Store in a cool, dry well-ventilated area away from heat or open flame.

EMERGENCY HANDLING: In case of contamination or decomposition, do not reseal container. If possible, isolate container in open and well ventilated area. Flood with large volumes of water. If fire occurs, extinguish fire by applying large quantities of water. Any unopened drums near the fire should be cooled by spraying with water.

PESTICIDE DISPOSAL: Pesticide wastes are acutely hazardous. Improper disposal of excess pesticide, spray mixture or rinsate is a violation of Federal Law. If these wastes cannot be disposed of by use according to label instructions, contact your State Pesticide or Environmental Control Agency, or the Hazardous Waste Representative at the nearest EPA Regional Office for guidance.

CONTAINER DISPOSAL: Triple rinse container. Then offer for recycling or reconditioning, or puncture and dispose of in a sanitary tandfill, or incineration, or, if allowed by state and local authorities, by burning. If burned, stay out of smoke.

ACCEPTED with COMMENTS FOR Letter Dated:

JUL 16 2004 VMC-6200 04/04

Under the Federal Insecticide, Fungicide, and Rodenticide Act as amended, for the pesticide, registered under EPA Reg. No. 5382—41