Measurements of Atmospheric NH₃, NOₓ/NO₂, and NO₂ and Deposition of Total Nitrogen at the Beaufort, NC CASTNET Site (BFT142)
Measurements of Atmospheric NH$_3$, NO$_y$/NO$_x$, and NO$_2$ and Deposition of Total Nitrogen at the Beaufort, NC CASTNET Site (BFT142)

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U.S. Environmental Protection Agency
Office of Research and Development
Washington, DC 20460
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1.0 Introduction

a. Background
The Clean Air Status and Trends Network (CASTNET) is a long-term environmental monitoring program that measures trends in ambient air quality and atmospheric dry pollutant deposition across the United States. CASTNET has been operating since 1987 and currently consists of 89 monitoring stations nationwide (Figure 1). The Environmental Protection Agency (EPA) operates a majority of the CASTNET sites. The National Park Service (NPS) currently operates 24 stations in cooperation with EPA. Other federal and state agencies participate in network operation. For example, the Bureau of Land Management (BLM) currently operates five CASTNET sites in Wyoming. For more information on CASTNET, see EPA’s CASTNET website: [http://java.epa.gov/castnet/](http://java.epa.gov/castnet/) (EPA, 2014).

Figure 1. CASTNET Monitoring Sites

b. Purpose of Study
CASTNET has a 25-year record of weekly atmospheric nitrogen, sulfur, and ozone measurements collected at rural/remote sites in the United States. CASTNET initially featured approximately 50 sites mostly in the eastern United States but has grown to currently include over 90 sites across the contiguous United States and Alaska. The ambient nitrogen measurements taken at CASTNET sites are nitric acid (HNO$_3$), particulate nitrate (NO$_3^-$), and ammonium (NH$_4^+$). In addition, particulate sulfate

1
(SO₂), sulfur dioxide (SO₃), and ozone concentrations are measured. CASTNET also produces estimated dry deposition fluxes of these compounds. However, some key contributors to the atmospheric nitrogen budget have not been measured. For example, ammonia (NH₃) and total reactive oxidized nitrogen (NOₓ) had not been measured routinely. To remedy this, many CASTNET sites began participating in the Ammonia Monitoring Network (AMoN) (Figure 2), which was initiated by the National Atmospheric Deposition Program (NADP) in 2007 to establish a nationwide network of passive NH₃ monitors. For more information, view the website: http://isws.illinois.edu/amon (NADP, 2013).

A few commercial off-the-shelf instruments are available to measure NOₓ (NO + NO₂ + NO₃) and NOx (NO + NO₂). NO₂ includes the nitrogen species HNO₃, nitrous acid (HNO₂), peroxyl acetyl nitrate (PAN), other organic nitrates, and particulate nitrates. In practice, NO₂ is calculated as NOₓ - NOx. Since 2005, CASTNET has measured NOₓ at Beltsville, MD (BEL116). During 2012, NOₓ measurements were added at Bondville, IL (BVL130) and Huntington Wildlife Forest, NY (HWF187). In 2013, the CASTNET NOₓ network further expanded to include Pinedale, WY (PND165), Cranberry, NC (PNF126), and Rocky Mountain National Park, CO (ROM206). Figure 3 displays the CASTNET sites with NOₓ measurements.

CASTNET has also operated other nitrogen measurement studies in the past several years to supplement and enhance the filter pack and AMoN measurements, including two projects at the Beaufort, NC site (BFT142) located in an agricultural area of coastal North Carolina. The results from these two studies are presented in this report.

**Figure 2. AMoN Monitoring Sites**
The first Beaufort study included NH₃ measurements based on the deployment of a weekly, dual annular denuder system (ADS) during designated measurement weeks. Additionally, the Beaufort ammonia study operated daily ADS sampling for comparison with the weekly integrated ADS sample. The daily ADS ran for a 24-hour period to collect standard CASTNET constituents and ammonia. Measurements were conducted from May 2011 through November 2012, operating on a two out of every six weeks schedule. The Beaufort studies also included combining model estimates of dry deposition velocities with the measured concentrations to estimate dry deposition of the nitrogen and sulfur species. The dry deposition fluxes were added to measured wet deposition rates to obtain total deposition. The results of the NH₃ measurement study are presented in Section 3. Estimates of deposition of atmospheric nitrogen and sulfur are presented in Section 4.0.

The second study at Beaufort focused on measuring NOₓ continuously from February 2012 through May 2013. Measurements of trace-level NOₓ were obtained using a commercially produced instrument with an optional configuration using a second converter that allowed for the collection of NOₓ in addition to NOₓ and NO. Using a difference method (NOₓ - NO), an NO₂ concentration was obtained. Similarly, NO₂ was calculated as the difference between measured NOₓ and measured NOₓ. These results are presented in Section 5.0. Section 6.0 presents a summary of the ADS and continuous measurements and estimates of nitrogen and sulfur deposition.
2.0 Site Characteristics and Monitoring Equipment: Beaufort, NC (BFT142)

The CASTNET monitoring site is located about 10 miles north of the Town of Beaufort, NC and is situated on western edge of Open Grounds Farm (OGF), a massive 57,000-acre farm northeast of Beaufort. OGF produces soybeans, corn, and some wheat and cotton. The sidebar summarizes the BFT142 site characteristics. Figures 4 and 5 illustrate the location of the monitoring site on Google aerial photographs. The NO\textsubscript{y} inlet with a NO\textsubscript{y} converter was located atop a 10-m tower, which is shown in Figure 6 along with the ADS, filter pack, and meteorological sampling towers. Figure 7 shows the NO\textsubscript{y} converter at 10 meters. A NO\textsubscript{x} converter was located in the shelter adjacent to the NO\textsubscript{y} instrument. Figure 8 shows the BFT142 site instruments. Figure 9 illustrates the NO-NO\textsubscript{x}-NO\textsubscript{y} sampling configuration. The BFT142 CASTNET site also hosts a NADP/NTN wet deposition monitoring system (NC06) and an AMoN (NC06) passive NH\textsubscript{3} monitoring device. An OGF-operated fertilizer and chemical plant is located about three miles east-southeast of BFT142 on Nelson Bay Road.

