		Shaughnessy #: 079407				
	-	EFB Logout Date: 18 APR 1984				
		Init.: # 5				
To:	George La Rocca Product Manager #15 Registration Division (TS-767)	A state of the sta				
From:	L.A. Richardson, Chief Environmental Chemistry Review S Exposure Assessment Branch Hazard Evaluation Division (TS-	Section #3				
Attach	ed please find the EAB review of	• • •				
Reg./F	ile No.: 11678-5					
Chemica	al: Endosulfan					
Type P	roduct: I	ti destretamente transcribita e anteriore, anticonomo, e proquente proprieta de partir de partir de partir de p				
Produc	t Name: Makhteshim-Agan Inc.					
Company	y Name: Response to RS (Adsorpt	ion/Desorption Study)				
Submis	sion Purpose:					
ZBB Coo	de:	ACTION CODE: 655				
Date In	n: 2/28/84	EFB # 4212				
Date Co	ompleted: 4/16/84	TAIS (level II) Days				
Deferra	als To:	1				
·	Ecological Effects Branch	Reviewer: Patricia Ott				
	Residue Chemistry Branch	Catriva Ott				
	Toxicology Branch					
		,				

Endosulfan Adsorption/Desorption Study

Reference:

Hoe 002671, adsorption/desorption in the soil/water system, submitted by Makhteshim-Agan Inc. Corporation. EPA Reg. No. 11678-5, cover memo dated January 9, 1984.

Conclusions:

This study partially fulfills the leaching requirement by providing information for adsorption/desorption for the parent compound, endosulfan. Endosulfan binds strongly to two sands (K_d 29-72) and a sandy loam (K_d = 33) containing from 0.8% to 2.58% organic carbon. Desorption was < 3%.

This study does not completely fulfill the registration requirement for the following reasons:

- 1. Workers failed to provide adsorption coefficients ($K_{\mbox{\scriptsize d}}$) for degradates.
- 2. No $K_{\mbox{\scriptsize d}}$ values for parent and degradates were provided for an aquatic sediment, which is required for the aquatic food use (watercress).

Materials and Methods:

Four concentrations of radiolabeled endosulfan in a 0.01 M CaCl₂ solution were equilibrated with each of three soils: two sands and a sandy loam. After centrifugation, the water phase was analyzed by liquid scintillation counting.

Reported Results:

Kd	Soil/Water (22°C)	K _{OC} Soil/Water (22°C)				
	20.10					
Sand (0.8% org. C	29 <u>+</u> 8 :)	3600 <u>+</u> 1000				
Sand (2.58% org.	72 <u>+</u> 20 C)	2800 <u>+</u> 800				
Sandy Loam	33 <u>+</u> 7	3300 <u>+</u> 700				

Concentration range: 0.04 umol-1.04 umol/1

Desorption: <3% K_{des} 7xK)

Discussion:

- 1. Two of the three soils selected were sands. Sand is an acceptable soil for one soil, as well as the sandy loam, which was the third soil studied. Instead of a second sand, another soil type, such as clay or clay loam should have been chosen. However, since endosulfan adsorbs appreciably to sand, this objection is not critical for endosulfan.
- 2. It is not good laboratory practice to evaporate pesticide samples to dryness because losses can occur, unless oil/fat is present.

REGISTRATION DIVISION DATA REVIEW RECORD Confidential Business Information — Does Not Contain National Security Information (E.O. 12065)										56 74 }	
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