

# ANNEX 5 Assessment of the Sources and Sinks of Greenhouse Gas Emissions Not Included

Although this report is intended to be a comprehensive assessment of anthropogenic<sup>97</sup> sources and sinks of greenhouse gas emissions for the United States, certain sources have been identified but not included in the estimates presented for various reasons. Before discussing these sources, however, it is important to note that processes or activities that are not *anthropogenic in origin* or do not result in a *net source or sink* of greenhouse gas emissions are intentionally excluded from a national inventory of anthropogenic greenhouse gas emissions, in line with guidance from the IPCC in their guidelines for national inventories.

Given a source category that is both anthropogenic and results in net greenhouse gas emissions, reasons for not including a source related to an anthropogenic activity include one or more of the following:

- Though an estimating method has been developed, data were not adequately available to calculate emissions.
- Emissions were implicitly accounted for within another source category (e.g., CO<sub>2</sub> from Fossil Fuel Combustion).

It is also important to note that the United States believes that the sources discussed below are very low in comparison with the overall estimate of total U.S. greenhouse gas emissions, and not including them introduces a very minor bias. In general, the emission sources described in this annex are for source categories with methodologies introduced in the 2006 IPCC Guidelines for which data collection has not been sufficient to pursue an initial estimation of greenhouse gases.

## **N<sub>2</sub>O from Caprolactam Production**

Caprolactam is a widely used chemical intermediate, primarily to produce nylon-6. All processes for producing caprolactam involve the catalytic oxidation of ammonia, with N<sub>2</sub>O being produced as a by-product.. More research is required to determine this source's significance because there is currently insufficient information available on caprolactam production to estimate emissions in the United States.

## **CO<sub>2</sub> and CH<sub>4</sub> from Calcium Carbide Production**

CO<sub>2</sub> is formed by the oxidation of petroleum coke in the production of calcium carbide. These CO<sub>2</sub> emissions are implicitly accounted for in the storage factor calculation for the non-energy use of petroleum coke in the Energy chapter. There is currently not sufficient data on coke consumption for calcium carbide production to estimate emissions from this source.

## **CO<sub>2</sub> from Graphite Consumption in Ferroalloy and Steel Production**

Emissions from "graphite," "wood," or "biomass" in calculating CO<sub>2</sub> emissions from ferroalloy production, iron and steel production or other "Industrial Processes" included in Chapter 4 of the inventory are not explicitly calculated. It is assumed that 100 percent of the C used in ferroalloy production is derived from petroleum coke and that all of the C used in iron and steel production is derived from coal coke or petroleum coke. It is also assumed that all of the C used in lead and zinc production is derived from coal coke. It is possible that some non-coke C is used in the production of ferroalloys, lead, zinc, and iron and steel, but no data are available to conduct inventory calculations for sources of C other than petroleum coke and coal coke used in these processes.

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<sup>97</sup> The term "anthropogenic," in this context, refers to greenhouse gas emissions and removals that are a direct result of human activities or are the result of natural processes that have been affected by human activities ("2006 IPCC Guidelines for National Greenhouse Gas Inventories").

1 Non-fuel uses of coal coke and petroleum coke are accounted for in the Industrial Process chapter, either directly  
2 for iron and steel, aluminum, ferroalloy, lead, zinc, and titanium dioxide production, or indirectly by applying a storage  
3 factor to "uncharacterized" non-fuel uses of petroleum coke and coal coke. Non-fuel uses of wood and biomass are not  
4 accounted for in the Energy or Industrial Process chapters, as all uses of wood and biomass are accounted for in the Land  
5 Use, Land-Use Change, and Forestry chapter. It is assumed for the purposes of the CO<sub>2</sub> emission calculation that no wood  
6 or other biogenic C is used in any of these industrial processes. Some biogenic C may be used in these industrial  
7 processes but sufficient data to estimate emissions are not available.

8 Consumption of either natural or synthetic graphite is not explicitly accounted for in the Industrial Process  
9 chapter. It is assumed that all of the C used in manufacturing C anodes for production of aluminum, ferroalloys, and  
10 electric arc furnace (EAF) steel are derived directly from petroleum coke and coal tar pitch (a coal coke byproduct), not  
11 from natural graphite or synthetic graphite sources. Some amount of C used in these industrial processes may be derived  
12 from natural or synthetic graphite sources, but sufficient data to estimate emissions are not currently available.

### 13 **Miscellaneous SF<sub>6</sub> Uses**

14 Sulfur hexafluoride (SF<sub>6</sub>) is used in several applications for which estimates have not been provided in this  
15 inventory. Sulfur hexafluoride may be emitted from the production, leakage, and dismantling of radar, tracer, and night  
16 vision equipment. Emissions from this source are believed to be minor, and no data were available for estimating the  
17 emissions. Sulfur hexafluoride may be used in foam insulation, for dry etching, in laser systems, for indoor air quality  
18 testing, for laboratory hood testing, for chromatography, in tandem accelerators, in loudspeakers, in shock absorbers, and  
19 for certain biomedical applications. Emissions from this source are believed to be minor, and no data were available for  
20 estimating the emissions. Sulfur hexafluoride may be emitted from the production, breakage, or leakage of soundproof  
21 double-glazed windows. Emissions from this source are believed to be minor, and no data were available for estimating  
22 the emissions. Sulfur hexafluoride may be emitted from applications involving the production of sport shoes, tires, and  
23 tennis balls. Emissions from this source are believed to be minor, and no data were available for estimating the emissions.  
24 Sulfur hexafluoride may be emitted from applications involving tracer gasses to detect leakage from pressure vessels and  
25 as a tracer gas in the open air. Emissions from this source are believed to be minor, and no data were available for  
26 estimating the emissions.

### 27 **CO<sub>2</sub> from Non-Hazardous Industrial Waste Incineration and Medical Waste Incineration**

28 Waste incineration is incorporated in two sections of the energy chapter of the inventory: in the section on CO<sub>2</sub>  
29 emissions from waste incineration, and in the calculation of emissions and storage from non-energy uses of fossil fuels.  
30 The former section addresses fossil-derived materials (such as plastics) that are discarded as part of the municipal  
31 wastestream and combusted (generally for energy recovery). The latter addresses two types of combustion: hazardous  
32 waste incineration of organic materials (assumed to be fossil-derived), in which regulated wastes are burned without  
33 energy recovery, and burning of fossil-derived materials for energy recovery. There additional categories of waste  
34 incineration that is not included in our calculus: industrial non-hazardous waste and medical waste incineration. Data are  
35 not readily available for these sources. Further research is needed to estimate the magnitude of CO<sub>2</sub> emissions, though they  
36 are believed to be very low in comparison with the overall emissions of waste incineration sources that are covered.