![Small Scale Google Aerial Photograph of BFT142, NC](image)
Figure 5. Large Scale Google Aerial Photograph of BFT142, NC and Open Grounds Farm

Figure 6. BFT142 Site Including NO$_y$ Tower

Note: 1 NADP/NTN precipitation sampling systems
2 AMoN sampler
3 Meteorological tower
4 NO$_y$ converter tower
5 Filter pack and ADS tower
The goals of the Beaufort studies were to continue and expand analyses that began during the Ammonia CASTNET CSN Study (ACCS; EPA, in press), and to add NO\textsubscript{y}-related measurements in an effort to estimate the components of total dry nitrogen deposition from nitrogen species not collected by the typical CASTNET filter pack. The Beaufort location expanded ACCS to include a coastal site located in an area of interest for ammonia monitoring because of its proximity to agricultural/crop production “i.e.”, OGF and animal production “e.g.”, Prestage Farms [“pork” and “turkey”] in eastern North Carolina.
Sampling Equipment

- Dual Denuder ADS for Measuring NH₃ and Other Species

The heart of the NH₃ sampling system consisted of nine dual denuder ADS (Figure 10) with a 3-stage filter pack arranged to collect seven daily samples, one field blank, and one weekly sample.

ADS featured:

1. A sodium carbonate (Na₂CO₃)-coated annular denuder for collecting gaseous HNO₃ and SO₂ analyzed by ion chromatography (IC) using deionized water (DIW) as the extraction fluid.
2. A phosphorus acid (H₃PO₃)-coated annular denuder for collecting gaseous NH₃ analyzed by automated colorimetry (AC) using DIW as the extraction fluid.
3. A 3-stage CASTNET-style filter pack consisting of:
   a. A Teflon filter for collecting SO₄²⁻, NO₃⁻, and NH₄⁺ analyzed by IC.
   b. A nylon filter for collecting SO₂ (as SO₄²⁻) and HNO₃ (as NO₃⁻) analyzed by IC. No measurable SO₂ was collected on the nylon filter.
   c. A H₃PO₃-impregnated filter for collecting any extra NH₄ either from denuder capture inefficiency or loss of NH₄⁺ through volatilization off the nylon filter (or both), which was analyzed by AC using DIW as the extraction fluid.

The ADS were installed and operated at 10 meters. AMEC personnel created a solenoid system and a data logger program that permitted unattended daily and weekly sampling. The BFT142 ADS was the same as used in ACCS. The ADS used a cyclonic separator for removing particles with mean aerodynamic diameter greater than 2.5 µm. The weekly ADS used a flow rate of 3.0 liters per minute (lpm) and the daily sampling was based on a flow of 16.7 lpm.

**Figure 10. Annular Denuder System (ADS) and Related Sampling Systems at BFT142**
The sampling schedule (Figure 11) was similar to ACCS:

- Daily ADS ran for one week, then again five weeks later.
- Weekly ADS ran for one week in conjunction with daily ADS then a second consecutive week for comparison with AMoN.
- ADS sampling occurred from May 2011 through November 2012.

**Figure 11. Sampling Schedule**

<table>
<thead>
<tr>
<th>CASTNET Filter Pack</th>
<th>Daily ADS</th>
<th>Weekly ADS</th>
<th>AMoN</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>1</td>
<td>2</td>
<td>3</td>
</tr>
<tr>
<td></td>
<td>4</td>
<td>5</td>
<td>6</td>
</tr>
<tr>
<td></td>
<td>7</td>
<td>8</td>
<td></td>
</tr>
</tbody>
</table>

**Figure 12. Radiello NH₃ Sampler**
• **AMoN Radiello Sampler for Measuring NH₃**

A Radiello NH₃ passive sampler was deployed as part of the NC06 AMoN sampling to collect 2-week NH₃ concentrations (Figure 12). The Radiello sampler was chosen by NADP for use in the AMoN based on comparisons with other samplers and annular denuder measurements (Puchalski et al., 2011).

• **Three-stage CASTNET filter pack**

A 3-stage CASTNET filter pack was deployed for each sampling period as part of the base operations of the core CASTNET program. The filter pack sampled gaseous SO₂ and HNO₃, particulate SO₄²⁻, NO₃⁻, and NH₄⁺; base cations; and chloride ion (Cl⁻). The CASTNET filter pack system did not utilize a cyclonic separator to remove large particles (mean aerodynamic diameter greater than 2.5 µm).

• **NO/NOx/NOy/ Measurements**

Trace-level NO₃ measurements were obtained using a Thermo Scientific (Thermo) Model 42i-Y NOy analyzer, which is a chemiluminescent trace-level gas analyzer that measures all reactive oxides of nitrogen in the ambient air. AMEC deployed an optional configuration at BFT142 using a second converter that permitted the collection of NOₓ in addition to NOy and NO. Thus NO₂ could be calculated using a difference method (NOₓ - NO). Continuous NOy measurements began in February 2012 and were completed in May 2013. The configuration produces hourly concentrations of NOy, NOₓ, and NO and calculated NO₂, and calculated NO₂. All measurements and model calculations made by AMEC during these studies followed the CASTNET Quality Assurance Project Plan (QAPP; AMEC, 2012).

• **Supporting Measurement Systems**

Hourly O₃ concentrations were collected at BFT142 as part of routine CASTNET operations. The continuous gaseous measurements were supported by a dynamic dilution calibration system and a zero air system. The sampling by all instruments was controlled by a Campbell Scientific Instruments (CSI) CR3000 data logger.

• **Meteorological Measurements**

Meteorological instruments were operated on a 10-m tower at the BFT142 site. The measurements included wind speed and direction, sigma theta (standard deviation of the wind direction), solar radiation, relative humidity, temperature at 9 m and 2 m, precipitation, and surface wetness. The meteorological measurements were archived as 1-hour averages. The data were used as input to the Multi-Layer Model (MLM) to estimate values of deposition velocity \( V_d \) (Meyers, 1998). The \( V_d \) values were used to estimate dry deposition of HNO₃, particulate NOₓ, particulate NH₄⁺, SO₂, and particulate SO₄²⁻. All meteorological measurements followed the CASTNET QAPP.
3.0 Results of the NH₃ Measurement Study

a. Overview

Figure 13 presents time series of daily, weekly, and daily-averaged-to-weekly NH₃ concentrations from BFT142 measurements. The figure also presents 2-week average NH₃ concentrations collected using the AMoN Radiello passive sampler at the collocated NC06 site. The data in this figure provide a good synopsis of the BFT142 19-month NH₃ measurement program. ADS measurements of weekly NH₃ concentrations were higher in the spring and summer of 2011. Table 1 summarizes the NH₃ measurements collected during the 19-month study. The weekly ADS NH₃ concentrations were typically higher than the average of the seven daily samples. There was no consistent bias between the 2-week average ADS concentrations and the AMoN samples. The average of weekly (twice per five weeks) BFT142 NH₃ values was 1.15 µg/m³, and the average (2-week values) for NC06 was 1.00 µg/m³. The weekly 2-week average BFT142 ADS values ranged from 0.43 to 2.89 µg/m³ with a standard deviation of 0.84.

Figure 13. Time Series of Measured NH₃ Concentrations (µg/m³)

Figures 14 and 15 provide maps of 2011 and 2012 annual mean NH₃ concentrations measured by AMoN. In 2011, nine sites measured annual mean concentrations greater than or equal to 2.0 µg/m³. The highest concentration (3.2 µg/m³) was measured at AMoN site CA83/CASTNET SEK430, CA. The 2011 mean concentration for NC06 was 2.3 µg/m³, which was higher than the mean AMoN value collected only during the BFT142 sampling period. The annual mean value at NC06 was affected by a concentration of 32.38 µg/m³, which was measured during the two weeks beginning 4/12/2011. Without the outlier, the annual mean concentration is 1.2 µg/m³. Twelve sites
measured concentrations greater than or equal to 2.0 µg/m³ in 2012. The highest 2012 concentration (15.1 µg/m³) was measured at Logan, UT (UT01), a site that was added to AMoN during late 2011 and is not included on Figure 14. NC06 measured a mean concentration of 0.7 µg/m³ during 2012.

b. **Comparison of ADS and AMoN NH₃ Concentrations**

The 2-week average of the weekly BFT142 values and the 2-week NC06 values shown in Figure 13 and Table 1 compare reasonably well. A linear regression based on ADS values aggregated to two weeks versus the 2-week AMoN concentrations is shown in Figure 16. The regression equation is

\[ y = 0.7792x + 0.0963, \]

with an R² value of 0.8178 where x = average weekly ADS concentration and y = AMoN bi-weekly concentration.

c. **Comparison of 24-Hour and Weekly ADS NH₃ Concentrations**

Similarly, the averages ("x" in Figure 13) of the seven 24-hour NH₃ concentrations compare well to the sampled weekly values (Figure 13). A linear regression based on the ADS values and corresponding 10 averages of the 24-hour concentrations is shown in Figure 17. The regression equation is

\[ y = 1.0527x - 0.1998, \]

with an R² value of 0.9265 where x = average weekly ADS concentration and y = average hourly ADS concentration.

**Table 1. 1-Week and 2-Week NH₃ Concentrations (µg/m³)**

<table>
<thead>
<tr>
<th>Date On</th>
<th>Average of 7 24-Hour ADS Values</th>
<th>Average of 7 Weekly ADS Values</th>
<th>AMoN 2-Week Sample</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td>1st Week Only</td>
<td>2nd Week Only</td>
<td>2-Week Average</td>
</tr>
<tr>
<td>05/10/11</td>
<td>2.54</td>
<td>2.38</td>
<td>NA</td>
</tr>
<tr>
<td>06/21/11</td>
<td>1.99</td>
<td>2.18</td>
<td>3.61</td>
</tr>
<tr>
<td>08/02/11</td>
<td>0.77</td>
<td>0.82</td>
<td>1.64</td>
</tr>
<tr>
<td>11/09/11</td>
<td>0.42</td>
<td>0.51</td>
<td>0.36</td>
</tr>
<tr>
<td>12/06/11</td>
<td>0.35</td>
<td>0.60</td>
<td>0.37</td>
</tr>
<tr>
<td>03/27/12</td>
<td>0.93</td>
<td>0.89</td>
<td>0.61</td>
</tr>
<tr>
<td>05/08/12</td>
<td>0.61</td>
<td>0.70</td>
<td>0.56</td>
</tr>
<tr>
<td>07/31/12</td>
<td>0.41</td>
<td>0.93</td>
<td>0.51</td>
</tr>
<tr>
<td>09/11/12</td>
<td>1.04</td>
<td>1.42</td>
<td>1.39</td>
</tr>
<tr>
<td>10/23/12</td>
<td>0.60</td>
<td>0.64</td>
<td>0.78</td>
</tr>
<tr>
<td>Mean</td>
<td>0.97</td>
<td>1.11</td>
<td>1.09</td>
</tr>
<tr>
<td>Standard Deviation</td>
<td>0.73</td>
<td>0.67</td>
<td>1.04</td>
</tr>
</tbody>
</table>

*Note: NA = not available*
Figure 14. AMoN Annual Mean NH₃ Concentrations (µg/m³) for 2011

Figure 15. AMoN Annual Mean NH₃ Concentrations (µg/m³) for 2012
Figure 16. NH$_3$ Concentrations Collected by BFT142 ADS and NC06 AMoN Samplers

\[ y = 0.7792x + 0.0963 \]
\[ R^2 = 0.8178 \]

Figure 17. NH$_3$ Concentrations Collected by BFT142 ADS Weekly and 24-Hour Samplers

\[ y = 1.0527x - 0.1998 \]
\[ R^2 = 0.9265 \]
d. Comparison of ADS and CASTNET Filter Pack Concentrations

In addition to the NH₃ concentrations discussed previously, the ADS, including its back-end filter pack component, and the CASTNET filter pack (FP) sampling system measured concentrations of gaseous HNO₃ and SO₂ and particulate SO₄²⁻, NO₃⁻, and NH₄⁺. HNO₃ and NO₃⁻ were summed to produce total NO₃ concentrations.

- **Particulate SO₄²⁻**

Because the measurement of atmospheric SO₄²⁻ is considered relatively straightforward, it represents a good quality control check for the measurement system. Satisfactory results indicate the likely success in measuring the other parameters. Figure 18 presents a time series of weekly ADS, daily ADS, daily averaged to weekly ADS, and weekly FP SO₄²⁻ data. Qualitatively, the four data sets show good agreement. A scatter plot of ADS versus FP weekly SO₄²⁻ concentration measurements is given in Figure 19 and provides a quantitative estimate of the relationship between the ADS and FP SO₄²⁻ data. The regression equation fit through the data is

\[ y = 0.9718x + 0.4025, \]

with an R² value of 0.8190.

Other than a couple of minor outliers, the comparisons are good over an order of magnitude range of SO₄²⁻ concentrations with a high correlation coefficient.

**Figure 18. Time Series of Measured SO₄²⁻ Concentrations (µg/m³)**
Figure 19. \( \text{SO}_2^{\text{A}} \) Concentrations (\( \mu \text{g/m}^3 \)) Measured by Weekly ADS/FP and CASTNET FP

![Graph showing the relationship between CASTNET FP Concentration and Weekly ADS Concentration.](image)

\[ y = 0.9718x + 0.4025 \]

\[ R^2 = 0.819 \]

Figure 20. \( \text{SO}_2^\text{A} \) Concentrations (\( \mu \text{g/m}^3 \)) Collected by ADS Weekly and 24-Hour Samplers

![Graph showing the relationship between Average Daily ADS Concentration and Weekly ADS Concentration.](image)

\[ y = 1.0339x - 0.0807 \]

\[ R^2 = 0.9384 \]
The weekly ADS SO$_4^{2-}$ measurements were also compared to the 24-hour concentrations aggregated over a week. The values ("x" in Figure 18) of the 10 24-hour SO$_4^{2-}$ concentration averages compare well to the sampled weekly values. A linear regression based on the ADS values and corresponding 10 averages of the 24-hour concentrations is shown in Figure 20. The regression equation is

\[ y = 1.0339x - 0.0807, \]

with an R$^2$ value of 0.9384.

Instead of presenting figures for all of the weekly ADS and 24-hour measurements aggregated over a week, Table 2 summarizes the linear regression statistics for SO$_4^{2-}$, NH$_3$, SO$_2$, NO$_3$, HNO$_3$, NH$_4^+$, and total NO$_3$ with x as the average weekly ADS concentration and y as the average hourly ADS concentration. The weekly and aggregated 24-hour concentrations compared reasonably well with the exception of HNO$_3$ and total NO$_3$. In these cases, the weekly ADS values were higher than the aggregated 24-hour measurements.

<table>
<thead>
<tr>
<th>Pollutant</th>
<th>Slope</th>
<th>Y-Intercept</th>
<th>Correlation</th>
</tr>
</thead>
<tbody>
<tr>
<td>SO$_4^{2-}$</td>
<td>1.0339</td>
<td>-0.0807</td>
<td>0.9384</td>
</tr>
<tr>
<td>NH$_3$</td>
<td>0.7792</td>
<td>0.0963</td>
<td>0.8178</td>
</tr>
<tr>
<td>SO$_2$</td>
<td>0.8099</td>
<td>0.0139</td>
<td>0.8192</td>
</tr>
<tr>
<td>NO$_3$</td>
<td>1.0298</td>
<td>0.0002</td>
<td>0.9413</td>
</tr>
<tr>
<td>HNO$_3$</td>
<td>1.3034</td>
<td>-0.0438</td>
<td>0.7577</td>
</tr>
<tr>
<td>NH$_4^+$</td>
<td>1.0337</td>
<td>0.0207</td>
<td>0.8865</td>
</tr>
<tr>
<td>Total NO$_3$</td>
<td>1.4705</td>
<td>0.6181</td>
<td>0.6609</td>
</tr>
</tbody>
</table>

- **SO$_2$**

Time series of daily, weekly, and daily-averaged-to-weekly SO$_2$ concentrations from BFT142 measurements are presented in Figure 21. The figure also presents weekly average SO$_2$ concentrations collected using the CASTNET filter pack at BFT142.

A scatter plot of ADS versus CASTNET FP weekly SO$_2$ concentration measurements is given in Figure 22. The regression equation fit through the data is

\[ y = 0.7332x + 0.1706, \]

with an R$^2$ value of 0.7862.
The scatter diagram of weekly ADS and CASTNET FP measurements shows reasonably good comparisons over the range of SO\textsubscript{2} concentrations. However, the SO\textsubscript{2} data show more scatter and a lower correlation coefficient than the SO\textsubscript{2} data.

**Figure 21. Time Series of Measured SO\textsubscript{2} Concentrations (µg/m\textsuperscript{3})**

**Figure 22. SO\textsubscript{2} Concentrations (µg/m\textsuperscript{3}) Measured by Weekly ADS/FP and CASTNET FP**
- Particulate NO$_3$

Figure 23 presents time series of daily, weekly, and daily-averaged-to-weekly particulate NO$_3$ concentrations from BFT142 measurements collected over a 19-month period. The figure also presents weekly average NO$_3$ concentrations collected using the CASTNET filter pack at BFT142.

Figure 24 presents a scatter plot of ADS versus CASTNET FP weekly NO$_3$ concentration measurements. The regression equation fit through the data is

$$y = 1.0262x + 0.5914,$$

with an $R^2$ value of 0.1750.

**Figure 23. Time Series of Measured NO$_3$ Concentrations (µg/m$^3$)**
The scatter diagram shows that, although the slope of the linear regression is near 1.0, the correlation coefficient is low indicating that the distribution is random. In contrast, the results for particulate SO$_2$ have a much higher correlation coefficient of 0.819 with a similar slope, primarily because SO$_2$ particles exist in the atmosphere in the submicron range. The results for NO$_3$ were likely caused by the lack of a cyclone separator on the CASTNET filter pack system and the subsequent sampling of large NO$_3$ particles (Lavery et al., 2009) in the form of NH$_4$NO$_3$ in the farm environment.

- **HNO$_3$**

Figure 25 presents time series of daily, weekly, and daily-averaged-to-weekly HNO$_3$ concentrations from BFT142 measurements collected over a 19-month period. The figure also presents weekly average HNO$_3$ concentrations collected using the CASTNET filter pack at BFT142.

Figure 26 presents a scatter plot of ADS versus CASTNET FP weekly HNO$_3$ concentration measurements. The regression equation fit through the data is

\[ y = 1.3034x - 0.0438, \]

with an $R^2$ value of 0.7577.
Figure 25. Time Series of Measured HNO₃ Concentrations (µg/m³)

Figure 26. HNO₃ Concentrations (µg/m³) Measured by Weekly ADS/FP and CASTNET FP
The scatter diagram shows good comparisons over the range of HNO₃ concentrations with a high correlation coefficient, even though the regression line suggests HNO₃ levels measured by the filter pack are typically higher than those measured by the ADS.

- **Particulate NH₄⁺**

Time series of daily, weekly, and daily-averaged-to-weekly NH₄⁺ concentrations from BFT142 ADS measurements are displayed in Figure 27. The figure also presents weekly average NH₄⁺ concentrations collected using the CASTNET filter pack at BFT142.

Figure 28 presents a scatter plot of ADS versus CASTNET FP weekly NH₄⁺ concentration measurements. The regression equation fit through the data is

\[ y = 0.3177x + 0.3153, \]

with an \( R^2 \) value of 0.2687.

**Figure 27. Time Series of Measured NH₄⁺ Concentrations (µg/m³)**
The scatter diagram shows poor comparisons over the range of $\text{NH}_4^+$ concentrations with a weak correlation coefficient. $\text{NH}_4^+$ concentrations are typically collected as $\text{NH}_4\text{NO}_3$ in the atmosphere. Consequently, any sampling issues with $\text{NO}_3^-$ concentrations on the CASTNET filter pack will affect $\text{NH}_4^+$ measurements. In particular, the results for $\text{HNO}_3$, $\text{NH}_4^+$ and $\text{NO}_3^-$ were likely caused by volatilization of $\text{NH}_4\text{NO}_3$ particles following collection on the Teflon filter. Volatilization results in some loss of $\text{NH}_4^+$ from the Teflon filter as $\text{NH}_3$. The volatilized $\text{NO}_3^-$ is then collected as $\text{HNO}_3$ on the nylon filter (Lavery et al., 2009).

- **Total $\text{NO}_3^-$**

Figure 29 shows time series of daily, weekly, and daily-averaged-to-weekly total $\text{NO}_3^-$ concentrations from BFT142 ADS measurements. The figure also presents weekly average total $\text{NO}_3^-$ concentrations collected using the CASTNET FP at BFT142. The time series plots show the FP estimates were consistently higher than the ADS/FP measurements over the lifetime of the sampling program.

Figure 30 presents a scatter plot of ADS versus CASTNET FP weekly total $\text{NO}_3^-$ concentration measurements. The regression equation fit through the data is

$$ y = 1.4705x + 0.6181,$$

with an $R^2$ value of 0.6609.
Figure 29. Time Series of Measured Total NO$_3$ Concentrations ($\mu$g/m$^3$)

![Time Series of Measured Total NO$_3$ Concentrations ($\mu$g/m$^3$)](image)

Figure 30. Total NO$_3$ Concentrations ($\mu$g/m$^3$) Measured by ADS/FP and CASTNET FP

![Total NO$_3$ Concentrations ($\mu$g/m$^3$) Measured by ADS/FP and CASTNET FP](image)

y = 1.4705x + 0.6181  
$R^2 = 0.6609$

Similar to the results for NO$_3$ sampling, the higher total NO$_3$ concentrations on the CASTNET FP were likely caused by the lack of a cyclone separator on the filter pack system and the subsequent sampling of large NO$_3$ particles (Lavery et al., 2009) in the environment around OGF.
4.0 Deposition of Atmospheric Nitrogen and Sulfur

Estimates of dry atmospheric deposition of nitrogen and sulfur species were obtained from measured concentrations and Multi-Layer Model (MLM)-modeled hourly deposition velocities \( V_d \), with the exception of \( V_d \) for NH\(_3\). The dry flux of NH\(_3\) was estimated from the concentrations measured at BFT142, NC and a \( V_d \) value of 0.81 centimeters per second (cm/sec) selected from a 2002 Community Multiscale Air Quality Modeling System (CMAQ) model run (EPA, 2011). The MLM calculations of \( V_d \) were based on onsite meteorological measurements and land use data approximated for an area within a 1 km radius around the BFT142 site. Wet deposition was based on NADP/NTN onsite measurements using the seven seasonal deposition values covering the study period averaged and scaled to the year.

Figure 31 provides a pie chart that illustrates the estimates of individual component contributions to total nitrogen (as N) deposition in kilograms per hectare per year (kg/ha/yr). Flux data used in the pie chart are taken from the entire study period and scaled to one calendar year. The small component labeled as being NH\(_x\) is from the H\(_3\)PO\(_3\)-impregnated backup filter and likely represents either NH\(_3\) gas not captured by the denuder or particulate NH\(_4^+\) that volatilized from the Teflon filter and was collected by the backup filter. Figure 32 shows the contributions to total sulfur (as S) deposition. The species contributions to total deposition and their percentages are summarized in Table 3. The dominant contributors to total nitrogen deposition at this site are dry NH\(_3\) deposition, which is particular notable because NH\(_3\) has not been routinely measured at CASTNET sites prior to inclusion of AMoN measurements, wet NH\(_4^+\) deposition, and wet NO\(_3^-\) deposition. The largest contributor to total sulfur deposition is wet SO\(_4^{2-}\) deposition.

**Figure 31. Components of Total Nitrogen Deposition (kg/ha/yr)**

![Pie chart showing contributions to total nitrogen deposition](image)
Figure 32. Components of Total Sulfur Deposition (kg/ha/yr)

Total S Deposition = 3.8 kg/ha/yr

Table 3. Contributions to Total Deposition

<table>
<thead>
<tr>
<th>Nitrogen (kg/ha/yr) as N</th>
<th>Sulfur (kg/ha/yr) as S</th>
</tr>
</thead>
<tbody>
<tr>
<td>Dry</td>
<td>Percent</td>
</tr>
<tr>
<td>HNO₃</td>
<td>0.48</td>
</tr>
<tr>
<td>NH₃</td>
<td>2.28</td>
</tr>
<tr>
<td>NH₄⁺</td>
<td>0.19</td>
</tr>
<tr>
<td>NHₓ</td>
<td>0.10</td>
</tr>
<tr>
<td>NO₃⁻</td>
<td>0.03</td>
</tr>
<tr>
<td>Wet</td>
<td></td>
</tr>
<tr>
<td>NH₄⁺</td>
<td>1.92</td>
</tr>
<tr>
<td>NO₃⁻</td>
<td>1.73</td>
</tr>
<tr>
<td>Total</td>
<td>6.74</td>
</tr>
</tbody>
</table>
5.0 Results of the NO$_y$ Measurement Program

a. Overview

The Beaufort Study collected hourly average NO, NO$_x$, and NO$_y$ concentrations from February 2012 through May 2013. The measurements were collected using a Thermo 42i-Y analyzer with two converters. One converter was located at 10 m for NO$_y$ and the second at the analyzer for NO$_x$. For every minute of data collection, 20 seconds were dedicated to the NO measurement, 20 seconds to a pre-reactor measurement, and 10 seconds each to NO$_y$ and NO$_x$ measurements. Performance of the data logger programs was verified at the AMEC field laboratory before deployment to BFT142. NO$_x$ was calculated as the difference between NO$_x$ and NO. All concentration data were archived as hourly averages.

b. Field Evaluation and Quality Assurance

Appendix 11 of the CASTNET QAPP describes the field evaluation methods used to control and check the operation of the trace-level gas instruments (AMEC, 2012). This appendix covers site operations, data collection, and QC requirements. All supplies used to conduct monitoring and sampling meet the specifications of EPA and Teledyne Advanced Pollution Instrumentation (TAPI). The calibration gases used for the gaseous criteria monitors are traceable to the National Institute of Standards and Technology (NIST) and are protocol gases. The analyzers are mounted inside a secure trailer, which is temperature controlled.

Zero, span, and precision (ZSP) checks of the NO$_y$ analyzers are performed automatically every other day based on the concentration levels listed in Table 4. TAPI multigas calibration and zero air systems, and protocol gas cylinders produce the calibration gases, and the CR3000 data logger controls the process. CASTNET data analysts review data, including NO$_y$ data, for the previous day for all EPA-sponsored CASTNET sites. A data analyst will note questionable values and enter observations per site into the CASTNET Problem Tracking System. Precision and span checks are judged successful if the results are within ±10 percent of the test values. Zeroes must be within 3 percent of full scale values. If the ZSP results exceed the criteria, data are invalidated back to the previous successful ZSP and forward to the next passing ZSP. Troubleshooting is performed to determine root cause, and appropriate corrective action is implemented. Instrument calibration is performed only if troubleshooting reveals that it is necessary.

Field calibrations are critical to achieving and maintaining data quality indicator (DQI) criteria. Every six months, AMEC or subcontractor technicians visit each site to perform routine calibration and maintenance of all sensors and instruments. AMEC personnel may calibrate the NO$_y$ sensors independent of the routine calibration visit. The concentration levels listed in Table 4 are used for the multipoint calibrations, again using the TAPI multigas calibration and zero air systems and the CR3000 data logger.
Table 4. Quality Control Checks for NCore Analyzers

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Zero (ppb)</th>
<th>Span (ppb)</th>
<th>Precision (ppb)</th>
</tr>
</thead>
<tbody>
<tr>
<td>SO₂</td>
<td>± 3.0</td>
<td>90 ± 10%</td>
<td>25 ± 10%</td>
</tr>
<tr>
<td>NO/NO₂/NPN</td>
<td>± 3.0</td>
<td>90 ± 10%</td>
<td>15 ± 10%</td>
</tr>
<tr>
<td>CO</td>
<td>± 40.0</td>
<td>1800 ± 10%</td>
<td>250 ± 10%</td>
</tr>
</tbody>
</table>

Notes: The n-propyl nitrate (NPN) QC checks are performed exclusively at span level at a frequency of every fourth day. Span concentrations are 90 percent of full scale. If the ZSP results exceed criteria, data are invalidated back to the previous successful ZSP and forward to the next passing ZSP.

<table>
<thead>
<tr>
<th>Parameter</th>
<th>Calibration Concentration Levels (ppb)</th>
</tr>
</thead>
<tbody>
<tr>
<td>SO₂</td>
<td>90 40 15 7 4 0.0</td>
</tr>
<tr>
<td>NO/NO₂/NPN</td>
<td>90 40 15 7 4 0.0</td>
</tr>
<tr>
<td>CO</td>
<td>1800 800 300 150 80 0.0</td>
</tr>
</tbody>
</table>

c. Data Evaluation and Completeness

Section 4.0 and Appendix 11 (NCore Air Monitoring Equipment) of the CASTNET QAPP describe the data validation procedures implemented for all CASTNET measurements. These apply to NO/NO₂ data as well. In general, continuous measurements undergo three levels of validation.

After daily polling of all CASTNET stations, Level 1 validation procedures are initiated. Level 1 validation consists of a set of automated screening protocols that initiate the transfer of data between tables, the translation of data status flags and data screening, and they create the data template, generate reports on the completeness of the data and the results of data screening and archive the data. Data status flags are listed in Table 4-8 in Section 4.0 of the CASTNET QAPP.

The purpose of Level 2 validation is to develop a complete database. The process involves a data analyst reviewing data at the end of a month and retrieving missing data using LoggerNet data collection software. Essentially, this step represents a double check of the daily review process. Level 2 validation is complete when the data for all time periods for all of the sampling sites have been accounted for, data have been recovered from the on-site data loggers and entered into the database, and sources of missing data are documented.

Level 3 validation involves a more detailed evaluation of the data. The site status report form (SSRF), site narrative log sheets, ZSP data, calibration data, and audit results are reviewed for each site. In addition, data are screened using tools that identify potential problems such as values greater than the expected range and invalid combinations of status flags, values, and spikes. All review and editing activities are documented both electronically and on hard copy forms.
Perhaps the most important assessment of NO/NO\textsubscript{x}/NO\textsubscript{y} measurements is the review of ZSP checks. If the ZSP results exceed criteria, troubleshooting is performed to determine root cause and appropriate corrective action is implemented. Unscheduled instrument calibration is performed as a corrective action only if the review reveals it is necessary. Routine 6-month field calibrations are also critical to achieving and maintaining DQI criteria. Results from the ZSP checks and the 6-month and unscheduled calibrations are used to assess NO/NO\textsubscript{x}/NO\textsubscript{y} data validity.

All BFT NO\textsubscript{y} and related data underwent the three levels of validation discussed above. These validation steps resulted in 87 percent valid data. Another criterion used specifically for the BFT database was to require both valid (called Category A) and suspect (flagged S) NO\textsubscript{y} data (called Category B) to be greater than or equal to NO\textsubscript{x} concentrations. These requirements resulted in a NO\textsubscript{y} data percent completeness of 54 percent. The low data completeness was likely caused by incomplete conversion of NO\textsubscript{y} at the 10-m converter or collection of NO\textsubscript{x} species in the NO\textsubscript{x} sample line and eventual release and sampling of these species as NO\textsubscript{x}. The valid NO\textsubscript{y} and related data were used to complete the analyses summarized in this section.

d. Comparison of Weekly Average NO\textsubscript{y} Concentrations with Weekly Filter Pack Total Nitrate Concentration Data

HNO\textsubscript{3} and particulate NO\textsubscript{3} are measured on CASTNET filter packs and the sum is defined as total NO\textsubscript{3}. To evaluate the components of NO\textsubscript{y} measurements, NO\textsubscript{y} concentrations were compared with CASTNET filter pack total NO\textsubscript{3} concentrations. Since NO\textsubscript{y} species include HNO\textsubscript{3} and particulate NO\textsubscript{3} the ratio of total NO\textsubscript{3} to NO\textsubscript{y} should always be less than or equal to 1.0. Figure 33 provides a time series of weekly total NO\textsubscript{3} and weekly average NO\textsubscript{y} concentrations collected at BFT142. Other than for one week, the NO\textsubscript{y} concentrations were consistently higher than the total NO\textsubscript{3} levels, as expected. Qualitatively, the two measurements follow similar temporal patterns throughout the study period. For this analysis, all valid NO\textsubscript{y} measurements were used regardless of their relationship to NO\textsubscript{x} measurements.
e. Time Series of Weekly NO\textsubscript{y}, NO\textsubscript{x}, NO\textsubscript{2}, NO\textsubscript{3}, NO, and O\textsubscript{3} Concentrations

Figure 34 provides graphs of valid weekly NO\textsubscript{y}, NO\textsubscript{x}, NO\textsubscript{2}, NO\textsubscript{3}, NO, and O\textsubscript{3} concentrations collected over the entire measurement period. The NO\textsubscript{2} and NO\textsubscript{3} concentrations are calculated values. The O\textsubscript{3} measurements were plotted only for those weeks that met a 60 percent completeness requirement. The seasonal pattern for O\textsubscript{3} and the nitrogen species is evident with ozone peaking in the warm summer months while the highest concentrations for the nitrogen species are evident during the winter months.

f. Time Series of Monthly NO\textsubscript{y}, NO\textsubscript{x}, NO\textsubscript{2}, NO\textsubscript{3}, NO, and O\textsubscript{3} Concentrations

Time series of monthly average nitrogen species concentrations and O\textsubscript{3} concentrations are given in Figure 35. Similar to Figure 34, the highest O\textsubscript{3} concentration occur during the summer season and the highest concentrations for the nitrogen species are seen during the winter months.
Figure 34. Weekly NO\textsubscript{y}, NO\textsubscript{x}, NO\textsubscript{2}, NO\textsubscript{z}, NO, and O\textsubscript{3} Concentrations (ppb)

Figure 35. Time Series of Monthly Average Nitrogen Species Concentrations and O\textsubscript{3} Concentrations (ppb)
g. Ratios of Weekly NO\textsubscript{x} and NO\textsubscript{y} Concentrations

Figure 36 provides time series of ratios of weekly NO\textsubscript{x} to NO\textsubscript{y} concentrations. The ratios range from about 0.58 to 1.0. For most of the study period, the ratio is constant in the middle of the range. Figure 37 presents a composite diurnal distribution of the ratios. The ratios peaked during the morning and evening rush hours when NO\textsubscript{x} was produced from fresh vehicular emissions. The NO\textsubscript{y}/NO\textsubscript{x} ratio decreased during the day as photochemistry depleted some of the NO\textsubscript{x}. However, the measurements indicate the NO\textsubscript{x} -- O\textsubscript{3} photochemistry was not depleted entirely, with sufficient NO\textsubscript{x} available to produce O\textsubscript{3} farther downwind.

**Figure 36. Time Series of Ratios of Weekly NO\textsubscript{x} to NO\textsubscript{y} Concentrations**
Figure 37. Composite Diurnal Distribution of Ratios of Weekly NO$_x$ to NO$_y$ Concentrations

![Composite Diurnal Distribution of Ratios of Weekly NO$_x$ to NO$_y$ Concentrations](image)

h. Diurnal Plots of Aggregated Nitrogen Species and O$_3$ Concentrations

Figure 38 shows six plots of composite NO$_y$, NO$_x$, NO$_2$, NO$_z$, NO, and O$_3$ concentrations. The curves were constructed by averaging all hours for each hour of the day for each parameter for the entire measurement period. The results show typical temporal distributions with NO values at 0.0 for much of the night due to oxidation by O$_3$ and conversion to NO$_2$. NO$_2$ levels showed little variation during the day. NO$_y$, NO$_x$, and NO$_2$ levels increased after sunrise with fresh NO emissions produced by vehicular traffic and other sources. O$_3$ concentrations peaked around 1500 local time. The growth of the convective boundary layer, photochemical transformations, and dry deposition decreased nitrogen concentrations during the day as O$_3$ levels increased. A secondary smaller peak in NO$_y$, NO$_x$, and NO$_2$ concentrations was observed in the evening. Estimated daytime NO$_2$ concentrations ranged from about 1.0 to 1.5 ppb, typical of rural settings. The typical daytime O$_3$ concentration increase was about 15 ppb for the Beaufort site.
6.0 Summary of Measurements

Atmospheric pollutant concentrations were collected using an ADS/FP system at the CASTNET BFT142 site near Beaufort, NC. ADS measurements generally compared reasonably well with AMoN and CASTNET filter pack data. The ADS data corroborate the use of 2-week passive Radiello samples to measure NH₃. The ADS/FP data also compared well with standard CASTNET FP measurements with the exception of particulate NO₂, particulate NH₄⁺, and total NO₃. Because the CASTNET filter pack system does not use a cyclone separator to collect large particles, CASTNET measurements of NO₂ and total NO₃ were higher than the corresponding ADS/FP measurements. The scatter diagram for NH₄⁺ shows fair comparisons over the range of NH₄⁺ concentrations with a weak correlation coefficient. These results were likely affected by the collection of larger NH₄NO₃ particles on the CASTNET filter pack. Also, any volatilization of NH₄NO₃ can result in loss of NH₄⁺ as NH₃ and collection of particulate NO₂ as HNO₃.

Total nitrogen and sulfur deposition were estimated for the Beaufort, NC site. Dry deposition was estimated from measured concentrations of nitrogen and sulfur species and modeled dry deposition velocities. Wet deposition rates were measured onsite and obtained from NADP/NTN. Dry and wet deposition fluxes were summed to obtain total deposition (Table 3). Dry NH₃, wet NH₄⁺, and wet NO₃ fluxes were the main contributors to nitrogen deposition. Wet SO₄²⁻ contributed more than 75 percent of total sulfur deposition.

Direct measurements of NO, NOₓ, NO₃, and O₃ concentrations were made at the Beaufort, NC site during the 19-month sampling program. Concentrations of NO₂ and NO₃ were estimated as
differences between the other nitrogen measurements. Apparent problems with the NOy converter atop the 10-m tower and other sampling issues resulted in 87 percent data completeness for the NOy measurements. Screening the data with an additional criterion requiring NOy concentrations to be greater than or equal to NOx values resulted in a data set with a completeness of 54 percent for use in this report. Subsequent analyses, e.g., those shown in Figure 33, suggest most of the problems were with the NOx measurements and that the NOy data were largely valid. NOx measurements may have been affected by the difference in inlet height from those used in typical NOx monitoring configurations. For this study, ambient air was sampled at the 10-m height of the NOy converter even though NOx monitoring is usually performed at inlet heights of less than 4 m. A vertical gradient in HNO3 concentrations with lower concentrations near the ground may have led to higher HNO3 being sampled by the NOx channel at 10 m. Also, sampling at 10 m required longer sampling tubing than a typical NOx installation. Adsorption and desorption of nitrogen species, such as HNO3, might have led to nitrogen components of NO2 being sampled as NOx. The various analyses presented in Figures 33 through 38 show that, following screening, the behavior of the nitrogen species and O3 concentrations were consistent with expected atmospheric behavior/models.

This study identifies a portion of the nitrogen deposition budget not typically characterized by routine CASTNET measurements. Specifically, as shown by Figure 34, information on the components of NOy, other than HNO3 and NO3, is gained by the addition of these nitrogen species measurements. These data provide a better estimate of the nitrogen burden in the atmosphere. The development of a NOy network within CASTNET will provide more data and increase understanding of the importance of these nitrogen compounds. The NOy study at BFT142 demonstrates that value can be added to existing NOy sites by the inclusion of converters specific to the measurement of other components of total nitrogen, such as NOx and, by difference, NO2. Future research should examine the artifacts produced by converters, such as the molybdenum converter used in this study, and investigate other options that may produce improved results, such as an LED-based photolytic converter for NOx.

The information in this document has been funded wholly under EPA contract EP-W-09-028, managed and collaborated by the United States Environmental Protection Agency’s Office of Research and Development and Office of Air Programs. It has been subjected to the Agency’s peer and administrative review and has been approved for publication as an EPA document.
7.0 References


